# MICROBIAL BIOTECHNOLOGY

### Progress and Trends









<sup>Edned by</sup> Farshad Darvishi Harzevili Hongzhang Chen

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# Contents

Preface	vii
Acknowled	gmentsix
Editors	xi
Contributor	s xiii
Chapter 1	Microbial biotechnology: An introduction
Chapter 2	<b>Screening of microbial metabolites and bioactive components9</b> <i>R. Baskaran, P.M. Mohan, S. Ganesamoorthy, and Ashok K. Rathoure</i>
Chapter 3	Metagenomics as advanced screening methods for novel microbial metabolites
Chapter 4	Genetic engineering of nonconventional yeasts for the production of valuable compounds
Chapter 5	Induced anhydrobiosis: Powerful method for preservation of industrial microorganisms
Chapter 6	Recent developments in solid-state fermentation: Chinese herbs as substrate
Chapter 7	Microbial biofuels production
Chapter 8	Microbial production of organic acids
Chapter 9	Microbial amino acids production
Chapter 10	<b>Probiotics:</b> The possible alternative to disease chemotherapy213 Adel M. Mahasneh and Muna M. Abbas

Chapter 11	<b>Microbial healthcare products</b> Zafar Alam Mahmood and Saad Bin Zafar Mahmood	239
Chapter 12	Microbial biomass production Ashok K. Rathoure	279
Chapter 13	<b>Microbial biofertilizers and their pilot-scale production</b> <i></i> Santosh Kumar Sethi, Jayanti Kumari Sahu, and Siba Prasad Adhikary	297
Chapter 14	<b>Bacterial biocontrol agents</b> <i>Rikita Gupta and Jyoti Vakhlu</i>	317
Chapter 15	<b>Microbial biodegradation of polycyclic aromatic hydrocarbons</b> Pankaj Kumar Jain	331
Index		351

# Preface

Microbial biotechnology is a technology based on microbiology. Microbial biotechnology is the use of microorganisms and their derivatives to make or modify specific products or processes. Microbial biotechnology is closely related to applied and industrial microbiology. Microbial biotechnology is sometimes considered synonymous with modern industrial microbiology.

The microorganisms used in industrial processes are natural, laboratory-selected mutants, or genetically engineered strains to obtain an economically valuable product or activity on a commercial and large scale. Natural and mixed microbial strains were used in ancient or traditional industrial microbiology, whereas pure or mutant microbial strains are used in classic industrial microbiology. The use of genetically engineered microbial strains began with modern industrial microbiology or microbial biotechnology.

Microbial biotechnology is an interdisciplinary field, and successful development in this field requires major contributions in a wide range of disciplines, particularly microbiology, biochemistry, genetics, molecular biology, chemistry, biochemical engineering, bioprocess engineering, and so on.

Recently, new methods of metabolic engineering, industrial systems biology, bioinformatics, and X-omics science such as genomics, metagenomics, transcriptomics, proteomics, metabolomics, fluxomics, and even nanobiotechnology, have been used to find and modify microorganisms with industrial capacity and their valuable products.

In general, the production of products in industrial microbiology and microbial biotechnology are typically investigated under upstream processes, fermentation processes, and downstream processes.

*Microbial Biotechnology: Progress and Trends* covers recent developments in some fields of microbial biotechnology. Chapter 1 reviews microbial biotechnology from its historical roots to its different processes. Chapters 2 through 5 discuss some of the new developments in upstream processes. Chapter 6 considers solid-state fermentation as an interesting field in fermentation processes. Chapters 7 through 12 argue about recent developments in the production of valuable microbial products such as biofuels, organic acids, amino acids, probiotics, healthcare products, and edible biomass. Chapters 13 through 15 discuss important microbial activities such as biofertilizer, biocontrol, biodegradation, and bioremediation.

The book is written in simple and clear text, and we also used many figures and tables to make the book easier to understand. Furthermore, case studies are included at the end of some chapters.

Overall, this book will serve as a suitable reference for students, scientists, and researchers at universities, industries, corporations, and government agencies interested in biotechnology, applied microbiology, bioprocess/fermentation technology, healthcare/

pharmaceutical products, food innovations/food processing, plant agriculture/crop improvement, energy and environment management, and all disciplines related to microbial biotechnology.

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> Farshad Darvishi Harzevili Editor-in-Chief

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**Dr. Farshad Darvishi Harzevili** earned a BSc in biology at the University of Guilan, Iran. He earned his MSc and PhD in industrial microbiology and microbial biotechnology from the University of Isfahan, Iran. He is currently a faculty member and head of the microbial biotechnology and bioprocess engineering (MBBE) group at the University of Maragheh, Iran. His main interest is in the biotechnological and environmental applications of yeasts, especially the use of agro-industrial wastes and renewable low-cost substrates in the production of biotechnologically valuable products such as microbial enzymes, organic acids, single-cell oils, biofuels, and so forth. He is also interested in the expression of heterologous proteins, metabolic engineering and synthetic biology of yeasts.

**Dr. Hongzhang Chen** earned his master's degree in microbiology from Shandong University in 1991, and his PhD in biochemical engineering from the Institute of Process Engineering, Beijing, China in 1998. He is currently a member of the Chinese Academy of Sciences and the vice director of the State Key Laboratory of Biochemical Engineering. His research is focused on ecological biochemical engineering and cellulose biotechnology. Currently, his research explores new types of solid-state fermentation techniques, devices and processes, and clean fractionation using steam explosion techniques for efficient pre-treatment in cellulose conversion to useful biorenewable materials.

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### chapter one

## Microbial biotechnology An introduction

### Farshad Darvishi Harzevili

Contents

1.1	Introduction	1
	A brief history	
	Nature of industrial microbiology and microbial biotechnology	
	Review of main processes in microbial biotechnology	
	1.4.1 Upstream processes in microbial biotechnology	
	1.4.2 Fermentation processes in microbial biotechnology	
	1.4.3 Downstream processes in microbial biotechnology	6
1.5	Conclusion	7
Refe	rences	8

### 1.1 Introduction

There are many definitions for biotechnology. Simply, biotechnology is a technology based on biology, or the use of living systems and organisms to make or develop useful products. The most comprehensive definition was given by the United Nations Convention on Biological Diversity at a meeting held in Rio de Janeiro, Brazil in 1992. In this convention, biotechnology was defined as "any technological application that uses biological systems, living organisms, or derivatives thereof, to make or modify products or processes for specific use." This definition was signed and accepted by 168 countries up to now.

The biological processes of microorganisms have been used to make and preserve useful food products for more than 6000 years. Microbial biotechnology or industrial microbiology is the use of microorganisms to obtain an economically valuable product or activity at a commercial or large scale. The microorganisms used in industrial processes are natural, laboratory-selected mutant or genetically engineered strains. Economically valuable products such as alcohols, solvents, organic acids, amino acids, enzymes, fermented dairy products, food additives, vitamins, antibiotics, recombinant proteins and hormones, biopolymers, fertilizers, and biopesticides are produced by microorganisms that are used in chemical, food, pharmaceutical, agricultural, and other industries. Biodegradation and biotransformation of complex compounds, domestic and industrial wastewater treatment, biomining, and enhanced oil recovery are examples of microbially valuable activities. According to the UN Convention on Biological Diversity, microbial biotechnology can be defined as any technological application that uses microbiological systems, microbial organisms, or derivatives thereof, to make or modify products or processes for specific use.

### 1.2 A brief history

Industrial microbiology has a long history and its roots can be traced back to the ancient times of human life, when microbiology had not yet been accepted and developed as a science. Approximately 7000 years BC, the Sumerians and Babylonians used yeast to convert sugar to alcohol. By 4000 years BC, the Egyptians used leaven containing yeast to improve bread quality. Moreover, the ancients knew how to use bacteria and molds for vinegar and cheese production.

The Chinese used molds as antibiotics for the treatment of purulent wounds approximately 500 years BC. This first period, from several thousand years to 150 years ago, is known as the ancient or traditional industrial microbiological period, in which mixed and impure cultures of microorganisms were used in nonsterile conditions to make products (El-Mansi et al. 2012; Soetaert and Vandamme 2010).

Approximately 150 years ago, Louis Pasteur proved the microbial source of fermentation and established industrial microbiology as a science based on scientific principles. During World War I, Chaim Weizmann used *Clostridium* bacterium for the production of acetone and butanol. Then *Aspergillus* mold was used to produce citric acid.

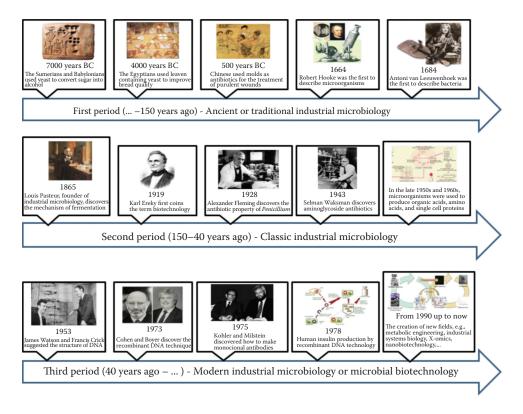
Following Alexander Fleming's discovery of the antibiotic properties of *penicillium* mold, Florey and Chain were able to prepare a pure form of penicillin during the Second World War. In the 1940s, Waksman discovered several aminoglycoside antibiotics such as streptomycin and neomycin. In the late 1950s and 1960s, microorganisms were used to produce amino acids and single-celled proteins, respectively. This second period, 150 years to 40 years ago, is known as the classic industrial microbiological period, in which pure cultures of microorganisms were used in sterile conditions for the manufacture of products. Furthermore, the microbial strains were improved by classic genetic methods such as protoplasm fusion and mutagenesis with physical and chemical mutagens (Glazer and Nikaido 2007).

The third period, which began in the 1970s and continues to the present, is known as the modern industrial microbiology or microbial biotechnology period. The prominent features of this period are its use of recombinant DNA or genetic engineering methods for the improvement of industrial strains and the production of recombinant proteins (Figure 1.1).

### 1.3 Nature of industrial microbiology and microbial biotechnology

In some sources, the term *biotechnology* was incorrectly substituted for genetic engineering or modification. This mistake originated in the United States where, several years ago, new genetic methods were considered as awful and demonic procedures. Therefore, the term biotechnology was used instead of genetic engineering and producing transgenic organisms to reduce worry and diversion of public opinion. Later, the term was used by the media and politicians, and thus it entered legislation and government documents.

Humans have been using genetic modification with the selective breeding of plants and animals for more productivity over tens of thousands of years. For more than 50 years, classic methods such as protoplasm fusion and mutagenesis have been used for the genetic modification of organisms. Genetic engineering and its equivalent terms, genetic modification or genetic manipulation, are advanced molecular biology techniques that have been used since the 1970s (Smith 2009).



*Figure 1.1* Timeline of industrial microbiology and microbial biotechnology.

The term biotechnology was first coined by Karl Ereky in 1917, when molecular genetics and genetic engineering had not yet been discovered. By 1919, in his book entitled *Biotechnology of Meat, Fat and Milk Production in an Agricultural Large-scale Farm,* Ereky described biotechnology as a technology based on converting raw materials into useful products.

Afterward, many definitions were proposed for biotechnology. The application of biological systems, living organisms, or derivatives thereof, to make or modify products is the most comprehensive definition for biotechnology. Biotechnology is not only a science or a set of methods but it is also the interdisciplinary science that encompasses microbiology, plant and animal science, biochemistry, cellular and molecular biology, genetic modification, and engineering fields with biological perspectives such as mechanics, electronics, information technology, robotics, and so on.

The European Federation of Biotechnology considers biotechnology in two categories— "traditional or old" and "new or modern" biotechnology. Several thousand-years–old traditional methods are used to produce beverages, foods, and dairy in traditional or old biotechnology, which is the equivalent of traditional industrial microbiology and classic industrial microbiology. New methods of genetic engineering, which were used from the 1970s to the early 1980s, began the development and evolution of traditional biotechnology to new or modern biotechnology, which is the equivalent of modern industrial microbiology or microbial biotechnology (Smith 2009). Today, the third wave of biotechnology, known as industrial biotechnology or white biotechnology, is expanding. It has made considerable progress in comparison with the second wave, namely, red biotechnology or medical biotechnology and the first wave, namely, green biotechnology or agricultural biotechnology.

Industrial biotechnology uses biological systems, especially microorganisms, in industrial fermentation processes to produce large quantities of pure materials and energy including alcohols, organic acids, amino acids, vitamins, solvents, antibiotics, biopolymers, biopesticides, enzymes, alkaloids, steroids, and others. Industrial biotechnology is founded on biological catalysts and fermentation technology, and it is associated with advances in molecular genetics, protein engineering, and metabolic engineering of microorganisms and cells.

Recently, new methods of metabolic engineering, industrial systems biology, bioinformatics, X-omics such as genomics, metagenomics, transcriptomics, proteomics, metabolomics, fluxomics, and even nanobiotechnology, have been used to find and modify microorganisms with industrial capacity and their valuable products (Soetaert and Vandamme 2010).

### 1.4 Review of main processes in microbial biotechnology

In the past, many diverse products were derived from natural sources or synthesized through chemical processes; nowadays, some of these products are commercially produced through microbial fermentation and biological conversion processes. The benefits of using microorganisms in such processes includes the ease of mass production, the high growth rate, and the use of cheap substrates which, in many cases, are considered waste products of some industries. In general, the products of industrial microbiology and microbial biotechnology are typically investigated under upstream processes, fermentation processes, and downstream processes (Stanbury et al. 1995).

### 1.4.1 Upstream processes in microbial biotechnology

The upstream processes consider the isolation and screening of microorganisms, strain improvement to produce cheap and abundant amounts of the desired product, industrial strain preservation, development of inoculum, substrate selection, optimization and development of appropriate media, and industrial sterilization of media (Figure 1.2).

Microorganisms (as producing agents) and raw materials (as substrates of fermentation) are important in the upstream processes. Strain improvement is critical in the development of most fermentation industries because it increases production efficiency and reduces production costs. In many cases, strain improvement is performed by conventional mutagenesis and recombinant DNA techniques. Then, a suitable strain for the industrial fermentation process is selected through mutant and recombinant strains.

The choice of medium depends on the scale of fermentation. In the small laboratory scale, mostly pure chemicals are used in media composition. This is not possible for large-scale fermentation due to the high costs. Hence, low-cost complex substrates are mainly used in large and commercial-scale fermentations. Most of these compounds are obtained from plant and animal residues, as well as from other industrial wastes that have variable compositions. The effects of fluctuation must be considered between each group of substrates. These effects on performance and product recovery are examined in the small scale with each group of substrate (Waites et al. 2009).

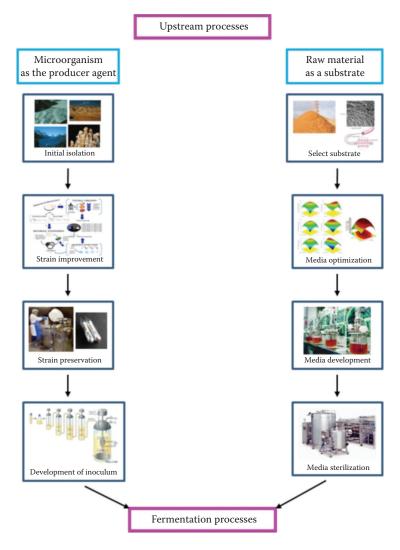


Figure 1.2 Upstream processes in microbial biotechnology.

### 1.4.2 Fermentation processes in microbial biotechnology

The root word of fermentation is derived from the Latin verb *fervere*, meaning to boil, which describes the boiling appearance of the action of yeast on extracts of fruit or grain during fermentation. Fermentation, from the viewpoint of an industrial microbiologist, is the production of a product by the mass culture of microorganisms under aerobic or anaerobic conditions. Fermentation is usually carried out in a fermentor or bioreactor set, whose main objective is to provide a suitable environment for organisms to produce biomass and metabolites. Fermentation system performance depends on many factors, but the main physicochemical properties that need to be controlled are temperature, pH, oxygen transfer, agitation, and foam level.

Fermentation is performed in simple to complex fermentors consisting of a tank equipped with (or without) an agitator controlled by an integrated computer system.

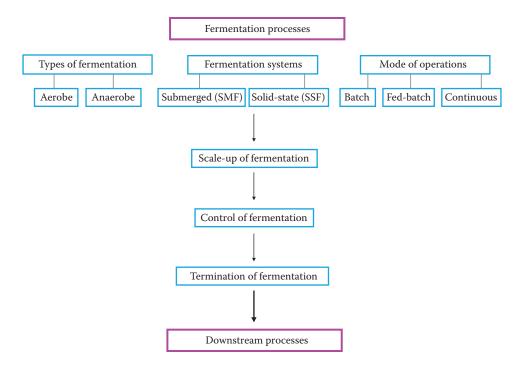


Figure 1.3 Fermentation processes in microbial biotechnology.

Industrial fermentation, depending on the type of end product, is operated by batch, semicontinuous, or continuous models (Stanbury et al. 1995).

The type of fermentation, fermentation system, mode of operation, fermentation scale-up, characteristics of the fermentor, physicochemical control conditions, addition of inoculum and additives to sterile medium, maintenance of sterile conditions during fermentation, and choice of a suitable time to stop fermentation are generally investigated in the fermentation processes (Figure 1.3).

### 1.4.3 Downstream processes in microbial biotechnology

Downstream processes involve all operations after fermentation, the main aim of which is to increase efficiency and ensure product recovery with the desired purity and biological activity. The primary process includes cell separation and cell wall disruption, followed by isolation and purification of the product from the cell extract or culture medium.

Biomass with insoluble bodies are separated by centrifugation or filtering from the liquid part of the medium. When the goal is the isolation of intracellular material, cells are disrupted using chemical, physical, and biological methods. The amount of desired product in the fermented medium is generally low, therefore a concentration process is required.

Afterward, purification methods such as different types of chromatography, dialysis, reverse osmosis, distillation, and solvent extraction are used depending on the type and desired purity level of the final product. In the final stages, evaporation, the addition of chemicals, or drying are used to make the powdered or crystal form of the product.

The downstream processes should use effective and rapid methods for the isolation and purification of the product. This is very important when impure products are

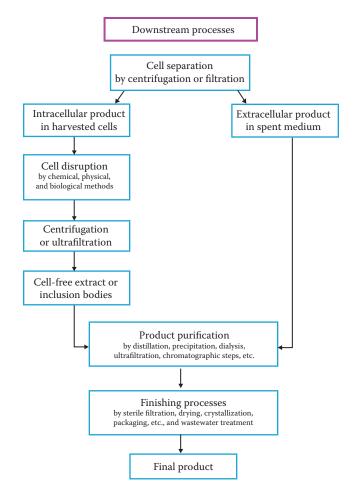


Figure 1.4 Downstream processes in microbial biotechnology.

unstable and affected by environmental conditions. In the case of some products, particularly enzymes, it is critical to maintain their biological activity during the downstream processes. Finally, any wastes produced during the fermentation process should be eliminated using cheap and safe methods (Waites et al. 2009).

In general, biomass separation, cell disruption to release intracellular metabolites, concentration, purification, quantity and quality control, packaging and storage of products, and sterilization and wastewater treatment (especially for industrial and recombinant microorganisms) are investigated in the downstream processes (Figure 1.4).

### 1.5 Conclusion

Microbial biotechnology is the use of microorganisms and their derivatives to make or modify specific products or processes. Microbial biotechnology is an interdisciplinary field, and successful development in this field requires major contributions from a wide range of other disciplines, especially microbiology, biochemistry, genetics, molecular biology, chemistry, biochemical engineering, bioprocess engineering, and so on. Typical microbial biotechnology processes involve upstream processes, fermentation processes, and downstream processes. Recent findings and new methods from other disciplines are used in the microbial biotechnology processes. For example, in the upstream processes, new methods in metabolic engineering, industrial systems biology, bioinformatics, omics science, and even nanobiotechnology are used to find and modify microorganisms with industrial capacity and their valuable products.

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### chapter two

# *Screening of microbial metabolites and bioactive components*

### R. Baskaran, P.M. Mohan, S. Ganesamoorthy, and Ashok K. Rathoure

### Contents

2.1	Intro	duction	9
2.2	Bioac	tive compounds from microorganisms	
		Bioactive compounds from bacteria	
		Secondary metabolites of fungi	
		Bioactive compounds from cyanobacteria	
2.3		study: Fatty acid analysis of antifungal active <i>Streptomyces parvulus</i>	
		MB D105 isolated from mangrove sediments of Andaman sea coast	
		Isolation of <i>Streptomyces</i> spp	
		Primary screening for antifungal activity	
		<i>In vitro</i> assay for antifungal activity	
		Analysis of fatty acids using the FAME method	
2.4		lusion	

### 2.1 Introduction

For many decades, synthetic chemicals have been used as drugs in the treatment of many life-threatening diseases. The pharmaceutical industry has synthesized more than 3 million new chemicals in their effort to produce novel drugs. Despite their success in developing drugs to treat or cure many diseases, the treatment of certain diseases such as cancer, AIDS, heart disease, and diabetes has not been addressed completely due to the complexity of these diseases. Over the centuries, people have been living in close association with the environment and relying on its flora and fauna as a source of food and medicine. As a result, many societies have their own rich plant pharmacopeias. In developing countries, due to economic concerns, nearly 80% of the population still depends on the use of plant extracts as a source of medicine. Natural products are playing an important role in the health care system. In 1816, the isolation of the analgesic morphine from the opium poppy, *Papaver somniferum*, led to the development of many highly effective pain relievers (Benyhe 1994).

The penicillin from the filamentous fungus *Penicillium notatum* was discovered by Fleming in 1929. It had a great effect on the investigation of nature as a source of new bioactive agents (Bennett and Chung 2001). For the past 50 years, antibiotics have transfigured medicine by providing cures for formerly life-threatening diseases. However, strains of bacteria and fungi have recently emerged as virtually unresponsive to antibiotics. Such multidrug resistance arising through antibiotic misuse is recognized as a lifethreatening problem worldwide. The condition is exacerbated by the fact that no novel class of antibiotics have been discovered in the last 20 years. Although many preexisting antibiotics have been customized to yield new derivatives, microorganisms have the capability to mutate their known resistance mechanism (Knowles 1997; Levy 1998).

Resistance to antibiotics is a phenomenon in which a microorganism is no longer affected by an antimicrobial compound to which it was previously sensitive to. It is the capacity of certain microorganisms such as bacteria, viruses, and parasites to neutralize the effect of the medicine. The resistance can come from the mutation of the microorganisms or from the acquisition of resistance genes. The infections caused by resistant microorganisms do not respond to ordinary treatment, which result in prolonged illness and the risk of death. A high percentage of the infections contracted in hospitals are caused by very resistant bacteria, such as methicillin-resistant Staphylococcus aureus (MRSA), as well as Enterococcus faecium and several microorganisms (that are) Gram-negative resistant to vancomycin. New mechanisms indicate that the ability of antimicrobial drugs to act against some bacteria has been completely cancelled. The World Health Organization (WHO) is focusing their efforts on standardizing-through alertness, technical support, generation of knowledge, alliances, anticipation, and control of certain illnesses such as tuberculosis, malaria, human immunodeficiency virus (HIV), proper illnesses of infancy, sexually transmitted diseases, and hospital infections-the quality, supply, and rational use of essential medicines. On the other hand, the struggle against antimicrobial-resistant compounds was one of the topics of the 2011 World Health Day. On this occasion, WHO sought to incorporate the growing issue of resistance to antimicrobial compounds by alerting the political bodies so that governments could initiate proactive steps to solve the problem (WHO 2011).

The importance of finding and using secondary metabolites can be justified in two ways: (1) to know the natural substances that could be beneficial for humans, and (2) to identify the organisms that produce these substances because these may be the only carriers of useful compounds to combat pathogenic microbes (Abbott 1925; Altschul et al. 1990). The screening of new bioactive compounds necessarily involves a vast study of microorganisms, their microbial habitat, and their life cycles. It is more convenient to design cultivation techniques for the different microbial genera present in a particular environment and to identify the organisms that produce bioactive secondary metabolites that are of interest. It will also help to sketch a plan of use and preservation for those species. These procedures represent a potential source of new drug development; especially those obtained from bacteria because bacteria have their own cultivation characteristics. Principally, it has attracted much attention on a global scale from a large number of investigators in search of new natural products with anticancer and antibiotic activity.

### 2.2 Bioactive compounds from microorganisms

The scientific investigations in the field of natural products have been emphasized particularly because of the following facts:

- To find biologically active compounds. Research plants or organisms are approved for bioactivity-driven isolation and identification of secondary metabolites if initial extracts (of higher plants and broths) showed useful biological activity
- 2. To screen transgenic organisms for a single metabolite of commercial importance. This is a natural extension of the first objective, with biotechnology skills playing a major role.
- 3. To reclassify a confused phylogeny. Secondary metabolites from each species of a genus are elaborately isolated, characterized, and compared with one another (Kopcke et al. 2002).

Microorganisms, including certain bacteria, fungi, and algae, which will produce secondary metabolites with some degree of bioactivity, either against another microorganism or acting against certain physiological states of a diseased body, are known as bioactive substances. New bioactive compounds may be prepared using chemical synthesis or it may be isolated from nature. The newly emerging field of combinatorial chemistry, besides the capacity to produce new compounds, has its own challenges in the synthesis of useful leads as new drugs. Berdy (2005) states that natural resources will provide structurally and mechanistically new molecules serving as useful direct drugs or lead compounds with better chances than any chemical approaches. At present, pharmaceutical manufacturers have synthesized several million (according to some estimations, approximately three to four million) new organic chemical structures, but only a negligible portion of them (no more than one in several ten thousand, ~0.001%) have become accepted as drugs.

From the tens of thousands of known microbial metabolites, approximately 150 to 160 (~0.2% to ~0.3%) compounds have been utilized commercially as therapeutic products. A comparison of a library of natural products (including bioactive microbial metabolites) with random libraries of synthetic and combinatorial chemistry shows that screening microbial product libraries is more efficient and economic than the discovery of new bio-active compounds. The inherent molecular diversity of natural products outweighs any chemical-combinatorial libraries (Feher and Schmidt 2003). Perhaps only a combination of the natural product libraries with these chemical libraries will provide more promising chances for additional target-based screenings to discover more active bioactive and useful derivatives of the natural products. Natural products usually result in new mechanisms of action and have a higher probability of exhibiting novel therapeutic activities compared with synthetic or combinatorial compounds.

The reason for the higher efficiency of natural products in contrast with random chemicals is that natural products are somewhat prescreened by nature and have already evolved specific biological interactions. Furthermore, the products of microbes have many similarities with the metabolites of other living systems, for example, mammalian systems. For a compound to have a higher chance of exhibiting its drug-like actions, it must be absorbable and metabolizable. Therefore, it would require minimal modification to develop as an effective, orally active, and easily marketable product. The overestimated capability of rational drug design or the structure–activity relationship approaches was thought of several years ago as a replacement for natural product research.

It has to be emphasized that combinatorial chemistry has a complementary role only and not a replacing role. It may assist in the development of favorable derivatives of lead compounds originally derived by screening from microbial or other natural sources. It will only help modify the structurally and mechanistically new, inherently active natural compounds with respect to absorption; transport and uptake make them effective *in vivo*. It is unquestionable that microorganisms represent the biggest opportunity for the development of useful compounds, which may serve not only as direct drugs or pesticides/herbicides but also as lead compounds and templates for rational drug design. It is an important point that microbes from higher plant and marine animal sources give better chances for satisfactory scale-up (Berdy 2005). Further, organic chemists could not have imagined the unprecedented architectures of these natural products. It is a strong belief that, in most respects, nature is the first; the chemists of today are still able to only copy nature (Berdy 2005). The approximate number of known natural products derived from the main types of plants, animals, and microorganisms are summarized in Table 2.1.

Source	No. of compounds	Bioactives	Antibiotics
Natural products	>1 million	200,000-250,000	25,000-30,000
Plant kingdom	600,000–700,000	150,000-200,000	~25,000
Microbes	>50,000	22,000–23,000	~17,000 (bacteria, 2900; actinomycetales, 8700; fungi, 4900)
Algae, lichen	3000-5000	1500-2000	~1000
Higher plants	500,000-600,000	~100,000	10,000-12,000
Animal kingdom	300,000-400,000	50,000-100,000	~5000
Protozoa	Several hundred	100-200	~50
Invertebrates	~100,000	_	~500
Marine animals	20,000-25,000	7000-8000	3000-4000
Insects/worms/etc.	8000-10,000	800-1000	150-200
Vertebrates (mammals, fishes, amphibians, etc.)	200,000–250,000	50,000–70,000	~1000

Table 2.1 A Data of Approximate Number of Known Useful Products from Natural Resources

Source: Berdy, J., The Journal of Antibiotics 58(1): 1–26, 2005.

### 2.2.1 Bioactive compounds from bacteria

The ocean is called the "mother of life" and occupies more than 70% of the Earth's surface. Due to its depth, it encompasses approximately three hundred times the habitable volume of the terrestrial habitats on Earth. In the past, the oceans had been considered as a rich source of extremely potent compounds, which represented a considerable number of drug candidates (Gochfeld et al. 2003; Haefner 2003). Although very little is known about the microbial diversity of marine sediments, similar to terrestrial soil, marine sediments also contain considerable amounts of readily available organic matter and carbon, which is present in complex form (i.e., cellulose and chitin). However, culture-independent methods have demonstrated that marine sediments contain a wide range of unique microorganisms that are not present in terrestrial environments (Ravenchlang et al. 1999; Stach et al. 2003). Although macroorganisms from the oceans have proven to be good sources of novel bioactive metabolites, large-scale production of these bioactive metabolites has been difficult (Bernan et al. 1997). Marine microorganisms such as bacteria and fungi have been reported to produce antibacterial, antifungal, antiviral, and antitumor substances (Rosenfeld and Zobell 1947; Bernan et al. 1997). The biodiversity of marine microbes and the versatility of their bioactive metabolites have not been fully explored. Due to the complex nature of the oceans, marine bacteria have developed sophisticated physiological and biochemical systems with which they uniquely adapt to extreme habitats and various unfavorable marine environmental conditions. They live in a biologically competitive environment with unique conditions of pH, temperature, pressure, oxygen, light, nutrients, and salinity, as well as being especially rich in chlorine and bromine elements.

Microbes can sense, adapt, and respond to their environment quickly and can compete for defense and survival by the generation of unique secondary metabolites. Even though these compounds are produced in response to stress, many have shown value in biotechnological and pharmaceutical applications (Wenzel and Müller 2005). The marine microbial metabolites exhibit unique biological activities compared with terrestrial bacteria (Blunt et al. 2004; Berdy 2005). Microorganisms that are present in different environments continue to provide pharmacologically important secondary metabolites, which are commercially explored as drugs and other products. The importance of marine sources for the discovery of novel natural products with a pharmaceutical potential has been proved during the last decade and has been highlighted in various reviewed articles (Faulkner 2000; Blunt et al. 2003, 2004; Hentschel et al. 2006).

Knowledge of the distributions of various bacterial groups in complex marine environments is essential for searching and developing a new chemical resource. However, studies on the distributions of marine bacteria in the marine habitat are limited. The actinobacteria or filamentous bacteria, which have been the single most important source of exciting metabolites from soil bacteria, are also members of the Gram-positive family of bacteria.

In past decades, many studies were carried out to understand the diversity of true marine bacteria in their natural marine environment (Jensen and Fenical 1996). Marine bacteria are generally involved in the mineralization of organic matter, nutrient cycling, and energy transfer in aquatic environments (Azam and Worden 2004). However, their potential to synthesize novel chemical compounds with antimicrobial properties was first recognized by Rosenfeld and Zobell (1947) and Grein and Meyers (1958). The bactericidal property of seawater was observed during the same period of study, and it was subsequently realized that this was due to the production of antibiotics by planktonic algae and marine bacteria, which were found abundantly distributed in the seawater (Baam et al. 1966; Baslow 1969).

Currently, the number of life-threatening infections caused by various microorganisms is rapidly increasing as the number of microorganisms that are found to be highly resistant to available antibiotics as well as other drugs has became a serious problem. Many strains of bacteria are resistant to one or more of the 100 antibiotics that are now in use, and these antibiotic-resistant bacteria also spreading globally. Microorganisms, particularly bacteria have tremendously influenced the development of medical science. More than 10,000 of these are biologically active and approximately 8000 are recognized as antibiotics and antitumor agents (Berdy 1989; Stierle et al. 1993; Tomasz 1995). Nearly a hundred microbial products have now found clinical applications as antibiotics, antitumor agents, and agrochemicals (Demain 1983; Clark 1996; Cragg et al. 1997). Despite these early observations, relatively little attention was directed toward the study of natural products from marine bacteria. This was due to the difficulties encountered during the isolation and cultivation of marine bacteria. Only a small percentage of the viable bacterial cells in marine samples ultimately grow under standard culture conditions. Not many studies about the diversity and distribution of marine bacteria have been conducted and only a small percentage has been explored. Hence, only few marine bacteria have been the subject of comprehensive chemical analysis and most of their structurally unique metabolites have been discovered through fermentation studies (Clark 1996; Cragg et al. 1997). This was mainly to establish significant pharmacological applications of the novel metabolites to replace failed antibiotics (Umezawa 1972, 1982).

Actinobacteria are Gram-positive bacteria that have characteristically high (>55%) G + C content (Goodfellow and Williams 1983). The name "actinomycetes" was derived from the Greek *aktis* (a ray beam) and *mykes* (fungus). This terminology was used for these beam-like organisms from the initial observations based on their morphology. The actinobacteria were originally considered to be an intermediate between bacteria and fungi but are now recognized as prokaryotes (Das et al. 2008). The majority of actinobacteria are free-living, saprophytic bacteria found widely distributed in soil, water, and colonizing plants. In the soil, they are involved in the decomposition of organic matter and in the

mineralization cycle through the production of extracellular enzymes, such as cellulases, chitinases, and lignin peroxides.

Genera of actinobacteria such as *Streptomyces, Sacchropolyspora,* and *Amycolatopsis* are adapted to survive in highly erratic and competitive soil environments. They are not only equipped with a wide array of enzymes for exploiting nutrients, but also to produce a broad range of bioactive metabolites of industrial and medical importance, for example, compounds with antibiotic activity for fungal and bacterial inhibition. Actinobacteria are recognized as a source of novel antibiotic and anticancer agents with unusual structure and properties (Jensen et al. 2005) and are also known to produce chemically diverse compounds with a wide range of biological activities (Bredholt et al. 2008). Therefore, actinobacteria hold a prominent position due to their diversity and proven ability to produce new compounds.

Streptomyces is the largest genus of actinobacteria belonging to the family Streptomycetaceae (Kampfer 2006). Approximately 500 species of Streptomyces have been described thus far (Euzeby 2008). It is predominantly found in soil and decaying vegetation. Most streptomycetes produce spores and are noted for their distinct "earthy" odor, which results from the production of a volatile metabolite, geosmin. They produce more than two-thirds of the clinically useful antibiotics (e.g., neomycin and chloramphenicol) of natural origin (Kieser et al. 2000). Members of the actinobacteria group are primarily recognized as organisms of academic curiosity because they are potential degraders and organisms of antibiotic producers. Actinobacteria are the main sources of clinically important antibiotics, most of which are too complex to be synthesized using combinatorial chemistry. Actinobacteria represent a ubiquitous group of microorganisms from various natural ecosystems around the world. In addition to its widespread distribution in terrestrial and aquatic habitats, it is also abound in extreme environments including marine ecosystems. It is well known that members of the genus Streptomyces produce numerous antibiotics and other classes of biologically active secondary metabolites. Approximately two-thirds of the known antibiotics are produced with actinobacteria and among them nearly 80% are made from members of the genus Streptomyces, with other genera trailing numerically. If we include secondary metabolites with biological activities other than antimicrobial systems, actinobacteria are still at the forefront, with more than 60% (again Streptomyces sp. account for 80% of these). Many reports on the effects of antagonistic actinobacteria against pathogenic bacteria and fungi are given in Table 2.2.

In recent years, marine microorganisms have gained important status in the pharmaceutical industry as they are potential sources of microbial products exhibiting antimicrobial, antiviral, antitumor, anticoagulant, and cardioactive properties. Although bacteriologists have long been fascinated with the identification of novel actinobacteria and novel compounds from diverse environments, the marine environment is a relatively recent source. Some of the industrially important strains of actinobacteria such as *Salinospora* have become a source of novel secondary metabolites. These molecules with potent biological activity and these marine actinobacteria represent an important future resource for small-molecule drug discovery. Compounds isolated from marine actinobacteria have a broad spectrum of biological activities such as antibiotic, antifungal, cytotoxic, neurotoxic, antimitotic, antiviral, and antineoplastic activities (Newman and Cragg 2007). Recently, new targets have been added to the general screening of actinobacteria producing bioactive compounds for AIDS, immunosuppression, anti-inflammation, Alzheimer disease, aging processes, and other tropical diseases. However, these new discoveries on marine actinobacteria are facing some unexpected problems. For many of the marine

Organism(s)	Activity against	Reference(s)
Actinobacteria	Fusarium oxysporum F. cubense, and Penicillium graminicolum	Meredith 1946
Actinobacteria	Mycobacterium tuberculosis	Johnstone 1947
Streptomyces spp.	<i>Trichophyton</i> sp., <i>Fusarium</i> sp., <i>Penicillium</i> sp., and certain bacteria	Leben et al. 1952
Actinobacteria	Yeast	Takahashi et al. 1993
Actinobacteria	S. aureus and E. coli	Krasilnikov et al. 1953
Streptomyces sp.	Soil fungi and bacteria	Lockwood 1959; Zeeck et al. 1987
<i>Streptomyces,</i> <i>Micromonospora,</i> and actinobacteria	Gram-positive and Gram-negative bacteria, yeast, and filamentous fungi	Pisano et al. 1989, 1992
Actinobacteria	S. aureus and B. subtilis	Choi and Park 1993
Microbispora spp.	S. aureus and A. niger	Hayakawa et al. 1995
S. aureus	H. oryzae, Curvularia lunata, and T. mentagrophytes	Chakrabarty and Chandra 1979
S. hydroscopicus	T. mentagrophytes and C. albicans	Gurusiddaiah et al. 1979
S. violaceoniger	Macrophomia phaseolina and C. albicans	Hussain and El-Gammal 1980
S. globus	Alternaria solani, A. niger, C. pallescens, T. rubrum, T. mentagrophytes, C. albicans, and Phytophthora sp.	Nair et al. 1994
S. arabicus	A. brassicae	Sharma et al. 1985
S. anandii	Bacteria	Christopher et al. 1987
S. roseiscleroticus (sultriecin) and S. hygroscopicus (yatakemycin)	C. albicans, Cryptococcus neoformans, A. fumigatus, F. moniliforme, T. mentagrophytes, Blastomyces dermatitides, and Petriellidium boydii	Ohkuma et al. 1992; Satoh et al. 1993; Zheng et al. 2000; Datta et al. 2001
<i>S. auermitilis</i> (Avermectins)	Antiparasitic—nematodes and arthropods	Ikeda and Omura 1993
<i>Streptomyces</i> sp. (nonpolyene)	<i>C. albicans, C. tropicalis, B. cinerea, A. fumigatus,</i> <i>F. solani, F. oxysporum, P. irregulare, Pseudomonas</i> sp., <i>Salmonella</i> sp. and Legionella sp.	Hacene et al. 1994; Yon et al. 1995
S. violaceusniger (new macrolide)	C. neoformans, C. albicans, C. tropicalis, C. parapsiolis, C. glabrata, A. fumigatus, A. flavus, T. mentagrophytes, T. rubrum, and M. canis	Kook and Kim 1995; Woo and Kamei 2003
Streptomyces sp.	Aeromonus hydrophila, A. sobria, Edwardsiella tarda, Vibrio alginolyticus, and V. harveyi	Patil et al. 2001a,b
S. bottropensis and S. griseoruber	Agrobacterium tumefaciensis and C. albicans	Saadoun and Al-Momani 1998, 2000
Actinomadura spp.	Bacillus subtilis, S. aureus, M. luteus, C. albicans, and fungi	Srivibool 2000
	~	(continued)

 Table 2.2 Antagonistic Activities of Bioactive Compounds Producing Actinobacteria

 Against Bacteria and Fungi

(continued)

Organism(s)	Activity against	References(s)
Streptomyces sp.	Bacteria and fungi	Ellaiah et al. 2002
<i>Streptomyces</i> sp.	B. subtilis, S. aureus, M. luteus, A. fumigatus, A. niger, M. hiemalis, P. roqueforti, P. rariotii, C. albicans, Cryptococcus humicolus, S. cerevisiae, P. cinnamoni, Pestalotiopsis sydowiana, B. cinerea, and Sclerotinia homoeocarpa	Frandberg et al. 2000; Moncheva et al. 2002; Narayana et al. 2004
<i>S. antibioticus</i> and <i>S. rimosus</i>	Gram-positive and Gram-negative bacteria and yeasts	Sahin and Ugar 2003
Actinobacteria	Gram-positive and Gram-negative	Kokare et al. 2004a,b
Streptomycetes	Phytopathogenic and human pathogenic fungi	Oskay et al. 2004
Streptomyces spp.	Rhizotonia solani	Cao et al. 2004
Marine actinobacteria	Salmonella sp., E. coli, Klebsiella sp., R. solani, Pyricularia oryzae, H. oryzae, and Colletotrichum falcatum	Kathiresan et al. 2005; Dhanasekaran et al. 2005a,b
S. albovinaceus	B. subtilis, B. pumilus, S. aureus, E. coli, and P. aeroginosa	Ellaiah et al. 2005

 Table 2.2 (Continued) Antagonistic Activities of Bioactive Compounds Producing Actinobacteria

 Against Bacteria and Fungi

actinobacteria, the taxonomy of the strain is poorly defined, so that binomial identification becomes cumbersome (Findlay et al. 1986).

In contrast with marine invertebrates, marine actinobacteria seem to be a promising source as a producer of drugs, for example, thiocoraline, a new anticancer drug produced by a marine Micromonospora strain (Romero et al. 1997). Also, the marine actinobacteria represent an important future resource for small-molecule drug discovery. Although the positive effects of actinomycete secondary metabolites on human health is clear, an intensive research by the pharmaceutical industry has exhausted the supply of compounds that can be discovered from this group of bacteria. This perception has been a driving force behind the recent shift away from natural products as a resource for the discovery of smallmolecule therapeutics toward other drug discovery platforms, including high-throughput combinatorial synthesis and rational drug design. The potential of secondary metabolites, derived from marine microorganisms to inhibit Plasmodium growth, and the effect of proteasome inhibitors such as salinosporamide A on in vitro and in vivo parasite development by a secondary metabolite of marine bacterium with significant antimalarial activity was studied (Prudhomme et al. 2008). Salinosporamide A, which is derived from a marine actinobacteria, showed inhibitory activity against parasite development in vitro (Plasmodium falciparum) and in vivo (Plasmodium yoelli) in which the 20S proteasome is the molecular target. This facilitates the discovery of an increased specificity of Plasmodium proteasome inhibitors in the future from some other sources from similar habitats (Prudhomme et al. 2008). The antiviral drug cyclomarin was also produced by a marine streptomycetes strain (Rosenberg et al. 2000). In both cases, fermentation of the microorganisms enabled the production of the amounts required for clinical applications.

The actinobacteria were first recognized by Gasperni (1890) as potential destroyers of fungi and bacteria. The actinobacteria antagonistic of *Pythium* from sugarcane was reported by Tims (1932). Waksman (1937) made a detailed survey of actinobacteria possessing antagonistic effects on the activity of other microorganisms in their studies on decomposition. The distribution of antagonistic activities of actinobacteria against fungi

was studied by Alexopolus (1941). Carvajal (1946) observed that only a few strains of *Streptomyces griseus* isolated from various sources were found to produce streptomycin, and these varied greatly in their ability to do so. Emerson et al. (1946) found that actinobacteria and molds isolated from the soil produced inhibitory substances.

Systematic screening of soil actinobacteria from different parts of India had already been carried out for the production of antibiotic compounds (Banerjee et al. 1954; Rangaswami et al. 1967; Agarwal et al. 1968). Nine percent of the marine actinobacteria isolated required seawater for normal growth and to produce strong antibiotic compounds for antibacterial and antifungal activities (Bredholt et al. 2008). Rizk et al. (2007) reported that 188 Streptomyces species, isolated from Egyptian soils, were screened for their antagonistic activity against 19 fungal and bacterial species. Substantial antagonistic activities were attained by five Streptomyces isolates against Candida albicans. Actinobacteria are capable of producing a large number of medically useful antibiotics (Nolan and Cross 1998). Due to the presence of a large genome and thousands of different genes that encode for a large number of proteins, actinobacteria are capable of utilizing a vast number of substrates. They occur in a multiplicity of natural and man-made environments (Goodfellow and Williams 1983) with different morphological, cultural, biochemical, and physiological characters. Thus, to survive in extreme natural environments, they are capable of producing various secondary metabolites and new antibiotics (Lechevalier and Lechevalier 1967). Nithya and Ponmurugan (2008) reported that strains such as KN-2 and KL-3 showed the production of secondary metabolites for antibacterial and antifungal activities. Muriru et al. (2008) reported that the concentration of antibiotic metabolites by partial purification and freeze-drying resulted in, respectively, 21.5% and 20.1% enhanced antimicrobial affinity. Pugazhvendan and coworkers (2010) reported 10 antimicrobial actinobacteria from the Tamil Nadu coastal area.

Ethyl acetate extract from the *Streptomyces* sp. isolate B865 delivered the trioxaccerains A to C (2a–2c) and additionally these new derivatives were designated as trioxacarcins D to F (2d–2F). All trioxacarcins showed high antibacterial activity, whereas some showed high antitumor and antimalarial activity. The structure of the new antibiotics were derived from mass, one-dimensional and two-dimensional nuclear magnetic resonance spectra, and confirmed by comparison. The absolute configuration of the trioxacarcicins was deduced from the x-ray analysis of gutingimycin (2g) and from known *Streptomyces* of the L-trioacarcinoses A and B (Markey et al. 2004). Some useful bioactive compounds of actinobacteria are given in Table 2.3.

The difficulty and high cost of isolating novel structures of antimicrobial agents with new modes of action led to the phase of decline in this field of research. However, in recent decades, the chemistry of natural marine products has emerged as a mature field after years of intensive research. As a result, studies on secondary metabolites from microorganisms are a rapidly growing field. Typically, primary metabolites are found across all species within broad phylogenetic groupings, and are produced using the same pathway or using similar pathways in all these species. However, microorganisms have received very little attention in the field of drug discovery. This is mainly due to the noncultivability of most bacteria (Hugenholtz et al. 1996). Several studies showed that marine bacteria are capable of producing unusual bioactive compounds that are not isolated from terrestrial sources (Fenical et al. 1993, 1997).

Many unusual compounds such as thermostable proteases, lipases, esterases, starch, and xylan-degrading enzymes have been isolated mainly from marine bacteria and hyperthermophilic archaea (Bertoldo 2002). Unusual Gram-positive bacteria from deep sea sediments, which have produced a series of new natural products, Macrolectin A to F of C24

Antibiotic	Production	Application(s)
Actinomycin D	Streptomyces spp.	Antitumor
Antimycin A	Streptomyces spp.	Telocidal
Avermectin	S. avermitilis	Antiparasitic
Bambermycin	S. bambergiensis	Growth promotant
Bialaphos	S. hygroscopicus	Herbicidal
Bleomycin	S. verticillus	Antitumor
Candicidin	S. griseus	Antifungal
Cephamycin C	Nocardia lactamdurans	Antibacterial
Chloramphenicol	S. venezuelae	Antibacterial
Chlortetracycline	S. aureofaciens	Antibacterial
Cycloserine	S. orchidaceus	Antibacterial
Daptomycin	S. roseosporus	Antibacterial
Daunorubicin	S. peucetius	Antitumor
Doxorubicin (adriamycin)	S. peucetius var. caesius	Antitumor
Erythromycin	Sac. erythraea	Antibacterial
FK506 (tacrolimus)	S. hygroscopicus	Immunosuppressant
Fosfomycin	Streptomyces spp.	Antibacterial
Gentamicin	Micromonospora spp.	Antibacterial
Hygromycin B	S. hygroscopicus	Antihelminthic
Kanamycin	S. kanamyceticus	Antibacterial
Lincomycin	S. lincolnensis	Antibacterial
Milbemycin	S. hygroscopicus	Antiparasitic
Mithramycin	S. argillaceus	Antitumor
Mitomycin C	S. caespitosus	Antitumor
Natamycin	S. nataensis	Antifungal
Neomycin	S. fradiae	Antibacterial
Nikkomycin	S. tendae	Antifungal; insecticidal
Nocardicin	Nocardia uniformis	Antibacterial
Nosiheptide	S. actuosus	Growth promotant
Novobiocin	S. niverus	Antibacterial
Nystatin	S. noursei	Antifungal
Oleandomycin	S. antibioticus	Antibacterial
Oxytetracycline	S. rimosus	Antibacterial
Phleomycin	S. verticillus	Antitumor
Polyoxins	S. cacaoi var. asoensis	Antifungal (Plant protection)
Pristinamycin	S. pristinaespiralis	Antibacterial
Puromycin	S. alboniger	Research
Rapamycin	S. hygroscopicus	Immunosuppressant
Ristocetin	Nocardia lurida	Antibacterial
Spectinomycin	S. spectabilis	Antibacterial
Spinosyns	Sac. spinosa	Insecticidal
Spiramycin	S. ambofaciens	Antibacterial
Streptogramins	S. graminofacients	Antibacterial
		(continued

Table 2.3 Bioactive Compounds of Actinobacteria

Antibiotic	Production	Application(s)
Streptomycin	S. griseus	Antibacterial
Teichoplanin	Actinoplanes teichomyceticus	Antibacterial
Tetracycline	S. aureofaciens	Antibacterial
Thienamycin	S. cattleya	Antibacterial
Thiostrepton	S. azureus	Growth promotant
Tobramycin	S. tenebrarius	Antibacterial
Tylosin	S. fradiae	Growth promotant
Validamycin	S. hygroscopicus	Plant protectant
Vancomycin	Amycolatopsis orientalis	Antibacterial
Virginiamycin	S. virginiae	Growth promotant

Table 2.3 (Continued) Bioactive Compounds of actinobacteria

Source: Kieser, T. et al., Practical Streptomyces Genetics, 2nd ed. Norwich, England: John Innes Foundation, 2000 (ISBN 0-7084-0623-8).

linear acetogen origin, have been isolated (Gustafson et al. 1989). The major metabolite, Macrolectin A, inhibits B16-F10 murine melanoma cells in *in vitro* assays, showing significant inhibition of mammalian herpes simplex virus (HSV type I and II) and protecting T lymphocytes against HIV replication (Carte 1996). In another study, a microbial metabolite from *Alteromonas* spp. with anti-HIV potential as reverse transcriptase inhibitor has been reported.

The first marine bacterial metabolite to be reported was the highly brominated pyrrole antibiotic. This was isolated by Burkholder and coworkers (1966) through fermentation studies of a Gram-negative bacterium obtained from the surface of the Caribbean sea grass Thalasia. The molecule showed impressive *in vitro* antibiotic properties against Gram-positive bacteria, with minimum inhibitory concentrations (MICs) ranging from 0.0063 to 0.2 pg/mL. The compound synthesized by this bacterium was known to be pentabromopseudiline and its antitumor properties were also reported. The Faulkner group in California isolated a purple-pigmented bacterium, which also proved to be a potent antibiotic. Chemical analysis showed that the *Alteromonas* sp. produces several antimicrobial compounds such as pyrrole, tetrabromopyrrole, hexabromo-2,2'-bipyrrole, and several simple phenolics including 4-hydroxybenzaldehyde, *n*-propyl-4-hydroxybenzoate. Tetrabromopyrrole showed moderate antimicrobial activity *in vitro* against *S. aureus, Escherichia coli, Pseudomonas aeruginosa*, and *C. albicans*. It was even more active against several groups of marine bacteria and showed autotoxicity against the producing *Chromobacterium* sp.

*Pseudomonas* sp. was reported to produce 6-bromoindolecarboxaldehyde, its analogue, and a mixture of 2-*n*-pentyl- and 2-*n*-heptylquinolinol (Wratten et al. 1977). The 2-heptylquinolinol is a known antibiotic produced by strains of *P. aeruginosa*. The most potent of these simple antibacterial agents was 2-*n*-pentylquinolinol, which showed its greatest activity against *S. aureus*. Another marine bacteria, *Alteromonas rubra*, was isolated and identified by researchers at the Roche Research Institute in Australia. This bacterium provided a novel new target for the isolation of unique natural products. Under a fermentation study, the bacterium produced a series of C16 aromatic acids, which are acetogenins of fatty acid synthetic origin. *P. magnesiorubra* isolated from the surfaces of the tropical marine green algae, *Caulerpa peltata*, also provided a novel

metabolite. Two antibiotic pigments, the magnesidins were isolated as 1-L mixtures of methylene homologues (4- and 6-methylene groups), and identified as minor metabolites of this marine bacterium (Gandhi et al. 1976). These unique pigments were considered to be oxidation products of prodigiosin, a common tripyrrolic pigment produced by marine as well as terrestrial bacteria. The seawater-derived bacteria studied thus far have been taxonomically very limited and seemingly driven by screening processes for new antibiotics.

The marine sediments are a more nutrient-rich microhabitat that varies greatly in organic content. Such a microhabitat provides a diversity of bacterial flora; usually not found in more nutrient-limited habitats. The presence of free surfaces in these microhabitats is known to stimulate and support enhanced bacterial colonization as well as the growth of chemically prolific marine bacteria. The antimicrobial agent producers present in marine sediments are generally Gram-positive bacteria, such as those from the genus *Bacillus* and the actinobacteria, and Gram-negative bacteria mainly *Pseudomonas* spp. and *Alteromonas* spp. The actinobacteria group comprises the major sources of antimicrobial compounds. Approximately 10% to 33% of the total bacterial community present in soil belongs to *Streptomyces* sp. and *Nocardia* sp., which are the most abundant actinobacteria in soil (Osborne et al. 2000; Pandey et al. 2002). These microbes exhibit a vast metabolic versatility. Hence, they have many physiological cycles that produce intermediate molecules such as enzymes or secondary metabolites with antibacterial, antifungal, and antiviral capabilities.

A slow-growing, Gram-positive bacterium C-237 was isolated from deep-sea sediment samples obtained from the coast of California. Fermentation studies in a salt-based medium yielded a series of novel cytotoxic and antiviral macrolides, the Macrolactins A to F (Gustafson et al. 1989). Under standard fermentation conditions at atmospheric pressure, this bacterium produced six macrolides and two open-chain hydroxy acids in varying amounts. Macrolactin A was produced as the major metabolite (4.8 mg/L) in most of the fermentations. The majority of the biological properties were due to Macrolactin A, which showed modest antibacterial activity, but was active against B16-FlO murine melanoma in vitro with  $IC_{50}$  values of 3.5 µg/mL. More importantly, Macrolactin A inhibited several viruses including HSV (IC<sub>50</sub> = 5.0  $\mu$ g/mL) and HIV  $(IC_{50} = 10 \ \mu g/mL)$ . Another group of Gram-positive bacteria, from the genus *Bacillus* sp. present in soil, also have the ability to produce antimicrobial agents with clinical and agricultural significance. Bacillus thuringiensis secrete certain toxins with insecticidal properties. This chemical has found wide application and is now used in the biocontrol of insects. In the case of B. thuringiensis serovar Israelensis (Bti), it produces a toxin that is used for the biocontrol of black flies and mosquitoes. Among the Gram-negative bacteria, Pseudomonas sp., Alteromonas sp., and Vibrios sp. are known to produce antimicrobial substances. Bioactive molecules produced by Pseudomonas sp. have found wide application mainly in the agriculture sector.

Another group of Gram-positive bacteria present in soil and responsible for the production of antimicrobial agents with clinical and agricultural importance is the genus *Bacillus*. This genus is characterized by being Gram-positive, and having sporeforming rods. It has been demonstrated that these microbes produce antimicrobial agents in various stages of their growth curve. For example, *Bacillus subtilis* 168 can produce nonribosomal oligopeptides with antifungal and antimicrobial properties such as surfactins, inturinics, and bacilysin (Oskay 2004). The antibiotics pyoluteorin (Plt), pyrrolnitrin (Prn), phenazine-1-carboxylic acid (PCA), and 2,4-diacetylphloroglucinol (Phl) are currently a major focus of research in biological control. Strains

from *Pseudomonas* sp. have been isolated from soils that exhibited suppressive effects against plant diseases such as Take-all of wheat, Black root rot of tobacco, *Fusarium* wilt of tomato, and damping off of tomatoes caused by *Rhizoctonia solani*.

Bacteria produce a variety of antimicrobial substances that are able to kill or inhibit other microorganisms. Bacterial antibiotics can be subdivided into two types on the basis of their chemical nature: (i) nonpeptide antibiotics and (ii) peptide antibiotics. One group comprises nonribosomally synthesized peptides produced on large enzymatic complexes such as surfactin (Carrillo et al. 2003). The second group comprises ribosomally synthesized peptides such as bacteriocins. Bacteriocins produced by Grampositive bacteria exhibit a number of characteristics that make them attractive for both the food industry and for biomedical applications (Bower et al. 2001; Chen and Hoover 2003). Several bacteriocins or bacteriocin-like substances (BLIS) produced by the genus Bacillus have been reported such as a bacteriocin of B. brevis, subtilin by B. subtilis, thruicin 7 by B. thuringiensis, and many others. The newly characterized bacteriocins include the lantibiotics ericin A (2986 Da) and ericin S (3342 Da) produced by B. subtilis; cerein 7A (3940 Da) and cerein 7B (4893 Da) from Bacillus cereus and the subclass IIa peptide SRCAM 1580 (3486 Da) produced by Bacillus circulans (Klein et al. 1992; Oscariz et al. 1999; Cherif et al. 2001; Hyung et al. 2001; Stein et al. 2002; Svetoch et al. 2005; Oskouie et al. 2006).

Brammavidhya and Usharani (2013) reported that a low-molecular weight peptide from sponge associated with *B. cereus* was inhibitory to human pathogens and foodspoilage bacteria, such as *C. albicans, Listeria monocytogenes, P. aeruginosa, B. cereus, B. subtilis,* and *Lactobacillus bulgaris* and *Proteus vulgaris*.

### 2.2.2 Secondary metabolites of fungi

In the last decade, marine fungi have become increasingly important sources of new bioactive natural products (Konig et al. 2006; Laatsch 2006; Ramaswamy et al. 2006; Shimizu and Li 2006). The biodiversity of fungi is enormous. Out of the estimated 1.5 million species of fungi recorded worldwide, approximately 4000 secondary metabolites of fungal origin are recognized to possess biological activities, the vast majority coming from the species of *Penicillium, Aspergillus, Acremonium*, and *Fusarium* (Hawksworth 1991; Dreyfuss and Chapela 1994; Onifade 2007).

The initial report of a bioactive natural product from a marine-derived fungus dates back to the 1940s, when the fungus *Acremonium chrysogenum* was isolated. This fungus was isolated from a sewage outlet in the Mediterranean Sea close to the island of Sardinia (Proksch et al. 2008). The fungus was the source of Cephalosporin C, the parent compound of modern cephalosporin antibiotics, which are indispensible for the treatment of numerous bacterial infections (Abraham and Loder 1972). Primarily, progress with rigorous evaluation of marine fungal metabolites was slow. This situation changed dramatically in the 1990s, when there was a sharp increase in interest in marine microbial metabolites, which continues until today. Up to the year 2002, 272 new natural products had been isolated from marine-derived fungi (Bugni and Ireland 2004). A dramatic increase in the number of elucidated marine fungal structures began afterward, illustrated by the fact that between 2002 and 2004, 240 additional compounds were described (Ebel 2006).

Marine-derived fungi as a new source of bioactive metabolites is now commonly accepted, there is still much debate on the nature of fungi that are isolated from various marine substrates, such as driftwood, algae, or invertebrates (Proksch et al. 2008). Because

numerous, if not most, of these fungi belong to genera already well known from the terrestrial environment, such as *Aspergillus, Penicillium, Cladosporium, Phoma*, and *Fusarium*, the true marine origin of these fungal strains is frequently doubted (Holler et al. 2000; Kohlmeyer and Volkmann-Kohlmeyer 2003). It is possible that several marine-derived fungi thus far investigated originated from terrestrial habitats (e.g., soil) from which they were washed to sea and survived (as spores) until they were recovered by a marine chemist looking for new compounds from the sea. On the other hand, in the last few years, more and more evidence has accumulated indicating an adaptation of these "ubiquitous" fungi to the marine environment (Geiser et al. 1998a,b; Alker et al. 2001; Duncan et al. 2002; Zuccaro et al. 2004).

Interestingly, sponges continue to be one of the most important sources for the isolation of metabolite-producing marine-derived fungi, even though the presence of fungal mycelia growing in sponges has not yet been proven. In contrast to sponge-derived natural products, such as jaspaklinolide L, theopederin C, or mycalamide A, which bear obvious structural resemblances to known bacterial metabolites and are assumed to originate from bacterial sponge symbionts, there are no examples of known fungal metabolites that have been isolated from sponges. For the time being, the true nature of sponge–fungal associations remains unclear and far more research needs to be devoted to this field (Holler et al. 2000; Jensen and Fenical 2002; Proksch et al. 2002; Bugni and Ireland 2004; Butzke and Piel 2006; Ebel 2006).

Proksch et al. (2008) reported that fungi isolated from sponge produced 19 different metabolites, including three new natural products, that were isolated and structurally identified. *Aspergillus ustus* yielded two sesquiterpenes, a drimane derivative, and deoxy-uvidin as well as a sesterterpene ophiobolin H. The drimane derivative had an ED<sub>50</sub> value against L5178Y cells of 1.9 mg/mL *in vitro. Penicillium* sp. yielded the largest number of metabolites. Viridicatin, viridicatol, cyclopenin, and cyclopenol suppressed larval growth of the polyphagous pest insect *Spodoptera littoralis* when incorporated into an artificial diet at an arbitrarily chosen concentration of 237 ppm (Proksch et al. 2008).

Screening of *Aspergillus gorakhpurensis* for the production of bioactive secondary metabolites results in the production of 4-(*N*-methyl-*N*-phenyl amino) butan-2-one and itaconic acid. Biological evaluation of the two compounds against test microorganisms showed strong inhibitory activity of 4-(*N*-methyl-*N*-phenyl amino) butan-2-one toward bacteria, fungi, and in the *Spodoptera litura* larvicidal bioassay. An investigation of the secondary metabolite from *Diaporthe helianthi* yielded the phenolic compound 2(-4 hydroxyphenyl)-ethanol (Tyrosol). Its antimicrobial reaction was tested and the ensuing antagonistic effects on the human pathogenic bacteria *Enterococcus hirae*, *E. coli, Micrococcus luteus, Salmonella typhi, S. aureus*, phytopathogenic *Xanthomonas campestris* pv. *phaseoli* and phytopathogenic fungi were demonstrated. Results show that bioactive compounds and Tyrosol produced by *D. helianthi* have a biotechnological potential (Specian et al. 2012). Recently, bioactive compounds such as an antihypertensive angiotensin I converting enzyme inhibitor into an antiangiogenic compound, antidementia β-secretase inhibitor and ginsenoside-Rg3 were produced and characterized to form *Saccharomyces cerevisiae* (Jeong et al. 2006; Lee and Lee 2007; Kim et al. 2009).

Previously, researchers have reported many secondary metabolites from marine fungi. These include *Cephalosporin* sp. and *Stachybotrys* sp. which were isolated from seawater and these were reported to produce Cephalosporins and Stachybotrins A and B, respectively. Some other fungal genera of marine origin such as *Aspergillus* sp., *Penicillin* sp., *Penicillium* sp., and *Aspergillus* sp. were reported to produce Gliotoxin, Epolactaene,

Acetophthalidin, and Aspergillamides A and B, respectively. Sponge-associated fungi such as *Trichoderma harzianum*, *Microascus longirostris*, *Microsphaeropsis olivacea*, *Exophiala pisciphila*, *Aspergillus niger*, *Gymnascella dankaliensis*, *T. harzianum*, *Paecilomyces* cf. *javanica*, *Trichoderma longibranchiatum*, *Microsphaeropsis* sp., *Conoithyrium* sp., and *Drechslera hawaiiensis* produced industrially important secondary metabolites such as Trichoharzin, Cathestatins A–C, Crebroside, Exophilin A, Asperazine, Gymnastatins A–E, Trichodenones A–C, Harzialactones A–B, *R*-mavalonnlactones, Deoxynortrichoharzin and Speciferones A and B, Speciferol A.

#### 2.2.3 Bioactive compounds from cyanobacteria

Cyanobacteria or blue-green algae are a charming group of primitive phototrophic prokaryotic organisms whose long evolutionary histories date back to the Proterozoic era. These organisms, endowed with tremendous genome plasticity, are distributed in all possible biotypes of the world. These organisms have tremendous potential in environmental management as soil conditioners. Due to their occurrence in diverse habitats, these organisms make excellent subject material for investigation by ecologists, physiologists, biochemists, pharmacists, and molecular biologists. Accordingly, looking for cyanobacteria with antimicrobial activity has gained importance in recent years. Biologically active substances were proved to be extracted by cyanobacteria (Borowitzka 1995; Kreitlow et al. 1999; Mundt et al. 2001; Volk and Furkert 2006). Various strains of cyanobacteria are known to produce intracellular and extracellular metabolites with diverse biological activities such as antialgal, antibacterial, antifungal, and antiviral activity (Moore et al. 1989; Jaki et al. 1999; Ghasemi et al. 2003, 2007; Isnansetyo et al. 2003; Kundim et al. 2003; Soltani et al. 2005).

Antibiotic resistance in bacteria is one of the emerging health-related problems in the world nowadays. Plants, among them algae, are valuable natural sources that are effective against infectious agents. Extensive efforts for the identification of bioactive compounds derived from natural resources have been made worldwide to develop safe, nontoxic, and efficient antimicrobial agents of valuable practice in pharmacology. Algae of marine and terrestrial origins have been the best choice among natural resources within the aquaculture and agriculture fields. Screening bioactivity of algal crude extracts is mandatory in biomedical practice, where antibacterial, antifungal, antiviral, and even more antialgal activity have been assessed to these metabolites. Emergence concerns have been raised to establish the structural and functional properties of the bioactive compounds described in algal crude extracts; thus far, more than 2400 bioactive metabolites have been isolated and identified from a diverse group of algal communities (Faulkner 2001; Hellio et al. 2002; Tang et al. 2002; Serkedjieva 2004; Tuney et al. 2006).

In a recent study, some biotechnologically important compounds from the crude extracts of some cynobacteria were reported (Al-Wathnani et al. 2012). These include *Phormidium autumnale* (1-hexyl-2-nitrocyclohexane; 91.7%), *Chlorella vulgaris* (3-methyl-2-butanol; 90.8%), *Spirulina platensis* (2-hexyl-1-nitrocyclohexane; 92.1%), *Nostoc nostoc* (octadecanal and aldehyde; 86.8%), *Dunaliella salina* (3-methyl-2-(2-oxopropyl) and furan; 90%), *Tolypothrix distorta* (boronic acid, ethyl, dimethyl ester; 83.9%), and last, *Microcystis aeruginosa*, which was reported to produce approximately 92% (*S*)-(+)-1-cyclohexylethylamine. Al-Wathnani et al. (2012) reviewed the potential compounds from the algae listed in Table 2.4.

Organism(s)	Bioactive component(s)	Target organism
M. aeruginosa	Lipid	Algaecide
M. aeruginosa	Micropeptin 478-A	Plasma inhibitor
	Micropeptin 478-B	Leucin aminopeptidase
	Microginin 299-A,B	Inhibitor
	Kawaguchipeptin B	Bactericide
M. aeruginosa	Aqueous extract	Antiviral (influenza A)
Microcystis viridis	Aeruginosin 102-A Aeruginosin 102-B	Thrombin inhibitor
Microcystis viridis	Micropeptin 103	Chymotrypsin inhibitor
M. aeruginosa (NIES-88)	Cyclic undecapeptide	S. aureus
Chlorella stigmatophora	N/R	S. aureus; B. subtilis
Synechococcus leopoliensis	N/R	B. subtilis
Laurencia chondrioides	Sesquiterpine	S. aureus; B. subtilis, M. flavus
Cystoseira tamariscifolia	Meroditerpenoid	A. tumafasciens, E. coli
Fucus vesiculosus	Polyhydroxylated Fucophlorethol	S. aureus, E. coli, P. aeruginosa
Scenedesmus sp.	N/R	Alternaria sp.
Scytonema julianum	Acetyl-sphingomyelin	Causes platelet aggregation
Scytonema varium	Scytovirin; Cyanovirin N	Anti-HIV protein
Haloferax alexandrinus	Canthaxanthin	Carotenoid
Microcystis PCC 7806	Cyanopeptolin 963A	Chymotrypsin inhibitor
Aphanizomenon flos-aquae	Phycocyanin	Antioxidant
M. aeruginosa	Zeaxanthin	Carotenoid
Oscillatoria redekei HUB 051	Unsaturated fatty acids	Antibacterial
Fischerella sp. CENA 19	Fischerellin A; 12-epi-hapalindole	Algaecide
Microcystis sp.	N/R	Antiviral (influenza A)
M. aeruginosa	Microviridin	Serum protease inhibitor
Schizotrix sp.	Schizotrin A	Antimicrobial

Table 2.4 Cyanobacterial Metabolites and the Species Responsible for Synthesis

Source: Al-Wathnani, H. et al., Journal of Medicinal Plants Research 6(18), 3425–3433, 2012.

# 2.3 Case study: Fatty acid analysis of antifungal active Streptomyces parvulus DOSMB D105 isolated from mangrove sediments of Andaman sea coast

The term mangrove, in a broader sense, refers to the highly adapted plants found in tropical intertidal forest communities or ecosystem. The term "mangrove" may have been derived from a combination of the Malay word *manggi-manggi*, for a type of mangrove tree (Avicenna) and the Arabic *el gurm* for the same, as *mang-gurm*. As an ecosystem, mangroves are one of the most productive wetlands on earth. The mangrove ecosystem does not have as many species of plants and animals as other ecosystems due to the harsh and constantly changing environment (Lee et al. 2005).

The mangrove ecosystem is one of the largest ecosystems, and one that is widely distributed throughout the world. This also includes forests that occupy several million hectares of coastal area worldwide; distributed in more than 112 countries and territories

comprising a total area of approximately 181,000 km<sup>2</sup> in more than one-fourth of the world's coastlines (Spalding et al. 1997; Alongi 2002). Mangroves are specialized ecosystems developed along estuarine seacoast and river mouths in tropical and subtropical regions of the world, mainly in the intertidal zone. Hence, the ecosystem and its biological components are under the influence of both marine and freshwater conditions. It has developed a set of physiological adaptations to overcome problems of anoxia, salinity, and frequent tidal inundations. This leads to the assemblage of a wide variety of plant and animal species with special adaptations suited to that ecosystem. The carbon fixed in mangroves is highly important in the coastal food webs, the litter from mangroves, and the subsequent formation of detritus and its tidal export have a profound effect on promoting the rich biodiversity. Mangroves are ecological habitats for a variety of microorganisms. They also play a valuable role in foreshore production, reducing erosion from cyclones and lessening the effects of storm surges.

Mangrove mud is rich in nutrients. Bacteria, fungi, and algae thrive on debris such as fallen mangrove leaves that are washed down by the river. These in turn will feed creatures that are present higher in the food chain. The roots themselves are covered by a vast range of sponges, anemones, marine algae, seaweed, oysters, and barnacles. These form the rich mangrove community. Mangrove microorganisms are an untapped source of metabolites and products with novel properties. Bacteria perform varied activities in the mangrove ecosystem such as photosynthesis, nitrogen fixation, methanogenesis, production of antibiotics and enzymes of specific activities (arylsulfatase, L-glutaminase, chitinase, L-asparginase, cellulase, protease, phosphatase), and others.

Chhiaki and coworkers (2007) stated that microorganisms that are present in the mangrove environment produce naturally and traditionally biologically active compounds, namely, bioactive substances. These natural organic compounds produced by microorganisms are important in screening the target for a variety of bioactive substances. Biologically active compounds produced by microorganisms in the mangrove may be a source for producing new bioactive compounds (Kokare et al. 2004a,b).

Bacteriological studies in the mangrove ecosystems are quantitative and a serious effort is necessary to identify the bacteria (Kathiresan et al. 1990). Most of the marine bacteria belong to the Gram-negative group, and the Gram-positive bacteria make up less than 10% of the total bacterial population (Zobell and Upham 1944). Now, evidence indicates that Gram-positive bacteria do occur at a higher percentage in sediments. Marine environments are a largely untapped source for the isolation of new microorganisms with the potential to produce active secondary metabolites (Baskaran et al. 2011).

It is now accepted that actinobacteria can be indigenous to the marine environment. It is promising to yield many unusual actinobacteria that have great potential as producers of novel antibiotics and other compounds. These actinobacteria are important in the field of pharmaceutical industries and also agriculture. Because of the discovery of novel antibiotic and nonantibiotic lead molecules through microbial secondary metabolite screening, it is becoming increasingly important.

Members of the marine actinobacteria are poorly understood and few reports are available regarding the existence of actinobacteria from mangroves (Lakshmanaperumalsamy 1978; Ratnakala and Chandtika 1993; Vikineshwari et al. 1997; Sivakumar 2001). Searching for novel actinobacteria constitutes an essential component in natural product-based drug discovery.

The marine sediment of India, especially in the Andaman and Nicobar Islands, have potentially enormous microbial biodiversity. Surprisingly, they have not been extensively explored for the registration of novel actinobacteria and its bioactive compounds. Keeping these in mind and recognizing the significance of marine actinobacteria as a source of novel bioactive compounds, the present study was made in an effort to isolate and screen for antifungally active *Streptomycetes* from the mangrove sediments collected from Andaman Islands, India.

The analysis of cellular fatty acid profiles in *Streptomyces* spp. has been well studied (Bowers et al. 1996; Seong et al. 2001; Magervey et al. 2004). Rapidly growing pathogenic aerobic actinobacteria was separated into two large groups based on the presence of a major amount of branched chain or of saturated or unsaturated straight-chain fatty acids (McNabb et al. 1997).

### 2.3.1 Isolation of Streptomyces spp.

The sediment samples were air-dried and ground aseptically with a mortar and pestle. The powdered samples were mixed thoroughly and passed through a 2-mm sieve to remove gravel and debris. The sample was kept at 70°C for 15 min (Hayakawa and Nonomura 1987; Hayakawa et al. 1991; Seong et al. 2001). Then, 10-fold serial dilutions of the sediment samples were made using sterile 50% seawater. The culture medium was prepared by using 50% seawater and contained the antibiotics cycloheximide and nalidixic acid (80 and 75 µg/mL, respectively; HiMedia, Mumbai) to prevent other bacterial and fungal growth (Baskaran et al. 2011). Approximately 0.1 mL of the serially diluted samples were spread over Kuster's agar containing glycerol (10 mL), casein (0.3 g), KNO<sub>3</sub> (2 g), K<sub>2</sub>HPO<sub>4</sub> (2 g), NaCl (2 g), MgSO<sub>4</sub>·7H<sub>2</sub>O (0.05 g), CaCO<sub>3</sub> (0.02 g), FeSO<sub>4</sub> (0.01 g), agar (18 g), seawater (500 mL), and distilled water (500 mL at pH 7  $\pm$  2).

The plates were incubated at  $28^{\circ}C \pm 2^{\circ}C$ . After 15 days, the actinobacteria colonies grown on Petri plates were counted and recorded. The actinobacterial colony was purified on yeast extract–malt extract agar medium ISP2 (International *Streptomyces* Project, as described by Shirling and Gottlieb 1966) by streak plate technique and were maintained in ISP2 agar slants for further investigation. The isolate was identified based on the colony morphology, aerial mycelia, substrate mycelia, and microscopic examination, which was subsequently used for further study.

# 2.3.2 *Primary screening for antifungal activity*

The Streptomycetes was screened using a cross streak method (Lemos et al. 1985) for antagonistic activity. The strain was streaked on potato dextrose agar medium and incubated at  $28^{\circ}C \pm 2^{\circ}C$ . After observing a good ribbon-like growth of the *Streptomyces* on the Petri plates, the fungal pathogens such as *A. niger, Aspergillus flavus, Aspergillus fumigates, Penicillium* sp. *Fusarium* sp., and *Candida magnolia* were streaked at right angles to the original streak of *Streptomyces* and incubated at  $28^{\circ}C \pm 2^{\circ}C$ . The inhibition zone was measured after 48 to 72 h of incubation.

# 2.3.3 In vitro assay for antifungal activity

Streptomyces was inoculated into a 500 mL conical flask containing 200 mL of production medium containing dextrose (20 g), soya bean (20 g), soluble starch (5 g), peptone (5 g),  $(NH_4)_2SO_4$  (2.5 g),  $MgSO_4 \cdot 7H_2O$  (0.25 g),  $K_2HPO_4$  (0.024 g), NaCl (4 g),  $CaCO_3$  (2 g), seawater (500 mL), and distilled water (500 mL, pH 7 ± 2) and incubated at 28°C ± 2°C for 7 days in rotary shaker at 250 rpm. The fermented broth was centrifuged at 10,000 rpm at 4°C for 20 min and the supernatant was filtered using a 0.45 µm membrane filter (Millipore).

An equal volume of ethyl acetate, ethanol, methanol, chloroform, and hexane was added separately to the cell-free culture filtrates and shaken for 12 h. The solvent extracts were evaporated in aseptic conditions and the crude powder was collected as the antimicrobial compound. This powder was mixed with dimethyl sulfoxide (DMSO) to evaluate its activity against pathogens (Sambamurthy and Ellaiah 1974). One yeast-like fungus and five filamentous fungi were inoculated into Sabouraud's dextrose broth and incubated for 24 to 48 h. The test organisms were swabbed over on the surface of Sabouraud's dextrose media. After solidification, wells were made using sterile cork borer. Then, the extracts were added separately in the wells in triplicate, and incubated for 48 to 72 h. After incubation, the zone of inhibition around the well was measured as millimeters in diameter. The maximum zone of inhibition obtained from ethyl acetate extract of *Streptomyces parvulus* DOSMB-D105 against *A. niger* was 25.66 mm, followed by *C. magnolia* (20.66 mm), *A. flavus* (20.33 mm), *A. fumigatus* (19.33 mm), *Penicilium* sp. (18.66 mm), and *Fusarium* sp. (17.33 mm; Figure 2.1).

Out of five different solvents used to extract the antimicrobial compounds of streptomycetes, only ethyl acetate extract possesses remarkable activity against all the pathogens tested. Whereas the methanol, chloroform, ethanol, and hexane extract exhibited moderate to minimum activity. They did not show inhibition to all the pathogens tested. Methanol extract of *S. parvulus* DOSMB-D105 showed moderate activity against 11 pathogens tested followed by chloroform, hexane, and ethanol (Table 2.5).

Ethyl acetate extract from both *Streptomyces* spp. showed maximum activity against all the pathogens when compared with standard antibiotic (erythromycin). The present

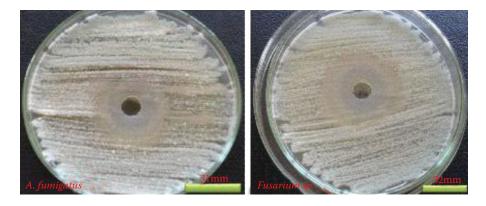


Figure 2.1 Antifungal activities of ethyl acetate extract of Streptomyces parvulus DOSMB-D105.

					Ethyl		Control
S. No	Pathogens	Methanol	Chloroform	Ethanol	acetate	Hexane	DMSO
1	A. niger	23	_	14.66	25.66	18.66	3.33
2	A. flavus	18.66	13.33	-	20.33	16	_
3	A. fumigatus	16.33	-	10.66	19.33	14.66	3.66
4	Penicillium sp.	15.66	13	14	18.66	13.33	_
5	<i>Fusarium</i> sp.	14.33	11	11.66	17.33	-	3
6	C. magnolia	17.66	12.66	12.66	20.66	14.33	_

Table 2.5 Antifungal Activity of S. parvulus DOSMB.D105 Extracted by Different Solvents

results suggest that the *Streptomyces* sp. DOSMB-D105 has potential antibiotic activity against human pathogens when compared with standard antibiotics. Therefore, the ethyl acetate extract of *Streptomyces* spp. were forwarded for GC-MS analysis for the detailed investigation of antimicrobial compounds.

Similarly, Vijayakumar and coworkers (2011) have reported on the antimicrobial activity of different solvent extracts of *Streptomyces* sp. (VPTSA18) against two species of fungi. The ethyl acetate extract was highly active against *C. albicans* (17 mm), the other solvent extracts had moderate to minimum inhibitory effects against the test pathogens. This also confirmed the efficacy of the extracted compounds from the present study.

Researchers evaluated the antimicrobial efficacies of the actinobacteria using separate solvents including *n*-butanol, chloroform, ethyl acetate, and methanol; ethyl acetate, methanol, chloroform, and alcohol; and petroleum ether, *n*-butanol, and ethyl acetate (Sahin and Ugur 2003; Thangadurai et al. 2004; Ilic et al. 2005; Taechowisan et al. 2005; Remya and Vijayakumar 2008; Vimal et al. 2009; Usha et al. 2010), and they have reported that the ethyl acetate solvent extract had promising activity against most pathogens.

Similar types of studies have also been reported by many researchers (Saadoun and Al-Momani 2000; Balagurunathan and Subramanian 2001; Bordoloi et al. 2001; Narayana et al. 2005). The above reports confirm the present approach of the ethyl acetate solvent for the extraction of antimicrobial compounds from marine actinobacteria.

### 2.3.4 Analysis of fatty acids using the FAME method

Relationships among the Streptomyces isolates was evaluated based on the analysis of cellular fatty acids (Kroppenstedt 1985). Streptomyces isolates were grown on starch casein agar for 7 to 10 days at  $28^{\circ}C \pm 2^{\circ}C$ . Spores were scraped from the surface of the medium and inoculated into 20 mL of starch casein broth in clean conical flask containing approximately 5.0 g of 3 mm diameter glass beads. The culture was incubated on a rotary shaker at 225 rpm at  $28^{\circ}C \pm 2^{\circ}C$  for 72 h. Mycelia were harvested by centrifugation (10 min at 4000 rpm on a benchtop centrifuge) and the growth medium was decanted. Approximately 200 to 220 mg wet weight of mycelia was placed in a screw cap test tube and the samples were processed. In this process, cellular fatty acids were saponified, and methylated to form fatty acid methyl esters (FAME). FAME was then extracted from the aqueous phase into an organic phase, washed, and stored in the freezer until analysis. FAME was separated using a Hewlett-Packard 5890, a gas chromatograph fitted with column DEGC flame ionization detector. Chromatography output of FAME peak was analyzed, named, measured, and expressed in milligrams of the total fatty acid content. The fatty acid profile was determined by gas chromatography. The fatty acid retention times, responses, area/ heights, response factors, equivalent chain lengths, peak names, and percentage shown by gas chromatography for both *Streptomyces* sp. is depicted in Figure 2.2 and Table 2.6.

*Streptomyces* sp. DOSMB-A105 had 26 different types of fatty acids in their cell. Among them, seven were saturated fatty acids and 19 were unsaturated fatty acids (Table 2.7). The strain DOSMB-D105 showed similar indices of 0.538 for *Streptomyces lavendulae*.

Analysis of the fatty acid profile to characterize taxa at the generic and suprgeneric level have certain limitations. However, quantitative analysis of fatty acid profiles provided useful taxonomic information at the species and, in some cases, at the subspecies level. Differences in fatty acid composition influenced membrane functions, particularly membrane permeability, which consequently could favor the synthesis of antimicrobial compounds (Gesheva and Rachev 1998, 2000; Kinked et al. 1998).

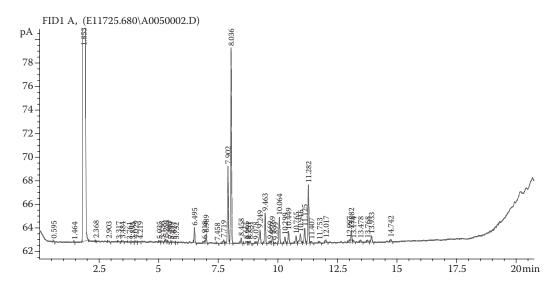


Figure 2.2 GC spectrum of potential strain DOSMB-D105 for fatty acid analysis.

RT	Response	Ar/Ht	RFact	ECL	Peak name	%	Comment 1	Comment 2
0.595	408	0.027	_	4.227		_	< min rt	
1.464	145	0.015	-	6.168		-	< min rt	
1.853	4.07E+8	0.029	-	7.036	SOLVENT PEAK	-	< min rt	
2.368	357	0.022	-	8.186		-	< min rt	
2.903	574	0.041	1.264	9.381	8:0 3OH	0.31	ECL deviates -0.004	
3.761	328	0.032	1.126	10.930	Sum In Feature 3	0.16	ECL deviates 0.002	unknown 10.928
5.291	1072	0.030	1.024	12.613	13:0 ISO	0.47	ECL deviates 0.001	Reference -0.003
5.384	595	0.032	1.020	12.701	13:0 ANTEISO	0.26	ECL deviates 0.000	Reference -0.004
5.499	200	0.027	-	12.809		_		
5.627	634	0.053	1.010	12.928	13:1 AT 12-13	0.27	ECL deviates –0.003	
6.495	6106	0.038	0.984	13.616	14:0 ISO	2.57	ECL deviates -0.002	Reference -0.005 (continued)
								(continueu)

Table 2.6 GC data of Fatty Acids Analysis of Streptomyces sp. DOSMB-D105

Table 2.6 (Continued) GC Data of Fatty Acids Analysis of Streptomyces sp. DOSMB-D105

RT	Response	Ar/Ht	RFact	ECL	Peak name	%	Comment 1	Comment 2
6.989	3712	0.040	0.971	14.000	14:0	1.54	ECL deviates 0.000	Reference -0.003
7.458	188	0.022	-	14.319		-		
7.719	2034	0.045	0.958	14.497	unknown 14.503	0.83	ECL deviates -0.006	
7.902	35,447	0.042	0.955	14.622	15:0 ISO	14.50	ECL deviates 0.001	Reference -0.002
8.036	87,321	0.042	0.953	14.713	15:0 ANTEISO	35.64	ECL deviates 0.002	Reference -0.001
8.458	2402	0.043	0.947	15.000	15:0	0.97	ECL deviates 0.000	Reference -0.003
9.078	814	0.034	0.940	15.386	16:1 ISO E	0.33	ECL deviates 0.000	
9.249	7860	0.051	0.939	15.492	Sum In Feature 3	3.16	ECL deviates 0.004	14:0 30H/16:1 ISO I
9.463	13,928	0.043	0.937	15.625	16:0 ISO	5.59	ECL deviates –0.001	Reference -0.004
9.669	1381	0.039	0.935	15.753	16:1 A	0.55	ECL deviates –0.004	
9.769	3872	0.043	0.934	15.815	16:1 CIS 9	1.55	ECL deviates –0.002	
9.899	1032	0.056	_	15.896		_		
10.064	13,131	0.046	0.932	15.999	16:0	5.24	ECL deviates –0.001	Reference -0.005
10.299	3518	0.048	0.931	16.138	15:0 ISO 3OH	1.40	ECL deviates 0.003	
10.449	5680	0.041	_	16.227		_		
10.765	4093	0.047	0.928	16.415	16:0 9? METHYL	1.63	ECL deviates –0.001	
10.946	6860	0.061	0.927	16.523	17:1 ANTEISO C	2.72	ECL deviates –0.002	
								( (; 1)

RT	Response	Ar/Ht	RFact	ECL	Peak name	%	Comment 1	Comment 2
11.125	8073	0.050	0.926	16.629	17:0 ISO	3.20	ECL deviates 0.000	Reference -0.005
11.282	28,577	0.046	0.926	16.722	17:0 ANTEISO	11.33	ECL deviates 0.000	Reference -0.005
11.407	862	0.039	0.925	16.796	17:1 CIS 9	0.34	ECL deviates 0.004	
	732	0.042	0.924	17.001	17:0	0.29	ECL deviates 0.001	Reference -0.004
12.017	1579	0.047	0.923	17.154	16:0 ISO 30H	0.62	ECL deviates 0.009	
12.997	1256	0.046	0.921	17.721	Sum In Feature 6	0.50	ECL deviates 0.001	18:2 CIS 9,12/18:0a
13.082	5733	0.050	0.921	17.771	18:1 CIS 9	2.26	ECL deviates 0.002	
13.174	773	0.032	0.921	17.823	Sum In Feature 7	0.30	ECL deviates 0.001	18:1 CIS 11/t 9/t 6
13.478	1200	0.039	0.921	18.000	18:0	0.47	ECL deviates 0.000	Reference -0.007
13.768	1123	0.039	0.921	18.168	17:0 ISO 30H	0.44	ECL deviates 0.004	
13.933	3562	0.051	_	18.263		_		
14.742	1377	0.043	0.920	18.732	19:0 ANTEISO	0.54	ECL deviates 0.003	
-	8189	-	-	-	Summed Feature 3	3.32	12:0 ALDE?	unknown 10.928
-	-	-	-	-		-	16:1 ISO I/14:0 3OH	14:0 30H/16:1 ISO I
-	1256	_	-	_	Summed Feature 6	0.50	18:2 CIS 9,12/18:0a	18:0 ANTEISO/ 18:2 c
-	773	_	-	_	Summed Feature 7	0.30	18:1 CIS 11/t 9/t 6	18:1 TRANS 9/ t6/c11

Table 2.6 (Continued) GC Data of Fatty Acids Analysis of Streptomyces sp. DOSMB-D105

*Note:* ECL deviation, 0.003; reference ECL shift, 0.004; number of reference peaks, 13; total response, 267,477; total named, 246,967; percent named, 92.33%; total amount, 233,575. Matches: library, ACTIN1 3.80; similar index, 0.538; and entry name, *Streptomyces lavendulae*.

Peak name/Fatty				
acid name	name Systemic name		Saturated/Unsaturated	
13:0 ISO	Isotridecanioc acid	+	Unsaturated	
13:0 ANTEISO	Anteisotridecanioc acid	+	Unsaturated	
14:0 ISO	Isotetradecanioc acid	+	Unsaturated	
15:0 ISO	Isopentadecanioc acid	+	Unsaturated	
15:0 ANTEISO	Anteisopentadecanioc acid	+	Unsaturated	
16:0 ISO	Isopalmitic acid	+	Unsaturated	
16:0	Palmitic acid	+	Saturated	
15:0 ISO 3OH	Trihydroxydecanioc acid	+	Unsaturated	
17:1 ANTEISO C	Anteisoheptadecanoic acid	+	Unsaturated	
17:0 ISO	Isoheptadecanoic acid	+	Unsaturated	
17:0 ANTEISO	Anteisoheptadecanoic acid	+	Unsaturated	
16:0 ISO 3OH	Isotrihydroxyshexadecanoic	+	Unsaturated	
	acid			
18:1 CIS 9	Oleic acid	+	Unsaturated	
18:0	Stearic acid	+	Saturated	
17:0 ISO 3OH	Isotrihydroxyheptadecanoic acid	+	Unsaturated	
19:0 ANTEISO	Anteisononadecylic acid	+	Unsaturated	
08:03 OH	Trihydroxyoctanoic acid	+	Unsaturated	
13:1 AT 12-13	Tridecenoic acid	+	Saturated	
15:0	Pentadecylic acid	+	Saturated	
16:1 ISO E	Isopalmitoleic acid	+	Unsaturated	
16:1 A	Palmitoleic acid	+	Unsaturated	
16:09 METHYL	9-Methyltetradecanoate	+	Saturated	
14:0	Myristic acid	+	Saturated	
17:1 CIS 9	cis-9-Heptadecenoic acid	+	Unsaturated	
17:0	Margaric acid	+	Saturated	
16:1 CIS 9	cis-9-Hexadecenoic	+	Unsaturated	

Table 2.7 Presence of Saturated and Unsaturated Fatty Acid in S. parvulus DOSMB-D105

# 2.4 Conclusion

New antibiotic compounds with novel mechanisms of action have become essential in response to the recent paradigm of increasingly resistant strains of bacteria. Secondary metabolites as natural and bioactive compounds from fungus, cyanobacteria, and bacteria such as actinobacteria serve as lead compounds for the development of pharmaceutical drugs. These drugs are widely used to fight bacterial, viral, and fungal infections, as well as cancer and immune system disorders.

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# chapter three

# *Metagenomics as advanced screening methods for novel microbial metabolites*

# Sakshi Sharma and Jyoti Vakhlu

# Contents

3.1	Introc	uction				
3.2	Metag	enomes as source of novel metal	oolites			
3.3						
	3.3.2 Metagenomic DNA extraction					
		3.3.2.2 Indirect DNA extractio	n			
	3.3.3	Screening metagenomes for nov	vel metabolites			
	3.3.4					
			entation of host strains or mutants			
		3.3.4.2 Phenotypical detection	of the desired activity			
			expression			
	3.3.5					
	3.3.6	Metagenomic data analysis		55		
	3.3.7					
Refe	erences					

# 3.1 Introduction

Metagenomics is the study of metagenomes, that is, microbial genetic material recovered directly from environmental samples, and is broadly referred to as environmental or community genomics. Metagenomics is used as a means of systematically investigating, classifying, and manipulating the entire microbial genetic material isolated from environmental samples (Zeyaullah et al. 2009). Metagenomics combines the power of genomics, bioinformatics, and systems biology. The term *metagenomics* was first used by Jo Handelsman and coworkers in 1998 and they defined it as "cloning and functional analysis of collective microbial genomes" (Handelsman et al. 1998). Recently, the definition has been modified by Kevin Chen and Lior Pachter at University of California, Berkeley as "the application of modern genomics techniques to the study of communities of microbial organisms directly in their natural environments, bypassing the need for isolation and lab cultivation of individual species" (Chen and Pachter 2005).

The emergence of metagenomics, as a field to study microbial diversity, was a consequence of the discovery of the fact that the microbial diversity in any environment is much greater than previously realized and accessed by cultivation. This unharvested diversity provides an unlimited source for novel microbial metabolites and some of them may find a place in the market. Considering that more than 99% of microorganisms in most environments are not amenable to cultivation, very little is known about their genomes, genes, and encoded enzymatic activities (Ferrer et al. 2005).

Thus, many of these microbes remain unnoticed and the potential of these microbes remains hidden, which limits the spectrum of search for novel metabolites. The metagenomic technologies developed are used to complement or replace culture-based approaches and bypass some of their inherent and established limitations. The isolation and analysis of environmental DNA has enabled the screening of microbial diversity, allowing access to their genomes, enabling the identification of protein coding sequences, and even the reconstruction of biochemical pathways and genomes, providing insights into the properties and functions of these organisms (Kennedy et al. 2008).

This rapidly growing research area provides new insights into microbial life and access to novel biomolecules. The construction and analysis of metagenomic libraries is thus a powerful approach to harvest and archive environmental genetic resources. It provides a way to identify which organisms are present, what they do, and how their genetic information can be beneficial to mankind. It can also be complemented with metatranscriptomic or metaproteomic approaches to describe expressed activities (Gilbert et al. 2008; Wilmes and Bond 2006). Metagenomics is also a powerful tool for generating novel hypotheses related to microbial functions; the remarkable discoveries of proteorhodopsinbased photoheterotrophy attest to this fact (Beja et al. 2000).

One of the main application areas of metagenomics is in the mining of metagenomes for genes encoding novel metabolites. Due to the complexity of most metagenomes, new sensitive and efficient high-throughput screening techniques that allow for fast and reliable identification of genes encoding beneficial novel metabolites from complex metagenomes have been invented. Screens of metagenomic DNA have been based either on nucleotide sequence as a sequence-driven approach or on metabolic activity as a functiondriven approach (Hall 2007; Kennedy et al. 2008; Li et al. 2009; Simon et al. 2009; Uchiyama and Watanabe 2007; Yun and Ryu 2005).

The key factor on which the success of the metagenomic process depends is the extraction of metagenomic DNA from an environmental sample (Purohit and Singh 2008). Once an appreciable amount of eDNA has been obtained, further analysis is based on two approaches: function-based screening and sequence-based screening. Both functionbased and sequence-based screening have individual advantages and disadvantages, and they have been applied successfully for the discovery of novel metabolites from the metagenome.

# 3.2 Metagenomes as source of novel metabolites

The total number of microbial cells on Earth is estimated to be 10<sup>30</sup> (Turnbaugh and Gordon 2008). Prokaryotes represent the largest proportion of individual organisms, comprising 10<sup>6</sup> to 10<sup>8</sup> separate genospecies (Sleator et al. 2008). They are the most ubiquitous organisms on Earth, and are represented in all habitats including soil, sediment, marine and terrestrial subsurfaces, animals, and plant tissues, which represent an enormous reservoir of novel valuable molecules for health or industry. Microbes harbor a far greater genetic and metabolic diversity than plants and animals. Microbial diversity is a precious source for modern biotechnology because it has the potential to provide innovative and sustainable solutions to a broad range of important and hard-to-solve problems of modern society. Because it is estimated that only 0.1% of the bacterial and 5% of the total fungal species has been reported, there exists a huge potential of finding new and important metabolites.

Moreover, among the microbes already studied, only a few have been explored with respect to metabolite production. Bioactive metabolites may be extracted from microbes isolated from diverse habitats. A crude or untreated extract from microbes typically contains novel and structurally diverse metabolites. The effort to search for these biologically active metabolites is known as *bioprospecting*. The probability of finding novel products from microbes depends on the number of strains being explored and the level of diversity between the explored strains, their uniqueness, and their ability to produce metabolites. These factors must be seriously thought of before screening metagenomes for bioactive metabolites. Because a large number of microbial products have already been discovered, exploiting microbes for novel metabolites is a challenging endeavor. However, microorganisms possess a remarkable ability to synthesize novel metabolites and variants of known metabolites under different conditions and environments. This ability of microbes results in a vast number of structurally original and potent bioactive compounds that are otherwise difficult or impossible to obtain through chemical methods. Biological and geographical diversity is responsible for the diverse nature of metabolites produced by microbes isolated from different habitats.

# 3.3 Prospecting from metagenomes

The techniques for the recovery of novel biomolecules from environmental samples comprise the selection of the environmental samples, extraction of metagenomic DNA, and cloning and construction of metagenomic libraries. This is followed by screening the libraries using various methods described later in the chapter. Figure 3.1 shows the overall process involved in the search of novel metabolites from microbial metagenomes.

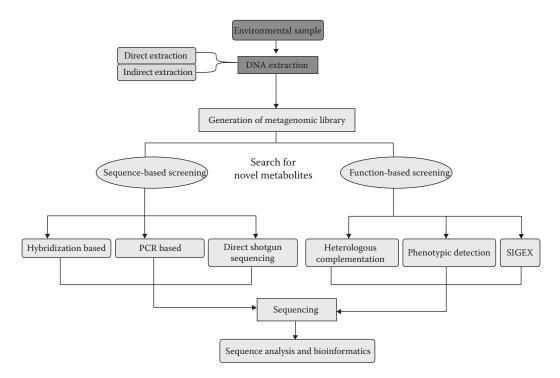


Figure 3.1 Different approaches for metagenome mining for new metabolites.

# 3.3.1 Sampling

Because there is hardly any place on Earth where microbes are not present, metagenomic sampling can be done from both abiotic as well as biotic habitats. Abiotic habitats include air, soil, or water including both marine and freshwater sources, whereas biotic habitats consist of animals, insects, plants, and others. Researchers are constantly looking forward to new unexplored environments that are likely to provide new insights into genetic and metabolic diversity. A step forward in this direction is the exploitation of extreme environments such as hot springs, geographical poles, very dry deserts, volcanoes, deep ocean trenches, cold habitats, and so on for the detection of extremophiles producing novel metabolites (Schiraldi and De Rosa 2002). It is now well established that microbes not only thrive in these previously considered hostile environments but are also metabolically active, producing various metabolites (Herbert 1992). To survive in such conditions, microorganisms have developed unique defenses and this has frequently led them to produce novel molecules. This capability had made them a mine of natural and novel metabolites (Simon and Daniel 2011).

Metagenomic studies fall into three main categories in terms of their selection of environments (Steele et al. 2008).

- i. The first category selects highly diverse environmental samples such as soil and marine waters. The sample is either directly subjected to DNA extraction or the sample is first enriched with microbes producing the target metabolite.
- ii. The second category selects environments naturally enriched for the target biocatalyst such as the search for xylanases in insect gut (Brennan et al. 2004).
- iii. The third category targets extreme environments for metabolites active under extreme conditions.

These metabolites may be active under extreme conditions, but the variation may be quite limited such as alkalophilic enzymes, which function only over a narrow alkaline pH range, because the extremophiles produce metabolites that are functional only in those particular environmental conditions. Highly genomically diverse environments, which are naturally less extreme and more heterogeneous in terms of being subjected to greater fluctuations on temperature, pH or salinity, may provide biocatalysts that show a greater level of stability and activity over a wider range of conditions (Steele et al. 2008).

# 3.3.2 Metagenomic DNA extraction

Microbial communities from any habitat are composed of a mixture of microbes having different cell wall characteristics and thus having variable susceptibility to lysis. This implies that some special methods are required for the lysis of diverse bacterial cells in a sample to extract their genomes simultaneously. The isolation of appreciable amounts of quality environmental DNA is one of the key factors responsible for the success of a metagenomic process (Kennedy et al. 2007; Sharma et al. 2007; Voget et al. 2003). The metagenomic DNA extracted should be representative of all cells present in the sample and should be manipulable at the molecular level. For successful isolation of DNA from diverse environmental samples, different isolation protocols have been developed and some of these protocols are commercially available as DNA extraction kits (Rondon et al. 2000; Purohit and Singh 2008). The methods vary with respect to purity and quantity vis-a-vis degradation and inhibitors present in the isolated DNA. All protocols rely on the basic principle of using enzymes or chemicals and hot detergent treatments to disrupt the cells (Rondon et al. 2000). Lysozyme and sodium dodecyl sulfate (SDS) are most commonly used to lyse the cells. Several protocols also recommend the application of mechanical forces generated by bead beating, freeze-thawing, and sonication methods to disrupt the rigid cell structure (Kennedy et al. 2007; Voget et al. 2003). Metagenomic DNA extraction from one site with more than one isolation protocol is recommended for the maximum representation of the microbial community (Delmont et al. 2011). The metagenomic extraction protocols can be categorized broadly into two types: direct extraction and indirect extraction (Gabor et al. 2003; Purohit and Singh 2008).

# 3.3.2.1 Direct DNA extraction

Direct DNA extraction is an *in situ* extraction, in which the cells are lysed within the environmental sample matrix. The DNA is then separated from the sample matrix and cell debris. This method provides a less biased extraction of microbial DNA with high nucleic acid yield, and is preferred when large quantities of nucleic acids are required. However, the resulting nucleic acid extracts are commonly sheared and contaminated. Furthermore, extracts often contain unknown amounts of extracellular or eukaryotic DNA (or both). Currently, two types of lysis methods are used either alone or in combination. These are the soft and harsh lysis methods (Purohit and Singh 2008).

The soft lysis method is based on the disruption of microorganism solely by enzymatic and chemical means. Many protocols based on enzymatic lysis have been developed. Lysozyme treatment is one of the most common (Niemi et al. 2001; Rochelle et al. 1992; Tebbe and Vahjen 1993). Another enzyme, achromopeptidase, was used to improve the lysis of the recalcitrant Gram-positive bacteria *Frankia*, whereas proteinase K has been used to digest contaminating proteins (Rochelle et al. 1992; Simonet et al. 1984; Zhou et al. 1996). Proteinase K is a broad-spectrum protease digesting most of the proteins. Chemical methods are extensively used either alone or in combination with physical methods. The most common chemical used is a detergent SDS, which dissolves the hydrophobic material of cell membranes. Detergents have often been used in combination with heat treatment and with chelating agents such as EDTA. The addition of cetyltrimethylammonium bromide or cetrimonium bromide (CTAB) and polyvinylpolypyrrolidone (PVPP) has also been reported. CTAB forms insoluble complexes with denatured proteins, polysaccharides, and cell debris. PVPP is used to remove polyphenols (Desai and Madamwar 2007; Verma and Satyanarayana 2011).

On the other hand, the harsh lysis approach involves mechanical cell disruption by bead beating, sonication, freeze-thawing, and grinding. Bead beating is an easy method based on the ballistic disintegration of the cells (Leff et al. 1995; Siddhapura et al. 2010). Tiny beads made of glass, ceramic, or steel are mixed with the sample already suspended in aqueous media, and subjected to agitation by stirring or shaking. The efficiency of cell disruption depends on the time of agitation and bead size. Sonication uses high-frequency ultrasound ranging from 20 to 50 kHz for disrupting the cells. The efficiency of sonication depends on the energy input (Siddhapura et al. 2010). Even under optimized conditions, harsh treatment may sometimes result in shearing of high molecular-weight DNA, low yields, and small fragment sizes.

# 3.3.2.2 Indirect DNA extraction

Indirect DNA extraction involves the separation of microbial cells from the sample matrix prior to cell lysis. Although time-consuming, indirect extraction methods are preferred for targeting prokaryotic communities when high DNA purity is required for inhibitor-sensitive methods, and when the recovery of high molecular-weight DNA is necessary (Robe et al. 2003). Indirect DNA extraction methods yielded 10-fold to 100-fold lower amounts of DNA compared with direct procedures (Steffan et al. 1988), but the bacterial diversity of DNA recovered by indirect means was distinctly higher (Gabor et al. 2003; Leff et al. 1995). The indirect DNA extraction protocols involve blending cation-exchange methods or other techniques such as the use of super paramagnetic silica–magnetite nanoparticles for the isolation and purification of DNA from the samples (Gabor et al. 2003; Gupta and Vakhlu 2012; Jacobsen et al. 1992).

Most DNA extraction protocols require standardization because the nature of different environmental samples varies with respect to their matrices, organic and inorganic compounds, and biotic factors. Environmental samples are heterogeneous and contain different substances, which may sometimes interfere with the extraction of quality DNA. Thus, it is quite obvious that the extraction procedures would have to be case specific. For example, soil is a complex matrix containing many substances such as humic acids, which can be coextracted during DNA isolation. It is thus essential to modify the extraction protocol in such a way so as to remove these acids from the extracted DNA before it can be further processed. Optimizing the DNA extraction protocols also reduces the bias caused by unequal lysis of different members of the microbial community. Improved DNA extraction techniques could help ensure a metagenomic library that efficiently represents the entire community's genome without inhibitory substances.

Inhibitor-free metagenomic DNA of high quality and high yield is essential for the success of metagenomics. Various inhibitory substances extracted along with the metagenome are a major problem because these compounds hinder the molecular biological manipulation such as DNA digestion, ligation, and polymerase activity and hence the DNA needs to be purified of such inhibitors. Some examples of such inhibitors include the humic acid and polyphenols present in soil samples. Samples from polluted environments consist of humic and fulvic acids, along with a large number of other organic pollutants, metal ions, and chemical impurities as inhibitors (Fortin et al. 2004; Hinoue et al. 2004; Sharma et al. 2007; Verma and Satyanarayana 2011). The phenolic groups in humic acids denature biological molecules by bonding to amides or are oxidized to form a quinone, which covalently bonds to DNA (Young et al. 1993). In most studies, cell lysis is followed by the purification of DNA using organic solvent extraction. This is done either by using phenol-chloroform or phenol-chloroform-isoamyl alcohol as solvents. This preliminary purification step is followed by the precipitation of DNA using ethanol, isopropanol, or polyethylene glycol (Steffan et al. 1988). These basic purification steps have been found to be insufficient to remove contaminants from soil metagenome. Many other purification methods have been tried by various workers to remove humic acids. These include gel elution, gel filtration (using Sephadex G150; Sepharose 2B, 4B, and 6B; and Biogel P100 and P200), ion exchange (using Q-Sepharose), and activated charcoal and amberlite resins, which are efficient removers of metal ions, organics, and humic acids (Harry et al. 2000; Jackson et al. 1997; More et al. 1994; Rochelle et al. 1992; Sharma et al. 2007; Verma and Satyanarayana 2011). Using all these methods for removing contaminants is time-consuming and laborious, hence new techniques need to be developed for obtaining inhibitor-free DNA.

One solution to this problem is the use of multiple displacement amplification (MDA) for whole genome amplification from single cells, which will allow the study of the entire biochemical potential of single uncultured microbes from complex microbial communities (Stepanauskas and Sieracki 2007; Woyke et al. 2009). In this approach, microbial cells are first sorted using fluorescence-activated cell sorting (FACS) and then collected. Then, MDA is used to amplify the entire genome from the sorted cells. The amplified genomic DNA can then be analyzed to access the full metabolic potential of the microorganism. This approach has great potential for the discovery of novel enzymes, as it affords easier access to rare microbiota.

#### 3.3.3 Screening metagenomes for novel metabolites

Once the manipulable metagenomic DNA is isolated, it can be analyzed for microbial community dynamics, construction of novel microbial genomes, and the isolation of novel genes and gene products for screening novel metabolites.

The focus of the present chapter is on the use of metagenomics as a tool for the isolation of novel metabolites. Hence, we restrict ourselves only to novel metabolite screening using metagenomics. For metabolite screening, cloning and expression of metagenomic DNA is compulsory in contrast to microbial diversity analysis or genome reconstruction. The extracted metagenomes can be screened for the presence of genes encoding novel metabolites using two different approaches: *sequence-based screening* and *function-based screening* (Hall 2007; Kennedy et al. 2008; Li et al. 2009; Simon et al. 2009; Uchiyama and Watanabe 2007; Yun and Ryu 2005).

Although sequence-based metagenomic approaches rely on comparing sequence data, which is obtained with sequences deposited in databases, functional metagenomics focuses on screening DNA library clones directly for a phenotype, that is, the genes are recognizable by their function rather than by their sequence. The power of such an approach is that it does not require the genes of interest to be recognized by already existing sequences. This ensures the identification of entirely new classes of genes for both known and indeed novel functions. Moreover, the results from a functional metagenomics approach are unambiguous. However, for the approach to be successful, specific and sensitive assays should be available and heterologous gene expression is still a bottleneck.

Sequence-based metagenomic approaches depend on both cloning-dependent as well as cloning-independent methods, whereas function-based approach is cloning-dependent. Cloning-based screening of metagenomes requires the generation of libraries. DNA and cDNA as well as specific amplicons can be used as inserts to generate libraries. These libraries are known as metagenomic, metatranscriptomic, and amplicon libraries, respectively. Table 3.1 shows some examples of the products derived from metagenomic libraries along with the screening methods.

Once the DNA has been isolated from the sample and purified, the next step is the construction of DNA libraries. Successful generation of DNA libraries require suitable cloning vectors and host strains and the selection of host–vector system depends primarily on the metabolite to be isolated. The most common approach used includes the generation of small insert libraries in a standard sequencing vector and in *Escherichia coli* as a host strain (Henne et al. 1999). The vector system is decided, depending on the desirable insert size, which in turn is based on the metabolite to be screened, DNA quality, targeted genes, and screening strategy. The insert size in small insert libraries is less than 10 Kb, and hence plasmid is mostly used as a vector. These small insert libraries suffer from the limitation of not being able to allow the detection of large gene clusters and operons. To overcome this limitation, many large insert libraries have been used by researchers. These libraries include cosmid DNA libraries with insert sizes ranging from 25 to 35 kb or bacterial artificial chromosome (BAC) libraries with insert sizes up to 200 kb and fosmids with inserts of 40 kb of foreign DNA (Beja et al. 2000; Entcheva et al. 2001; Rondon et al. 2000).

Metagenomic libraries constructed using DNA extracted directly from naturally occurring bacterial populations are now used extensively to screen for clones that have the

Sampling site	Target	Vector used	No. of clones screened	Screening technique	Reference
Soil from a meadow, sugar beet field, and river valley (Germany)	Na+/H+ antiporters	Plasmids	1,480,000	Heterologous complementation	Majernik et al. 2001
Crude oil- contaminated groundwater microbial flora (Japan)	Aromatic hydrocarbon catabolic operon fragments	Plasmids	152,000	SIGEX	Uchiyama et al. 2005
Collection of soil samples (United States and Costa Rica)	Oxidative coupling enzyme (OxyC)	Cosmids	10,000,000	Sequence-based	Banik and Brady 2008
Soil from a botanical garden (Germany)	Blue light photoreceptor	Cosmids	2500	Sequence-based	Pathak et al. 2009
Groundwater	Phenol degradation	Plasmids	152,000	SIGEX	Uchiyama et al. 2005
Copper waste– exposed sediment	Copper P-type ATPase	_	_	PCR homology	Iglesia et al. 2010
Soil metagenome	<i>N</i> -acyl- homoserine lactones	Phagemid	7392	Functional	Bijtenhoorn et al. 2011
Antarctic soil	Esterase	Fosmids	10,000	Functional	Hu et al. 2012
Mountain soil	Thioesterase	Plasmid	10,000	Functional	Sudan and Vakhlu 2013
Woodland soil	Cellulose	-	-	PCR homology	Cucurachi et al. 2013

Table 3.1 Recent Examples for Metagenome-Derived Biocatalysts and the Screening Strategy Used

genetic capacity to produce new biocatalysts as well as small molecule products (Healy et al. 1995). Screening of metagenomic libraries unraveled many important biological products that had previously gone unnoticed. Banik and Brady explored 10,000,000 unique cosmid clones and successfully recovered glycopeptide-encoding gene clusters from soil metagenomic libraries (Banik and Brady 2008). In 2005, two separate groups reported the cloning and heterologous expression of biosynthetic gene clusters for the cyanobactins patellamide from metagenomic libraries of uncultured cyanobacterial symbionts associated with marine *Didemnidae* sponges (Long et al. 2005; Schmidt et al. 2005). Cyanobactins display interesting cytotoxic activities. Heath and coworkers screened 100,000 fosmid clones from a metagenomic library of Antarctic desert soil and identified a novel alkaliphillic esterase active clone (Hu et al. 2012).

Pathak and coworkers developed a metagenomic library of soil from a botanical garden in Germany using cosmid vectors and screened 2500 clones to detect the presence of a blue light photoreceptor (Pathak et al. 2009). Avneet and coworkers isolated a novel thioesterase gene from the metagenome of a mountain peak in the Himalayas (Sudan and Vakhlu 2013). All these studies suggested that exploring metagenomic libraries have led to the discovery of many novel products and that these libraries can still further be explored for the same.

# 3.3.4 Function-based screening

Unculturable bacteria are predicted to be a significant reservoir of novel molecules. One means to access the biosynthetic potential contained within the genomes of uncultured bacteria is functional metagenomics (Riesenfeld et al. 2004b). Function-based screening of novel genes encoding novel metabolites is based on the metabolic activities of metagenomic library containing clones. This method can be thought of as the only approach that has the potential to identify novel classes of genes encoding known or novel functions because it does not depend on the sequence information or sequence similarity to known genes (Daniel 2005; Ferrer et al. 2009; Riesenfeld et al. 2004a).

The success of function-based screening, that is, the probability of identifying a certain gene depends on various parameters. These parameters include the host, vector system, size of the target gene, its abundance in the source metagenome, the assay method, and the efficiency of heterologous gene expression in a surrogate host (Uchiyama and Miyazaki 2009).

*E. coli* is still the most preferred host for the cloning and expression of any metagenomederived genes. Although quite a large number of genes derived from Enterobacteriaceae will be readily expressed in the most common *E. coli* host, many genes from more distantly related organisms may not be expressed. This is because the promoter regions of these genes are not recognized by the *E. coli* transcriptional machinery or are expressed at low levels due to differences in codon usage (Simon et al. 2009).

Even where transcription and translation of foreign genes results in efficient protein expression, additional problems can arise when proteins need to be posttranslationally modified or exported for activity. It has been observed that significant differences exist in expression modes between different taxonomic groups of prokaryotes and only 40% of the enzymatic activities may be detected by random cloning in *E. coli* (Gabor et al. 2004). Thus, additional hosts, such as *Streptomyces* spp., *Thermus thermophilus*, *Sulfolobus solfataricus*, and diverse Proteobacteria have been used to expand the range of detectable activities in metagenomic screens (Albers et al. 2006; Angelov et al. 2009; Craig et al. 2010; Wang et al. 2000).

Although functional metagenomic-based approaches have proven to be successful for the discovery of novel enzymes, it is quite likely that many novel enzymes from rare microbes in complex communities are poorly represented in metagenomic libraries. In addition, function-driven screening often requires the analysis of more clones than sequence-based screening for the recovery of a few positive clones. The major advantage of a function-based screening approach is that only full-length genes and functional gene products are detected (Simon et al. 2009).

The following three different types of function-driven approaches have been used for the screening of metagenomic libraries. All the three methods have their own benefits and limitations.

#### 3.3.4.1 *Heterologous complementation of host strains or mutants*

It is a simple and a fast screening technique in which clones expressing the desired function are identified. The technique is based on the principle of heterologous complementation of host strains or mutants of host strains that require targeted genes for growth under selective conditions. Only recombinant clones harboring the targeted gene and producing the corresponding gene product in an active form are able to grow. The method is highly selective as no false-positives occur because the clones that do not contain the target genes are not able to express. One recent example is the identification of DNA polymerase-encoding genes from metagenomic libraries derived from microbial communities present in glacier ice. The host used in the study was an *E. coli* mutant carrying a cold-sensitive lethal mutation in the 5'-3' exonuclease domain of the DNA polymerase I. At a growth temperature of 20°C, only recombinant *E. coli* strains complemented by a gene conferring DNA polymerase-activity were able to grow (Simon et al. 2009).

#### 3.3.4.2 Phenotypical detection of the desired activity

This method uses chemical dyes or chromophore-containing derivatives of the enzyme substrates in the growth medium. The target gene or the metabolite can be identified by visualizing the change on the growth medium associated with them. Hence, the specific metabolic capabilities of the clones can be registered. A recent example of such an activity-driven screen targeted genes is encoding bacterial  $\beta$ -D-glucuronidases, which are part of the human intestinal microbiome. These enzymes have putatively beneficial effects on human health. A metagenomic library comprising 4600 clones derived from bacterial DNA extracted from pools of feces was screened using an *E. coli* strain that is deficient in  $\beta$ -D-glucuronidase activity. Nineteen positive clones were detected, with one that exhibited strong  $\beta$ -D-glucuronidase activity after cloning of the corresponding gene into an expression vector (Gloux et al. 2011).

#### 3.3.4.3 Substrate-induced gene expression

The technique was first proposed by Uchiyama and coworkers in 2005, and is based on the fact that catabolic genes require some environmental stimulus for their expression. In many cases, expression is usually controlled by regulatory elements situated close to the genes to be expressed.

Substrate-induced gene expression (SIGEX) screens the clones harboring the desired catabolic genes that are expressed in the presence of substrates but are not expressed in the absence of substrates.

The SIGEX procedure is composed of the following four steps:

- i. Construction of a metagenomic library in liquid culture. The metagenome to be analyzed is restriction digested. The metagenome fragments are then cloned upstream of *gfp* gene in the operon-trap p18GFP vector, thereby placing green fluorescent protein (GFP) expression under the control of promoters in the metagenomic DNA. A library is subsequently constructed in a liquid culture by transforming suitable host, for example, *E. coli*.
- ii. Removal of clones containing self-ligation plasmids and those expressing GFP constitutively. All clones that fluoresce in the absence of the substrate are discarded by FACS after cultivation and induction with isopropyl-β-D-thiogalactopyranoside (IPTG). This procedure greatly improves the efficiency for selecting positive clones because false-positive clones can almost completely be eliminated.
- iii. Selection of clones expressing GFP in the presence of a target substrate. Cells selected in step ii are grown in liquid cultures in the presence of a target substrate and subjected to FACS to select for fluorescent cells.

iv. Colony isolation of the sorted cells on agar plates. A sorted-cell fraction is spread on agar plates to isolate positive clones as colonies. Isolated colonies are then grown under the same conditions as step iii and subjected to FACS to check if they really are positive. Positive clones are then used for sequence analyses and expression of enzymatic activities.

SIGEX has many advantages in metagenomic library screening. The most important advantage is that it saves time, labor, and expenses because it is semiautomated. Moreover, it can be considered as a high-throughput approach because it is aided by FACS, which allows the cloning of many different genes in a comparatively shorter time. SIGEX can also detect catabolic genes for which other screening methods such as colorimetry or on-plate screening have not been developed. In addition, SIGEX does not require modified toxic and costly substrates for the detection of catabolic genes. SIGEX enable us to access genes that are otherwise difficult to obtain (Uchiyama and Watanabe 2007; Yun and Ryu 2005).

Despite the great advantages that make SIGEX a highly attractive technique for screening metagenomic libraries to identify novel metabolites, the technique has many limitations as well. In this approach, the catabolic genes are expressed only when the transcriptional regulator encoded in the SIGEX fragment interacts effectively with the transcriptional factors in host cells. The affinity of interaction may be low when the SIGEX fragment is derived from an organism distantly related to the host organism. This implies a possibility that SIGEX selection may bias itself toward genes in organisms closely related to the host. Even if the regulatory genes and catabolic genes interact efficiently, the success is still not sure because they are not always proximate to each other. It has also been observed that the expression of catabolic genes is not always induced by their substrates, which also limits the success rate of SIGEX. Substrates that do not migrate to the cytoplasm cannot be used with SIGEX. Genes for catabolic enzymes obtained using SIGEX may sometimes be partial when they are situated at the end of a cloned genome fragment (Uchiyama and Watanabe 2007).

Finally, the gate setting in FACS and the media conditions containing the inducer are critical for discriminating false-positive and false-negative results. Therefore, when SIGEX is applied, these drawbacks should be considered carefully. Despite the limitations of SIGEX, which can only detect genes that are correctly induced in a heterologous host, the method is extremely high-throughput because a large number of clones can be screened in relatively short timescales (Yun and Ryu 2005). SIGEX can be a very powerful tool for the isolation of novel metabolites only when the conditions are appropriate, that is, compatible substrate and target genes are selected and the gate settings of FACS are optimized.

# 3.3.5 Sequence-based screening

In sequence-based screening, the isolation of novel proteins is based on homology searches. It is an efficient method of isolating metabolites with different levels of similarity from previously identified genes. Generally, it is based on the conserved DNA sequences of target genes. It provides information on the distribution of functions in a community, linkage of traits, genomic organization, and horizontal gene transfer. Approaches typically involve either the sequencing of random clones to accumulate vast stores of sequence information or the identification of clones based on methods that detect a particular sequence. With both of these approaches, phylogenetic markers are sought on the clone of interest to link cloned sequences with the probable origin of the DNA. The discovery of novel metabolites or genes encoding novel metabolites by sequence-based screening can be carried out either by exploring metagenomic libraries or by directly analyzing the community DNA as such.

Cloning-based approaches require the generation of metagenomic libraries as discussed previously. The screening of these libraries is based on a common procedure of designing oligonucleotide probes or degenerate primers, which are derived from conserved regions of already-known genes or protein families (Schmidt et al. 1991). The library is then screened for the genes of interest using these probes or primers in combination with polymerase chain reaction. Subsequently, the gene products are subcloned, expressed, and characterized. Alternatively, the metagenomic libraries can be directly sequenced by direct shotgun sequencing method.

The sequence-based screening of metagenomic libraries generally experience two major limitations. First, primer/probe design is purely dependent on known sequences of enzymes and the reference sequence is of the proteins that have mostly been purified and cloned from easily cultured laboratory isolates of microorganisms. It is, therefore, possible that the designed primers and probes are biased toward genes that are similar to known genes and thus eliminate less homologous genes encoding novel enzymes. The second major drawback is that PCR amplification results in only partial fragments of target genes. Hence, the recovery of whole genes necessary for their activities may require additional steps.

In contrast to cloning-based screening, cloning independent approach escapes the most tedious step of generating metagenomic libraries. Instead, cloning independent strategies directly exploit the advantages of sequencing methods. The total DNA extracted from an environment is subjected to sequencing. The metagenomic DNA is either sequenced as such or it is first subjected to 16S-based PCR amplification, the amplicons purified, and then sequenced. The sequences obtained are then subsequently analyzed to detect the presence of any genes, functions, or novel traits and metabolites. Currently, two main sequencing strategies are used in metagenomics: shotgun sequencing and metagenomics analysis using next generation sequencing (NGS) technology.

Advances in high-throughput sequencing technology and lower cost sequencing technologies have made random shotgun sequencing of environmental DNA economically feasible. In this method, the metagenomic DNA is broken up randomly into numerous small segments using different restriction enzymes. These fragmented segments are then sequenced using a chain termination method to obtain reads. Multiple overlapping reads for the target DNA are obtained by performing several rounds of this fragmentation and sequencing. The overlapping ends of different reads are then assembled into a continuous sequence with the help of various computer programs, and therefore the whole metagenome can be analyzed. This approach generates vast amount of data (Carola and Daniel 2009), reveals millions of novel genes, and can be used to deduce metabolic pathways from uncultured bacteria. However, the technique is costly and requires considerable efforts. Moreover, the assembled sequence is a result of DNA from different organisms and hence may contain chimeric regions. There has been a great demand for a low-cost technique for sequencing, which led to the development and popularization of high-throughput sequencing methods. These techniques produce thousands or millions of sequences simultaneously (Church 2006; Hall 2007). In ultrahigh-throughput sequencing, as many as 500,000 sequencing-by-synthesis operations may be run in parallel (Bosch and Grody 2008).

NGS is based on the same principle as that of capillary electrophoresis-based Sanger sequencing. In both these methods, a small fragment of DNA is resynthesized from a template strand and the base sequence is identified depending on the signals emitted. NGS differs from traditional Sanger's method in that the sample can consist of a population of DNA molecules that do not require clonal purification and the whole process can be completed within hours, regardless of genome size. Whereas Sanger's sequencing method requires library preparation in most cases. Moreover, the sample must contain a

single template that requires purification from single bacterial and yeast colonies or phage plaques (Ronaghi et al. 1996). These advances enable rapid sequencing of large stretches of DNA fragments in a short period. Developed instruments, which produce hundreds of gigabases of data in a single sequencing run, are like a boon adding to the benefits of these advances. NGS methods are high-throughput methods involving single-molecule real-time sequencing, ion semiconductors, pyrosequencing, sequencing by synthesis, and sequencing by ligation.

Sequence-based strategies have led to the successful identification of a large number of genes encoding novel enzymes. Some of them include dimethylsulfoniopropionatedegrading enzymes, dioxygenases, nitrite reductases, [Fe-Fe]-hydrogenases, and chitinases (Bartossek et al. 2010; Hjort et al. 2010; Lee et al. 2011; Varaljay et al. 2010; Zaprasis et al. 2010). The genomic sequencing of *Streptomyces avernitilis* was done using a shotgun sequencing technique and revealed the presence of more than 20 biosynthetic gene clusters for metabolites; before sequencing, it was known to produce three to four secondary metabolites only (Omura et al. 2001; Schmidt et al. 2010).

# 3.3.6 Metagenomic data analysis

Analyzing metagenomic data is not easy as the data is much more complex than what has previously been seen in genomics. Metagenomic sequence data has lower sequence redundancy, lower sequence quality, short read lengths, increased polymorphisms, and relative abundance. Moreover, the size of the data increases rapidly. The scientific community has already seen the size of these data sets quickly move from mega base pairs (Mbps) to giga base pairs (Gbps) and now terabases, which require significant computational resources and expertise as well.

Sufficient computational technologies are available and the raw data can be subjected to various different approaches. The results depend on the analysis path taken and the tools used. Each approach has its own strengths and limitations. The bioinformatic tools available for metagenomic analysis are still evolving.

The metagenomic data obtained is subjected to a series of bioinformatic tools to generate valuable information (Figure 3.2). The data generated by metagenomics experiments

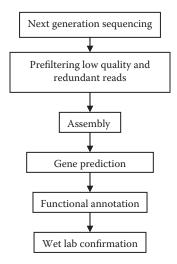


Figure 3.2 Analysis of the metagenomic data.

are both enormous and inherently noisy, containing fragmented data representing as many as 10,000 species (Wooley et al. 2010). The first step in metagenomic data analysis requires certain prefiltering steps, which include the removal of redundant, low-quality sequences and sequences of probable eukaryotic origin (Balzer et al. 2013). The tools available for prefiltering include *Eu-Detect* and *DeConseq* (Schmeider and Edwards 2011).

Prefiltering is followed by data assembly, which, in case of metagenomics, is difficult and unreliable because the data is highly redundant and is error prone due to short reads (Handelsman et al. 2007). Many assembly programs are available such as *Phrap* or *Celera* assembler and *Velvet* assembler.

Assembly of the metadata is followed by gene prediction. Metagenomics uses two approaches for predicting the gene. The first approach is to identify genes based on homology with genes that are already available in sequence databases. This is usually done using simple BLAST searches. MEGAN4 is usually used to implement such approach (Huson et al. 2011). The second approach is an *ab initio* method that uses intrinsic features of the sequence to predict coding regions based on gene training sets from related organisms. This is the approach taken by programs such as *GeneMark* and *GLIMMER* (Wenhan et al. 2010).

Once the gene involved in the metabolite production has been predicted, the next step is the functional annotation of the gene. All the possible products that can be produced by a particular gene are identified using bioinformatic tools. If the gene does not correspond to any of the previously available products, it can be a novel metabolite. All these results can then be confirmed by wet laboratory experiments such as gene expression. A few bioinformatics programs have been established for analyzing metagenome sequences, gene prediction, and annotation. The European Union-funded "MetaFunctions" project covers the development of "metagenomes Mapserver." It is a data-mining system that correlates genetic patterns in genomes and metagenomes with contextual environmental data. Nevertheless, more innovative and sophisticated bioinformatics tools must be devised to assure continued valuable progress in the field of metagenomics.

### 3.3.7 *Future prospects*

Metagenomics is one of the fastest growing research areas and has proven effective for isolating novel metabolites from the environment. The field has made great progress in its scale and scopes, which have expanded since the concept was first introduced. Initially, metagenomics was applied only to the microbial communities of temperate environments, but with the advances made in DNA isolation methods, cloning strategies, and screening techniques, microbial communities of extreme environments have also been analyzed for novel metabolites. However, the whole metagenomic process still needs to be improved. Development of standard protocols for isolating sufficiently purified metagenomic DNA from environmental samples is required. Screening strategies need to be revised because each strategy has its own limitations. Conventional screening methods are time-consuming and laborious and hence it remains necessary to develop more effective and economic strategies. SIGEX is a good alternative but better high-throughput screening methods need to be developed. In addition, we need to understand the utility and limitations of each screening method to facilitate the selection of the appropriate one for a desired gene or set of genes. Despite all these shortcomings, metagenomics is still an important and indispensable tool for the identification of novel biomolecules.

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## chapter four

# *Genetic engineering of nonconventional yeasts for the production of valuable compounds*

Andriy Sibirny, Catherine Madzak, and Patrick Fickers

## Contents

4.1	Intro	luction	.64			
4.2	Pichia	pastoris	.64			
	4.2.1	,				
	4.2.2	•				
	4.2.3	Bioprocess operations and optimization				
	4.2.4					
	4.2.5	Production of recombinant proteins for industrial uses				
4.3	Yarroa	via lipolytica	.72			
	4.3.1					
	4.3.2					
		4.3.2.1 Promoters	. 74			
		4.3.2.2 Targeting signals for secretion and surface display	.77			
		4.3.2.3 Targeting to microbodies				
		4.3.2.4 Vectors, related methods, and amplification strategies				
		4.3.2.5 Strains	. 81			
	4.3.3	Examples of processes and products of recombinant Y. lipolytica strains	. 82			
		4.3.3.1 <i>Y. lipolytica</i> as a high-throughput expression platform for				
		protein engineering	. 82			
		4.3.3.2 Glycoengineering of <i>Y. lipolytica</i> for the production of				
		humanized pharmaceutical proteins	. 82			
		4.3.3.3 Y. lipolytica as whole-cell biocatalyst for SCO production				
		4.3.3.4 Y. lipolytica as whole-cell biocatalyst for carotenoid production	.83			
		4.3.3.5 Future prospects	.84			
4.4	Hanse	nula polymorpha	.84			
	4.4.1	Available strains				
	4.4.2	Genetic methods and genome research				
	4.4.3	3 Genetic control of methanol induction and glucose catabolite repression in <i>H. polymorpha</i>				
	4.4.4	.4 H. polymorpha as a system for the expression of own and heterologou				
		proteins	.87			
	4.4.5	Examples of processes and products based on the recombinant strains				
		of H. polymorpha				
		4.4.5.1 Production of heterologous protein	. 89			

4.4.5.2	Production of low-molecular weight biologically active	
	compounds in <i>H. polymorpha</i>	
	1 1 2 1	
1.0101010000		

## 4.1 Introduction

Yeasts are suitable host organisms for the production of heterologous proteins and other valuable compounds because they combine ease of genetic manipulation with rapid growth on inexpensive medium at high cell densities, and have the ability, in contrast with bacteria, to perform complex posttranslational modifications. Yeasts also offer the possibility of secreting proteins, thus avoiding toxicity from intracellularly accumulated materials. Protein secretion also simplifies the subsequent purification process. Heterologous protein production tools were first developed in Saccharomyces cerevisiae in the 1980s. However, plasmid instability, low production yield, and hyperglycosylation phenomena have limited the development of profitable industrial processes (Buckholz et al. 1991; Romanos 1995). In addition, S. cerevisiae produces proteins with N-linked glycosylation patterns ending with an α-1,3-linked mannose motif that is known to be allergenic (Çelik et al. 2012). All of these led to the development of alternative expression systems in other yeasts including *Pichia* pastoris, Hansenula polymorpha, Kluyveromyces lactis, Schizosaccharomyces pombe, Arxula ade*ninivorans*, and *Yarrowia lipolytica*. In this chapter, the production of recombinant proteins and other valuable compounds using the nonconventional yeasts P. pastoris, Y. lipolytica, and *H. polymorpha* is discussed.

## 4.2 Pichia pastoris

### 4.2.1 Historical background and main characteristics

Only a limited number of yeast species are able to use methanol as a primary carbon source (Hazeu et al. 1972). In that regard, the Phillips Petroleum Company (PPC) was the first to develop media and fermentation protocols for growing *P. pastoris* in continuous cultures at high cell densities to produce single-cell proteins (SCP) for feedstock from methanol (Cereghino et al. 2000). Unfortunately, the oil crisis of the 1970s rendered this process unattractive due to the high cost of methanol. In the following decade, PPC and the Salk Institute Biotechnology/Industrial Associates Inc. (SIBIA, La Jolla, CA) developed tools for heterologous protein production in *P. pastoris*. As a result, recombinant expression vectors, methods for transformation, selectable markers, and fermentation processes were developed to exploit the productive potential of this yeast (Cregg et al. 1993; Rosenfeld 1999). Expression systems developed for *P. pastoris* were patented by Research Corporation (Carlsbad, CA). Actually, more than 1000 proteins have been cloned and expressed in *P. pastoris* (Cregg et al. 2000; Damasceno et al. 2012). *P. pastoris* has also been selected by several protein production platforms for genomics programs (Yokoyama 2003).

The *P. pastoris* genome is organized into four chromosomes with a total estimated size of 9.7 Mbp by pulsed-field gel electrophoresis (Ohi et al. 1998) and 9.43 Mbp after genome sequencing of strain GS115 (De Schutter et al. 2009). Genome sequence analysis led to the identification of 5.313 protein-coding genes with a GC content of 41.6%. Protein encoding genes represent 80% of the genome sequence. A secretion signal peptide was predicted in

9% of these genes. Overall, the codon usage is similar to that of *S. cerevisiae*. The *S. cerevisiae* orthologue gene *MNN1*, encoding the  $\alpha$ -1,3-mannosyltransferase, which is responsible for the incorporation of the immunogenic terminal  $\alpha$ -1,3-mannosyl glycotopes, was not detected in GS115 genome. The genome sequence analysis allowed the identification of vacuolar and secreted proteases, which will ease the development of protease-deficient strains (De Schutter et al. 2009).

*P. pastoris* prefers a respiratory growth mode and thus does not produce fermentation by-products such as acetic acid or ethanol (Cereghino et al. 2002). This enables cultures to reach high cell densities (up to 200 g/L dry weight; Heyland et al. 2010). Another advantage of *P. pastoris* is its capacity to secrete large amounts of heterologous proteins into the culture medium. Because *P. pastoris* secretes its native (i.e., nonrecombinant) proteins at low levels, secretion provides an effective method of separating heterologous proteins from the bulk of intracellular host proteins and other cellular materials (Romanos 1995).

Methylotrophic yeasts such as *P. pastoris* are able to metabolize methanol as their sole carbon source for energy through a specific metabolic pathway involving several unique enzymes. The initial steps of this pathway take place in the peroxisomes where methanol is converted into formaldehyde by two specific alcohol dehydrogenases, which is in turn converted into dihydroxyacetone and glyceraldehyde-3-phosphate by a dihydroxyacetone synthase. In the peroxisome, a catalase converts the oxygen peroxide generated in the first steps of methanol oxidation into oxygen and water. The subsequent steps of methanol assimilation are localized in the cytosol. Extensive reviews on methanol metabolism in *P. pastoris* could be found in the works of Cereghino et al. (2000) and Gellissen (2000).

#### 4.2.2 Genetic and molecular tools

All P. pastoris expression strains are derived from the wild-type strain NRRL-Y 11430 (Northern Regional Research Laboratories, Peoria, IL). Auxothrophic mutants could carry one auxotrophy such as GS115 (*his4*), GS190 (*arg4*), or JC254 (*ura3*); two auxotrophies such as GS200 (arg4 his4) or JC227 (ade1 arg4); three auxothrophies such as JC300 (ade1 arg4 *his4*); and even four auxotrophies such as JC308 (*ade1 arg4 his4 ura3*; Cereghino et al. 2000). Protease-deficient strains are also available such as SMD1163 (his4 pep4 prb1), SMD1165 (his4 prb1), and SMD1168 (his4 pep4). In addition, there are three types of P. pastoris host strains available that differ with regard to their ability to metabolize methanol. These are the wild-type strain (Mut<sup>+</sup>) and those resulting from AOX1 deletion (Mut<sup>s</sup>, methanol utilization slow) or from the double AOX1-AOX2 deletion (Mut-, methanol utilization minus). The Mut<sup>+</sup> phenotype is mainly used for heterologous protein production, although the Mut<sup>s</sup> phenotype has been used in some cases. Molecular genetic manipulations of *P. pas*toris such as DNA-mediated transformation, gene targeting, gene replacement, and cloning by functional complementation are similar to those described for other yeasts such as S. cerevisiae (Cereghino and Cregg 2000). P. pastoris can be transformed by electroporation  $(10^5 \text{ transformant/}\mu\text{g DNA})$  or by using the classic lithium chloride  $(10^2 \text{ transformant/}\mu\text{g})$ DNA) or polyethylene glycol (PEG<sub>1000</sub>,  $10^3$  transformant/µg DNA) methods (Cregg et al. 1985, 1998). P. pastoris has a propensity for homologous recombinations between native and exogenous DNAs. However, gene replacement occurs at a lower frequency compared with S. cerevisiae and requires longer flanking sequences to direct integration (Cregg et al. 1998).

The success of *P. pastoris* as a platform for the production of heterologous proteins is related to the strong and tightly regulated promoter of the *AOX1* gene encoding alcohol oxidase I. As stated above, alcohol oxidases are the first enzyme of the methanol catabolism pathway. In *P. pastoris*, there are two alcohol oxidase genes, *AOX1* and *AOX2*, with

*AOX1* representing 90% of the total alcohol oxidase activity in the cell (Cos et al. 2006). In appropriate culture conditions (i.e., methanol feeding at a growth-limiting rate), the Aox1 level can represent more than 30% of the total soluble proteins (Cregg et al. 2000). Therefore, *AOX1* promoter (*pAOX1*) is mainly used to drive and control the synthesis of heterologous proteins. *pAOX1* is repressed in the presence of glucose, glycerol, and ethanol and is strongly induced in the presence of methanol. Other carbon sources such as sorbitol have no significant repressive effect on *pAOX1* induction (Çalık et al. 2010; Niu et al. 2013). Indeed, sorbitol does not repress cell growth or heterologous protein production at less than 50 g/L (Çelik et al. 2009).

Despite pAOX1-based expression presenting several advantages, it also has some drawbacks. First, methanol is a fire hazard and therefore its storage in large quantities is undesirable. Second, as a derivative of petroleum sources, it is unsuitable for use in the production processes of certain foods and additives. Third, its monitoring during the production process is often difficult due to the unreliability of on-line probes and the time-consuming methods of off-line measurements (Macauley-Patrick et al. 2005). Fourth, methanol accumulation in the culture broth could result in a decrease of cell viability and alter heterologous protein stability.

To circumvent the use of methanol or at least to reduce its utilization, efforts have been made to develop novel alternative promoters to pAOX1. Several inducible or constitutive promoters have been developed. For instance, the promoter of the gluthatione-dependent formaldehyde dehydrogenase (FLD) was used successively to drive the production of heterologous proteins. *FLD1* is a key enzyme of methanol metabolism and certain alkylated amines such as methylamine as nitrogen source (Zwart et al. 1980). Its promoter is strongly and independently induced by either methanol/ammonium sulfate or methylamine/glucose mixtures. Furthermore, induction levels obtained with either methanol or methylamine are comparable with those obtained with pAOX1 in the presence of methanol (Shen et al. 1998).

Waterham and coworkers developed an expression vector based on the promoter of glyceraldehyde 3-phosphate dehydrogenase gene (pGAP). Promoter studies, using  $\beta$ -lactamase as a reporter, showed that pGAP is constitutively active. However, its strength varies depending on the carbon source used for cell growth. Expression levels in glucose media are significantly higher than those obtained with pAOX1 in the presence of methanol. pGAP has been successfully used for the synthesis of heterologous proteins from bacterial, yeast, insect, and mammalian origins (Waterham et al. 1997). In addition, Sears et al. (1998) developed a moderate promoter from the YPT1 gene that encodes a small GTPase involved in secretion. Expression levels from YPT1 promoter are 10-fold to 100-fold lower than those from the GAP promoter. This promoter is particularly well suited for expressing genes that would be toxic for the cells when overexpressed. A similar approach for moderate expression level was based on the promoter from the PEX8 gene encoding a peroxisomal matrix protein that is essential for peroxisome biogenesis (Liu et al. 1995). The use of the AOX2 gene promoter (full or truncated version) has also been reported in the literature (Mochizuki et al. 2001). Despite its expression level being much lower than pAOX1, Kobayashi et al. (2000b) demonstrated that the addition of small amounts of oleic acid (0.01%) led to a twofold increased protein production. This could be associated with the presence of an oleic acid-responsive element within the pAOX2. Menendez and coworkers (2003) cloned the ICL1 gene encoding an isocitrate lyase and developed expression vectors based on its promoter (pICL1). The authors reported that high levels of heterologous gene expression could be obtained in the presence of low glucose concentrations or in the presence of ethanol as the sole carbon source. However, higher expression levels were obtained

in the absence of glucose than after induction with ethanol. An extensive review of the promoters used for heterologous protein production in *P. pastoris* and their regulation has been published recently (Vogl et al. 2013).

Although molecular genetic techniques are well developed for P. pastoris, few selectable markers have been described in the literature. These are biosynthetic markers from histidine biosynthetic pathways, namely, HIS1, HIS2, HIS4, HIS5, and HIS6; arginine biosynthetic pathways, namely, ARG1, ARG2, ARG3, and ARG4; or uracil biosynthetic pathways, namely, URA3 and URA5 (Cereghino et al. 2001; Cregg et al. 1985; Nett et al. 2003, 2005). Others markers such as ADE1 (PR-amidoimidazolsuccinocarboxamide synthase; Cereghino et al. 2001), MET2 (homoserine-O-transacetylase; Thor et al. 2005), and FLD1 (formaldehyde dehydrogenase, see below; Sunga et al. 2004) have been described. Besides biosynthetic markers, other selectable markers, based on bacterial antibiotic resistance genes, could be used in *P. pastoris* (Romanos et al. 1992). These are the *Sh ble* gene (Zeo<sup>R</sup>), from the bacterium Streptoalloteichus hindustanu, which confers resistance to the bleomycinlike drug zeocin (Drocourt et al. 1990; Higgins et al. 1998), the blasticidin S-deaminase gene from Aspergillus terreus, which confers resistance to blasticidin (Kimura et al. 1994), and the kanamycin resistance gene (Kan<sup>R</sup>), which confers resistance to the eukaryotic antibiotic G418 (Scorer et al. 1994). The use of these markers allows multicopy integrations of an expression vector to be obtained when the selection of transformants is performed at high antibiotic concentrations (Scorer et al. 1994). A disadvantage of these multicopy integrants is the integration of bacterial antibiotic resistance genes within the P. pastoris genome. Indeed, this could represent a potential DNA hazard such as the dissemination of these resistance genes into the environment. To circumvent this drawback, an amplification marker based on the FLD1 gene was developed (Sunga et al. 2004). As stated previously, P. pastoris can metabolize methanol as its sole carbon source and certain alkylated amines such as methylamine and choline as sole nitrogen sources. Formaldehyde is a common toxic intermediate in the methanol and methylamine pathways that could be further oxidized to formate by formaldehyde dehydrogenase Fld1 and to carbon dioxide by formate dehydrogenase (Sunga et al. 2004). Thus, this oxidative pathway may help the cell in protecting itself from the toxic effects of an excess of formaldehyde (Sibirny et al. 1990). Sunga and coworkers (2004) isolated a P. pastoris fld1 mutant with increased sensitivity to formaldehyde and demonstrated that the level of resistance to formaldehyde was proportional to the number of *FLD1* gene copies present in that disrupted strain. Thus, the utilization of this selectable marker in combination with this *fld1* mutant strain allows multicopy integrations of expression vectors without any potential biohazard problem.

Several methods for unmarked genetic modifications in *P. pastoris* have been developed. One uses either *URA3* or *URA5* as a counterselectable marker (Nett et al. 2003). However, uracil auxothrophic strains grow very slowly even in medium supplemented with uracil, which renders their utilization difficult and time-consuming (Cereghino et al. 2001). To overcome this limitation, alternative counterselectable markers that can be used in any *P. pastoris* strain have been developed. One of these is the T-*urf*13 gene from the mitochondrial genome of male sterile maize used in combination with the insecticide Methomyl (Soderholm et al. 2001). Indeed, the introduction of T-*urf*13 gene renders the strain sensitive to the insecticide. The problem with this method is that the toxicity of T-*urf*13 gene may cause conditional lethality with certain gene *mazF*, which is used as a counterselectable marker (Yang et al. 2009a). The main disadvantage of this method is that gene disruptions require the construction of plasmids and the performance of time-consuming selectable marker rescue, which leaves undesired repeat sequences in the *P. pastoris* genome. However, Pan et al. (2011) have developed a rapid method for unmarked gene deletion and marker rescue based on the Cre/mutated *lox* system and a zeocin resistance gene. The disruption cassette "5' target region-*lox71*-Cre-ZeoR-*lox66-3*' target region" is obtained by PCR, rendering plasmid construction unnecessary. The Cre gene expression is regulated by pAOX1 and thus is expressed in the presence of methanol. After gene disruption, transient expression of the Cre recombinase allows *lox71* and *lox66* to recombine with excision of the Cre-ZeoR cassette. The resulting hybrid *lox* site (known as *lox72*) in the genome displays a strongly reduced binding affinity for the Cre recombinase. This method could be used sequentially to disrupt genes without the introduction of any marker in the cell genome and without the possibility of genome rearrangement.

*P. pastoris* can produce heterologous proteins intracellularly or extracellularly. Protein secretion allows several steps of purification such as cell lysis and subsequent clarifications to remove cell debris to be avoided. The *S. cerevisiae*  $\alpha$ -factor prepro-peptide or the *P. pastoris* acid phosphatase (*PHO1*) signals are the most widely used secretion signals (Cereghino et al. 2000). However, other secretion signals have also been used successfully. These are signal peptides from *Rhizopus oryzae*  $\alpha$ -amylase (Li et al. 2011), *Trichoderma reesei* class 2 hydrophobins (Kottmeier et al. 2011), human serum albumin (Xiong et al. 2008), K28 yeast virus toxin (Eiden-Plach et al. 2004), *Pichia acacia* killer toxin (Crawford et al. 2002), *Phaseolus vulgaris* phytohemagglutinin (Raemaekers et al. 1999), and the *P. pastoris* endogenous signal peptides is protein-dependent and must be investigated experimentally to isolate the most suited one. In some cases, the native signal sequence of the heterologous protein is used.

In contrast with bacteria, yeasts are able to produce glycosylated proteins. However, glycosylation patterns in yeasts are different from those of higher eukaryotes such as mammalian cells. P. pastoris is capable of adding O-linked and N-linked carbohydrate moieties to secreted proteins (Cregg et al. 2000). However, O-oligosaccharides in P. pastoris are composed of mannose residues compared with mammals in which O-oligosacharides are made up of mainly sialic acid, galactose, and N-acetylgalactosamine (Daly et al. 2005). Moreover, and in contrast with most eukaryotes that add O-linked saccharides onto hydroxyl groups of serine and threonine, P. pastoris has no preferred amino acids for O-glycosylation (Cregg et al. 2000). Glycosylation of proteins in *P. pastoris* could modify the functions and characteristics of the heterologous proteins produced, which could become an issue for some applications (Eckart et al. 1996). Moreover, the absence of sialic acid in glycoproteins leads to their rapid clearance from the bloodstream, which can also be a disadvantage in some applications (Khandekar et al. 2001). In the last decade, there have been great advances in humanizing the N-glycosylation pathway in P. pastoris. Vervecken and coworkers used a strategy that consists of the disruption of the OCH1 gene coding for  $\alpha$ -1,6-mannosyltransferase Och1p, which is responsible for hypermannolysation in yeasts (Jacobs et al. 2008; Vervecken et al. 2004). In a second step, they stepwise introduced heterologous glycosylation enzymes into the endoplasmic reticulum or Golgi complex. These enzymes are an  $\alpha$ -1,2-mannosidase and two chimeric glycosyltransferases, namely, Ker2-GnTI and Ker2-β-1,4-galactosyltransferase. For that purpose, they developed specific vectors known as GlycoSwitch vectors, which allow the conversion of any wild-type P. pastoris strain into strains that modify their glycoproteins with Gal2GlcNAc2Man3GlcNAc2Nglycans (Jacobs et al. 2008). Other groups also succeeded in humanizing glycosylation pathways in P. pastoris. Hamilton and coworkers (2006) constructed strains capable of secreting terminally sialylated complex and biantennary glycoproteins. After knocking out four genes to eliminate yeast-specific glycosylation, they introduced 14 heterologous

genes, allowing them to replicate the sequential steps of the human glycosylation pathway. In addition, Li et al. (2006) were able to optimize the production of humanized IgGs in a glycoengineered *P. pastoris* strain that presented antibody-mediated effector functions.

#### 4.2.3 Bioprocess operations and optimization

Bioprocess parameters such as medium composition, pH, temperature, aeration rate, induction, and feeding strategies are of utmost importance because it directly affects the production yield. These parameters vary according to the strain used or the heterologous protein produced. However, there are some guidelines that allow high productivity yields to be attained. The most commonly used medium for heterologous protein production in *P. pastoris* is the basal salt medium (BSM), which was developed by Invitrogen Co. (Invitrogen 2000). Despite this medium being considered as the standard, it presents some negative points including unbalanced composition, precipitate formation, ionic strength, and others. Therefore, alternative media were developed (d'Anjou et al. 2000; Stratton et al. 1998). All defined media are used in combination with the trace salt solution PTM1 from Invitrogen Co. (Invitrogen 2000). Ammonium hydroxide is usually used as the nitrogen source and also to maintain pH at the setup value. Various combinations of glycerol and methanol are used as carbon and energy sources. Most of the heterologous proteins are produced under the control of the AOX1 promoter in Mut<sup>+</sup> or Mut<sup>S</sup> P. pastoris strains according to a protocol proposed by Invitrogen Co., which is mainly derived from the works of Brierley et al. (1990). This includes a glycerol batch phase (GBP), a glycerol-fed batch phase (GFBP), a transition phase (TP), and finally a methanol induction phase (MIP).

The objective of the GBP is to accumulate cells in the least amount of time possible. Glycerol is used as a carbon source because it is nonfermentable and allows higher maximum specific growth rate compared with methanol (0.18 vs. 0.14 h<sup>-1</sup>; Cos et al. 2006). In GBP, glycerol concentration is limited to 40 g/L because higher values could inhibit cell growth (Invitrogen 2000). When glycerol from GBP is exhausted, as indicated by a sharp increase of the dissolved oxygen, the GFBP phase starts. Glycerol is added at a constant rate usually equal to the maximum glycerol-specific consumption rate (0.0688 g (gh)<sup>-1</sup>; Cos et al. 2006). The primary objective of this phase is to further increase cell concentration without repressing growth. At the end of this phase, methanol is added at an increasing stepwise mode to acclimate cells and to derepress the *AOX1* promoter (TP phase). Finally, in the MIP, methanol is added in fed-batch mode at a specific feeding rate that varies according the *P. pastoris* strain genotype and the heterologous protein produced. Indeed, it has been shown that the specific growth rate on methanol influences heterologous protein production (Çelik et al. 2009; Kobayashi et al. 2000a; Zhang et al. 2005).

Methanol is a high-degree reductant with high heat of combustion (Niu et al. 2013). This leads to one major challenge during a large-scale MIP. Because heat production is almost linearly correlated with oxygen consumption, the challenge is how to reduce oxygen consumption without affecting protein productivity. One possible answer is to use a cosubstrate such as glycerol (Jungo et al. 2007a; Zalai et al. 2012; Zhang et al. 2003), glucose (Jorda et al. 2012; Paulová et al. 2012), or sorbitol (Çelik et al. 2009; Gao et al. 2012; Inan et al. 2001; Ramón et al. 2007; Wang et al. 2010; Zhu et al. 2011) to partially replace or at least complement methanol during the protein production phase. The most promising cosubstrate is sorbitol because it is a low-degree reductant and a nonrepressing carbon source for pAXO1 (Inan et al. 2001). The benefits of mixed feeding of sorbitol and methanol have been widely characterized with the aims of reducing oxygen consumption without loss of protein productivity. For instance, Jungo et al. (2007b) developed a fed-batch strategy

based on a methanol/sorbitol mixture (43:57, C-mol C-mol). However, very few studies have been performed to quantitatively analyze the cellular physiology during cofeeding (Celik et al. 2010; Jungo et al. 2007b). Most of them focused only on the final product (i.e., heterologous protein) without any insights into pAOX1 regulation. Niu et al. (2013) reported on the quantitative characterization of *P. pastoris* cell metabolism, with special emphasis on the quantification of pAOX1 induction during a methanol/sorbitol cofeeding process using the LacZ reporter gene, transient continuous culture, and metabolic flux analysis (MFA). Their results demonstrated that cell-specific oxygen consumption  $(qO_2)$ could be reduced by decreasing the methanol fraction in the feeding media. More interestingly, optimal pAOX1 induction was achieved and maintained in the range of 0.45 to 0.75 C-mol/C-mol of the methanol fraction. In addition, the  $qO_2$  was reduced by 30% at most in those conditions. Based on a simplified metabolic network, an MFA was performed to quantify intracellular metabolic flux distribution. Cofeeding of sorbitol can reduce the oxidation flux in the peroxisome leading to less oxygen consumption and heat production. At the same time, sorbitol in the mixture produces energy through the TCA cycle and provides carbon source for biomass synthesis. According to their MFA model, 61% of the sorbitol goes through the TCA cycle whereas the rest is used for biosynthesis at 0.5 C-mol/C-mol methanol.

Most of the processes of heterologous protein production by *P. pastoris* are performed at 30°C, which is the optimal growth temperature. However, Dragosits and coworkers demonstrated that a 10°C decrease led to a threefold increased heterologous protein yield together with a reduced flux through the TCA cycle, reduced levels of protein involved in oxidative stress response, and a lower cellular level of molecular chaperones. All these indicate that folding stress is generally decreased at lower cultivation temperatures, enabling more efficient heterologous protein secretion in *P. pastoris* host cells (Dragosits et al. 2009).

The pH of the culture medium also plays an important role because many parameters such as growth rate, enzyme stability, and proteolytic degradation are pH dependent. Although pH values in the range of 3.5 to 5.5 were reported to have a limited effect on cell growth of *P. pastoris* Mut<sup>+</sup> strain (Inan et al. 1999), different optimal pH values have been reported for the production of heterologous proteins (Macauley-Patrick et al. 2005).

#### 4.2.4 Production of recombinant proteins for biopharmaceutical and medical uses

Antibodies are glycoproteins with high affinities to specific molecules. They are considered as a group of biopharmaceuticals with potent applications for the treatment of cancer as well as inflammatory and infectious diseases (Jeong et al. 2011). *P. pastoris* is the host cell of choice for antibody or antibody fragment production. It allows high protein titers to be obtained, free of bacterial endotoxins or potential viral contaminations associated with mammalian cell culture (Barnard et al. 2010). During the 1990s, only antibody fragments, single-chain IgG fragments (scFv), or fusion ScFv were produced in *P. pastoris* (Goncalves et al. 2013). The first success of full-length IgG production was published in 1999 (Ogunjimi et al. 1999). Despite this, IgG was biologically active, it was produced in *P. pastoris* strain SMD1168 (*pep4 his4*), which was unable to achieve human-like glycosylations. The glycan moiety of IgG is an important feature not only for its stability and solubility but also for its efficient binding to the target receptor. With the advances in strain glycoengineering such as GlycoFi (Beck et al. 2010) or GlycoSwitch (Jacobs et al. 2008), high titers of full-length antibodies with humanized glycan chains have been reported (Li et al. 2006; Potgieter et al. 2009; Ye et al. 2011).

Other proteins of medical interest have been produced in *P. pastoris* at various yields. These are insulin precursor (1.5 g/L; Wang et al. 2001), cytokines such as interleukin 2 (4 g/L; Cregg et al. 1993), human serum albumin (6 g/L; Cregg et al. 1993), tumor necrosis factor (10 g/L; Sreekrishna et al. 1989), tetanus toxin fragment C (12 g/L; Clare et al. 1991), coagulation inhibitors such as hirudin variant HV2 (1.5 g/L; Rosenfeld et al. 1996), ovine pregnancy recognition hormone interferon  $\tau$  (0.28 g/L; Van Heeke et al. 1996), cytomegalovirus antigenic protein pp52 (0.1 g/L; Battista et al. 1996), 5HT<sub>5A</sub> serotonin receptor (22 pmol/mg membrane protein; Weiß et al. 1995), and human  $\mu$ -opioid receptor (400 fmol/mg protein; Talmont et al. 1996).

#### 4.2.5 Production of recombinant proteins for industrial uses

*P. pastoris* has been utilized for the production of lytic enzymes widely used in the pulp and paper, textiles, detergent, agro-food, and chemical industries. Some examples are listed below:

- Xylanases such as *Aspergillus niger* endo-β-1,4-xylanase (based on pAOX1 and *S. cerevisiae* invertase signal peptide, 60 mg/L; Berrin et al. 2000), thermostable *Thermomyces lanuginosus* IOC-4145 β-1,4-xylanase (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 148 mg/L; Damaso et al. 2003), or yeast-like fungus *Aureobasidium pullulans* acidophilic endo-1,4-β-xylanase (based on pAOX1 and native signal peptide, 178 mg/L; Tanaka et al. 2004).
- Cellulases such as acidic and thermostable carboxymethyl cellulase from the yeast *Cryptococcus* sp. S-2 (based on pAOX1 and HAS signal peptide; Thongekkaew et al. 2008) or the edible straw mushroom *Volvariella volvacea* endo-β-1,4-glucanase EG1 (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 65 mg/L; Ding et al. 2002).
- Proteases such as *Mucor pusillus* aspartic protease (based on p*AOX1* and *S. cerevisiae* invertase signal peptide; Montesino et al. 1999), *Aspergillus oryzae* alkaline protease (based on p*AOX1* and *S. cerevisiae*  $\alpha$ -factor signal peptide, 513 mg/L; Guo et al. 2008), or *Thermomonospora fusca* YX thermostable serine protease (based on p*GAP* and *S. cerevisiae*  $\alpha$ -factor signal peptide, 135 units/L; Kim et al. 2005).
- Lipases such as *Y. lipolytica* Lip2p (based on p*AOX1* and *S. cerevisiae*  $\alpha$ -factor signal peptide, 630 mg/L; Yu et al. 2007), *Burkholderia cepacia* lipase (based on p*GAP* and *S. cerevisiae*  $\alpha$ -factor signal peptide, 184 units/L; Jia et al. 2010), or *Candida antarctica* lipase B (based on p*GAP* and *S. cerevisiae*  $\alpha$ -factor signal peptide, 44 mg/L; Larsen et al. 2008).
- Amylases such as mouse salivary  $\alpha$ -amylase (based on pAOX1 and pGkl killer protein signal peptide, 0.24 mg/L; Kato et al. 2001), alkalophilic *Bacillus*  $\alpha$ -amylase (based on pAOX1 and PHO1 phosphatase signal peptide, 50 mg/L; Tull et al. 2001), or *Thermobifida fusca*  $\alpha$ -amylase (based on pGAP and *S. cerevisiae*  $\alpha$ -factor signal peptide, 510 units/L; Yang et al. 2010).
- Phosphatases such as human placental alkaline phosphatase (based on pAOX1 and PHO1 phosphatase signal peptide, 2 mg/L; Heimo et al. 1998), soybean root nodule phosphatase (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 10 mg/L; Penheiter et al. 1998), or *Drosophila* type 1 serine/threonine phosphatase (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 37 mg/L; Szöőr et al. 2001).
- Pectinase such as *Fusarium solani* pectate lyase (based on pAOX1 and PHO1 phosphatase signal peptide; Guo et al. 1995) or *Bacillus subtilis* pectate lyase based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 100 units/L; Zhuge et al. 2008).

- Phytases such as *A. niger* PhyA (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 63 units/mL; Han et al. 1999), *Peniophora lycii* 6'-phytase (based on pAOX1 and synthetic MF4I signal peptide, 10.540 units/mL; Xiong et al. 2006), or *A. niger* SK-57 phytase (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 865 units/mL; Xiong et al. 2005).
- Laccases such as *Pycnoporus sanguineus* laccase (based on pAOX1 and native signal peptide, 28 mg/L; Lu et al. 2009), or *Trametes* sp. laccase B (based on pAOX1 and *S. cerevisiae* α-factor signal peptide, 32 units/mL; Li et al. 2007).

*P. pastoris* has also been used for the production of other compounds such as antifreeze proteins (Loewen et al. 1997), human gelatin (Williams et al. 2008), or human cathepsin K protease inhibitor (Linnevers et al. 1997).

## 4.3 Yarrowia lipolytica

#### 4.3.1 Main characteristics and industrial applications

Y. lipolytica is an oleaginous yeast from the Saccharomycetes/Hemiascomycetes class, which exhibits remarkable lipolytic and proteolytic activities. This yeast appears ubiquitously in natural environments (soils, mycorrhizae, marine, and hypersaline waters) as well as in food, particularly in fermented dairy and meat products (e.g., cheeses and dry sausages), as reviewed in the article by Groenewald et al. (2014). Because Y. *lipolytica* is able to efficiently degrade hydrocarbons, especially alkanes, it is also frequently isolated from oil-polluted environments, hence its potential for bioremediation (reviewed in an article by Bankar et al. 2009). Y. lipolytica is a biosafety class 1 microorganism (Groenewald et al. 2014), which means that it is considered as "safe-to-use": several applications of this yeast have been given a "generally regarded as safe" (GRAS) status by the U.S. Food and Drug Administration. The complete sequence and annotation of Y. lipolytica genome by the Genolevures consortium (CNRS, France; Dujon et al. 2004) has been made available on the Genolevures website (http://www.genolevures .org/). This data, together with the recent design of the first genome-scale metabolic model (Loira et al. 2012), provided new tools for the study and valorization of this yeast, which has recently been the subject of volumes 24 and 25 from Springer's Microbiology Monographs series (Barth 2013a,b). For more than 50 years, Y. lipolytica has called the attention of scientists and manufacturers due to its potential for industrial applications. Oriented at first toward producing biomass or metabolites with commercial value from wild-type or traditionally improved strains, industrial use of this yeast could benefit from several decades of progress in genetic engineering, which would allow for modifications in its metabolism or its use as a host for heterologous protein production. In the 1950s, industrial use of Y. lipolytica was pioneered by the British Petroleum Company (UK), which exploited the alkane-degradating properties of this yeast to produce SCP for livestock feeding (Toprina G). This application was discontinued in 1978 as a consequence of the oil crisis (Groenewald et al. 2014). In the 1970s, Pfizer Inc. (US) developed industrial citric acid production from Y. lipolytica, a technology acquired in 1990 by Archer Daniels Midland Company (US), which is presumed to be still exploiting it nowadays (Groenewald et al. 2014). These large-scale industrial applications for producing SCP or citric acid have driven extensive safety and efficacy studies (GRAS status) and produced a large amount of data on Y. lipolytica cultivation. More recent applications of wild-type or traditionally improved Y. lipolytica strains include the production of erythritol for use as a food additive (Baolingbao Biology Co., China), the use of biomass as fodder yeast (Skotan SA, Poland), and the production of a starter for the bioremediation of lipid-rich wastewaters, composed of freeze-dried cells and extracellular lipase (Artechno, Belgium). The Polish company Skotan SA is also developing prebiotic and probiotic applications for the food industry (Groenewald et al. 2014). Because genetic and molecular tools have been developed, *Y. lipolytica* has emerged as an efficient heterologous production host for pharmaceutical or industrial proteins and enzymes (reviewed in the works of Madzak et al. 2004, 2013). Since the development by Pfizer Inc. (US) of a transformation method for *Y. lipolytica* (Davidow et al. 1985), the scientific literature has reported the successful expression of more than 120 proteins in this yeast, originating from approximately 75 species of various origins (reviewed in the works of Madzak et al. 2013).

The success of this research domain is largely due to the high secretion capacity of Y. lipolytica and to the predominance of the cotranslational secretion pathway in this yeast (Boisramé et al. 1998), a characteristic shared with higher eukaryotes (e.g., mammals), which optimizes the folding of complex heterologous proteins. The relatively recent development of different surface display systems now allows Y. lipolytica to be used as an arming yeast, further increasing its potential as a microbial cell factory (Yue et al. 2008; Yang et al. 2009b). However, most applications proposed in the scientific literature remain only prospective, their development rarely passing the exploratory stage. Indeed, there are currently only a few commercial or industrial applications for genetically modified (GM) strains of Y. lipolytica, which are described in Table 4.1 (data from Groenewald et al. 2014, from the companies' websites or personal communications). A commercial kit for heterologous expression/secretion is currently on the market (Yeastern Biotech Co., Taiwan) and Y. lipolytica has been used as a protein manufacturing platform (Protéus, France; Oxyrane, UK); both use technologies developed at INRA (French National Institute for Agricultural Research). Two categories of valuable compounds are presently produced using GM Y. lipolytica: carotenoids (Microbia, acquired by DSM in 2010) and eicosapentaenoic acid (EPA)rich products (DuPont). DuPont has recently disclosed all information about Y. lipolytica metabolic engineering for the production of EPA-rich single-cell oils (SCO; Xue et al. 2013; see Section 4.3.3.3). However, if their EPA-rich Y. *lipolytica* biomass is currently used as an ω-3 supplement in fish feed (harmoniously raised salmon, Verlasso), the marketing of EPArich SCO for use as a dietary supplement (New Harvest) has been recently discontinued (personal communication from DuPont). New Harvest was only advertised as a vegetarian alternative to fish-based  $\omega$ -3 oils; therefore, one can only wonder if public perception of its actual GMO-derived nature had something to do with DuPont's decision to stop its commercialization. In addition, two projects using recombinant enzymes, produced in engineered Y. lipolytica strains, for enzyme replacement therapies (ERTs), are now nearly on the marketing stage: human lysosomal enzymes, for the treatment of lysosomal storage diseases, for example, Pompe disease (Oxyrane, Belgium; Tiels et al. 2012; see Section 4.3.2.3) and overexpressed homologous Y. lipolytica lipase (Pignède et al. 2000) for the treatment of exocrine pancreatic insufficiency (Mayoly Spindler, France).

#### 4.3.2 *Genetic and molecular tools*

The development of tools for engineering *Y. lipolytica* dates back nearly 30 years. This yeast was at first distinguished for its high secretion yield of numerous proteins, especially of alkaline extracellular protease (AEP), encoded by *YlXPR2* gene. Consequently, this gene provided the first (and the most frequently used) elements necessary for heterologous

		-		
Years of activity	Company	Country	Product or use	Intellectual property, tools, safety, and efficacy data
Since 2004	Yeastern Biotech	Taiwan	YLEX Kit for expression of heterologous proteins in <i>Y. lipolytica</i>	INRA License, expression (pYLEX1) and secretion (pYLSC1) vectors + Po1g recipient strain
Since 2005	Protéus/PCAS Group	France	Use of <i>Y. lipolytica</i> as expression host in protein manufacturing platform	INRA License
2005–2007	Oxyrane	South Africa	Enantiomerically pure pharmaceuticals	INRA License, use of Y. lipolytica as expression platform for enantioselective epoxide hydrolases
Since 2007	Oxyrane	UK/ Belgium/ US	Idem + human lysosomal enzymes for ERTs of lysosomal storage diseases	INRA License, engineered Y. <i>lipolytica</i> strains for efficient targeting of therapeutic enzymes to lysosomes (high levels of mannose-6- phosphate); start of clinical trials for a treatment targeting Pompe disease
Since 2005	Microbia/DSM	US/ Netherlands	Carotenoids for use in food and feed	GRAS self-affirmation for $\beta$ -carotene
Since 2006	Mayoly Spindler + Protea Biosciences (since 2011)	France/US	Recombinant <i>Y. lipolytica</i> lipase for ERT	INRA License, overproduction of homologous LIP2p, clinical phase II trials for chronic pancreatitis
2010–2013	DuPont Applied Biosciences	US	EPA-rich biomass for use in fish feed, EPA-rich SCO for use as dietary supplement	SCO safety and efficacy studies, GRAS status (GRN 355), marketing of New Harvest SCO discontinued in 2013

Table 4.1 Commercial and Industrial Applications of GM Strains of Y. lipolytica

production in *Y. lipolytica*, namely, a promoter, a secretion signal (optional), and a terminator. The main genetic elements described in the literature for use in *Y. lipolytica* expression systems are listed in Table 4.2.

#### 4.3.2.1 Promoters

Despite its strength and historical importance, the *XPR2* promoter (*pXPR2*) was shown to have a complex regulation (Ogrydziak et al. 1977), which hindered its industrial use,

Type of element (most frequently used in boldtype)	Characteristics (reference) (when unspecified: Barth et al. 1996)
Selection	markers
LEU2, URA3, LYS5, ADE1	Auxotrophy complementation
ura3d4	Idem, defective allele (Le Dall et al. 1994)
E. coli Phleo <sup>R</sup> , <b>hph (Hyg<sup>R</sup>)</b>	Antibiotic resistance: dominant
S. cerevisiae SUC2	Sugar utilization: dominant (Nicaud et al. 1989)
Promoters (s	ource gene)
p <i>LEU2</i> (β-isopropylmalate dehydrogenase)	Inducible (leucine precursor)
pXPR2 (alkaline extracellular protease)	Inducible (peptones), complex regulation
<b>pPOX2</b> , pPOT1 (acyl-CoA oxidase 2,3-oxo-acyl-CoA thiolase)	Inducible (fatty acids and derivatives, alkanes)
pICL1 (isocitrate lyase)	Inducible (fatty acids and derivatives, alkanes, ethanol, and acetate)
pPOX1, pPOX5 (acyl-CoA oxidases 1 and 5)	Weakly inducible (alkanes)
p <i>G3P</i> (glycerol-3-phosphate dehydrogenase)	Inducible (glycerol)
p <i>MTP</i> (metallothioneins 1 and 2)	Bidirectional, inducible (metallic salts)
$pFBA1$ and $pFBA1_{IN}$ [+native intron] (fructose 1,6-biphosphate aldolase)	Inducible (glucose), very strong especially when <i>FBA1</i> intron is included (Hong et al. 2012)
<b>pTEF1</b> , pRPS7 (translation elongation factor-1α, ribosomal protein S7)	Constitutive (Müller et al. 1998)
<b>hp4d</b> (hybrid promoter derived from p <i>XPR2</i> )	Growth phase-dependent, based on four iterations of p <i>XPR2</i> UAS1B sequence (Madzak et al. 2000)
Hybrid promoters constructed on the "hpnd" model (derived from p <i>XPR2</i> )	Very strong, tunable, based on <i>n</i> iterations of UAS1B with <i>n</i> as high as 32 (Blazeck et al. 2011)
Secretion signals (processir	ng enzyme for maturation)
XPR2 prepro	157 aa, KR cleavage site (Xpr6p endoproteinase)
XPR2 pre	13 aa, LA cleavage site (signal peptidase complex)
LIP2 prepro	32 aa, KR cleavage site (Xpr6p endoproteinase)
<i>LIP2</i> pre + dipeptides track	21 aa, XA or XP cleavage site (diamino-peptidase)
GPI anchor signals	
C terminal of YICWP1p	110 aa (Yue et al. 2008)
C terminal of S. cerevisiae FLO1p	1097 aa (Yang et al. 2009b)
C terminal of YICWP3p or YICWP6p	121 and 139 aa, respectively (Yuzbasheva et al. 2011)
Termir	nators
XPR2t, LIP2t, PHO5t	430, 150, and 320 bp, respectively, of noncoding 3'
minimal XPR2t	100 bp of noncoding 3' + flanking restriction sites (obtained by PCR; Swennen et al. 2002)

Table 4.2 Genetic Elements Used in the Design of Y. lipolytica Expression System

*Note:* aa, amino acids; A, alanine; K, lysine; L, leucine; P, proline; R, arginine; X, any aa.

prompting the search for new promoters. A functional dissection of pXPR2 showed that one of its upstream activating sequences, UAS1B, was only weakly affected by environmental conditions (Blanchin-Roland et al. 1994; Madzak et al. 1999). This UAS1B element was used to design the recombinant hp4d promoter, composed of four tandem copies of UAS1B upstream from a TATA box (minimal pLEU2; Madzak et al. 2000). Patented as a strong promoter whatever the pH and composition of culture medium, hp4d is not however a constitutive promoter because it has retained some unidentified elements that confer a growth phase-dependent expression (Madzak et al. 1996, 2000). Namely, hp4d-driven expression is limited during cell growth and increases at the beginning of the stationary phase, naturally ensuring a partial dissociation of growth and expression phases. This very peculiar characteristic is particularly interesting for heterologous protein production because it maximizes productivity and alleviates possible toxicity problems. Consequently, hp4d is by far the promoter most frequently used for such applications (>40% of cases reported in the scientific literature, as reviewed in the works of Madzak et al. 2013). Two strong constitutive promoters have been isolated, for the purpose of expression cloning, from Y. lipolytica TEF1 and RPS7 genes (Müller et al. 1998). Although pTEF1 has also been used for heterologous production per se, these fully constitutive promoters are not recommended for such use because early expression of heterologous genes could impair cell growth. The use of inducible promoters responding to an easily controllable factor would ease industrial applications. Several laboratories have isolated such inducible promoters, among which the most interesting seem to be pICL1, pPOT1, and pPOX2, isolated from key genes from fatty acid metabolisms, which are particularly well adapted for use in engineering Y. lipolytica hydrophobic substrate pathways (Juretzek et al. 2000). More recently, several promoters have been isolated from glycolytic genes and described in patents on metabolic engineering of Y. *lipolytica* (reviewed in the works of Madzak et al. 2013). The most interesting of these glucose-induced promoters seems to be pFBA1, not only because its activity is particularly high but also because heterologous production can be further increased (by a fivefold factor) when retaining the upstream region of the FBA1 gene, carrying its 102 bp intron, in the recombinant construct (Hong et al. 2012). As demonstrated in the case of  $pFBA1_{INV}$  the presence of the intron was able to simultaneously enhance promoter activity and increase mRNA stability, a phenomenon previously reported in various eukaryotes as intron-mediated enhancement (Hong et al. 2012). Blazeck and coworkers (2011) have recently performed an important improvement in the design of strong and tunable Y. lipolytica promoters based on the concept of the recombinant hp4d promoter. These authors constructed a large set of recombinant promoters, in which various copy numbers (from 1 to 32) of UAS1B were inserted upstream of either a minimal pLEU2, or different fragments of pTEF1. The strongest of these "hpnd-like" promoters exhibited unprecedented efficiency, eightfold higher than that of any known natural Y. lipolytica promoter (Blazeck et al. 2011). These strong tunable promoters constitute promising tools, but the genetic stability of those with a high number of tandem repeats may reveal an inability to comply with the very strict standards required for industrial processes. Specifically, Good Manufacturing Practice (GMP) guidelines (http://www.gmp -compliance.org/eca\_link\_navigator.html) state that producing strains should be stable over hundreds of generations, when a "hp12d-like" promoter was shown to exhibit recombinations in 17 out of 20 progeny isolates after only 36 generations (Blazeck et al. 2011). When inserting either eight or 16 UAS1B upstream from different fragments of pTEF1 (from minimal to larger than full-length), the authors observed enhancing effects for all pTEF1 variants. Thus, this work suggests that endogenous promoters are

enhancer-limited and that this limitation can be alleviated through the addition of new UASs. Moreover, transcription factor availability does not seem to be a limiting factor, even for the strongest promoters constructed. Consequently, this work paved the way for designing fine-tuned promoters, in which added UASs could increase the expression level of endogenous promoters that could be selected from developing genomic and transcriptomic data.

#### 4.3.2.2 Targeting signals for secretion and surface display

The *XPR2* prepro region has been the most widely used secretion signal, until it was shown that the *XPR2* pre region alone was sufficient to drive efficient heterologous secretion and occasionally ensured a more complete maturation (Swennen et al. 2002). As a consequence, this smaller signal is now preferably used for directing secretion. Alternative secretion signals have also been used, such as *LIP2* prepro region and a *XPR2/LIP2* prepro hybrid (Nicaud et al. 2002). More recently, it was also shown that the *LIP2* pre region with its dipeptide stretch performed better than the larger *LIP2* prepro (Gasmi et al. 2011). In addition, native secretion signals, especially those originating from fungi or plants, were often shown to be efficient in *Y. lipolytica* (reviewed in the works of Madzak et al. 2013).

Surface display systems have been developed rather recently in *Y. lipolytica*. They are based on transcriptional fusion of the heterologous gene to a sequence encoding the C-terminal part of a cell wall protein, a region acting as a signal for the addition of a glyco-sylphosphatidylinositol (GPI) anchor. A GPI anchor is a posttranslational modification that fastens the modified secreted protein into the cell wall of eukaryotic cells. The first report of surface display in *Y. lipolytica* was by Yue and coworkers (2008), using the GPI anchor domain of YICWP1p, a cell wall protein previously described in an article by Jaafar et al. (2004). Alternative GPI anchor domains have also been used: a heterologous one, from *S. cerevisiae* FLO1p (Yang et al. 2009b), and those from newly described *Y. lipolytica* cell wall proteins, YICWP3p and YICWP6p (Yuzbasheva et al. 2011).

#### 4.3.2.3 Targeting to microbodies

The possibility of targeting heterologous proteins to cellular microbodies, such as peroxisomes or lysosomes, is of particular interest in metabolic engineering. Targeting to Y. lipolytica peroxisomes of a polyhydroxyalkanoate synthase from Pseudomonas aeruginosa has been obtained by fusion to the modified C-terminal region (34 amino acids) of glyoxysomal isocitrate lyase from Brassica napus (Haddouche et al. 2010). The C-terminal SKM tripeptide from this latter sequence, previously shown to direct efficient peroxisomal targeting in S. cerevisiae, was modified into AKI to fit the supposed Y. lipolytica tripeptide consensus, deduced from that found in Candida tropicalis, a more closely related yeast (Haddouche et al. 2010). More recently, a C-terminal SKL tripeptide was shown to be efficient for targeting green fluorescent protein (GFP) to Y. lipolytica peroxisomes (Xue et al. 2013). Targeting to Y. lipolytica lysosomes of recombinant human lysosomal enzymes, for ERTs of lysosomal storage diseases, has been obtained by Oxyrane (Belgium; see Table 4.1). They glycoengineered Y. lipolytica (through heterologous expression of a glycosidase from *Cellulosimicrobium cellulans*) for the production of lysosomal enzymes bound to high levels of mannose-6-phosphate, a specific sugar structure enabling efficient targeting of these therapeutic enzymes to the lysosomes through interaction with cation-independent mannose-6 phosphate receptors (Tiels et al. 2012). Recently, several heterologous proteins have been targeted to and anchored on the surface of Y. lipolytica oleosomes through their fusion to the C-terminal domain of a

sesame oleosin (Han et al. 2013). Namely, plant oleosins are naturally embedded into the phospholipid monolayer membrane of plant oleosomes, and their heterologous expression as fusion proteins in Y. lipolytica allows the design of functionalized oleosomes that can be easily purified using floating centrifugation. These nanoparticles (mean diameter, 200–300 nm) were found to be highly stable, which allows their application as scaffolds for protein immobilization and display. The proof of concept of codisplaying multiple proteins on Y. lipolytica oleosomes was demonstrated by using the high-affinity interaction between cohesin and dockerin domains from the cellulosome scaffolding system of Clostridium cellulolyticum (Fierobe et al. 1999). Oleosomes displaying a cohesin domain can be incubated with different dockerin-fusion proteins to form a self-assembled multifunctional nanoparticle able to target a specific cell type and to bring along new catalytic functions to its surface. For proof of concept, cohesin-displaying oleosomes interacted with (i) a GFP-dockerin fusion protein, for cell-targeting by binding to a GFP-antibody surface-displayed on Y. lipolytica cells; and (ii) a dockerin-bearing CelA endoglucanase from C. cellulolyticum, whose activity was followed by the detection of a red fluorescent compound. These engineered oleosomes were shown to be functional for both cell-targeting and reporting activities (Han et al. 2013). Such tunable multifunctional oleosomes could be useful for various applications, for example, targeted drug delivery, pathogen detection, or self-assembly of nanofactories (functionalized supramolecular nanostructures).

#### 4.3.2.4 Vectors, related methods, and amplification strategies

Similar to other yeasts, the expression/secretion vectors used for transforming Y. lipolytica are shuttle vectors containing a bacterial moiety (plasmidic backbone) and a yeast moiety composed of a selection marker, an expression/secretion cassette, and an element for transformation and maintenance into yeast cells. The latter can be either an autonomously replicating sequence (ARS) in replicative vectors, or a sequence that will target or promote integration into the genome in integrative vectors. In Y. lipolytica, replicative vectors are not very attractive for heterologous protein production due to their low copy number and high loss frequency, both linked to the colocalization of centromeric and replicative functions in ARSs (Fournier et al. 1993). However, they are useful when transient expression is required, for example, when using marker rescue for performing multiple gene deletions. Namely, an efficient gene knockout method based on Cre-lox recombination system consists of integrating a promoter-terminator gene disruption cassette, carrying a selection marker flanked by *loxR–loxP* sequences, which are able to recombine during transient expression of Cre recombinase, allowing efficient marker rescue (Fickers et al. 2003). This method constitutes a powerful tool for genetic engineering of metabolic pathways in Y. lipolytica. Integrative vectors are widely used for heterologous production in Y. lipolytica. Their integration is generally targeted by linearizing the vector within a region homologous to the genome and occurs by single crossover. Their transformation efficiency can be very high (in the range of 10<sup>5</sup> transformants per microgram of DNA) when using chemical (lithium acetate) methods (Chen et al. 1997; Madzak et al. 2005; Xuan et al. 1988). INRA has developed autocloning vectors from which the bacterial moiety can be removed prior to transformation, enabling the resulting recombinant strain to retain its GRAS status (Nicaud et al. 2000). These autocloning vectors use zeta sequences, which are the long terminal repeats (LTRs) from Ylt1 retrotransposon for targeting or promoting integration into the Y. lipolytica genome. In Ylt1-carrying strains, zeta sequences provide at least 100 targeting sites per genome (Schmid-Berger et al. 1994), constituting a powerful tool for multiple homologous integration when combined with

a defective selection marker. Surprisingly, zeta sequences also exhibit the unexpected property of enhancing nonhomologous integration when introduced into *Y. lipolytica* strains devoid of Ylt1 retrotransposon, thus providing a method for random genomic integration (patented by Nicaud et al. 2000). Some examples of ready-to-use vectors for intracellular expression, secretion, or surface display of heterologous proteins are shown in Table 4.3.

Some expression/secretion vectors and optimized strains constructed at INRA have been commercially available since 2006, such as the YLEX Expression Kit (see Table 4.1), which can be purchased from Yeastern Biotech Co. (Taiwan) or from their retailer (for Europe, the United States, and Canada), Gentaur (Belgium). The preferred strategy for increasing the copy number of a heterologous gene in Y. lipolytica is to perform multiple integrations of a vector (or expression cassette) into the genome, which can only be obtained by using a defective selection marker, such as the *ura3d4* allele (Le Dall et al. 1994). It is noteworthy that the use of *ura3d4* for selection dramatically decreases the transformation efficiency to approximately 10<sup>2</sup> transformants per microgram (Juretzek et al. 2001). High copy number integrants can be obtained not only when using repeated sequences (rDNA or zeta) as targets but also with a single integration site (Juretzek et al. 2001), because multiple homologous integrations almost always occur in tandemly repeated copies, at only one or two sites, irrespective of the number of potential targets (Le Dall et al. 1994). This phenomenon could unfortunately lead to genetic instability problems in the producing strains. In this context, the development of a nonhomologous integration system, by using autocloning vectors carrying zeta sequences into Y. lipolytica strains devoid of Ylt1 retrotransposon, has offered new possibilities: randomly integrated copies are more dispersed in the genome, leading to a better stability of high copy number integrants (Nicaud et al. 2000). This method, however, has its own drawbacks: transformation efficiency is reduced further, to approximately 10 transformants per microgram (Juretzek et al. 2001), when random integration effects require the testing of numerous transformants for selecting a good producer strain. Despite such difficulties, this method has been successfully used to produce numerous heterologous proteins, as reviewed in the works of Madzak et al. (2013). However, some producing strains developed at INRA, using random multiple integrations, have recently failed to obtain marketing authorizations due to an insufficient stability level. Even if these strains were able to roughly maintain their copy number, some changes in integration sites were observed over time (Nicaud et al. 2009); the regulations regarding GMO from GMP guidelines require that genomic modifications of the producing strains should be described precisely, and be stable over hundreds of generations. To alleviate these problems, a new strategy for obtaining stable multicopy transformants was recently patented (Nicaud et al. 2009). This new approach provides a genetically engineered strain, deleted for major homologous secreted proteins (to alleviate metabolic burden), in which several loci have been selected for targeting integrative vectors. This strain could be sequentially transformed by a set of vectors carrying different auxotrophic or dominant markers, with the possible use of marker rescue. Successive integrations could be checked by the inactivation of different reporter genes, which generates new phenotypes detectable in plate assays. This method allows us to obtain three to 10 stable copies of a gene of interest, or to coexpress the same number of different new functions for the engineering of Y. lipolytica metabolic pathways (Nicaud et al. 2009). Finally, another strategy for increasing heterologous production or for coexpressing several heterologous genes is to insert multiple expression cassettes into a single vector. The limitations of this method are the vector size (probably limited to between 12 and 14 kb) and possible recombinations between repeated sequences if the same promoter/targeting/terminator or even gene sequences are

sequence

zeta

sequence

zeta

sequence

(Nicaud et al. 2002)

(Madzak et al.

submitted)

		of Heterologous F1		Joiyiicu				
		Plasmid type: char	acteristics/dr	awbacks				
		Expression	Selection	Targeting				
Function	Name	cassette	marker	sequence	Reference			
		vectors for precise ta						
transform	nation efficien	cy, growth-phase rela		on (hp4d prom	oter)/presence of			
		bacterial	sequences					
Intracellular expression	pINA1269	hp4d/MCS/ XPR2t	LEU2	pBR322 backbone	(Madzak et al. 2000)			
Secretion	pINA1296	hp4d/XPR2 pre/ MCS/XPR2t	LEU2	pBR322 backbone	(Madzak et al. 2000)			
Mono/low co	py auto-cloni	ng vectors for targeti	ng at zeta seq	uences or rand	om integration into			
Mono/low copy auto-cloning vectors for targeting at zeta sequences or random integration into Ylt1-devoid strains: high transformation efficiency, devoid of bacterial DNA, constitutive								
		hase related expressi						
Intracellular expression	pINA1312	hp4d/MCS/ XPR2t	ura3d1	zeta sequence	(Nicaud et al. 2002)			
Intracellular expression	JMP62	pPOX2/MCS/ LIP2t	ura3d1	zeta sequence	(Nicaud et al. 2002)			
Intracellular expression	pKOV96	p <i>TEF1/MCS/</i> <i>LIP2</i> t	ura3d1	zeta sequence	(Labuschagne et al. 2007)			
Secretion	pINA1317	hp4d/XPR2 pre/ MCS/XPR2t	ura3d1	zeta sequence	(Nicaud et al. 2002)			
Secretion	JMP61	pPOX2/LIP2 prepro/MCS/ LIP2t	ura3d1	zeta sequence	(Nicaud et al. 2002)			
Surface display	pINA1317- YICWP110	hp4d/XPR2 pre/ MCS/Y1CWP1GPI anchor/XPR2t	ura3d1	zeta sequence	(Yue et al. 2008)			
Multicopy	auto-cloning	vectors for targeting	at zeta seque	nces or randon	n integration into			
		id of bacterial DNA,						
induction (pPOX2), amplification of expression/very low transformation efficiency								
Intracellular expression	pINA1292	hp4d/MCS/ XPR2t	ura3d4	zeta sequence	(Nicaud et al. 2002)			
Intracellular expression	JMP64	pPOX2/MCS/ LIP2t	ura3d4	zeta sequence	(Nicaud et al. 2002)			
Secretion	pINA1297	hp4d/XPR2 pre/	ura3d4	zeta	(Nicaud et al. 2002)			

 Table 4.3 A Selection of Ready-to-Use Integrative Vectors for Expression/Secretion/Surface Display of Heterologous Proteins in Y. lipolytica

*Note:* MCS, multiple cloning sites; pINA1269 and pINA1296 are, respectively, pYLEX1 and pYLSC1 in the YLEX kit (Yeastern Biotech Co., Taiwan).

ura3d4

ura3d4

MCS/XPR2t

pPOX2/LIP2

prepro/MCS/

LIP2t

hp4d/XPR2 pre/

MCS/YICWP1GPI

anchor/XPR2t

Secretion

Surface

display

JMP63

pINA1297-

YICWP110

used for different cassettes. Vectors with two expression cassettes are currently used for marker rescue (Fickers et al. 2003), although for transient expression only. More recently, *Y. lipolytica* fatty acid pathways were engineered by the coexpression of two heterologous desaturases, using a tandem double cassette vector with the same hp4d promoter (Chuang et al. 2010; see Section 4.3.3.3). Finally, triple and even quadruple expression cassettes in a single vector were described in several patents (using different promoters and terminator sequences) and, very recently, in an article by Celińska et al. (2013). These authors chose to coexpress three heterologous genes involved in glycerol metabolism using the same regulatory elements: a glycerol-induced *G3P* promoter and an *XPR2*-like terminator. The expression cassettes were shown to be stably maintained in the genome throughout more than 40 passages in selective medium, lyophilization followed by reviving process, and bioreactor cultivation without selective pressure (Celińska et al. 2013).

#### 4.3.2.5 Strains

The E150 strain (MatB, his1, leu2-270, ura3-302, and xpr2-322) was chosen for the Y. lipolytica genome-sequencing project (Dujon et al. 2004). Y. lipolytica strains used for heterologous production were selected empirically at first for their growth and secretion capacities, and genetically modified to facilitate their use or to improve yields. A comprehensive list of such Y. lipolytica strains was published by Madzak et al. (2004). The most frequently used for heterologous expression are E129 (MatA, lys11-23, leu2-270, ura3-302, and xpr2-322) and essentially Pold (MatA, leu2-270, ura3-302, and xpr2-322) and its derivatives. Pold presents many interesting features: a high level of secretion, the deletion of AEP, which was a potent threat for secreted heterologous proteins, two auxotrophies due to nonrevertant mutations and the production of recombinant invertase (ura3-302 allele corresponds to ura3::pXPR2:SUC2; namely, disruption of URA3 by SUC2 from S. cerevisiae under the control of pXPR2), which allows it to metabolize sucrose (Nicaud et al. 1989). These characteristics have been retained in a series of derivatives that were further adapted for heterologous protein production (Madzak et al. 2000, 2004): Polf, Polg, and Polh were deleted for acid extracellular protease ( $\Delta AXP$ : *axp1-2* allele), the other source of secreted proteasic activity. Polg received an integrated pBR322 docking platform to ease the integration of pBR-based vectors. Finally, Po1g and Po1h retain only one auxotrophy, respectively, Leu<sup>-</sup> and Ura<sup>-</sup>. Some Y. lipolytica strains, like E129 and E150, carry Ylt1 retrotransposons in their genome, when others, like the Po1 series of strains, are devoid of this transposable element. All these strains are available from INRA's CIRM (International Center for Microbial Resources) Yeasts Library (http://www7.inra.fr/cirmlevures/page.php?page=home&lang=en). Other derivatives of Pold have been optimized for applications linked to lipid metabolism, such as the MTLY60 strain, deleted for extracellular lipases (MatA, ura3-302, leu2-270, xpr2-322,  $\Delta lip2 \Delta lip7$ , and  $\Delta lip8$ ), which has been further fitted with a zeta docking platform to produce the JMY1212 strain, used in high-throughput expression platforms (Bordes et al. 2007; see Section 4.3.3.1). More recently, "obese" strains were derived from Po1d by deleting YlGUT2, encoding glycerol-3-phosphate dehydrogenase and, optionally, the acyl-coA oxidase genes YIPOX1 to POX6 (implicated in the  $\beta$ -oxidation pathway), have been patented at INRA (Nicaud et al. 2000). These mutant strains exhibit a threefold and fourfold increase, respectively, in lipid accumulation, which makes them interesting tools for the production of valuable lipid products (Beopoulos et al. 2008; see Section 4.3.3.4). Finally, some wild-type strains have been selected for their interesting properties (high lipid or protein content) from a library of 78 marine Y. lipolytica strains (Ocean University of China, Qingdao) and have been genetically engineered for use as SCO (Zhao et al. 2010), SCP (Cui

et al. 2011), or citric acid producers (Liu et al. 2010), illustrating the potentialities offered by exploiting natural *Y. lipolytica* biodiversity.

## 4.3.3 Examples of processes and products of recombinant Y. lipolytica strains

As already exemplified in the commercial and industrial applications presented in Table 4.1, genetic engineering of *Y. lipolytica* for the production of valuable compounds has developed in two directions: on the one hand, using it as a host for the production of heterologous proteins of interest and, on the other hand, modifying its metabolism for use as a whole cell biocatalyst (microbial factory). Moreover, the relatively recent development of surface display and lysosome targeting methods opened new possibilities of using arming *Y. lipolytica* or derived nanoparticles for molecular recognition applications. For example, a *Y. lipolytica* strain harboring hemolysin from *Vibrio harveyi* could be used as a live vaccine for the fish-farming industry (Yue et al. 2008). However, it remains critical to know if, despite its GRAS status, the use of GM *Y. lipolytica* for whole cell applications will be socially acceptable, especially in the environmental domain and in the food industry.

# 4.3.3.1 Y. lipolytica as a high-throughput expression platform for protein engineering

Industrial processes generally require optimizing enzymatic properties using protein engineering and directed evolution methods. For this purpose, a high-throughput expression platform was developed for *Y. lipolytica* using the JMY1212 strain (INRA/INSA Toulouse/Toulouse University, France collaboration; Bordes et al. 2007). It was applied to the engineering of *C. antarctica* lipase B (CalB), one of the enzymes most widely used in industrial biocatalysis: large libraries of CalB mutants were created and screened, identifying new variants with higher catalytic efficiencies (Emond et al. 2010).

# 4.3.3.2 *Glycoengineering of* Y. lipolytica for the production of humanized pharmaceutical proteins

Yeasts modify glycoproteins with high mannose-type N-glycans, which can reduce the protein half-life in vivo and could be immunogenic. Since the genetics of N-glycosylation of proteins in Y. lipolytica were developed (Barnay-Verdier et al. 2004, 2008), some research groups from South Korea and Belgium focused on engineering the Y. lipolytica metabolism to provide "humanized" strains that are able to produce glycoproteins with humancompatible N-linked oligosaccharides for therapeutic applications. Song and coworkers (2007) have shown that a strain deleted for  $\alpha$ -1,6-mannosyltransferase YlOCH1 synthesized only the core oligosaccharide Man8GlcNAc2, whereas wild-type strains synthesized oligosaccharides with heterogeneous sizes (up to Man12GlcNAc2). More recently, the same South Korean research groups showed that YIMPO1 was necessary for mannosylphosphorylation of N-linked oligosaccharides in secreted proteins (Park et al. 2011). They constructed a  $\Delta Y loch1 \Delta Y lmpo1$  double mutant as a host for producing glycoproteins lacking yeast-specific hypermannosylation and mannosylphosphorylation. Another strategy was developed by the Belgian research groups, which engineered N-glycan biosynthesis in Y. lipolytica so that it produces N-glycoproteins homogenously carrying a Man3GlcNAc2 core. After patenting their work, the steps involved were recently fully described (De Pourcq et al. 2012a): (i) disruption of YIALG3 α-1,3-mannosyltransferase, (ii) overexpression of YIALG6 glucosyltransferase, (iii) overexpression of the heterodimeric Apergillus niger glucosidase II, and (iv) overexpression of a Y. *lipolytica*-optimized ER-targeted T. reesei  $\alpha$ -1,2-mannosidase. Interestingly, the resulting core Man3GlcNAc2 is common to all mammalian N-glycan structures, and it can be further modified *in vitro* into any complex type of *N*-glycan, using the appropriate glycosyltranferases and sugar-nucleotide donors (De Pourcq et al. 2012a). The same research group also engineered *Y. lipolytica* to produce human-type *N*-glycoproteins homogenously carrying either Man8GlcNAc2 or Man5GlcNAc2. For this purpose, they inactivated both yeast-specific Golgi  $\alpha$ -1,6-mannosyltransferases YlOch1p and YlMnn9p, yielding a strain producing homogeneous Man8GlcNAc2 *N*-glycans. Further introduction of a *Y. lipolytica*-optimized ER-targeted *T. reesei*  $\alpha$ -1,2-mannosidase yielded a strain producing proteins homogeneously glycosylated with Man5GlcNAc2 (De Pourcq et al. 2012b). These new expression platforms will allow the development of *Y. lipolytica* as a competitive heterologous host for pharmaceutical applications.

#### 4.3.3.3 Y. lipolytica as whole-cell biocatalyst for SCO production

Genetic engineering of Y. lipolytica for use as a whole-cell biocatalyst has essentially focused on its lipid metabolism pathways. This oleaginous yeast possesses a very effective lipid uptake system, enabling the efficient transport of hydrophobic substrates, from protrusions on the cell surface to catalytic sites within the cells through specific channels (Fickers et al. 2005). Y. lipolytica can accumulate lipids to more than 50% of its cell dry weight and, among oleaginous yeasts, it has the highest proportion of linoleic acid (more than 50% of fatty acids; Beopoulos et al. 2009). There is a strong economic interest in the production of SCO enriched in essential fatty acids, namely, unsaturated fatty acids that are not synthesized by mammals but are essential for health and must be supplied by the diet. Engineering Y. lipolytica  $\omega$ -3/ $\omega$ -6 biosynthetic pathways target the commercial niche of SCO enriched with  $\omega$ -3 fatty acids, such as EPA, docosahexaenoic acid (DHA), and  $\alpha$ -linolenic acid (ALA), or  $\omega$ -6 fatty acids such as arachidonic acid (AA) and  $\gamma$ -linolenic acid (GLA), for use as dietary supplements in feed or food. Damude and coworkers (2006) have obtained a GM Y. lipolytica strain accumulating ALA up to 28% of its cell dry weight, by expressing a bifunctional  $\Delta 12/\omega 3$  desaturase from *Fusarium moniliformis*. More recently, efficient GLA production, accounting for 20% of the total lipids and 44% of triacylglycerol fraction, was obtained by coexpression of  $\Delta 6$  and  $\Delta 12$  desaturases from *Mortierella alpina* under the control of hp4d promoter (Chuang et al. 2010). After patenting their process for the production of EPA-rich SCO (see Table 4.1), DuPont has recently described the numerous steps involved in the metabolic engineering of their production strain, which accumulated EPA to 15% of its dry cell weight and 57% of its lipids (Xue et al. 2013). Briefly, to produce EPA from endogenous LA whereas avoiding GLA build-up, the  $\Delta$ -9 pathway was selected by overexpressing several heterologous enzymes ( $\Delta$ -9 elongases,  $\Delta$ -8 desaturases,  $\Delta$ -5 desaturases,  $\Delta$ -17 desaturases,  $\Delta$ 12 desaturases, and C16/18 elongases) and further improvements were brought by knocking out several genes (YIPEX10, involved in the import of peroxisomal matrix proteins; YlLIP1, encoding lipase 1; and YlSCP2, encoding a sterol carrier protein).

#### 4.3.3.4 Y. lipolytica as whole-cell biocatalyst for carotenoid production

Microbia (see Table 4.1) and DuPont are independently developing *Y. lipolytica* as a production host for carotenoids, organic pigments used as natural coloring, and antioxidant agents for food and feed. Ye and coworkers (2012) from DuPont disclosed some information on *Y. lipolytica* engineering for the production of lycopene (an important intermediate in the biosynthesis of carotenoids), involving the expression of bacterial *crtE*, *crtB*, and *crtI* genes. Very recently, the production of unprecedented lycopene levels for a eukaryotic host was reported by a group from Dresden University. These authors used an "obese" strain ( $\Delta YIGUT2$ ,  $\Delta YIPOX1$  to POX6) developed at INRA, in which they expressed *crtB* and

*crtI* genes from *Pantoea ananatis* and overexpressed the rate-limiting genes for isoprenoid biosynthesis in *Y. lipolytica, YIGGS1*, and *YIHMG1* (Matthäus et al. 2014).

#### 4.3.3.5 Future prospects

The production of biofuel, namely, of biosynthetic hydrocarbons from engineered microorganisms, in replacement of petroleum-derived fuels, constitutes a developing research area. In that context, Blazeck and coworkers (2013) recently engineered Y. lipolytica to produce pentane from linoleic acid by strongly overexpressing a soybean lipoxygenase (Blazeck et al. 2013). This work constitutes the first report of pentane microbial production and demonstrates that Y. lipolytica is a promising host for short-chain n-alkane synthesis. This yeast is also expected to become a source for many other valuable lipid-derived compounds such as wax esters (industrial lubricants), polyhydroxyalkanoates (PHAs), and free hydroxylated fatty acids (HFAs, used as chiral building blocks in fine chemicals synthesis). PHAs are linear polyesters produced by bacterial fermentation of sugar or lipids, which are biodegradable and can be used for producing bioplastics. Recombinant Y. lipolytica strains expressing the PHA synthase gene from *P. aeruginosa* in their peroxisomes have been shown to produce PHAs in various yields, depending on their POX (acyl-CoA oxidases) genotype (Haddouche et al. 2010). It was further shown that redirecting the fatty acid flux toward  $\beta$ -oxidation, by deleting neutral lipid synthesis pathways (YlLRO1, DGA1, DGA2, and ARE1 acyltransferase genes), increased PHA levels (up to more than 7% of cell dry weight; Haddouche et al. 2011). Developing Y. lipolytica into a versatile and highthroughput microbial factory for the production of industrially valuable lipid-derived compounds (wax esters, isoprenoid-derived compounds, PHAs, and HFAs) constitutes the purpose of the European Union-sponsored LipoYeasts project (Sabirova et al. 2011). This consortium used specific enzymatic pathways from hydrocarbonoclastic bacteria, and from other hydrocarbon and lipid-degrading bacteria from various lipid-contaminated environments, to redirect Y. *lipolytica* versatile lipid metabolisms toward high-value lipid products. The main metabolic routes to be optimized will be  $\beta$ - and  $\omega$ -oxidations, together with lipid accumulation. The results of this work can be found on the consortium web page (http://www.lipoyeasts.ugent.be/).

## 4.4 Hansenula polymorpha

*H. polymorpha* is a thermotolerant species capable of growing at record high temperatures of up to 50°C (Cabeç-Silva et al. 1984; Ishchuk et al. 2009; Reinders et al. 1999). The maximal growth temperature of this species is the absolute recorded among all known yeast species. It is only 10°C below the maximal growth temperature of all known eukaryotic organisms (Middelhoven 2002; Reinders et al. 1999). Cells of *H. polymorpha*, grown under increased temperatures, accumulate trehalose as a thermoprotector (Reinders et al. 1999). Thermotolerance could be further elevated by the disruption of the *ATH1* gene, coding for acid trehalose, or by the overexpression of *HSP16* and *HSP104* genes coding for two heatshock proteins (Ishchuk et al. 2009).

*H. polymorpha* is widely used by scientists and in industrial biotechnology, reaching in this respect *P. pastoris* among the methylotrophic yeasts. Along with this latter, *H. polymorpha* is the favorite organism in yeast cell biology, especially for studying peroxisome biogenesis and degradation. *H. polymorpha* is the only species among the methylotrophic yeasts used to produce bioethanol from lignocellulosics and glycerol (Dmytruk et al. 2013; Hong et al. 2010; Ishchuk et al. 2009; Ryabova et al. 2003). Additionally, *H. polymorpha* is known to be the most efficient yeast producer of glutathione (Ubiyvovk et al. 2011a). The most popular strains of *H. polymorpha* are DL-1, CBS4732, and NCYC495. There have been lengthy discussions on the reclassification of *H. polymorpha*. Alternative names have been proposed, such as *Pichia angusta* (Barnett et al. 2000; Kurtzman et al. 1998), *Torulopsis methanothermo, Hansenula angusta*, or *Ogataea polymorpha* (Middelhoven 2002; Sudbery 2003; Suh et al. 2010; Teunison et al. 1960; Yamada et al. 1994). It has to be pointed out that strains CBS4732 and NCYC495 are closely related and intercrossed whereas strain DL-1 is unable to cross with either of the abovementioned strains. According to the latest classification, strains CBS4732 and NCYC495 belong to the species *O. polymorpha*, whereas strain DL-1 belongs to the separate species *Ogataea parapolymorpha*. Additional species are also known (*Ogataea angusta*; Suh et al. 2010; Kurtzman 2011; Naumova et al. 2013). In this review, the commonly accepted name *H. polymorpha* is used.

#### 4.4.2 Genetic methods and genome research

Using an inbreeding program, a genetic cell line of *H. polymorpha* derived from strain CBS4732 was isolated. This presents an efficient hybridization and massive sporulation of four-spored asci, thus, allowing, tetrad analysis (Lahtchev et al. 2002). Auxotrophic and other types of mutants have been isolated from this genetic cell line. Most of the markers analyzed showed Mendelian segregation in tetrad analysis. Strain DL-1, although unable to hybridize, seems to be suitable for molecular genetics experiments. It is characterized by high growth rate and exhibits a high frequency of homologous recombination (Kang et al. 2001). Transformation of *H. polymorpha* was originally developed for *ura3* mutants with URA3 gene as selectable marker (Merckelbach et al. 1993). Besides this, leu2 mutant and heterologous LEU2 gene from S. cerevisiae have been proposed (Tikhomirova et al. 1986, 1988). Several dominant markers, such as geneticin (G418) and zeocin resistance for transformant selection have been adapted for H. polymorpha (Dmytruk et al. 2008a,b). Transformation is carried out using the LiCl-based method, protoplasts in the presence of polyethylene glycol (Tikhomirova et al. 1988) or electroporation (Faber et al. 1994). Methods for gene disruption and knock out in *H. polymorpha*, including those with the use of modified CreloxP system, have been described (Kang et al. 2002; Qian et al. 2009). Methods of multicopy gene integrations in the genome of *H. polymorpha* have also been developed (Agaphonov et al. 1999; Krasovska et al. 2013; Sohn et al. 1996, 1999).

Pulse field electrophoresis of *H. polymorpha* displays six chromosomes in both strains (CDS4732 and DL-1), with sizes ranging from 0.9 to 1.9 Mb, although the electrophoretic patterns of chromosomes in both strains is quite different (Ramezani-Rad et al. 2003; Kunze et al. 2009). Pulse electrophoresis of chromosomes from nine different strains of *H. polymorpha* showed a strong difference in chromosome numbers, which ranged from two to six depending on the strains (Marri et al. 1993).

Until now, the genomes of the three abovementioned strains have been sequenced. However, data from strains CDS4732 and DL-1 are not public and belong to a German private company, Rhein Biotech GmbH (Ramezani-Rad et al. 2003) and the Korean Institute KRIBB (Kang et al. 2005, personal communication). Data on the genome sequence of strain NCYC495 is publicly available (http://genome.jgi-psf.org/Hanpo1/Hanpo1.home.html). The genome of *H. polymorpha* NCYC 495 is 8.97 Mb in size and contains 5162 open reading frames.

In strain CBS4732, 8.73 Mb of DNA is sequenced, which contains 5848 open reading frames coding for proteins larger than 80 amino acids. In addition, 389 open reading frames were identified as coding for proteins containing less than 100 amino acids and 4771 open reading frames (81.6%) showed homology to known proteins. Average gene density is one gene per 1.5 Kb, whereas the protein contains 440 amino acids on average. In total, 91 introns were identified in the *H. polymorpha* genome. Functional annotation showed the following gene distribution depending on putative functions: 4% encode genes involved in energy metabolism, 3% in signaling, 6% in protein synthesis, 4% in stress response, 9% in transport, 9% in cell cycle and DNA metabolism, 17% in protein processing, 13% in transcription, and 19% in metabolism.

# 4.4.3 Genetic control of methanol induction and glucose catabolite repression in H. polymorpha

In H. polymorpha, and similarly to other methylotrophic yeasts, methanol strongly induces synthesis of the enzymes involved in primary methanol metabolism localized both in cytosol (formaldehyde and formate dehydrogenases, dihydroxyacetone kinase, and fructose-1,6-bisphosphatase) and in peroxisomes (alcohol oxidase, catalase, and dihydroxyacetone synthase). Methanol also induces peroxisome growth and proliferation (Eggeling et al. 1980; van der Klei et al. 2006; Veenhuis et al. 1983). Shifting methanol-grown cells to glucose or ethanol-based media represses the synthesis of these enzymes and additionally causes autophagic peroxisome degradation or pexophagy (Dunn et al. 2005; van der Klei et al. 2006). Transcriptome analysis of *H. polymorpha* showed that 2 h after shifting glucosegrown cells to methanol, 1184 genes out of approximately 6000 showed a twofold increased expression level, whereas 1246 other genes presented a twofold reduction of their expression level (van Zutphen et al. 2010). Maximal induction was observed for the genes of the central regulator of methanol metabolism MPP1 (394-fold) and FMD, coding for formate dehydrogenase (347-fold). A strong induction was also observed for genes coding for enzymes of fatty acid  $\beta$ -oxidation, glyoxylic shunt and carnitine mitochondrial transporter (111-fold increase). The MPP1 gene identified belongs to the family of  $Zn(II)_2Cy_{6}$ zinc cluster transcriptional activators (Leão-Helder et al. 2003). The mutant  $\Delta mpp1$  failed to grow on methanol as the sole carbon and energy source, and was characterized by a strong decrease in the level of peroxins and enzymes of methanol catabolism. However, the mechanism of transcription activation by Mpp1p remains unknown. H. polymorpha MPP1 shows a close similarity with transcription factors from other species of methylotrophic yeasts, namely, P. pastoris MXR1 (Lin-Cereghino et al. 2006) and Candida boidinii TRM1 (Sasano et al. 2008), which are also specifically required for the expression of genes involved in methanol metabolism and peroxisome biogenesis.

Carbon catabolite repression is the process of differential inhibition of enzymes of less favored carbon substrates (which support lower growth rate) in the medium containing more favored (supporting higher growth rate) substrates (Schuller 2003; Stasyk et al. 2003). In *S. cerevisiae*, catabolite repression is controlled by transcription repressors *MIG1* and *MIG2*, as well as transcription activators and different protein kinases (Gancedo 2008; Turcotte et al. 2010). An important role in catabolite repression in *S. cerevisiae* belongs to the *HXK2* gene encoding hexokinase II (Kraakman et al. 1999). In *H. polymorpha*, catabolite repression is induced by glucose and ethanol. The latter compound is involved only in the repression of enzymes of methanol metabolism whereas glucose additionally represses enzymes involved in the catabolism of alternative substrates, such as maltose (Alamäe et al. 2003; Liiv et al. 2001). There are several publications on *H. polymorpha* mutants that are capable of synthesizing enzymes of methanol metabolism and peroxins in medium

containing glucose as the sole carbon source (Alamäe et al. 1998; Hodgkins et al. 1993; Parpinello et al. 1998; Roggenkamp 1988; Stasyk et al. 1997a). Only some of the corresponding genes and mutations were identified. In one such mutant, the corresponding mutation was recessive and monogenic, and was designated as *glr1* (Parpinello et al. 1998). These mutations led to the synthesis of enzymes of methanol metabolism, corresponding mRNAs and peroxisome proliferation in medium containing glucose but not methanol. Repression by ethanol and pexophagy were not affected. A double mutant of *H. polymorpha*, which was defective in hexokinase and glucokinase, was unable to catabolize glucose and showed defects in catabolite repression of enzymes from methanol metabolism and maltase. Thus, sugar phosphorylation is necessary for repression (Karp et al. 2003; Kramarenko et al. 2000; Laht et al. 2002).

Mutants defective in glucose catabolite repression were also isolated among strains resistant to 2-deoxyglucose in a methanol medium (Stasyk et al. 1997b). Genetic analysis led to the identification of the monogenic recessive mutation gcr1. Subsequently, the wild-type GCR1 allele and mutant gcr1-2 allele were cloned and sequenced. Additionally, knock-out mutant  $\Delta gcr1$  was isolated (Stasyk et al. 2004). These mutants were characterized by pleiotropic defects of metabolism, namely, by constitutive synthesis of peroxisomal enzymes of methanol metabolism (alcohol oxidase, catalase) and constitutive presence of peroxisome in glucose medium (but not in ethanol medium), a decrease in the intracellular level of glycolysis intermediates, and defects in catabolite repression of cytosolic enzymes formaldehyde, formate dehydrogenases, and  $\alpha$ -glucosidase. GCR1 gene displayed strong homology to yeast and fungal genes coding for hexose transporters and sensors. Gcr1 protein is localized on the cytoplasmic membrane. It was suggested that Gcr1 protein fulfills transport and sensing functions in glucose catabolite repression but not pexophagy (Stasyk et al. 2004). Subsequently, two new genes (HXS1 and HXT1), coding for glucose sensors and glucose transporters, respectively, have been identified in H. polymorpha (Stasyk et al. 2008). Thus, H. polymorpha possesses several hexose sensors and transporters involved in catabolite repression but not in pexophagy. To identify the other components involved in catabolite repression in *H. polymorpha*, orthologues of transcriptional repressor S. cerevisiae genes MIG1, MIG2, and TUP1, as well as their knockout mutants, were isolated. Surprisingly, it was found that in single  $\Delta mig1$ , double  $\Delta mig1 \Delta mig2$  as well as in  $\Delta tup1$  mutants, glucose catabolite repression was only slightly affected (Stasyk et al. 2007). At the same time, the abovementioned mutations strongly impaired pexophagy induced by both glucose and ethanol. These data suggest substantial differences in mechanisms of catabolite repression between S. cerevisiae and H. polymorpha. It was also suggested that in H. polymorpha catabolite repression is mediated by orthologues of S. cerevisiae transcriptional regulators MIG1, MIG2, and TUP1.

# 4.4.4 H. polymorpha as a system for the expression of own and heterologous proteins

As mentioned previously, *H. polymorpha* is, along with *P. pastoris*, a favorite system for the expression of heterologous proteins. For expression, mitotically stable integrative vectors are used (Gellissen et al. 1997). Some strains allow multicopy integration with maximally up to 100 copies of integrated plasmids per genome represented as tandem repeats (Gatzke et al. 1995; Gellissen 2000; Kang et al. 2002). In other strains, multicopy integration does not occur spontaneously. Two methods of multicopy integrant selection in such strains have been developed based on the use of heterologous yeast auxotrophic genes, namely, the

S. cerevisiae URA3 gene and the P. pastoris ADE1 gene with a shortened native promoter. Sequential use of both selection markers produced stable transformants containing up to 30 integration cassettes with HBsAg gene encoding hepatitis B surface antigen (Krasovska et al. 2013). In this case, vectors contained FMD, MOX, TPS1 promoters of formate dehydrogenase, alcohol oxidase, and trehalose-6-phosphate synthetase genes. Also, recombination in the corresponding chromosomal genes occurred (Kunze et al. 2009). It is interesting to note that the production of proteins, especially of secreted ones, often shows no correlation with the copy number of the integrated gene (Gellissen et al. 1992). For instance, maximal expression of the urinary plasminogen activator u-PA and of human serum albumin HSA in DL-1 strain was achieved by integration of one to two copies of expression vector (Kang et al. 2001), whereas for oversynthesis of other proteins integration of high copy number of the expression cassette is necessary (Krasovska et al. 2013). Directed integration in *H. polymorpha* needs much longer homologous sequences as compared with *S. cerevisiae* (González et al. 1999; van Dijk et al. 2001). Vectors containing a set of several subtelomer ARS sequences seemed to be capable of homologous genome integration resulting in the integration of single or multiple tandem repeats into corresponding telomere genome regions (Sohn et al. 1999; Kim et al. 2003). A set of vectors for the integration of heterologous sequences into the ribosomal DNA locus of H. polymorpha was constructed (Steinborn et al. 2005, 2006; Kunze et al. 2009).

Several strong constitutive and regulatory promoters have been proposed for heterologous expression (Gellissen et al. 1997; Kang et al. 2002; Song et al. 2003). *MOX* and *FMD* promoters are strong, inducible by methanol, and repressible by glucose. *TPS1* promoter is constitutive, however, its expression in regulated by temperature (Amuel et al. 2000). In combination with the high copy number of integrated cassettes, the abovementioned promoters could drive the production of large amounts of heterologous proteins. For example, *FMD* promoter-regulated expression of secreted phytase provided a yield of 13.5 g/L (Mayer et al. 1999). Rarely, *MAL1* promoter induced by maltose (Liiv et al. 2001; Alamäe et al. 2003); nitrate-inducible promoters *YNT1*, *YNI1*, and *YNR1* (Avila et al. 1998); *PHO1* promoter from acid phosphatase gene repressed by phosphate (Phongdara et al. 1998); or *FLD* promoter induced by methanol (Baerends et al. 2002) are used. The list of constitutive promoters used includes *ACT* (Kang et al. 2001), *GAP* (Heo et al. 2003; Voronovsky et al. 2005; Dmytruk et al. 2008a,b), *PMA1* (Cox et al. 2000), and heterologous strong constitutive promoter *TEF1* from *A. adeninivorans* (Steinborn et al. 2006).

For the secretion of heterologous protein and its delivery to certain cell compartments, different signal sequences were successfully exploited. To provide efficient secretion, several leader sequences have been used, such as that of *PHO1* gene (Phongdara et al. 1998), *Schwanniomyces occidentalis GAM1* (Brake et al. 1984; van Dijk et al. 2000). However, the most popular one is the MFα1 pre-pro-leader sequence of *S. cerevisiae* (Brake et al. 1984; Gellissen 2000; Krasovska et al. 2007). To deliver proteins to peroxisomes, *PTS1* and *PTS2* targeting signals were used (van Dijk et al. 2000). Glycosyl phosphatidyl inositol motifs were used for the delivery of heterologous proteins, such as glucose oxidase, on the cell surface (Kim et al. 2002). Expression vectors of *H. polymorpha* contain prokaryotic and eukaryotic DNA fragments (Gellissen et al. 1997). Native circular or linearized vectors are used for transformation. The following loci were successfully used for homologous integration: *MOX/TRP1* (Agaphonov et al. 1995), *URA3* (González et al. 1999), *LEU2* (Agaphonov et al. 1999), *GAP* promoter region (Heo et al. 2003), ARS sequence (Sohn et al. 1996; Agaphonov et al. 1999), and rDNA cluster (Klabunde et al. 2002, 2003). The known vector system CoMed, with broad host spectrum, contains individual modules consisting of expression cassettes with

efficient promoters, selective markers, and rDNA sequences for integration (Steinborn et al. 2006).

# 4.4.5 *Examples of processes and products based on the recombinant strains of* H. polymorpha

#### 4.4.5.1 Production of heterologous protein

*H. polymorpha* is used for the industrial production of hepatitis B surface antigen (HBsAg), human interferon IFNa-2a, insulin, hexose oxidase, and phytase (Böer et al. 2007). Design and optimization of the fermentation procedure depend significantly on the characteristics of the host cell, targeted localization of the aimed product and, most importantly, on the promoter used. Most often, medium with simple synthetic components including trace elements and suitable nitrogen source, which provide efficient expression of the gene of interest and high biomass yield, are used. Total fermentation time varies from 60 to 150 h. Based on the characteristics of the most often used promoters (FMD and MOX), glycerol, methanol, glucose, and their combinations are normally used as carbon sources. For the expression of secretory heterologous proteins in *H. polymorpha*, such as hirudin, glycerol was used as a carbon source (Avgerinos et al. 2001; Weydemann et al. 1995). The recombinant strain contained 40 copies of an expression cassette containing  $MF\alpha 1$  pre-pro secretory sequence and hirudin structural gene expressed under control of MOX promoter. Hirudin production was activated by a decrease in the initial glycerol concentration and by maintaining it at an optimal level by pO<sub>2</sub>-controlled glycerol feeding. Fermentation started by the addition of 3% glycerol and after its consumption, pO<sub>2</sub>-controlled glycerol feeding started to maintain its concentration in the range of 0.05% to 0.3%, thus leading to derepression of the MOX promoter.

For the production of HBsAg, a nutrition regime based on the changes between two carbon sources was used (Brocke et al. 2005). The producing strain possessed a high copy number of expression cassettes containing the coding sequence for small surface antigen (S-antigen) under control of methanol-regulating promoters. Fermentation was conducted in 50 L fermenters. Cells with high antigen concentrations were obtained using a twofermenter cascade, with a first seeding fermenter of 5 L and a production fermenter of 50 L. The first stage was similar to that used for hirudin production. Cultivation started in glycerol medium using a fed-batch regime. Then, methanol was added leading to an HBsAg accumulation to up to several grams per liter. It was suggested that methanol is necessary during the production phase, in particular, for lipoprotein synthesis and membrane proliferation. To isolate HBSAg, cells were disrupted and inclusion bodies were purified using a multistep procedure involving adsorption, ion-exchange chromatography, ultrafiltration, and ultracentrifugation (Brocke et al. 2005; Schaefer et al. 2007). The assumption that methanol is specifically necessary for HBsAg synthesis (due to the activation of lipoprotein synthesis and membrane proliferation) contradicts recent observations that AOX promoter-controlled HBsAg synthesis in gcr1 mutant is activated by glucose in the absence of methanol (Krasovska et al. 2013).

*H. polymorpha* seems to be an especially productive organism for phytase synthesis (Mayer et al. 1999; Papendieck et al. 2005). A production yield of 13.5 g/L of phytase was obtained with a strain expressing the phytase gene under control of the *FMD* promoter cultivated in a medium with glucose starvation. The secreted product was purified by centrifugation, filtration, and ultrafiltration, which provided high yields (up to 92%; Mayer et al. 1999).

Known systems of heterologous expression in H. polymorpha based on methanol-inducible promoters are not free of serious drawbacks. Thus, the use of toxic and highly inflammable methanol for promoter induction is a real challenge. Methanol concentrations have to be maintained at a strictly defined level because high concentrations of this alcohol are toxic for the cell (Gellissen et al. 2005). In addition, methanol utilization requires intense aeration rates due to the high oxygen requirement of alcohol oxidase (Jenzelewski 2002). It is known that the MOX promoter of *H. polymorpha* is not strictly dependent on methanol because of its sufficient derepression when glycerol or glucose are limited in the medium (Hellwig et al. 2005). However, for maximal promoter activity, methanol is necessary. The mutants H. polymorpha gcr1 and their derivatives EAO, producing alcohol oxidase constitutively in glucose medium without methanol, were isolated. The alcohol oxidase activity in EAO mutants was much higher than that in methanol medium (Gonchar et al. 1998; Moroz et al. 2000; Stasyk et al. 1997a, 2004). The nature of mutations leading to the EAO phenotype remains unidentified. Expression of heterologous genes of A. niger glucose oxidase GOD, S-antigen HBsAg and secreted mini-proinsulin under control of the MOX promoter in the EAO strain of *H. polymorpha*, resulted in transformants producing recombinant proteins in the medium without methanol. Moreover, the highest promoter induction was achieved in the medium containing xylose (Krasovska et al. 2007). Expression of HBsAg in a strain with multicopy-integrating cassettes with this gene further increased the production level. Deletion of the PEX3 gene coding for peroxine involved in the early step of peroxisome formation substantially increased the production of HBsAg in glucose medium as compared with the parental EAO strain. Maximal production of HBsAg in the  $\Delta pex3$  strain was nearly in the range of 8% to 9% of total cell protein (Krasovska et al. 2013). Optimization and scaling-up for heterologous protein synthesis in the mentioned system has not yet been carried out.

In the case of the production of toxic proteins or small peptides, which undergo rapid cytosolic degradation, the products could be targeted to peroxisomes using peroxisometargeting signals PTS1 or PTS2. In such a way, transformants of *H. polymorpha* accumulating in peroxisomes' small functional peptides as insulin-like human growth factor II (IGF-II) and *Xenopus laevis* magainin II have been obtained (Faber et al. 1996).

Some efforts have been made to improve heterologous protein secretion. It was shown that an increase of *HpCNE1* gene expression, coding for membrane chaperone protein synthesis, led to an improvement in the secretion of several tested glycoproteins and of one nonglycosylated protein by twofold to fourfold (Klabunde et al. 2007; Qian et al. 2009). Improvements in the secretion of the urokinase type of plasminogen activator were achieved with mutations in *PMT1* and *PMR1* genes coding for protein *O*-mannosyl transferase and Golgi ion pump, respectively (Agaphonov et al. 2005, 2007). Secretion of this protein was also activated by mutations in the *RET1* gene coding for  $\alpha$ -COP, the sub-unit COPI of the protein complex involved in protein translocation in the Golgi apparatus (Chechenova et al. 2004), as well as genetic defects in the vacuole protein sorting (vps) pathway due to mutations in the genes *PEP3*, *VPS8*, *VPS10*, *VPS17*, and *VPS35* (Agaphonov et al. 2005).

Sometimes, genetic manipulations led to the secretion of normally intracellular proteins such as alcohol oxidase, which is normally a peroxisomal enzyme. However, expression of the alcohol oxidase gene AOX under control of its own promoter and fused with  $MF\alpha 1$  pre-pro-secretory resulted in the accumulation of monomeric protein in the endoplasmic reticulum. However, substitution of the  $MF\alpha 1$  signaling sequence for *S. cerevisiae ISS* invertase signaling sequence provided synthesis of the intact enzymatically active protein secreted into the cultivation medium (van Der Heide et al. 2007). High levels of protein secretion were obtained by increasing the copy number of expression cassettes. Thus, the heterologous *ISS* signal appeared to be stronger than the native PTS1 signal for peroxisome targeting.

Normally, in *H. polymorpha* and similarly to *S. cerevisiae* and other yeast species, hyperglycosylation of glycoproteins occurs as compared with the glycosylation level in animal and human cells. *H. polymorpha* strains lacking hyperglycosylation similar to those for *P. pastoris* (Hamilton et al. 2007; Li et al. 2006) have been isolated. For this, the *HpALG3* gene, coding for dolychol phosphomannose-dependent  $\alpha$ -1,3-mannosyl transferase was deleted in the background of the *Hpoch1* mutant with a defect in yeast-specific synthesis of the outer mannose chain, which additionally expresses the heterologous gene of  $\alpha$ -1,2mannosidase from *Aspergillus saitoi* (Oh et al. 2008). Such recombinant strains produced glycoproteins containing *N*-glycan of trimannosyl type (Man<sub>3</sub>GlcNAc<sub>2</sub>), which is the typical core for different human glycans.

#### 4.4.5.2 *Production of low-molecular weight biologically active compounds in* H. polymorpha

In addition to heterologous proteins, *H. polymorpha* seemed to be a promising producer for some low-molecular weight compounds, both native and heterologous. These compounds include antibiotic penicillin, tripeptide glutathione, and ethanol. The corresponding producers were constructed by a combination of methods of classic selection and metabolic engineering.

4.4.5.2.1 *Penicillin*. The market for β-lactam antibiotics, penicillins, and cephalosporin consists of approximately 40% of the total antibiotic market, with an overall market of nearly 55 billion US dollars (Demain 2007; Kresse et al. 2007). Several years ago, the market for penicillin was at 8.2 billion US dollars (Demain 2007). Industrial production of penicillin is based on specially constructed strains of mycelial fungus Penicillium chrysogenum (Kresse et al. 2007). Use of a multicellular mycelial organism generates some difficulties for largescale cultivation; therefore, introduction and efficient expression of penicillin biosynthesis genes into cells of an evolutionarily related organism, the unicellular yeast H. polymorpha, has undoubted scientific and applied interest. Penicillin biosynthesis involves four enzymatic steps, the first two being located in the cytosol whereas the last two occur in peroxisomes (Evers et al. 2004). First, the P. chrysogenum enzyme successfully expressed in H. polymorpha was the peroxisomal isopenicillin N-acyl transferase catalyzing the penultimate reaction of penicillin biosynthesis (Lutz et al. 2005). The enzyme was localized in yeast peroxisomes. Subsequently, the last enzyme of the pathway, phenylacetyl-CoA ligase (which is also located in yeast peroxisomes) was expressed (Gidijala et al. 2007). Finally, strains of H. polymorpha producing heterologous first and second enzymes of penicillin biosynthesis were constructed (Gidijala et al. 2008, 2009). A strain of H. polymorpha that expressed all four enzymes of penicillin biosynthesis was constructed. The amount of penicillin secreted by this strain was similar to that of the wild-type P. chrysogenum NRRL1951 strain (nearly 1 mg/L; Gidijala et al. 2009). This work opens up opportunities to increase penicillin production levels in *H. polymorpha* up to industrial levels as well as to construct on the base of *H. polymorpha* from the producers of other  $\beta$ -lactam antibiotics and pharmacologically important peptides. Thus, one may assume that, in the future, H. polymorpha will become an industrial producer of antibiotics and immunosuppressive and cytostatic peptides.

4.4.5.2.2 *Glutathione*. Glutathione (γ-glutamylo-L-cysteinyloglycine) is a natural tripeptide and simultaneously is the main nonprotein thiol in all eukaryotic and many

prokaryotic organisms (Penninckx 2000). Glutathione is the main redox buffer because the ease of interconversion between oxidized and reduced forms of glutathione maintains cell redox status (Ostergaard et al. 2004). In addition, glutathione plays a major role in cell response to stress induced by substrate starvation, heavy metals, xenobiotics, and free radicals as well as participating in sulfur storage, regulation of gene expression, and cell signaling (Bachhawat et al. 2009; Pócsi et al. 2004; Sies 1999). More than 90% of glutathione is located in cells in reduced form (Pócsi et al. 2004). Both forms of glutathione, reduced thiol and oxidized disulfide, are widely used in medicine, the cosmetic industry, and as a food additive (Li et al. 2004). Being an active ingredient of food, medicines, and cosmetics, glutathione relieves the effects of harmful oxidizers, removes toxic products, and enhances the action of cosmetics regenerating skin cells. The oxidized form of glutathione is used as a cryoprotector and immunomodulator (Chatterjee et al. 2001; Kozhemyakin et al. 2008). It was found that an increase in glutathione cell content biosynthesis in *H. polymorpha* due to overexpression of GSH2 and MET4 genes coding for the first enzyme of glutathione biosynthesis and the central regulator of sulfur metabolism, respectively, strongly activates glucose (but not xylose) alcoholic fermentation (Grabek-Lejko et al. 2011).

Methylotrophic yeasts are considered to be a rich source of glutathione because the metabolism of the products of alcohol oxidase reaction, formaldehyde, and hydrogen peroxide depends on glutathione. Thus, oxidation of glutathione in cytosol is catalyzed by NAD-dependent and glutathione-dependent formaldehyde dehydrogenase (Sahm 1977), whereas detoxification of hydrogen peroxide depends on the activity of Pmp20p, glutathione peroxidase, localized in the vacuolar membrane (Horiguchi et al. 2001; Yano et al. 2009). It is known that methanol induces an increase in intracellular glutathione concentration (Ubiivovk et al. 1986), which is especially noticeable in *H. polymorpha* mutants defective in formaldehyde dehydrogenase (Sibirny et al. 1990). However, the regulatory mechanisms involved in glutathione synthesis and degradation in methylotrophic yeasts remained unknown for a long time. Glutathione auxotrophic mutants of H. polymorpha, divided into two complementation groups designated as gsh1 and gsh2, have been isolated (Ubiivovk et al. 1999). Using a gsh2-1 mutant, the existence of glutathione-dependent and independent systems of xenobiotic detoxification were demonstrated (Ubiyvovk et al. 2003). The gene GSH2 was cloned by functional complementation of gsh2-1 mutant. It seems that this latter encodes  $\gamma$ -glutamylo-L-cysteine synthetase, in other words, this is a homologue of the GSH1 gene of S. cerevisiae (Ubiyvovk et al. 2002). The H. polymorpha GSH1 gene was also cloned by functional complementation and was found to be the homologue of the gene S. cerevisiae MET1, coding for S-adenosyl-L-methionine uroporphyrinogen III methyltransferase, which is involved in siroheme biosynthesis (Ubiyvovk et al. 2011b). The gene *H. polymorpha GGT1*, which seemed to be the homologue of the *S. cerevisiae CIS2* gene, coding for  $\gamma$ -glutamyl transpeptidase was cloned and a  $\Delta ggt1$  knockout strain defective in glutathione degradation was isolated (Ubiyvovk et al. 2006). It was found that  $\Delta ggt1$  and  $\Delta met1$  ( $\Delta gsh1$ ) mutants are unable, in contrast with the *H. polymorpha* wild-type strain, to accumulate  $Cd^{2+}$  ions inside cells, whereas the cells of the  $\Delta gsh^2$  mutant accumulated increased amounts of this ion (Blazhenko et al. 2006). It was found that H. polymorpha synthesizes only one type of cadmium-binding chelator, namely, glutathione, and does not synthesize compounds of the phytochelatine type. It is suggested that the gene GSH1/ MET1 participates in maturation, and gene GSH2 is involved in metabolism of the intracellular complex of Cd<sup>2+</sup> ions with glutathione.

To construct *H. polymorpha* mutants overproducing glutathione, the strains overexpressing either *GSH2*, coding for the first enzyme of glutathione synthesis, or *MET4*, which is the central positive regulator of glutathione biosynthesis, were constructed (Kang et al. 2002; Thomas et al. 1992; Ubiyvovk et al. 2011a). The production of intracellular and extracellular glutathione was studied in the constructed strains depending on cultivation parameters (dissolved oxygen pressure, pH, and mixing rate) and on the type of carbon source (methanol or glucose). Under optimal conditions during cultivation in fed-batch mode with glucose medium, recombinant strains accumulated 900 to 2300 mg of total intracellular glutathione per liter of culture broth. Strain cultivation in the same fed-batch mode in methanol medium resulted in the accumulation of 250 mg of extracellular glutathione per liter of culture broth (Ubiyvovk et al. 2011a). The obtained titers exceed maximal data for glutathione synthesis in *S. cerevisiae*, even after the addition of exogenous amino acids to the medium (Wang et al. 2007). Thus, *H. polymorpha* seems to be a promising organism for the industrial production of this important tripeptide.

4.4.5.2.3 H. polymorpha as a promising producer of fuel ethanol from lignocellulose. H. polymorpha is a promising organism for high-temperature alcoholic fermentation. All eight strains of *H. polymorpha* that were tested were capable of fermenting glucose, xylose, mannose, maltose, and cellobiose, whereas galactose and L-arabinose practically did not support the growth of this yeast (Dmytruk et al. 2013; Ryabova et al. 2003). The optimal temperature for fermentation was 37°C to 40°C, although fermentation was quite intensive even at 45°C to 48°C, which is the absolute temperature recorded for eukaryotic alcoholic fermentation. At 50°C, fermentation is strongly suppressed. However, an increase of intracellular trehalose content due to knockout of the acid trehalase gene ATH1 or overexpression of the heat shock proteins Hsp16 and Hsp104 allowed normal xylose fermentation at 50°C (Ishchuk et al. 2009). Fermentation was most active under limited aeration and starvation for flavins, which are necessary for cell respiration. H. polymorpha was more tolerant of ethanol compared with the most effective natural xylose-fermenting yeast, Pichia stipitis; however, it was more susceptible relative to S. cerevisiae (Ryabova et al. 2003). Ethanol tolerance could be further increased due to overexpression of the heterologous gene MPR1 (Ishchuk et al. 2010) or ETT1's own gene (Ishchuk et al. in preparation; Abbas et al. 2013). Using an auxonographic selection method, H. polymorpha mutants accumulating increased amounts of ethanol in xylose medium were isolated (Grabek-Lejko et al. 2006). Recombinant strains of this organism, which express amylolytic and xylanolytic enzymes and directly ferment starch and xylan to ethanol, have been isolated (Voronovsky et al. 2009).

H. polymorpha could be a promising organism for simultaneous saccharification and fermentation (SSF) processes. In the SSF process, pretreated lignocellulose is hydrolyzed by cellulases and hemicellulases and, to avoid product inhibition of hydrolysis by the liberated monosaccharides, the vessels also contain microorganisms that efficiently ferment produced sugars to ethanol (Olofsson et al. 2008). Because cellulases and hemicellulases express maximal activity under temperatures in the range of 50°C to 60°C, the microorganisms used in the SSF process, in addition to effective alcoholic fermentation of major sugars of lignocellulosic hydrolysates, have to survive and actively ferment at high temperatures (Merino et al. 2007). H. polymorpha is perfectly suited to this requirement as it belongs to the very few thermotolerant yeast species capable of xylose fermentation and could be the organism of choice. However, it is not free from several drawbacks. Most importantly, ethanol yield and productivity from xylose in wild-type strains of H. polymorpha are very low. However, these features could be improved through classic selection and metabolic engineering. Three approaches have been used previously for the construction of *H. polymorpha* strains with improved characteristics of ethanol production from xylose. In one line of investigation, XYL1's own gene coding for xylose reductase (XR), and two paralogs of xylitol dehydrogenase (XDH), XYL2A and XYL2B, were deleted in the collection strain CBS4732, and bacterial xylA gene from E. coli or Streptomyces coelicolor were expressed (Voronovsky et al. 2005). The corresponding transformants possessed the activity of xylose isomerase and grew on xylose, but the amount of accumulated ethanol was very low (both transformants and wild-type cells accumulated a maximum of 0.15 g/L of ethanol). Overexpression of *E. coli xylA* together with own *XYL3* coding for xylulokinase (XK) increased ethanol production. However, maximal ethanol accumulation did not exceed 0.6 g/L at 48°C (Dmytruk et al. 2008a,b). In another work, XR was engineered by site-specific mutagenesis to reduce affinity toward NADPH, according to the approach developed for Candida tenuis (Petschacher et al. 2005). Consequently, genes coding for modified XR (XYL1m), native XDH (XYL2), and XK (XYL3) were overexpressed in strain CBS4732 and resulted in twofold higher ethanol accumulation in corresponding transformants reaching 1.3 g/L of ethanol (Dmytruk et al. 2008a,b). In another line of investigations, the wild-type strain NCYC495 (the currently sequenced strain) was used initially because we showed that it seemed to be a more efficient xylose fermenter relative to strain CBS4732. The mutant 2EthOH<sup>-</sup>, unable to use ethanol as a sole carbon source, was isolated from strain NCYC495 and characterized by threefold elevated ethanol accumulation. Subsequently, PDC1 gene coding for pyruvate decarboxylase (PDC) was cloned and overexpressed in strain 2EthOH<sup>-</sup>. The best-selected transformants accumulated 2.5 g/L of ethanol at 48°C (Ishchuk et al. 2008). However, the ethanol production achieved is still very low and has to be substantially increased before the strain meets the requirements for an SSF process.

Recently, more efficient *H. polymorpha* high-temperature ethanol producers from xylose have been isolated (Kurylenko et al. submitted for publication). For this, a combination of the methods of metabolic engineering and classic selection were applied. The mutant 2EtOH<sup>-</sup> (Ishchuk et al. 2008) was used for overexpression of the genes *XYL1m*, *XYL2*, *XYL3*, and *PDC1*. The best-selected transformant was used for isolation of the mutants that were resistant to the anticancer drug 3-bromopyruvate, which is known to inhibit glycolysis (Cardaci et al. 2012; Ganapathy-Kanniappan et al. 2010; Shoshan 2012). The best obtained strain showed a 15-fold enhancement in ethanol synthesis from xylose as compared with the wild-type strain, accumulating up to 10.0 g/L of ethanol at 45°C, which, after correction for ethanol evaporation, corresponds to the calculated concentration of 20 g/L.

Thus, accumulation of ethanol from xylose increased by more than 15 times as compared with the *H. polymorpha* wild-type strain. The maximal observed level of ethanol produced from xylose by the best-isolated strains (near 10 g/L at 45°C, which approximately corresponds to the 20 g/L value calculated after correction for evaporation, resulting in ethanol yields of 0.18 and 0.35 g/g xylose, respectively) makes *H. polymorpha* quite close to known promising organisms for use in the SSF process. Still, these results are lower than ethanol production in mesophilic xylose-fermenting organisms such as *P. stipitis* (0.35–0.44 g/g xylose) and *Spathaspora passalidarum* (0.42 g/g xylose; Jeffries et al. 2007; Long et al. 2012) but higher than those in the best engineered strain of thermotolerant yeast *Kluyveromyces marxianus* (0.31 g/g xylose under anaerobic conditions at 45°C; Wang et al. 2013). To be industrially feasible, ethanol yield in *H. polymorpha* has to be further increased to be close to the theoretical maximum. It is desirable to manipulate with new targets for metabolic engineering (e.g., xylose transport, pentose phosphate pathway), which could be successful for the further increase of ethanol yield from xylose during high-temperature alcoholic fermentation of this promising organism.

#### 4.5 Conclusion

In this chapter, we provide information and data that clearly highlights that the nonconventional yeasts *P. pastoris*, *Y. lipolytica*, and *H. polymorpha* present many advantages over other model organisms such as *E. coli* or even mammalian cells for the production of heterologous proteins or other valuable compounds. Indeed, these yeasts are able to grow at high cell densities in low-cost culture media, to express heterologous genes at high levels, to synthesize and secrete the corresponding proteins at high yield, and to perform humanlike posttranslational modifications.

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### chapter five

## Induced anhydrobiosis Powerful method for preservation of industrial microorganisms

#### Armando Hernández García

#### Contents

5.1	Introduction	113	
5.2	Anhydrobiosis in microorganisms		
5.3	Mechanisms of desiccation damage		
	Mechanisms of desiccation tolerance		
5.5	Induced anhydrobiosis as a powerful method for preservation of		
	microorganisms	119	
5.6	Technologies used in the drying of microorganisms		
5.7	Stability of desiccated microorganisms		
5.8	Case study: Anhydrobiotic cells of a biocontrol agent		
5.9	Conclusion		
Refe	References		

#### 5.1 Introduction

The preservation of microorganisms has been a challenge since the beginning of mankind. Some strains of beneficial microorganisms are involved in the production of dairy, bakery, spirits, alcohol, vaccines, antibiotics, enzymes, silage, vinegar, and others (Uzunova-Doneva and Donev 2005). There are several methods for the preservation of industrial microorganisms, for example, subcultivation, use of mineral oils, water–salt solutions, cryogenic conservation, and drying (Uzunova-Doneva and Donev 2005). Among these methods, the preservation of microorganisms by desiccation has been the preferred method for long-term storage (Morgan et al. 2006).

Nowadays, there are big culture collections that depend on drying methods for preserving their microorganisms. In addition, there are many industrial applications of microorganisms preserved in aliquots. Microbiological quality control is necessary in the pharmaceutical, food, and beverage industries. It is also evident in clinical microbiology, which is performed by using samples of microbial strains acting as positive controls. These strains are preserved through drying, which increases their stability for transportation at room temperature. Moreover, emerging fields such as the probiotic and biocontrol industries require the conservation of certain microorganisms (Garcia 2011; Morgan et al. 2006).

Anhydrobiosis is the state at which an organism stops its vital functions temporarily due to partial or total desiccation. This state is characterized by the extreme reduction of mensurable metabolism (Rebecchi et al. 2007). A quantitative definition of the total desiccation states that the water content in an anhydrobiote is less than 0.1 g of free water per gram of dry cell weight (Alpert 2005).

Concerning the industrial importance of anhydrobiosis, the worldwide market in the stabilization of cells and cell products is approximately \$500 billion (Garcia 2011; Potts et al. 2005). Moreover, induced anhydrobiosis is applied in the preservation of reference strains used in the quality control of pharmaceutical and food industries, the conservation of reference strains from culture collections, and the generation of biotechnological products in the emerging fields of probiotics and biocontrol (Morgan et al. 2006).

In the fields of probiotics and biocontrol, an intense amount of work has been performed during the last decade using the technologies of freeze-drying and spray-drying (Ananta et al. 2005; Arunsiri et al. 2003; Carvalho et al. 2004; Corcoran et al. 2006; Fravel 2005; Hernández et al. 2006, 2007; Kuang et al. 2010; Lauten et al. 2010; Nag and Das 2012; Strasser et al. 2009; Ziadi et al. 2005). These technologies show relevant results in the use of bacterial dehydration as a powerful tool for future applications in agriculture, food industry, and medicine.

According to recent reports, to achieve high survival rates, a combination of physiology and technology should be considered (Garcia 2011; Fu and Chen 2011). Thus, sample preparation and optimal drying conditions should be rigourosuly selected for long-term stabilization of desiccated cells (Garcia 2011; Julca et al. 2012).

Considering the technological approach, although freeze-drying is the most widely used technology, it is, however, the most expensive one (Fu and Chen 2011; Hernández et al. 2007). Thus, nowadays, spray-drying and fluidized bed drying are being evaluated by many reseachers worldwide as potential drying techniques to be used as alternatives for freeze-drying in the preservation of several microbial species (Fu and Chen 2011; Morgan et al. 2006).

With respect to the mechanisms of desiccation tolerance, most studies have been performed mainly on prokaryotes (Potts 1994). Although, up to date, little is known about such mechanisms. Research suggests that small changes of energy are involved in the decrease of viability, and that those cells sensitive to desiccation could lose an important fraction of free water (Potts et al. 2005).

Among the methodologies used to improve desiccation tolerance, we found: (1) selection—training through multiple exposure to stress; (2) accumulation of intracellular protective agents—through uptake or through genetic engineering; (3) use of extracellular protective agents—as additives or secretion through genetic engineering; and (4) manipulation of cell metabolism/physiology—before exposure to stress (Potts et al. 2005).

In this chapter, we focus our attention on the induced anhydrobiosis phenomenon, and the practical implications of the drying processes in the preservation of microorganisms.

#### 5.2 Anhydrobiosis in microorganisms

Desiccation tolerance is generally defined as the ability of an organism to reach desiccation up to a state of equilibrium with moderately or extremely dry air, and then the recovery of its normal functions after rehydration (Alpert 2005). According to França et al. (2007), the term *desiccation tolerance* is not the same as *drought tolerance*. Drought is related to the moderate dehydration of a microorganism, whereas desiccation is related to the extreme dehydration of cells when the hydration shell of molecules is gradually lost (França et al. 2007).

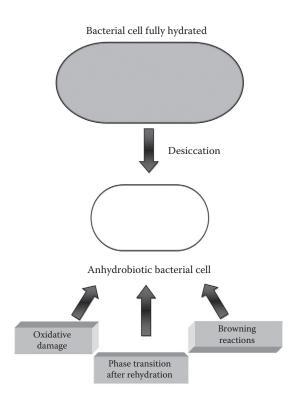
Most microorganisms with industrial applications are bacteria; as a consequence, the anhydrobiosis phenomenon has been studied mainly in prokaryotes (Garcia 2011; Potts et

al. 2005). The minimal water content or hydric potential for survival in these desiccationtolerant cells is extremely low. Thus far, a decrease of up to 2% (w/w) in the water content of bacterial cells has been reported (Potts 1994).

#### 5.3 Mechanisms of desiccation damage

Microbial cells, under desiccation, are subjected to several stress processes once they reach the anhydrobiosis state. One of the most important processes related to the desiccated state is oxidative stress. Little is known in relation to the damage caused by oxidative stress as a consequence of dehydration. Although oxygen is a vital compound for aerobic organisms, this substance could be harmful in excess (França et al. 2007). Figure 5.1 shows the mechanisms of desiccation damage. The chemical damage caused by free radicals is one of the main reasons for desiccation lesions. The water stress increases reactive oxygen species (ROS) formation, which produces lipid peroxidation, protein denaturation, and damage in nucleic acids with drastic consequences in the general metabolism (Hansen et al. 2006).

Aerobic organisms use oxygen as electron acceptors. Nevertheless, during respiration, this compound could be partially reduced by forming ROS, such as superoxide anions  $(O_2^-)$ , hydrogen peroxide  $(H_2O_2)$ , and hydroxyl radicals  $(OH\bullet)$ . The balance between the ROS production and the cellular defenses determines the degree of oxidative stress (França et al. 2007). According to Pereira et al. (2003), yeast cells increase the oxidation process by more than 10-fold when they are dehydrated.



*Figure 5.1* Main mechanisms of desiccation damage in a microbial cell. A bacterial cell is shown as the model.

Changes in the water content of the cells could produce disfunctions in specific enzymes, or promote the development of chemical reactions that normally would not occur in a fully hydrated cell. Thus, under normal metabolic conditions, free radicals would be trapped by the antioxidant defense systems. However, under water stress, these mechanisms could be inactive (França et al. 2007).

Once the cell membranes are dehydrated, they are more susceptible to ROS attack (Crowe et al. 1989). These radicals frequently cause an extensive peroxidation and deesterification of membrane lipids for dehydration levels of up to 50% (Senaratna et al. 1987). Furthermore, an increase in the packing density of the polar heads of the membrane phospholipids takes place, leading to the strengthening of the van der Waals interactions between the carbon chains (Crowe et al. 1992). Consequently, the phase transition temperature ( $T_m$ ) increases significantly, and lipids are in gel phase at room temperature. Then, when these lipids are rehydrated, they undergo a phase transition producing extensive leakage and cell death (Potts 1994).

Desiccation-tolerant cells have mechanisms for decreasing the  $T_{\rm m}$  value, and thus preventing phase transition. A decrease in the  $T_{\rm m}$  value could be reached by increasing the unsaturation degree of fatty acid membranes (Hoekstra et al. 2001). However, desiccated cells using this mechanism are more susceptible to lipid oxidation (França et al. 2007; Hoekstra et al. 2001). In addition, the polar heads of the phospholipids generate ROS through Fe<sup>2+</sup> auto-oxidation (Motta and Sechi 1992).

Protein dehydration induces significant conformational changes that were detected by Fourier transform infrared spectroscopy (Prestrelski et al. 1993). The drying, by itself, could produce protein denaturation, which causes a loss of biological activity after rehydration (França et al. 2007). On the other hand, protein oxidation catalyzed by metals is an additional consequence of cellular desiccation. Oxygen reduction directly produces  $H_2O_2$  or superoxide as an intermediary, which reacts with  $Fe^{2+}$  to yield  $Fe^{3+}$  and oxygen. The ion  $Fe^{2+}$  binds a metal-binding site in the protein, and the complex protein- $Fe^{2+}$  reacts with  $H_2O_2$  to generate ferryl ions, leading to reactions with protein side chains (França et al. 2007; Potts 1994).

After oxidative modification, proteins become more sensitive to proteolysis and may be inactivated or decrease their biological activity (Potts 1994). Some amino acid residues are converted to carbonyl derivatives and the carbonylated proteins accumulate in the desiccated cells (Fredrickson et al. 2008; Yin and Chen 2005). In fact, most recent experimental evidences indicate that protein oxidation seems to be mainly responsible for the desiccation damage in bacterial cells (Fredrickson et al. 2008). Protection of proteins against ROS damage is equally important (or more critical) for maintaining ancient bacteria in a viable state than DNA repair.

According to Potts et al. (2005), drastic changes occur in bacterial populations exposed to air desiccation. Among these changes, the most remarkable are related to increases in surface and saline precipitation, as well as changes in shape, color, and texture. Also, important changes occur at the cellular level, affecting the bacterial cell physically and chemically. Consequently, physiological changes will be observed. For instance, the thickness of the cell wall is decreased due to the packing of the peptidoglycan layers after dehydration (Hernández 2009). Simultaneously, the intracellular salt concentration is increased, producing an agglomeration of biopolymers. In addition, intracellular viscosity is increased causing low rates of diffusion, enzymatic activity is reduced, and finally, growth stops.

For the survival of microbial cells, DNA must be fully hydrated, which ensures its chemical stability (Potts 1994; Potts et al. 2005). Thus, changes in the DNA hydration patterns could affect several molecular processes, for example, replication, transcription, and protein synthesis, leading to the disruption of cellular equilibrium. After the desiccation

process, many harmful reactions occur; browning reactions could affect the stability of DNA and proteins. In fact, the Maillard reaction produces melanoidins through condensation between carbonyl groups of saccharides and the amino group of proteins and nucleic acids (Potts et al. 2005). This chemical change in DNA could generate abnormalities that inhibit the copying process performed by polymerases. Other undesired reactions such as Haber–Weiss and Fenton reactions produce OH•, this radical is the most reactive species among ROS, causing oxidation of pyrimidines, enhancing mutation rates, and leading to cell death (Kranner and Birtic 2005).

#### 5.4 Mechanisms of desiccation tolerance

Some radiation-tolerant bacterial species, for example, *Deinococcus radiodurans* and cyanobacteria such as *Nostoc commune* and *Chroococcidiopsis* spp. are considered anhydrobiotes (Mattimore and Battista 1996; Potts et al. 2005; Rebecchi et al. 2007). *N. commune* did not show significant damage in DNA molecules after 13 years of desiccation (Shirkey et al. 2003). Also, tolerant cyanobacteria in the desert of Negev can recover up to 50% of its normal activity in photosystem II in only 2 min after rehydration (Harel et al. 2004). Recently, Slade and Radman (2011) have illustrated the remarkable correlation between desiccation tolerance and resistance to ionizing radiation in *D. radiodurans*. Considering the data presented by these authors, the survival curves for both gamma radiation and desiccation show identical shapes for survival rates between 100% and 10%. This evidence suggests the hypothesis that desiccation tolerance and radiation resistance could be molecularly regulated by similar mechanisms.

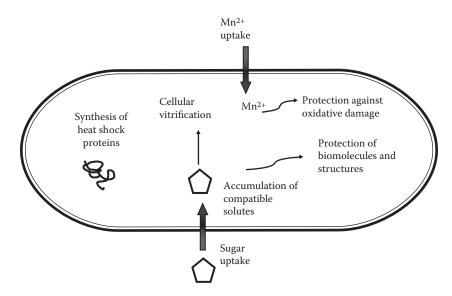
In general, Gram-positive bacteria are the most desiccation-tolerant bacteria (Potts et al. 2005). Many Gram-positive species accumulate Mn<sup>2+</sup>, have a high Mn<sup>2+</sup>/Fe<sup>2+</sup> ratio, and are resistant to radiation (Daly et al. 2004). On the other hand, cyanobacteria, despite having a similar cell wall to Gram-negative bacteria, show phylogenetic affinity to Gram-positive species, being markedly tolerant of both desiccation and gamma radiation (Billi et al. 2000; Potts 1994). Recently, the intracellular accumulation Mn<sup>2+</sup> has been proposed as the most probable mechanism involved in the protection of proteins from oxidative damage after desiccation (Fredrickson et al. 2008). These researchers found a strong correlation between the intracellular levels of Mn and the desiccation tolerance in several bacterial genera.

In Gram-negative bacteria such as *Escherichia coli* and *Pseudomonas putida*, the induction of desiccation tolerance was achieved using trehalose and hydroxyectoine (de Castro et al. 2000; Manzanera et al. 2002, 2004; Tunnaclife et al. 2001). In addition, high radioactivity tolerance has lead to the discovery of new bacterial species resistant to desiccation in radioactive work areas (Venkateswaran et al. 2003).

In relation to the mechanisms of desiccation tolerance (Figure 5.2), during the drying process, loss of water is balanced by reversible interactions with other molecules (Rebecchi et al. 2007). Thus, biomolecules and structures are protected, and they retain their native conformation after rehydration (Wolkers et al. 2002).

One of the survival strategies against desiccation is the accumulation of nonreducing disaccharides such as sucrose and trehalose (Clegg 2001; Rebecchi et al. 2007). These sugars have a double function in the desiccation-tolerant organisms. First, the sugar molecules protect the cells and biomolecules by replacing the water normally forming hydrogen bonds. Second, the saccharides are involved in the formation of a vitreous cytoplasmatic matrix (Wolkers et al. 2002).

Although the importance of nonreducing disaccharides has been emphasized for both unicellular and pluricellular organisms, the molecular mechanisms of anhydrobiosis are not clearly understood (Rebecchi et al. 2007). It seems that trehalose does not induce anhydrobyosis



*Figure 5.2* Main mechanisms of desiccation tolerance in a microbial cell. A bacterial cell is presented as the model.

by itself. In fact, in some anhydrobiotes, trehalose is not relevant to desiccation and they need other adaptations (Clegg 2001; Kshamata et al. 2003; Tunnacliffe and Lapinski 2003).

In this context, the attention has been focused on the definition of other molecular adaptations required for anhydrobiosis. Several families of stress proteins seem to be the key for understanding the anhydrobiotic mechanisms. During drying, the response to stress is associated with a rapid synthesis of stress proteins (Potts et al. 2005; Rebecchi et al. 2007). Proteins commonly known as "heat shock proteins" (HSP), which are regulated through thermal stress, and seem to have some function in desiccation tolerance (França et al. 2007; Goyal et al. 2005).

The HSP genes are highly conserved in most species in which they have been studied (Feder and Hofmann 1999). As in any biochemical system, these proteins act like molecular chaperones, even in nonstressed cells. They have important functions such as protein biosynthesis, folding, assembly, intracellular localization, secretion, and degradation of other proteins (Feder and Hofmann 1999).

The intracellular proteins are able to compensate for the loss of water through the formation of hydrogen bonds to other molecules. It is very probable that proteins and saccharides interact by hydrogen bonds in the cytoplasm of anhydrobiotes. Thus, stress proteins and sugars could play an important role in the molecular organization of the desiccated cell once they are present in the same cellular compartment (Wolkers et al. 2002).

In bacteria, the regulation of the genes encoding HSP in response to desiccation has been observed, for example, in the cyanobacterium *Anabaena* sp. 7120 (Katoh et al. 2004). In addition, the overexpression of Hsp20 has been detected in *D. radiodurans* (Tanaka et al. 2004).

On the other hand, the so-called late embryogenesis abundant (LEA) proteins play an important role in anhydrobiotes, mainly in plants (Hoekstra et al. 2001). LEA proteins are usually classified into five groups on the basis of amino acid sequence and conserved motif. Recently, LEA proteins have been found in nonplant organisms (Rebecchi et al. 2007). In addition, several genes encoding LEA-like proteins have been identified from the genome analysis of *D. radiodurans*. In fact, the inactivation of the gene encoding for the group-3 LEA

reduces the desiccation tolerance of this bacterium (Battista et al. 2001). The group-3 LEA proteins are characterized by 11 amino acid tandem repeats (TAQAAKEKAGE; Ried and Walker-Simmons 1993). More recent studies have shown the high correlation between the expression of proteins with hydrophilic low complexity (LC) regions and desiccation tolerance in several bacterial species, including *D. radiodurans* (Krisco et al. 2010). Presumably, these proteins provide resistance to water stress conditions by avoiding aggregation. However, the exact mechanism remains unclear.

Any damage caused by air-drying is manifested during cell rehydration (Potts 1994). Apparently, there are similarities between the kinds of damage produced during desiccation and during the subsequent rehydration stage. Certainly, during both desiccation and rehydration, a dynamic change in the proteins and nucleic acid contents takes place (Shirkey et al. 2000). Therefore, a question arises in relation to the probable parallelism between the response to some types of stress and the poststress recuperation (Potts et al. 2005).

In this sense, oxidative damage is considered as an important consequence of desiccation stress. For example, Aldsworth (1999) proposed that bacterial cells perceive any stress as an oxidative stress. The decoupling between growth and metabolism generates free radicals, which are lethal to cells (Potts et al. 2005). Under such conditions, the role of superoxide dismutase (SOD) in response to stress is critical, and it could explain why some SODs are excreted by cells (Cannio et al. 2001).

A protein complex Fe-SOD (SodF) is the third more abundant in the desiccated cells of *N. commune*. In this cyanobacterium, SodF could protect cells from the free radicals generated when the extracellular polysaccharide is exposed to UV radiation (Shirkey et al. 2000).

The aconitases are involved in the rapid initial response to oxidative stress in *E. coli* and *Bacillus subtilis* by acting as mRNA binding proteins (Alén and Sonenshein 1999). In this mechanism, the cofactor sensor is the cluster iron-sulfur (redox sensor), which constitutes a positive posttranscriptional self-regulatory switch (Potts et al. 2005).

# 5.5 Induced anhydrobiosis as a powerful method for preservation of microorganisms

In general, the studies related to desiccation tolerance indicate that the stress processes induce several survival strategies offering protection to microorganisms subjected to desiccation. The production of stress proteins during bacterial growth could protect the cells, preparing them from further desiccation processes (Morgan et al. 2006). Thus, the design of experiments to evaluate drying resistance in bacterial strains is important for the industry. Moreover, the genomics of the microorganism subjected to desiccation should be studied to characterize the genes associated with the drying tolerance, which allows the development of strategies for the manipulation of external conditions to maximize the expression of such genes (Potts et al. 2005).

Another approach to induce anhydrobiosis in microorganisms is related to the use of protectants. The protective agents could be added during microorganism growth, or just before the drying process. The type of protectant depends mainly on the microorganism; nevertheless, some protectants seem to be universal (Morgan et al. 2006). These include skimmed milk, serum, trehalose, glycerol, betaine, adonytol, sucrose, glucose, lactose, and polymers such as dextrane and polyethylene glycol (Hubalek 2003).

According to Morgan et al. (2006), the protectant additives could be classified into two groups: (1) those forming amorphous glasses, and (2) the crystallized eutectic salts. The first group includes substances such as carbohydrates, proteins, and polymers. The term "glass"

is related to a supersaturated liquid that is thermodynamically unstable and highly viscous (Garcia 2011). In this sense, the glassy additives are the best protectant agents during freezedrying processes. The glassy state formation induces enough viscosity inside and outside the cells to reduce the molecular mobility to a minimum value (Morgan et al. 2006).

Trehalose and sucrose have been reported as the best membrane stabilizers and proteins in drying several microorganisms through a mechanism of water replacement (Rudolph and Crowe 1985). Both disaccharides preserve the structure and function of the proteins during drying, through stabilization by hydrogen bonds, keeping the tertiary structure in the absence of water (Leslie et al. 1995).

Several studies suggest that trehalose provides higher survival rates than sucrose (Crowe et al. 1998; Gómez Zavaglia et al. 2003; Leslie et al. 1995; Streeter 2003). The scientific explanation for this evidence is focused on the high glass transition temperature ( $T_g$ ) of the trehalose. The glass transition temperature is the temperature at which glass is transformed to the corresponding liquid phase. Hence, the higher the  $T_g$  value, the higher the stability of the dried matrix when the temperature is increased (Morgan et al. 2006). Particularly, anhydrous trehalose has  $T_g = 110^{\circ}$ C, keeping a high viscosity at high temperatures. Otherwise, sucrose has  $T_g = 65^{\circ}$ C, reducing its potential use with respect to trehalose (Crowe et al. 1998). However, sucrose is cheaper than trehalose.

Previous studies have established the correlation between the high  $T_g$  value of trehalose, and the increase in the protection to desiccation by comparison to other protecting carbohydrates with lower  $T_g$  values (Sun and Davidson 1998). In addition, trehalose has the ability to form a crystalline dihydrate. Consequently, in the formation of the dehydrated crystal, a small amount of water is absorbed and the matrix stays in a glassy state without decreasing the  $T_g$  of the system (Crowe et al. 1998).

Although it is believed that sugars are the major components in glass formation, it has been shown that polypeptides could significantly alter the vitreous properties of sugars. Buitink et al. (2000) found that proteins were more stable above their  $T_g$  values than sugars. This fact suggests that proteins contribute higher extension to glass formation in comparison with sugars. Moreover, it could explain why milk serum and skimmed milk are such good protectors against desiccation (Hubalek 2003).

Therefore, it seems that the most efficient protector against desiccation is a mixture of proteins and sugars. Desmond et al. (2002) found that the survival of *Lactobacillus paracasei* significantly increased up to 1000-fold when acacia gum (10% w/w) was added to reconstituted skimmed milk (10% w/v).

On the other hand, rehydration is the most important stage once the microbial cells are subjected to the desiccation process. Rehydration is critical because, according to the rehydration protocol, the survival of the cells will be more or less affected. Hence, the rehydration medium plays a key role. For example, it has been observed that a protective solution used before the drying is a good rehydration solution (Garcia 2011; Julca et al. 2012; Potts et al. 2005). In addition, a complex medium generally has higher osmotic pressure, which diminishes the negative effect of osmotic shock in anhydrobiotic cells (Abe et al. 2009; Ahi et al. 2010; Fu and Chen 2011; Morgan et al. 2006).

The rehydration temperature is another variable to be considered, mainly because rehydration at a temperature below  $T_{\rm m}$ , generally leads to phase transition with the subsequent leakage of intracellular material and lethal damages (Garcia 2011; Julca et al. 2012; Potts et al. 2005). Besides, the rehydration speed should be controlled. Some authors have shown that the survival rate is increased once the cells are slowly rehydrated (Potts 1994; Potts et al. 2005). Nevertheless, if rehydration is extremely slow, the process is not economically feasible (Morgan et al. 2006).

#### 5.6 Technologies used in the drying of microorganisms

There are several methods used in the drying of bacteria, such as freeze-drying, spray-drying, and fluidized bed drying (Morgan et al. 2006). Spray-drying is the predominant process in the dairy industry and could be used for producing large amounts of milk by-products at relatively low cost in comparison to freeze-drying (Lian et al. 2002; Morgan et al. 2006). Using the spray-drying technology, it is possible to obtain granulated dry powders from a cream. This process implies the spraying of the cream into the drying chamber. The air in the chamber is injected at high speed with temperatures up to 200°C (inlet temperature; Fu and Etzel 1995; Silva et al. 2004, 2005; To and Etzel 1997). The little drops formed during the atomization of the cream are dried in a rapid manner through direct contact with the hot air, forming particles or powder granules that reach the final temperature (outlet temperature).

In the last decade, several studies have shown the possibility of using this relatively cheap and highly productive technology in the preservation of probiotic bacteria (Corcoran et al. 2004; Desmond et al. 2002; Gardiner et al. 2000; Silva et al. 2002). Furthermore, some biocontrol bacterial agents have been formulated by spray-drying for their use in agriculture, mainly as wettable powders of small particle size (Arunsiri et al. 2003; Bashan et al. 2002; Hernández et al. 2007).

Table 5.1 shows several examples of the use of spray-dried bacteria in medicine, agriculture and the food industry. For instance, microencapsulation of probiotic bacteria by spray-drying is the most commonly used method for the formulation of such products

Drying technology	Use in medicine	Use in agriculture	Use in food industry
Spray- drying	Probiotics to reinforce the body's natural defense mechanisms (Anal and Singh 2007; Corcoran et al. 2006) Therapeutic use of live bacterial cells (pathobiotechnology; Culligan et al. 2009; Prakash and Jones 2005)	Biological control of insects (Arunsiri et al. 2003; Zhou et al. 2004) Biological control of nematodes (Hernández et al. 2007, 2009)	Starters for the production of polysaccharides (Boza et al. 2004) Probiotic food (Ross et al. 2005; Simpson et al. 2005)
Freeze- drying	Treatment of gastrointestinal diseases (e.g., Crohn's disease with human IL-10 expressed in <i>L. lactis</i> ) (Huyghebaert et al. 2005) Vaccines based on attenuated bacteria (Edwards and Slater 2008; Rexroad et al. 2002)	Biofertilizers for nitrogen fixation (Arraes Pereira et al. 2002; Vriezen et al. 2006) Encapsulation of plant growth-promoting bacteria (Bashan et al. 2002)	Starter cultures for food production (Kiviharju et al. 2005; Ziadi et al. 2005)
Fluidized bed drying	Encapsulation of probiotic bacteria for therapeutic uses (Anal and Singh 2007; Lopez-Rubio et al. 2006)	Biopesticides containing heat-sensitive bacteria (Slininger et al. 2010)	Optimization in the drying of Lactobacilli strains (Mille et al. 2004) Improvement in the preservation of lactic acid bacteria (Strasser et al. 2009)

Table 5.1 Application of Anhydrobiosis in Bacteria to Several Fields

(Anal and Singh 2007; Kailasapathy 2002). These probiotic bacteria have beneficial effects on patients with gastrointestinal disorders, mainly by improving the immunological system (Anal and Singh 2007; Huyghebaert et al. 2005).

Fluidized bed drying is a method for drying granulated solids. An advantage of this technology is the incorporation of a spray-dryer to the fluidized bed apparatus, which allows the conversion of liquids in powders and the subsequent drying of the granulated powders. This method is recommended for drying heat-sensitive bacterial strains, mainly by microencapsulation (Anal and Singh 2007; Lopez-Rubio et al. 2006). In the drying chamber, temperatures of approximately 40°C are reached, which produces a gentle drying without thermal stress.

Although freeze-drying is the favorite method of drying in the microbiological industry, there are losses in cell viability (Morgan et al. 2006). One of the main causes affecting the loss of viability is the need to freeze the samples to be desiccated. The freezing process and especially the freezing speed could damage the cells during drying (Uzunova-Doneva and Donev 2000). The formation of large ice crystals could lead to damage in the cell membranes, causing mortality in the processed biomass.

The freeze-drying process involves two key stages: primary and secondary drying. Primary drying takes place at low temperatures, when the frozen mixture is sublimed by reduction of the chamber pressure below the ice pressure within the product (Morgan et al. 2006). The secondary stage takes place once all frozen water has been sublimed, and the chamber temperature is slowly increased, producing the sublimation of bound water. The end point of the secondary drying is usually determined by analysis of the residual water content in the final product.

Always the freeze-drying method is cited considering the long-term storage of cellular suspensions containing more than 10<sup>8</sup> cells/mL (Costa et al. 2000; Miyamoto-Shinohara et al. 2006). Thus, creams usually have cell concentrations above this value to ensure high survival rates after the drying process (Garcia 2011; Morgan et al. 2006). The freeze-drying technology has been used in the preservation of bacterial strains for several applications (Table 5.1), mainly in the production of probiotics for the treatment of gastrointestinal disorders.

#### 5.7 Stability of desiccated microorganisms

The shelf-life time of a product based on desiccated cells will depend on the method of packing and storage. In desiccated bacterial cells, the mechanisms inducing cell death by oxidation have been acting since the beginning of the drying process (França et al. 2007; Hernández et al. 2009; Potts 1994).

Concerning the analysis of stability in anhydrobiosis state, there are only isolated reports (Achour et al. 2001; Hernández et al. 2007, 2009; Ziadi et al. 2005; Morgan et al. 2006). Most researchers describe the stability of anhydrobiotic cells by using first-order models of microbial death (Achour et al. 2001; Ananta et al. 2005; Ziadi et al. 2005). This means that, according to classical kinetic mechanisms, bacterial inactivation is considered a first-order reaction. However, each bacterial species has its particularities and for both desiccated cells and liquid-phase cells, the thermal death patterns could be dissimilar.

van Boekel Martinus (2002) reported the use of the Weibull model (Equation 5.1) to describe the inactivation of bacterial strains in liquid phase. In this model, *S* is the survival rate, and  $\alpha$  and  $\beta$  are parameters or constants. The parameter  $\alpha$  is called the scale parameter, and  $\beta$  is called the shape parameter. If  $\beta < 1$ , an upward concavity will be observed.

Otherwise, if  $\beta > 1$ , a downward concavity will be observed. Then, a straight line will be obtained when  $\beta = 1$ .

$$\log(S) = -\frac{1}{2.3} \left(\frac{t}{\alpha}\right)^{\beta} \tag{5.1}$$

In general, the formation of tails and arms has been observed in thermal death curves for microorganisms exposed to heat in liquid phase; however, this phenomenon has not been reported for anhydrobiotic cells. In the few reports concerning the stability of desiccated cells, most authors fitted the viability data to linear curves, assuming  $\beta = 1$ , without considering the concavities appearing in survival curves (Achour et al. 2001; Ananta et al. 2005; Ziadi et al. 2005). This could cause mistakes in the estimation of survival for particular zones in the curve.

Recently, Hernández et al. (2009) have reported that the introduction of the term "anhydrobiosis quotient" (Equation 5.2):

$$\varepsilon = \log \frac{Xv}{Hr} 100 \tag{5.2}$$

To characterize the stability of anhydrobiotic cells, where *Xv* is the viability or viable cells (colony-forming units [CFU]/mL); and *Hr* is the residual moisture (%). This function is advantageous because it allows the estimation of stability by using the nonlinear model (Equation 5.3):

$$\varepsilon = \varepsilon_0 - (\alpha . t)^\beta \tag{5.3}$$

Considering the concavities of the curves, moreover, it seems to give a reasonable prediction for the stability of desiccated bacterial cells. Nevertheless, attention should be focused on further mathematical modeling of the desiccated state.

#### 5.8 Case study: Anhydrobiotic cells of a biocontrol agent

Considering the effect of biopesticides in agriculture and the environment, here we present an experimental work related to the obtainment of a solid formulation containing the nematocidal bacterium *Tsukamurella paurometabola* C-924. This microorganism was desiccated by spray-drying technology, and the stability of anhydrobiotic cells was evaluated at several temperatures (Hernández et al. 2007, 2009).

Cells of *T. paurometabola* C-924 were cultured in a 300 L fermentor, and then the biomass produced was subjected to spray-drying at several outlet temperatures between 50°C and 62°C to evaluate the effect of outlet temperature on survival rate using sucrose 10% (w/w dry biomass) as a protective agent. All the experiments were done in triplicate; in total, 27 runs were carried out with a Mobile Minor spray dryer. The cream (pH 6) was subdivided in 1-L samples per run. The inlet temperature was 120°C. According to the experimental design, outlet temperature was between 50°C and 60°C.

Survival rate was determined as follows: 0.1 g of powder was suspended by vortexing in 10 mL of NaCl at 9 g/L, and serial dilutions were made to plate. The survival rate was calculated as follows (Equation 5.4):

Survival (%) = 
$$\frac{N}{N_0}$$
 100 (5.4)

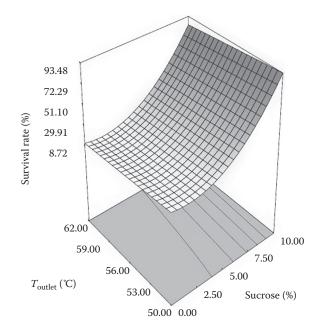
The moisture content of the powders was determined in triplicate by oven-drying at 105°C.

From an analysis of trends in Figure 5.3, it suggests that by increasing sucrose concentration, cells are more protected against thermal stress during dehydration. Data in Figure 5.3 were well-fitted to a reduced quadratic model (Equation 5.5;  $r^2 = 0.9992$ , F = 2118.82, p < 0.0001):

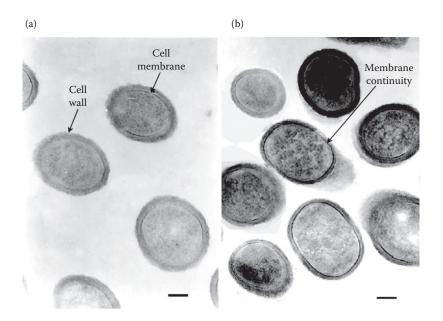
Survival = 
$$57.19 - 3.12 \times \text{sucrose}(\%) + 1.5 \times \text{sucrose}(\%)^2 - 0.74 \times T_{\text{outlet}}$$
 (5.5)

All the coefficients were significant (p < 0.05); mainly, sucrose concentration had a remarkably positive effect on survival. As expected, outlet temperature exerted a negative contribution to survival. High survival rates reflect tolerance to desiccation of *T. pauro-metabola* C-924 in the presence of sucrose. From evaluation of the model, sucrose exerts a more significant effect on survival than temperature. For example, at 10% (w/w), sucrose survival decreases by only 9% when temperature changes from 50°C to 62°C, meanwhile at 62°C, survival is increased in 73.5% when sucrose concentration changes from 0% to 10% (w/w). In all the experimental points, moisture was less than 10% (w/w), which must ensure high stability of formulation according to Garcia (2011).

From this result, the selected experimental conditions were  $T_{\text{inlet}}$  120°C,  $T_{\text{output}}$  62°C, and sucrose concentration 10% (w/w). Using these conditions, survival rates higher than 80% were obtained. Cell membranes remained undamaged after desiccation, as evidenced by transmission electron microscopy (Figure 5.4), contributing to cell survival after spray-drying.



*Figure 5.3* Response surface for the survival rate during the spray-drying of *T. paurometabola* C-924. A three-level factorial design was performed using sucrose as an additive. Sucrose concentration was set between 0% and 10% (w/w). Outlet temperature was set between 50°C and 62°C. Data were analyzed using Design Expert 6.0.1.



*Figure 5.4* Electron micrographs of *T. paurometabola* C-924 cells. (a) Cells before the spray-drying process; (b) cells after desiccation by spray-drying. Bar = 100 nm.

A stability assay of anhydrobiotic cells was performed. Samples were incubated at 4°C, 25°C, and 32°C (under vacuum and without vacuum), respectively, and the inactivation curves were obtained by plotting survival rate versus time.

Shelf-life time (SLT) and half-life time (HLT) are the most useful parameters to describe the stability of a formulation. SLT has been defined as the period in which a biological product can be stored before its biological activity decays (Hernández et al. 2007).

On the other hand, HLT of a bioproduct is the time at which the biological activity of the formulation decreases by 50%. In practice, the most used parameter is SLT. In our case, SLT was defined as the time value at which the cell concentration per gram of the powder (CFU/g), decreases to  $1 \times 10^{12}$  CFU/g. The starting point of viability was  $4 \times 10^{12}$  CFU/g; therefore, in terms of survival, SLT is the time at which survival decreases by 75%.

It was expected that the highest stability was at 4°C for both treatments, and the SLT value was ninefold higher for the powders stored at vacuum. With respect to the powders stored at room temperature (25°C) in both conditions (under vacuum and without vacuum), SLT was twofold higher for powders under vacuum; whereas at 32°C SLT was threefold higher for the powders stored under vacuum. These findings are very similar to those previously obtained by authors who have been working with other bacteria (Achour et al. 2001).

Considering the values of the cell death constant, according to classical kinetics, the Arrhenius plot was performed (Achour et al. 2001). Plotting ln(k) versus (1/T) was well-fitted to a straight line for the powders stored at vacuum in the interval 4°C to 32°C, where k is the cell death constant and T is the absolute temperature.

$$\ln(k) = -4423 \quad \frac{1}{T} + 12.45 \text{ with } r^2 = 0.9435 \tag{5.6}$$

The activation energy calculated from the slope was  $\Delta E^{\#} = 8.75$  kcal/mol, which could be considered as an energy related to diffusion controls, in other words, the energy related to those reactions that are not diffusion-limited reactions in a glassy cytoplasm (Potts 1994).

Non-Arrhenius behavior was obtained for the powders exposed to air (data not shown); this phenomenon could be explained on the basis of the glass transition. Vitrification is an extensively studied process (Crowe et al. 1998), and dried bacterial cells in the presence of sugars could form bacterial glasses (Potts 1994). In the glassy state, rates are low, but Arrhenius kinetics nevertheless still apply. When the system is over the glass transition temperature ( $T_g$ ), non-Arrhenius kinetics are observed, and the dependence of viscoelastic properties on temperature is described by the Williams–Landel–Ferry theory (Potts 1994), which relates the *k* value to *T* according to Equation 5.7; where  $k_g$  is the death constant at  $T_g$ :

$$\log \frac{k_{\rm g}}{k} = \frac{-C_1(T - T_{\rm g})}{C_2 + (T - T_{\rm g})}$$
(5.7)

Consequently, deviations from Arrhenius law will be found experimentally when the interval of temperatures includes  $T_g$ . In addition,  $T_g$  decreases when moisture is increased. In these experiments, the bags containing the powders were sealed in the presence of air with high moisture content (>80%); as a consequence, residual moisture of powders was increased, and likely  $T_g$  decreased. Based on this analysis, we could hypothesize that  $T_g$  value is in the experimental interval of temperatures, explaining the non-Arrhenius behavior obtained.

#### 5.9 Conclusion

Induced anhydrobiosis is one of the most powerful methods used in the preservation of microorganisms. It has been widely applied in the preservation of microbial strains with industrial, agricultural, and biomedical applications. Among the most frequently applied technologies for the stabilization of microbial cells by drying, freeze-drying is the most popular and the most expensive. Although a common methodology for evaluating the stability of anhydrobiotic microbial cells has been developed, further research on mathematical modeling of the desiccated state in microorganisms should be performed to predict the stability of such biological systems.

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# *chapter six*

# *Recent developments in solid-state fermentation*

Chinese herbs as substrate

## Hongzhang Chen and Guanhua Li

## Contents

6.1	Introduction						
6.2	Solid-	Solid-state fermentation versus submerged fermentation					
6.3	Histor	story of Chinese herb SSF processing133					
6.4			hinese herb SSF processing				
	6.4.1	Natura	l SSF of herbs	135			
	6.4.2	SSF app	blied to a concoction of Chinese herbs	135			
	6.4.3	SSF app	blied to the extraction of Chinese herbs	136			
	6.4.4	Advant	ages of Chinese herb SSF processing	137			
6.5	Proces		ey technologies in Chinese herb SSF process				
	6.5.1	Pretrea	tment process of Chinese herbs	139			
		6.5.1.1	Selecting and washing	139			
		6.5.1.2	Cutting process technology	140			
		6.5.1.3	Drying process technology	140			
	6.5.2	SSF mic	croorganisms	141			
		6.5.2.1	Medicinal fungi	142			
			Photosynthetic bacteria				
	6.5.3		hnologies				
			Preparation culture medium with Chinese herbs				
			Bioreactors of SSF				
6.6	Prosp	ects of C	hinese herb SSF processing	146			
	6.6.1	Overall	multistage conversion and directional fractionation of Chinese herb	s146			
	6.6.2	Novel Chinese herb SSF processing based on steam explosion technology 146					
	6.6.3 Process engineering of Chinese herb SSF and its ecoindustrial integration			ı 148			
6.7	7 Conclusions						
Ackı	Acknowledgments						
Refe	rences.			150			

## 6.1 Introduction

Chinese herb solid-state fermentation (SSF) processing has a long history; numerous Chinese medicines such as Shenqu and Hongqu are all produced using SSF. In this chapter, the essence, history, and advantages of SSF are introduced, and the application of SSF in the concoction and extraction of herbs is also stated. The key technology used in Chinese herb SSF processes such as pretreatment, microorganisms, and SSF technology

are systematically summarized. Finally, the idea of process engineering for Chinese herbs SSF and its ecoindustrial integration was proposed and explained.

## 6.2 Solid-state fermentation versus submerged fermentation

Before discussing Chinese herb SSF processing, detailed comparisons between SSF and submerged fermentation (SMF) are provided in Table 6.1. From the table, the many advantages of SSF over SMF can be revealed:

- 1. The culture conditions in SSF are more similar to their natural habitat, which leads to higher product yields (Castilho et al. 2000).
- 2. The fermentation media per mass of substrate in SSF is smaller than that in SMF, which results in smaller reactors and lower capital costs yet higher product yields.

		SSF	SMF
Components	Solid phase	More than 70%	Less than 5%
	Liquid phase	Important part	Main body
	Gas phase	Free oxygen, playing an important role Gas phase	Dissolved oxygen, limit factor Liquid phase
Substrate	Medium	Culture media are simple and low-cost	High quality requirements and high cost
	Nutrients	Nutrients are dissolved in water that is absorbed on substrate, and the distribution is uneven	Nutrients are dissolved in water, and the distribution is uniform
Microorganisms	Growth state	Microorganisms adsorb on or penetrate into the solid substrate	Microorganisms suspend in the culture system
	Strains	Natural enrichment or artificial breeding strains	Pure strains
	Inoculation size	More than 15%	Less than 10%
	Water activity	Low	High
	Environment	Similar to their natural habitat	Not conducive to microbial growth
Fermentation process	Determination and control	Hard	Easy
	Uniformity	Heterogeneity	Homogeneity
	Heat removal	Hard	Easy
Bioreactor	Characteristics	Simple, open	Complex, sealed
	Cost	Operation cost is high	Investment cost is high
	Gas circulation Scale-up	Oxygen supply and heat removal Hard	Oxygen supply Easy
Product	Concentration	High	Low
	Yield	High	Low
	Wastewater	Seldom	Large
	Extraction	Complicated	Easy

Table 6.1 The Detailed Comparisons between SSF with SMF

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- Aeration is much easier and the aeration efficiency is higher in SSF, which is beneficial for those fermentation processes demanding high levels of oxygen (Nagao et al. 2003).
- 4. Diversity in products, especially for anaerobic mixed SSF (Sindhu et al. 2006).
- 5. Lower demand on sterility due to the low water activity used in SSF (Hölker et al. 2004).
- 6. Less water is needed in SSF.

The major challenges in SSF that need to be overcome can be summarized as follows:

- 1. Scale-up in SSF has long since been a limiting factor because of its inefficient heat and mass transfer.
- 2. Determination and separation of biomass are also challenges in SSF.

Although there are some challenges such as high cost of product recovery and purification in SSF, an economic evaluation of the overall process should be done to determine its feasibility for a specific purpose (Rodriguez-Couto and Sanroman 2005; Singhania et al. 2009). This system is suitable for the production of high-value products such as medicines and enzymes, especially those applications in which concentrated products with high titers are required rather than a high degree of purity.

## 6.3 History of Chinese herb SSF processing

In a contemporary Chinese dictionary, traditional Chinese medicine (TCM) is defined as the sum of medicines that are used by herbalist doctors, which includes plants, animals, and a few minerals. For this definition, folk medicine, ethnic medicine, modern Chinese herbs, and even ordinary western medicine were all included. Lei (1998) defined TCM as the generic term for all medicines that have been used by Chinese people in the long history of that nation, which excludes modern Chinese herbs and western herbs. Gao (2007) pointed out that TCM represents not only the medicines but also the traditional syndrome differentiation theory. In conclusion, we thought that TCM were the medicines that were used for the prevention, diagnosis and treatment, and rehabilitation of health, under the guidance of syndrome differentiation theory. Because of the kernel of syndrome differentiation theory of Chinese medicine, TCM cannot be deemed as an empirical health care system. The various aspects of TCM includes Chinese materia medica, acupuncture, and moxibustion, among others. Here, Chinese herbs will be introduced in detail.

Solid-state fermentation originated in ancient China and developed into a vibrant industry in traditional Chinese liquor, rice wine, soy sauce, and vinegar. As early as 4500 years ago, Chinese herbs were processed by using solid fermentation technology to change the original quantity, improve the medicinal efficacy, and expand medication performance. For example, medicated leaven, fermented soybean, and red fermented rice are all fermented products that have been and are still used up to now (Table 6.2). The SSF of Chinese herbs is the cultivation process of beneficial microbes on pretreated herbs; the optimum temperature is approximately 30°C to 37°C, and relative humidity is 70% to 80%.

Solid-state fermentation of Chinese herbs can be divided into two categories:

- Herbs mixed with other raw materials are used as culture medium, which could provide adequate carbon and nitrogen sources for the microorganisms. At the same time, the microorganisms produce metabolites or transform the herbs' ingredients into beneficial active substances.
- 2. Medicinal fungi are cultivated on agricultural materials that provide either nutrients or inducers to obtain active metabolites.

Period	Content
Ancient times	Herb was washed and cut, and also roasted when fire was discovered, to reduce toxicity and enhance the efficacy
Xia and Zhou	It was the beginning of fermentation of Chinese herb, for example, rice wine, soy sauce, and vinegar were produced
Qin and Han	Several medicinal fungi such as <i>Ganoderma lucidum</i> , <i>Poria cocos</i> , and <i>Phellinus baumii</i> were discovered and their medicinal effects were recorded in "Shen Nong Materia Medica." The theory of TCM came into being
Han and Jin	Many processing technologies and medicinal functions of TCM such as medicated leaven and fermented soybean were recorded in "Febrile Disease"
The North Wei	The SSF technology was developed further
Tang	Pretreatment methods of Chinese herbs were summarized; the theory of TCM was concluded and spread to East Asian countries
Song	The effects of auxiliary materials on Chinese herbs during the SSF were studied and utilized
Jing	Mongolian medicine and Tibetan medicine were absorbed into TCM
Ming	The Ming era was the golden era of TCM research, and the theory of fermentation and concoction had been established completely. The TCM and traditional syndrome differentiation theory were combined more closely
Qing	The theory of fermentation and concoction was enriched and developed

Table 6.2 Chronology of Important Events Related to TCM

Source: Chen, H.Z. Medicinal Plants, Process Engineering and its Ecological Industry Integration. Science Press, Beijing, China, 2010; Gong, Q.F. The Concoction of Chinese Traditional Medicine. Chinese Medicine Press, Beijing, China, 2007.

Here, we introduce two fermented Chinese herbs:

(a) Shenqu—also known as medicated leaven or fry bran koji, was first recorded in the Han Dynasty, and is a traditional Asian fermented medicine. Medicated leaven is a type of SSF product made from flour and food fungus. In China, the documentation recommends the use of medicated leaven to help digestion, eliminate phlegm, and so on.

The traditional SSF methods were summarized as follows: (1) *Prunusmandshurica Koehne* and *Phaseolus calcaratus Roxburg* were ground to powders and mixed with flour; (2) *Artemisia annua* juice, *Xanthium sibiricum* juice, and *Polygonum flaccidum* juice were added; (3) after that, the powders were mixed and kneaded, and formed into a paste; (4) the fermentation proceeded at 30°C to 37°C for 7 to 8 days; and finally, (5) the Shenqu should be kept in a cool and dry place (Nan and Zhu 1992).

(b) *Hongqu*—also known as red fermented rice, red yeast rice, or anka, was first recorded in the Zhou Dynasty, and is a traditional Asian fermented food and folk medicine. Hongqu is a type of SSF product made from steamed rice and food fungus from the *Monascus* genus. Hongqu is rice-shaped, purple or brown-red, crisp, friable, and slightly tartish or tasteless. In China, the documentation recommends the use of Hongqu as a colorant for cooking, Chinese medicine, and fermentation starters to brew red wine.

The traditional SSF methods were summarized as follows: (1) A 1-m-deep pit is made in red soil, and a mat is put around this pit; (2) rice (preimmersed for 2 h in water at 30°C) and *Monascus* are placed into the pit; (3) a big stone is placed over the fermentation medium; (4) fermentation proceeds at approximately 28°C for 3 to

4 days; (5) the shell of the rice turns red, and the core of the rice turns purple-red; (6) the red fermented rice is put into a pot and roasted until the shell becomes black and the core becomes yellow; and finally, (7) Hongqu should be kept in a cool and dry place (Chen and Hu 2005).

## 6.4 Principle of Chinese herb SSF processing

### 6.4.1 Natural SSF of herbs

In ancient times, people occasionally vomited or even lost consciousness after eating some plants and animals. However, these symptoms might be alleviated or relieved by eating some other foods. People probably gained the first cogitation of medicine from these daily practices. In other circumstances, people consciously preserved some excess fruits during harvest time, which would begin to ferment if the environment was suitable. The sugars in the fruits were first fermented into ethanol by microbial metabolism, and subsequently, the ethanol was partially transformed into acetic acid. Then, the reaction of ethanol with acetic acid produced some special flavor substances such as ethyl acetate, which would bring in a magical taste. In fact, this is similar to SSF, which occurred during the storage process of Chinese herbs. The herbs would be invaded by some microbes, especially molds such as *Penicillium* and *Aspergillus*, which would produce enzymes and secondary metabolites using it as a carbon and nitrogen source. Some active ingredients in the herbs were modified or transformed by enzyme or microbes during the process. Therefore, the fermented herbs could be used as medicine, and the herbs resulting from different microbes would produce different medical efficacies.

## 6.4.2 SSF applied to a concoction of Chinese herbs

Herbs should be concocted before being used as medicine, which is characteristic of traditional Chinese herbs, and is also a major feature of the theory behind Chinese medicine. The effects of concoction can be summarized by two aspects: (1) destruction or alteration of the chemical structure of the noxious ingredients in the herbs to release toxicity; and (2) inactivation of the coexisting enzymes in the herbs to ensure the clinical efficacy of the medicine.

The effects of concoction caused by microbes on Chinese herbs can be mainly stated as follows:

1. *Biotransformation of herbs*. The ingredients in herbs can be modified or even transformed into new compounds by various enzymes such as cellulase, laccase, lignin peroxidase, and lipase, which is produced from microbes at room temperature. For example, Pu-erh tea is a well-known traditionally fermented product that is derived from the dry leaves of *Camellia sinensis* var. *assamica* in Yunnan, China. Several investigations have recently confirmed the beneficial effects of Pu-erh tea on health, such as its antioxidant, antimicrobial, and anticancer effects, as well as lowering cholesterol, blood pressure, and blood sugar levels (Hou et al. 2009). Wang et al. (2011) studied the SSF of Pu-erh tea using *Foxia* as a microbe. The SSF proceeded for 20 days at 45°C and 70% relatively humidity. The results showed that the theabrownin, total carbohydrate, polysaccharide, amino acid, and protein contents all increased, whereas the tea polyphenol content decreased sharply. Kluge and coworkers (2012) reported that stereoselective benzylic hydroxylation and C1–C2 epoxidation of

alkylbenzenes were catalyzed by heme-thiolate peroxygenase produced from *Agrocybe aegerita*. Benzylic hydroxylation led exclusively to the (*R*)-1-phenylalkanols. Epoxidation of straight-chain and cyclic styrene derivatives gave a heterogeneous picture. Cranberry pomace contains large amounts of phenolic glycosides, which are important sources of free phenolics that are used as antioxidants, flavorings, and nutraceuticals. Zheng and Shett (2000) used  $\beta$ -glucosidase produced by *Lentinus edodes* to hydrolyze glycosides. The results showed that after 50 days of cultivation, the yield of total free phenolics could reach 0.5 mg/g. The major free phenolics produced from cranberry pomace were gallic acid, chlorogenic acid, *p*-hydroxybenzoic acid, and *p*-coumaric acid. It can be concluded that the enzymes produced by micro-organisms were closely related to the compositional changes occurring as well as the quality of the herbs used.

- 2. *Biological detoxification of herbs.* Some toxic substances in the herbs can be decomposed by the microbes during the SSF. Brand et al. (2000) studied the detoxification of coffee husk in SSF using *Rhizopus, Phanerochaete*, and *Aspergillus*. The results showed that *Rhizopus arrizus* can degrade 87% of caffeine and 65% of tannins when fermentation proceeded for 6 days at pH 6.0 and at 60% moisture. When *Phanerochaete chrysosporium* was used, the maximum degradation of caffeine and tannins were 70.8% and 45%, respectively. When fermentation proceeded for 14 days at pH 5.5 and at 65% moisture, *Aspergillus* was used; the best detoxification rates could achieve 92% for caffeine and 65% for tannins (Brand et al. 2000). The results demonstrated that microbes had the ability to detoxify herbs.
- 3. The formation of new active compounds.
  - a. Several secondary metabolites could be produced by medicinal fungus
  - b. The secondary metabolites formed may interact with the active ingredients of herbs to generate new compounds
  - c. Compared with a conventional culture medium (straw or starch), herbs are full of active ingredients, such as polyphenol, which may promote the growth of microbes

Treviño-Cueto and coworkers (2007) used *Larrea tridentata* powder as the sole carbon source and inducer to cultivate *Aspergillus niger*. The results indicated that tannase production reached values of 1040 units/L, and a high content of tannins and gallic acid were obtained during fermentation for 43 h. Lin and Chiang (2008) used *Radix astragali* as the medium to culture *Cordyceps militaris*, which would produce a substance with antitumor activity. These results showed that the antitumor activity of the crude fermentation broth can be enhanced by microorganisms. Besides cordycepin, several other unknown compounds were also responsible for the antitumor activity, which indicated that the novel antitumor activity mechanisms were caused by the addition of *R. astragali*.

#### 6.4.3 SSF applied to the extraction of Chinese herbs

The distributions of active ingredients in Chinese herbs are very complex. Active ingredients in different parts of Chinese herbs may have different pharmacological effects. Therefore, the extraction of active ingredients is an important step. A basic requirement in the extraction process is the maximization of the yield of active ingredients from the herbs while keeping the yield of harmful and ineffective substances at a minimum. Traditional

Types	Mechanism/Characteristic
Ultrasound-assisted extraction (UAE)	The mechanism is generally attributed to mechanical, cavitation, and thermal effects. These effects can result in the disruption of cell walls, reduction in particle size, and enhancement of mass transfer. The implosion of cavitation bubbles accelerates the eddy diffusion and internal diffusion (Cabaleiro et al. 2013)
Supercritical fluid extraction (SFE)	SFE is based on the use of an extractant in its supercritical state; gas can penetrate into porous solid materials more effectively than liquid. Carbon dioxide is commonly used for this purpose, because it easily reaches supercritical conditions, and moreover, it displays low toxicity (Lang and Wai 2001) and easy separation
Microwave-assisted extraction (MAE)	MAE uses microwave energy to enhance the extraction efficiency. The electric field causes heating via two simultaneous mechanisms, dipolar rotation and ionic conduction. This oscillation produces collisions with surrounding molecules and thus the liberation of thermal energy into the medium (Kaufmann and Christen 2002)

Table 6.3 Modern Technology Applied to the Extraction of Herbs

Source: Chen, H.Z. Process Engineering in Plant-Based Products. Nova Science Publishers, Inc., New York, 2009.

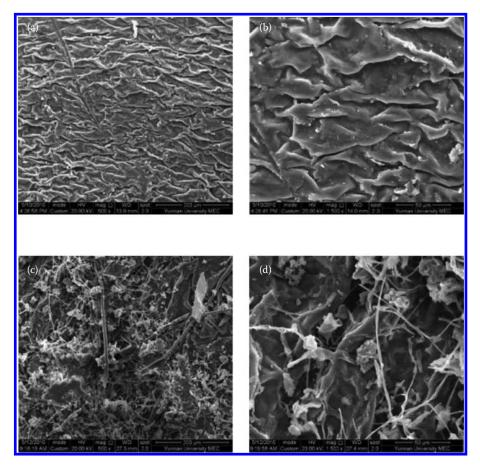
extraction methods for herbs include boiling, soaking, percolation, refluxing, sublimation, and others. Although these classic methods have their own advantages, some issues still need to be addressed, such as being time-consuming, having low efficiency, using large amounts of solvent, and having a complicated extraction procedure. Consequently, researchers are still exploring more effective extraction methods. With the development of science and technology, more and more new technologies are being applied to the extraction process of herbs (Table 6.3; Chen 2009).

Active ingredients in herbs wrapped by cellulose, hemicellulose, lignin, and others could cause difficulties in extraction. The dense structure of herbs can be destroyed by the microbe's invasion or effectively degraded by enzymes, which are beneficial for further extraction processes. The SSF of Pu-erh tea was done by Wang and colleagues (2011). The results demonstrated that the surfaces of untreated tea appeared to be smooth and intact, and the structures of the cell, cellulose, and lignin were complete. Yet, after several days of fermentation, their surfaces were covered by microorganisms and the structures of the cells were largely disrupted. The scanning electron micrograph (SEM) morphology also suggests that the structural changes in the tea leaves were related to the growth of microorganisms (Figure 6.1).

#### 6.4.4 Advantages of Chinese herb SSF processing

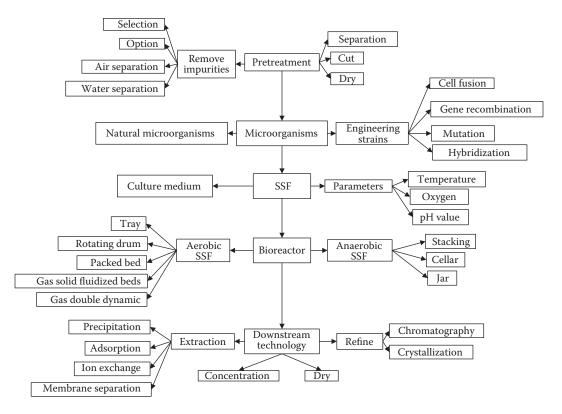
Compared with other methods, the advantages of SSF on Chinese herb processing can be stated as follows:

1. Mild reaction condition. SSF is often carried out under relatively mild conditions such as ambient temperature and pressure; as a result, the active ingredients in the herbs can be kept from destruction as effectively as possible, especially for heat-sensitive aromatic volatile substances, vitamins, and other active ingredients.



*Figure 6.1* SEM of tea. (a, b) SEM of untreated tea. (c, d) SEM of fermented tea. Magnification: (a, c) ×500; (b, d) ×1500. (From Wang, Q.P. et al., *Journal of the Science of Food and Agriculture*, 91: 2412–2418, 2011.)

- 2. Formation of new Chinese medicine compounds. Some microbes have been identified as traditional folk medicines for their special active metabolites. Microbes could interact with herbs during the SSF. The microbial growth and metabolism could be promoted, and the ingredients in herbs could be modified. As a result, the medicinal effects of Chinese medicine compounds were improved.
- 3. Being convenient for assimilation and transformation of active ingredients. The macromolecule substances can be degraded into micromolecule substances by microbial metabolism. Compared with macromolecule substances, small active molecule substances easily pass through the blood–brain barrier and combine with the protein of human cells. Thus, they can be faster and more completely assimilated and used by humans.
- 4. Multilevel utilization of Chinese herbs. The ingredients of herbs are complex; in the traditional process, only several special components have been used, yet many other parts were discarded, which resulted in a waste of resources. During the fermentation process, herbs were repeatedly used and most of the nutrients can be converted. Therefore, SSF is beneficial for comprehensive utilization of herb resources.
- 5. Last but not least, the fermentation products can be used without any postprocessing.



*Figure 6.2* SSF on Chinese herb processing. (From Chen, H.Z., Medicinal Plants, Process Engineering and Its Ecological Industry Integration. Science Press, Beijing, China, 2010; Gong, Q.F., The Concoction of Chinese Traditional Medicine. Chinese Medicine Press, Beijing, China, 2007.)

## 6.5 Process and key technologies in Chinese herb SSF process

There are five sections involved in SSF of herbs, which are summarized as follows (Figure 6.2).

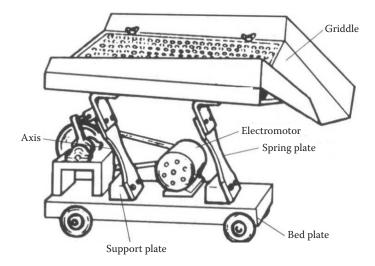
#### 6.5.1 Pretreatment process of Chinese herbs

#### 6.5.1.1 Selecting and washing

Herbs are often mixed with sediment impurities, and invaded by insects and fungus during the collection, pretreatment, storage, and transportation processes. Therefore, herbs need to be pretreated to reach the standard of medicinal purity and to facilitate the subsequent processes. Selection is the first step of herb processing. A complete herb usually consists of roots, stems, leaves, flowers, and fruit. According to the different ingredients contained in different parts of one herb, the herb should be selectively fractionated to reach uniformity, such as the removal of roots, residual stems, cores, and hull.

The sand and impurities can be separated from herbs, after being passed through different specification sieves. On the other hand, the different parts of herbs can be fractionated based on their size, which is suitable for subsequent fermentations and concoctions (Figure 6.3; Gong 2007).

Air separation is one of the most common methods used in herb processing. Herbs and impurities can be blown off, shaken, and separated by wind according to their different



*Figure 6.3* Shock screening machine. (From Gong, Q.F., *The Concoction of Chinese Traditional Medicine*. Chinese Medicine Press, Beijing, China, 2007.)

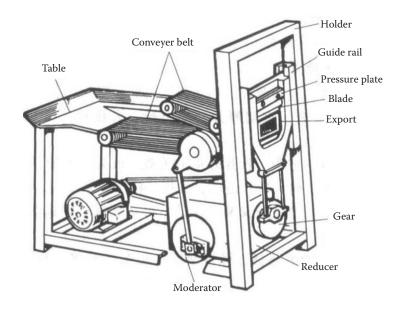
specific gravities. If the sediments, salt, or other impurities cannot be removed by air separation, the herbs should be washed in water.

#### 6.5.1.2 Cutting process technology

The particle size of the herbs is also a major factor affecting the performance of SSF. Large particle size can hamper the penetration of fungi into cellulosic biomass and also prevent the diffusion of air, water, and metabolite intermediates into the particles. However, the reduced particle size with a decreased size of interparticle channel may adversely affect interparticle gas circulation (Zadrazil and Puniya 1995). Cutting is the process in which herbs are cut into special specifications to facilitate identification, fermentation, concoction, and extraction. Dried herbs are relatively hard, so soaking pretreatment is the first step that needs to be done before cutting. During the soaking process, herbs are soaked to soften, as a result, herbs are easy to cut and the toxic effects of the herbs can be alleviated. The cutting process can be divided into hand-cutting and machine-cutting. Hand-cutting is flexible, but less efficient, thus it is suitable for fine herb processing. Machine-cutting is fast, highly efficient, and less labor-intensive, thus it is suitable for bulk herb processing. Commonly used machines include chopping knife, cutting machines, and rotary cutting machines (Gong 2007; Figure 6.4).

#### 6.5.1.3 Drying process technology

After the cutting process, Chinese herbs should be dried for long-term preservation. The basic requirements being that the herbs should retain their shape, color, smell, and taste. The drying process is usually divided into natural drying and machine drying. For natural drying, herbs are placed in the sun or in a cool but ventilated place. Traditional natural drying technology does not need special equipment, but the drying efficiency is often affected by the weather. For machine drying, herbs are dried by equipment such as a direct hot air dryer, electric dryer, or far infrared dryer, which is clean, weatherproof, and highly efficient.



*Figure 6.4* Rotary cutting machine. (From Gong, Q.F., *The Concoction of Chinese Traditional Medicine*. Chinese Medicine Press, Beijing, China, 2007.)

#### 6.5.2 SSF microorganisms

Microorganisms constitute the most abundant and diverse set of organisms on earth, and are widely distributed in the soil, water, and air. Microorganisms play a dominant role in performing key biochemical reactions essential to sustaining the biosphere. Generally, it is accepted that more than 80% of the total microorganisms are unknown; however, reactions mediated by both the known and the unknown microorganisms are already used in SSF (Baxter and Cummings 2006). Regardless of the exact nature of the fermentation and concoction process, all fermentation techniques depend on the right microorganisms in the right place with the right environmental conditions for degradation or biotransformation to occur (Iranzo et al. 2001). The right microorganisms are those microbes that have the physiological and metabolic capabilities to utilize and degrade the herbs. Compared with other environments, the conditions in soil are more suitable for the growth of microorganisms, thus the most industrial strains are isolated from the soil. The vast majority of wild-type microorganisms cannot produce commercially acceptable yields of the desired product, so the isolated strains are often modified.

Selection of suitable microorganisms can be deemed as one of the most important criteria in SSF. The requirements of SSF for strains can be stated as follows:

- 1. A unique characteristic of SSF is having no free water in the medium, so the microbes can grow in low concentrations of moisture
- 2. The microorganisms are capable of producing a range of extracellular enzymes required for the hydrolysis of complex, polymeric raw solid substrates
- 3. Growth cycle is as short as possible
- 4. The culture condition such as temperature, pH, and oxygen can be easily met

- 5. The microorganisms should grow fast and can inhibit other microbes' reproduction
- 6. They cannot be pathogenic microorganisms or gene-engineering strains
- 7. For the microorganisms used in SSF of herbs, there are several additional requirements that need to be stated
  - a. The growth of microorganisms cannot be inhibited by the active ingredients in the herbs, and the active ingredients in the herbs cannot be destroyed by microbes either
  - b. The microorganisms do not produce harmful substances to humans, and can easily be inactivated

#### 6.5.2.1 Medicinal fungi

In "Shen Nong Materia Medica," plenty of fungi were recorded as medicine, which had been used for thousands of years in East Asia, especially in China. Medicinal fungus belonging to higher Basidiomycetes and Ascomycetesares are an immensely rich yet largely untapped resource of useful, accessible, and natural compounds with various biological activities. The medicinal ingredients produced by fungus have been isolated and identified from the fruit bodies, culture mycelium, and culture broth. These various cellular components and secondary metabolites (polysaccharides, proteins and their complexes, phenolic compounds, polyketides, triterpenoids, steroids, alkaloids, nucleotides, etc.) have provided significant opportunities for the treatment of patients, for example, using drugs with antitumor, antioxidant, immunomodulatory, antibacterial, antiparasitic, radical scavenging, antihypercholesterolemia, cardiovascular, antiviral, detoxification, and hepatoprotective effects, among others (Elisashvili 2012). For example, the best known antibiotic, penicillin, discovered from a species of *Penicillium* in the early 1940s, has thus far played an important role in human life.

The active ingredients in medical fungus can be divided into five types: polysaccharides, alkaloids, sterols, terpenes, and polypeptides (Table 6.4).

Wasser (2010) summarized that approximately 700 species of higher Heterobasidiomycetes and Homobasidiomycetes could produce fungal polysaccharides. Approximately 126 medicinal compounds produced by fungus have been reported, and several of these compounds have been proceeded through Phase I, II, and III clinical trials and used extensively and successfully in Asia.

*Phellinus* Quél. is an important genus in Basidiomycota; approximately 220 species belong to this genus (Larsen and Cobb-Poulle 1990). Dai and coworkers summarized that approximately 70 species of *Phellinus* have been reported in China, and 26 species are recognized as medicinal fungi. Polysaccharides and polyphenols of the *Phellinus* genus are two of the main kinds of medicinal metabolites, and their medicinal functions can be summarized as antitumor, improving immunity, and antioxidant (Dai and Yang 2008; Dai et al. 2010).

#### 6.5.2.2 Photosynthetic bacteria

Photosynthetic bacteria (PSB), the common microorganisms in the natural environment, have been applied in the fields of environmental protection and energy production, as well as in the medical, fertilizer, and food industries. Photosynthetic bacteria can produce relatively large amounts of physiological active substances such as vitamin  $B_{12}$ , ubiquinone, 5-aminolevulinic acid, porphyrins, and RNA, some of which have been prepared and commercialized (Sasaki et al. 2005; Table 6.5).

Active ingredients	Medicinal metabolites	Functions	Species
Polysaccharide	Hispidin,1,1- distyrylpyrylethan 3,14'-bihispidinyl, hypholomine	b, c	Inonotus xeranticus, Phellinus linteus (Jung et al. 2008)
	51	i, l, n, o	P. baumii (Hwang et al. 2007)
	Intracellular polysaccharide, triterpenoid	e, h, k	(Shih et al. 2006)
	Intracellular polysaccharides, Exopolysaccharides	С	<i>Laetiporus sulphureus</i> Lung and Huang (2012)
	Polysaccharide fractions	a, g	G. lucidum (Zhao et al. 2010)
	Heteropolysaccharide maitake Z-fraction	g	<i>Grifola frondosa</i> (Masuda et al. 2010)
Alkaloid	Isoindolinone	g	<i>Hericium erinaceum</i> (Noh et al. 2012)
	Sesquiterpenes	g	<i>Flammulina velutipes</i> (Xu et al. 2013)
Sterol	Ascorbic acid, flavonoids	с, е	<i>Phellinus igniarius</i> (Lung et al. 2010)
	Geranylated aromatic compound	с	<i>Hericium erinaceum</i> (Yaoita et al. 2012b)
	Vitamin D <sub>2</sub>	g, h, j	<i>Agaricus bisporus</i> (Koyyalamudi et al. 2011)
	Vitamin D <sub>4</sub>	j	(Phillips et al. 2012)
	Cerevisterol	g	(Froufe et al. 2012)
Terpene	Lanostane triterpenoids	d	<i>Ganoderma orbiforme</i> (Isaka et al. 2013)
	Sesquiterpenoids	d	<i>Russula sanguinea</i> (Yaoita et al. 2012a)
	(-)-(2S,8R)-8,12-dihydroxy- isolongifolanol	d	Aspergillus niger (Sakata et al. 2011)
Polypeptide	Low-molecular weight protein fraction	g	<i>Grifola frondosa</i> (Kodama et al. 2010)
		g	(Liang et al. 2011)
		h	G. lucidum (Yeh et al. 2010)

	Table 6.4	The Acti	ve Ingre	edients I	Produced	from	Medical	Fungi
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Notes: Medicinal functions of fungus: (a) activation of macrophage cell; (b) alleviating septic shock; (c) anti-oxidation; (d) antibacterial; (e) anti-inflammatory; (f) antiplasmodial activity; (g) antitumor activities; (h) improving immunity; (i) lowering serum lipids; (j) preventing and treating cardiovascular diseases; (k) promoting blood circulation; (l) restoring liver; (n) treating anemia; and (o) treating human diabetes.

Zhu et al. (2010) studied active extracts from Liuweidihuang that was fermented by PSB. The results showed that PSB could lead to an increase in endogenous antioxidant enzyme activity and a reduction in the risks of lipid peroxidation accelerated by age-induced free radicals. Li added certain amounts of extract of Chinese herbs such as liquorice, baikal skullcap, and trichosanthis into the medium to cultivate PSB. The results showed that the growth of PSB can be greatly prompted (Sun et al. 2007).

Medicinal metabolites	Medicinal functions	Species
Vitamin B <sub>12</sub>	Treating anemia and	R. spheroids (Cauthen et al. 1967)
	neuritis, or used as an	R. spheroids, R. rubrum (Ohmori et al. 1971)
	eye lotion	Propionibacterium (Sasaki et al. 1998)
Coenzyme Q10	Antioxidative	<i>R. rubrum</i> (Tian et al. 2010)
	supplement, enhancing	R. palustris (Fang et al. 2005)
	immune system, easing hypertension	Rhodobacteria sp. (Lai et al. 2006)
5-Aminolevulinic acid	Treatment of skin cancer	R. sphaeroides (Tanaka et al. 1994)
Porphyrin	Treatment of skin cancer	Photosynthetic bacteria (Mongin et al. 1999)

Table 6.5 Active Substances Produced from Photosynthetic Bacteria

#### 6.5.3 SSF technologies

#### 6.5.3.1 Preparation culture medium with Chinese herbs

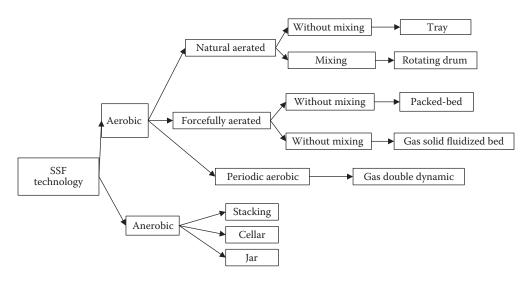
Nutrition is the process in which microorganisms get energy and nutrients from the external environment in their lifetime, which also plays basic physiological functions that provide structure substances, energy metabolism regulation substance and the necessary physiological environment to metabolisms (Zhou 2002). The addition of nutrients to the herb medium can generally increase the formation of the fungal biomass and also facilitated fungal colonization in the deeper areas. Microbial basic nutritional elements can be divided into six categories: carbon sources, nitrogen sources, energy, mineral, water, and growth factors (Chen 2013). Carbon sources are major nutrients for microorganisms, and includes organic carbon and inorganic carbon sources. In SSF processes, the carbon source substances can often be used both as nutrients and as inert carrier materials that maintain the microorganisms' growth. Consequently, it is essential to go into the characteristics of the solid substrate during the SSF process, especially for the amplification process. Chen (2013) has paid much attention to the pretreatment of the herbs, which will be discussed in later sections. Nitrogen sources mainly provide nitrogen elements for microbial growth, which are used to synthesize important live protein materials and nucleic acid. Common raw protein materials mainly include bean substances, such as soybean peas, soybean meal, bran, urea, peptone, cicada chrysalis powder, and others. Energy mainly provides initial energy sources for nutrition or for microbial organisms. Inorganic salts and growth factors are two other substances needed for microbial growth. One common characteristic is their lower demand; they are essential to the growth of the microorganisms yet they cannot be synthesized by the microbial organisms themselves. These mainly include a vitamin base, amine, and small-molecule fatty acids. Inorganic salts refer to K, P, S, Ca, Mg, and others. During the fermentation process, the additional carbon and nitrogen sources are often mixtures that include combinations of ingredients. For example, straw is rich in K, P, S, Ca, Cu, Mg, and others, which can provide enough inorganic salt for the growth of microbes (Yu and Chen 2010). Water is a necessary nutritional element for microbial growth and is an important part of the organism itself. The initial water content of the substrate is important for fungal establishment and growth, which also affects the secondary metabolism of the fungus. Previous studies suggested that an initial water content ranging from 70% to 80% was the optimal level for most white rot fungi (Asgher et al. 2006; Wan and Li 2010) Water can assist microbes in transferring metabolic nutrients from the outside into the cell. On the other hand, water molecules also play a role in the maintenance of macromolecular stability, and provide a relatively stable microenvironment. In aerobic SSF, water can be divided into bound water and free water, and water content is an important factor that influences heat and mass transfer. Bound water exists as a thin water film layer, and plays a role in the absorption of nutrients and the desorption of metabolic substances. Free water, commonly expressed by water activity, can be defined as the ratio of solvent fugacity and pure solvent fugacity (Chen 2013; van den Doel et al. 2009).

Compared with ordinary solid substrates, the ingredients in herbs are more complex, and the interactions between herbs are more diverse. Therefore, some of the additional requirements of a culture medium for microbial growth should be stated: (1) to meet the basic growth requirements of the microorganisms, (2) the changes of active ingredients in herbs caused by the microorganisms during fermentation, (3) the effects of the active ingredients on the growth of microbes, (4) the conditions of pretreatment of herbs, and (5) the drying process of the products.

#### 6.5.3.2 Bioreactors of SSF

Various bioreactor types have been used in SSF processes, based on mixing and aeration. They can be divided into four general categories (Figure 6.5; Mitchell et al. 2000).

- 1. *Unmixed beds without forced aeration*—the bed is static and air is circulated around the bed, but not blown forcefully through it (i.e., tray bioreactors). Each tray contains a thin layer of substrate (~5 to 15 cm deep).
- 2. Unmixed beds without forced aeration—the bed is continuously mixed or mixed intermittently with a frequency of minutes to hours, and air is circulated around the bed, but not blown forcefully through it. The substrate bed is held within a horizontal or near-horizontal drum, which may or may not have baffles yet continuously rotates, such as a rotating drum.
- 3. *Unmixed beds with forced aeration*—the bed is static and air is blown forcefully though the bed (i.e., packed-bed bioreactors). There is a static bed on top of a perforated plate through which conditioned air is blown, although one interesting variation is the



*Figure 6.5* The classification of SSF technology used in Chinese herb processing. (From Mitchell, D.A. et al., *Process Biochemistry*, *35* (10): 1211–1225, 2000.)

introduction of air through a perforated rod inserted into the center of the bed, such as a packed bed.

4. *Mixed beds with forced aeration*—the bed is agitated and air is blown forcefully through the bed (Chen 2013).

Chen and coworkers designed gas double-dynamic SSF technology. In 1998, the largescale solid-state pure culture fermentation demonstration plant was built. The 110 m<sup>3</sup> bioreactors were used to produce biopesticides. The results showed that economic indicators of this technology were better than traditional SMF. On this basis, the gas double-dynamic SSF technology gradually developed as one of the modern SSF technologies (Chen and Li 2007; Chen 2013; Chen and He 2013).

## 6.6 Prospects of Chinese herb SSF processing

## 6.6.1 Overall multistage conversion and directional fractionation of Chinese herbs

Chinese herbs, as treasures of the Chinese civilization, play important roles in human life, yet most of them are rare species in nature and should be preserved. Due to overexploitation and utilization, the wild resources of Chinese herbs have suffered serious damage. Therefore, the utilization and protection of Chinese herbs should be coupled together; ecological, social, and economic benefits should aim for harmony and unity, which are the core requirements for the modernization of the TCM industry.

With respect to the low extraction efficiency and high separation cost in the traditional Chinese herb processing, the overall multistage conversion and directional fractionation of Chinese herb processing was proposed by Chen (2010). This idea contains two aspects. First, for various kinds of herbs or exploitation targets, different transformation processes should be established. Second, all different active ingredients in herbs should be separated and used based on their characteristics.

#### 6.6.2 Novel Chinese herb SSF processing based on steam explosion technology

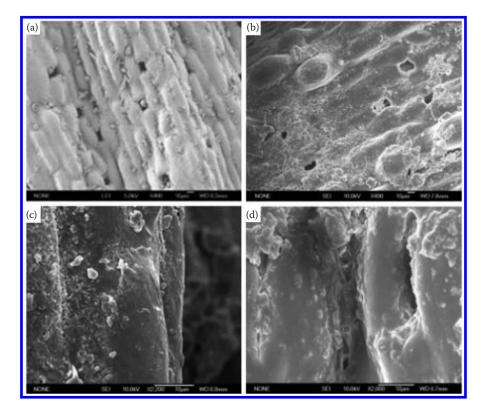
The herbs are enclosed by a thick and rigid phellem layer, and most active ingredients in herbs are combined with the lignin in the cell wall that is wrapped by the cellulose and hemicellulose, which causes the difficulty in the extraction of the active ingredients. Based on the essential characters of the herbs, Chen and others developed the unpolluted low-pressure steam explosion technology, which has been successfully applied to Chinese herbs processing, clean pulping, and xylan production. For this technology, the pressures are lower than 1.5 MPa, and no chemicals are added (Chen and Liu 2007a; Zhang et al. 2012). Raw herbs are cooked by vapor under certain pressure and temperature conditions, the cell wall is broken, starch and hemicellulose are hydrolyzed into soluble sugars, and the lignin is softened and partially degraded. On the other hand, vapor can penetrate into the pores of materials, and then the pressure is discharged immediately, resulting in a mechanical crushing effect. That is to say, the steam explosion technology is coupled with chemical hydrolysis and physical crushing (Chen and Liu 2007b).

The effects of steam explosion technology on the extraction of ephedrine from the ephedra herb were studied. The optimum steam explosion condition was 35% moisture at 1.5 MPa for 3 min, and then the pressure was discharged immediately. Under these conditions, the extraction yield of ephedrine was 0.345%, which was 243% higher than the control. The pretreated ephedra herbs were observed by electron microscopy; the results

showed that the holes of the untreated ephedra surfaces were scarce and small, yet the holes of the pretreated ephedra surfaces were large and dense, and cellulose fibers were exposed and became sparse (Figure 6.6; Chen 2006; Chen and Peng 2012).

Resveratrol (3,5,4'-trihydroxystilbene) is a naturally occurring phytoalexin produced by some spermatophytes, such as *Polygonum cuspidatum*. As a phenolic compound, resveratrol contributes to the antioxidant potential and may play a role in the prevention of human cardiovascular diseases. Resveratrol has also been shown to modulate the metabolism of lipids, and to inhibit the oxidation of low-density lipoproteins and the aggregation of platelets. Moreover, as a phytoestrogen, resveratrol may provide cardiovascular protection (Adrian and Jeandet 2012). *P. cuspidatum* is a perennial herb, its roots are full of resveratrol and is commonly known in TCM. The combination pretreatment of steam explosion with biological fermentation was established to transform polydatin into resveratrol. The steam explosion condition was 0.8 to 1.6 MPa for 1 to 6 min, the pretreated sample underwent enzymatic hydrolysis for 36 h, and the fermentation condition was 28°C for 5 days (Chen and Peng 2010).

Steam explosion is conducive to the extraction of the active ingredient from herbs and the growth of microorganisms in SSF. Therefore, the unpolluted low-pressure steam explosion technology has broad application prospects in the extraction, concoction, and detoxification of herbs.



*Figure 6.6* SEM of ephedra herb (a) The untreated ephedra herb (magnification, ×400); (b) the pretreated ephedra herb (magnification, ×400); (c) the untreated ephedra herb (magnification, ×2200); (d) the pretreated ephedra herb (magnification, ×2000). (From Chen, H.Z. and Peng, X.W., *Progress in Chemistry*, 24 (9): 1857–1864, 2012.)

#### 6.6.3 Process engineering of Chinese herb SSF and its ecoindustrial integration

In traditional Chinese herb processing, the characteristics and interactions between operating units were often neglected. Each operating unit was relatively independent, lacking coordination and complementarity, which lead to low efficiency and high production costs. For example, in the basic process of traditional concoction, only one active ingredient was utilized and the other parts of the herb were often discarded as waste, or used for mushroom cultivation or feed.

From the perspective of engineering and processing, Chen proposed an ecological industry chain thinking that was based on steam explosion technology and SSF technology. The key points of ecological industry chain thinking were selective fractionation and economic functional utilization. Namely, based on the characteristics of the herb, its components were fractionated and used. Various technologies were introduced, each operating unit was analyzed deeply, and multiple processes were coupled (Chen 2009, 2010). In this process, resource utilization, energy conversion, and environmental protection were organically combined; the ecological industry chain, including clean production, technology integration, and product diversification, was established ultimately.

Pueraria (*Radix puerariae*) is a perennial leguminous plant, which was first recorded in Shennong's Chinese Materia Medica. It is rich in starch and isoflavones. *R. puerariae* is listed as both food and medicine by the Ministry of Public Health of China. More than 30 types of Pueraria are found around the world, and most of them are distributed in China, Japan, and Southeast Asia.

*R. puerariae* is composed of starch, cellulose, protein, isoflavones, and a small amount of fat, pectin, tannins, and alkaloid. *R. puerariae* can be used for the extraction of flavones, or as a starch-type energy source plant to produce ethanol. Consequently, *R. puerariae* has become one of the focuses of economic crop development projects in China, and its planting area is huge (Figure 6.7).

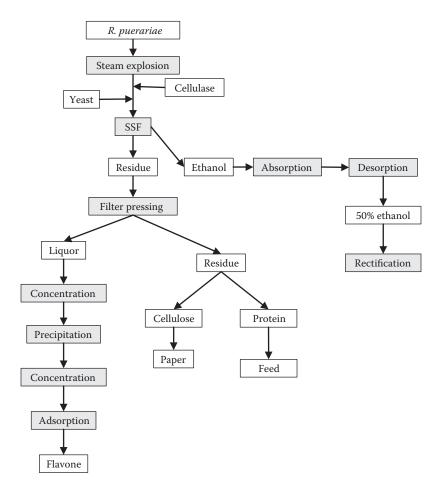
Chen invented a new process for coproducing ethanol and *R. puerariae* flavones as well as fibers from *R. puerariae*. Finally, a classified transformation and a comprehensive and clean utilization of *R. puerariae* components technology was established. This technology is composed of the following steps: (1) *R. puerariae* was pretreated by steam explosion; (2) ethanol was produced through simultaneous saccharified SSF; (3) *R. puerariae* flavones were extracted from the fermentation residue; and (4) fibers were separated from residues (Chen et al. 2007; Chen and Qiu 2010; Chen and Li 2013).



Figure 6.7 Pueraria (R. puerariae).

The concrete steps can be stated as follows (Figure 6.8):

- 1. *R. puerariae* was steam explosion pretreated under a steam pressure of 0.5 to 1.0 MPa for 2 to 4 min
- 2. SSF for ethanol production
  - a. The steam-exploded *R. puerariae* was sterilized in the SSF reactor. Then the glucoamylase, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, KH<sub>2</sub>PO<sub>4</sub>, and the activated yeast were added to the sterilized materials.
  - b. The motor of the airlock was turned on; the sterilely inoculated *R. puerariae* substrate was slowly delivered into the fermentation pot. When the materials were fed for fermentation,  $CO_2$  entered through the gas inlet and replaced the air inside the pot. The temperature was maintained at  $35 \pm 1^{\circ}C$  in the fermentation pot.
  - c. The gas was circulated by the CO<sub>2</sub> entering through the gas inlet at the bottom of the gas extraction pot, and ethanol was carried away from the substrate.
  - d. The mixed gas of CO<sub>2</sub> and ethanol passed into active carbon absorbing columns.



*Figure 6.8* The process of a technical system for cleanly coproducing fuel ethanol, flavones, and fibers by steam explosion and SSF of *R. puerariae*. (From Chen, H.Z. et al., Process for comprehensively utilizing steam exploded *radix puerariae* and device therefor. US Patents: US8333999 B2, 2007.)

- e. The adsorbed columns were heated to 90°C for recovery of ethanol.
- f. The ethanol was recovered and obtained after condensation.
- g. The *R. puerariae* residue was filter-pressed.
- 3. 95% ethanol was added to the residue for the extraction of isoflavones.
- 4. The extraction liquid was filtered, and the concentrated liquid was passed onto macroporous adsorbing resins for separation of flavonoids.

Fibrous residues were dried, followed by mechanical classification into a fiber portion or a protein portion with a ratchet.

## 6.7 Conclusions

Solid-state fermentation has been widely applied to the concoction and extraction of Chinese herbs due to the advantages of mild reaction condition, formation of new compounds, and multilevel utilization. Microorganisms, especially medicinal fungus, and modern SSF technologies should be introduced into the processing of Chinese herbs. Process engineering of Chinese herb SSF and its eco-industrial integration based on SSF and steam explosion were proposed, and will become one of the main directions of traditional Chinese herb development in the future.

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## chapter seven

# Microbial biofuels production

#### Anoop Singh and Poonam Singh Nigam

#### Contents

7.1	Introc	luction	. 155	
7.2 Microbial biofuels				
		Bioalcohols		
	7.2.2	Biodiesel	. 158	
		Biohydrogen		
	.3 Conclusion			

## 7.1 Introduction

Increasing energy demand due to the steep rise in industrialization and motorization leads to an increase in crude oil prices, which is directly affected by global economic activity (He et al. 2010). Today, fossil fuels take up 80% of the primary energy consumed in the world, of which 58% is consumed by the transport sector alone (Escobar et al. 2009). The sources for these fossil fuels are becoming exhausted; in addition, the consumption of fossil fuels to fulfill this energy demand has also been found to be a major contributor to greenhouse gas (GHG) emissions (Prasad et al. 2007a; Singh et al. 2010a). It has many negative effects including climate change, receding of glaciers, rise in sea levels, loss of biodiversity, and others, which led to moves toward alternative, renewable, sustainable, efficient, and costeffective energy sources with fewer emissions (Gullison et al. 2007; Prasad et al. 2010b; Singh et al. 2010b). Among many energy alternatives, biofuels are the most environmentfriendly energy source (Nigam and Singh 2011). A worrying statistic is that, the global oil and gas production is approaching its maximum production level and the world is now finding one new barrel of oil for every four it consumes (Aleklett and Campbell 2003). Hence, as an alternative to fossil fuels, biofuels have been portrayed as a future leading supplier of energy sources that have the ability to increase the security of supply, reduce the amount of vehicle emissions, and provide a steady income for farmers.

Biofuels are referred to gas, liquid, and solid fuels predominantly produced from biomass. A variety of fuels can be produced from biomass such as ethanol, methanol, biodiesel, Fischer–Tropsch diesel, hydrogen, and methane (Demirbas 2008). Renewable and carbonneutral biofuels are necessary for environmental and economic sustainability (Nigam and Singh 2011). Biofuels are important because they replace petroleum fuels. Between 1980 and 2005, worldwide production of biofuels increased by an order of magnitude from 4.4 to 50.1 billion liters with further dramatic increases in the future (Armbruster and Coyle 2006; Licht 2008; Murray 2005). Different countries have adopted different measures to introduce biofuels. The economics of each fuel vary with location, feedstock, and several other factors. Political agendas and environmental concerns also play a crucial role in the production and utilization of biofuels. There are a number of technologies that exist and several that are under development for the production of biofuels such as fermentation of sugar substrates, catalytic technology to convert ethanol to mixed hydrocarbon, hydrolysis of cellulose, biobutanol by fermentation, transesterification of natural oils and fats to biodiesel, hydrocracking of natural oils and fats, pyrolysis and gasification of various biological materials, and others. This chapter is an effort to summarize the microbial biofuel production system.

## 7.2 Microbial biofuels

Advancements in current research have shown that some microbial species such as yeasts, fungi, and microalgae can be used as potential sources of biomass for the production of biodiesel, which can be biosynthesized to store large amounts of fatty acids (Xiong et al. 2008). The microbial lipids, similar to vegetable oils, mainly contain palmitic, stearic, oleic, and linoleic acid with unsaturated fatty acids amounting to approximately 64% of the total fatty acid content (Nigam and Singh 2011; Singh et al. 2011a,b). The accumulation of lipid within microbial cells could be efficiently enhanced with the addition of various sugars to the pretreated molasses. The lipid content increased to more than 50% of the cell mass (Chen et al. 1992; Fakas et al. 2007).

Recent interest in the use of oleaginous microalgae as a biodiesel feedstock has grown considerably. This is mainly due to high oil yields from algal biomass (5000–100,000 L/ha per year), the ability to capture CO<sub>2</sub>, and the capability to grow algae on abandoned or unproductive land using brackish, salt, or wastewaters instead of freshwater (Levine et al. 2010). Some microalgae respond to certain chemical and physical stimuli through the accumulation of intracellular triglycerides (Hu et al. 2008). A biodiesel production process that obviates biomass drying and organic solvent use for oil extraction could lead to significant energy and cost savings (Singh and Olsen 2011). Attempts to combine extraction with acid-catalyzed transesterification in one step have been successful with dry algal biomass, but the reaction is severely inhibited by water (Ehimen et al. 2010; Johnson and Wen 2009).

Microbial oil can be produced from sulfuric acid-treated rice straw hydrolysate (SARSH) by cultivation of the microorganism *Trichosporon fermentans*. Fermentation of SARSH without detoxification gave a poor lipid yield of 0.17% w/v (1.7 g/L; Huang et al. 2009). They also worked on the improvement process to increase the yield and found that the detoxification pretreatment, including overliming, concentration, and adsorption by Amberlite XAD-4 improved the fermentability of SARSH significantly. The pretreatment process helped in increasing the lipid yield by removing the inhibitors in SARSH. Moreover, besides SARSH, *T. fermentans* was also found to be capable of metabolizing other sugars such as mannose, galactose, or cellobiose available in the hydrolysates of other natural lignocellulosic materials used as the single carbon source. Huang and coworkers concluded that the organisms studied could be used as a promising strain for microbial oil production. Levine et al. (2010) have developed a two-step, catalyst-free biodiesel production process involving intracellular lipid hydrolysis coupled with supercritical *in situ* transesterification.

Zhu et al. (2008) have conducted a study on the production of microbial biofuel from waste molasses and concluded that lipids produced in microbial biomass can be used for biodiesel production. In this study, they have optimized the growth medium components for culture cultivation and studied the effects of culture conditions on microbial biomass and lipid production by a microbial strain of *T. fermentans*. The optimal nitrogen source, carbon source, and C/N molar ratio for the best lipid yields were found to be peptone,

glucose, and 163, respectively. The most favorable initial pH of the cultivation medium and temperature were 6.5 and 25°C, respectively. *T. fermentans* could be cultivated in a medium consisting of waste molasses from the sugar industry. Zhu and coworkers recorded 12.8 g/L lipid yield from the bioconversion of waste molasses consisting of 15% total sugar concentration (w/v) at pH 6.0 by *T. fermentans*. The microbial oil with an acid value of 5.6 mg KOH/g could be effectively transesterified to produce biodiesel following a process of base catalysis after the removal of free fatty acids, and subsequently, a high methyl ester yield of 92% could be obtained (Zhu et al. 2008). The ability of yeast to grow well on pretreated lignocellulosic biomass could efficiently enhance the lipid accumulation, and provides a promising option for the production of economically and environmentally sound microbial oil from agricultural residues (Prasad et al. 2007a,b; Nigam and Singh 2011).

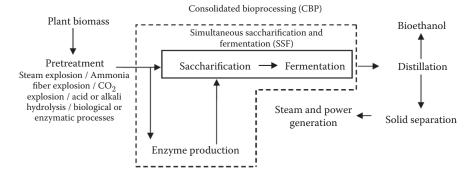
Multiple approaches are currently being researched for the use of microorganisms in the production of various biofuel, for example, bioalcohols, biohydrogen, biodiesel, and biomethane from multiple feedstocks (Elshahed 2010).

#### 7.2.1 Bioalcohols

Bioalcohols are produced mainly using two approaches, that is, direct fermentation and indirect fermentation. Direct fermentation involves the identification of starting plant material, isolation and development of bacterial and fungal strains, and design of appropriate protocols for efficient conversion of plant material to sugar monomers, and then sugars are converted to ethanol by yeasts or genetically engineered bacterial strains. Indirect fermentation involves pyrolysis of the feedstock, followed by the conversion of the produced gas to alcohol using acetogenic bacteria (Elshahed 2010; Klasson et al. 1992).

Different feedstocks are used for the production of bioalcohols (mainly ethanol). The composition of plant materials varies among plant species. In general, the major polymers within lignocellulosic biomass is cellulose (35%–50%), followed by hemicellulose (20%–35%), and lignin (10%–25%; Liu et al. 2008). Therefore, different kinds of microorganisms, enzymes, incubation conditions, and engineering schemes are required for efficient depolymerization (Elshahed 2010). Celluloses and hemicelluloses can be degraded anaerobically, but lignin requires pretreatment for better degradation. Pretreatment is also required to increase the surface area of cellulose and hemicellulose exposed for microbial or enzymatic degradation. Various pretreatment approaches are employed, such as the use of alkaline peroxidases, concentrated acids, dilute acids, alkali, alkali peroxidases, wet oxidation, steam explosion, ammonia fiber explosion, liquid hot water, or organic solvent treatments (Prasad et al. 2007a; Wyman 1994).

The general steps for producing ethanol include pretreatment of substrates, saccharification process to release the fermentable sugars from polysaccharides, fermentation of released sugars, and finally a distillation step to separate ethanol (Figure 7.1). Pretreatment is designed to facilitate the separation of cellulose, hemicellulose, and lignin, so that complex carbohydrate molecules constituting the cellulose and hemicellulose can be broken down by enzyme-catalyzed hydrolysis into their constituent simple sugars (Nigam and Singh 2011). The degradation of starch, cellulose, or hemicellulose of any plant material yields hexoses and pentoses that need to be fermented to ethanol. Several fermentation schemes, for example, mixed acid fermentation by enteric bacteria and hetereolactic acid fermentation by some lactic acid bacteria are known to produce ethanol. Two groups of microorganisms, yeast *Saccharomyces cerevisiae* and bacterium *Zymomonas mobilis*, naturally produce 2 mol of ethanol per mole of hexose during fermentation (Elshahed 2010).



*Figure 7.1* Various processes for the production of bioethanol from biomass. (From Nigam, P.S. and Singh, A., *Progress in Energy and Combustion Science* 37, 52–68, 2011.)

The use of *S. cerevisiae* for ethanol conversion from hexoses is one of the best perfected industrial processes. The capability of strains for efficient simultaneous uptake of multiple sugars can be improved through genetic manipulation of sugar transporters. This could direct laboratory evolution results in near-stoichiometric production of ethanol from glucose (Liu et al. 2008). The availability of the genome sequences of *S. cerevisiae* and the advancement of genetic systems allows for continuous genetic manipulations and strain improvements (Goffeau et al. 1996). Because hemicellulose is always present in plant material with cellulose, a microorganism capable of efficiently and simultaneously metabolizing both sugars is the best option to degrade the plant material.

The genetically modified strains efficiently degrade xylose and are shown to work well with pure substrates as well as sugars released by enzymatic treatment of plant materials (Ho et al. 1998; Kuyper et al. 2005; Prasad et al. 2007a; Sedlak and Ho 2004; Wisselink et al. 2009).

The indirect fermentation approach pyrolyzed (burned) the plant material to produce Syngas, which consists primarily of CO, CO<sub>2</sub>, and hydrogen, and converted this to ethanol by acetogenic bacteria (Elshahed 2010). Acetogens convert C1 compounds into C2 products, mainly acetate (Müller 2003). Acetogens are usually Gram-positive bacteria belonging to the class Clostridia within the phylum *Firmicutes* (Leadbetter et al. 1999). Several acetogenic *Clostridia* are very efficient for anaerobic fermentation when grown on hexose sugars and produce various products, for example, acetate, butyrate, and ethanol. All plant material or nonplant wastes that could be pyrolized could theoretically be used in such an approach (Gulati et al. 1996). The approach makes use of all plant components, including lignin, which is generally not utilized in direct fermentation approaches, and can use mixed plant flora within a batch (Tilman et al. 2006). The key technical difficulties include relatively low growth rates and low product concentration in the aqueous phase when compared with yeast fermentation (Tanner 2008).

#### 7.2.2 Biodiesel

Biodiesel could be produced from plant lipids, animal fats, and used cooking oils by esterification of triglycerides with methanol (Fukuda et al. 2001). Microbial species such as yeasts, fungi, and algae can be used as potential sources for the production of biofuels. They are rich in oil content and provide multiple advantages over the plant lipids because these microorganisms can grow extremely rapidly and oil content in microalgae can exceed 80% by weight of dry biomass (Table 7.1; Metting 1996; Singh et al. 2011c; Spolaore et al. 2006). Microalgae commonly double their biomass within 24 h, and biomass doubling times during exponential growth are commonly as short as 3.5 h (Elshahed 2010).

Microalgae can produce lipids, proteins, and carbohydrates in large amounts over short periods of time and these products can be used for the production of biofuels. Microalgae have the ability to take up  $CO_2$  from the atmosphere as well as discharge gases and soluble carbonates. Microalgae can tolerate and use substantially higher levels of  $CO_2$ (up to 150,000 parts per million by volume) compared with higher plants (Brown 1996; Wang et al. 2008). Algal cells are veritable miniature biochemical factories, and appear more photosynthetically efficient than terrestrial plants due to their very efficient  $CO_2$  fixation. The ability of algae to fix  $CO_2$  has been proposed as a method of removing  $CO_2$  from flue gases from power plants, and thus can be used to reduce the emission of greenhouse gases simultaneous with biodiesel production (Singh et al. 2011c, 2012; Singh and Olsen 2011).

Microalgae that are able to survive heterotrophic and exogenous carbon sources offer prefabricated chemical energy, which the cells often store as lipid droplets (Ratledge 2004). Naturally, lipid accumulation increases under certain conditions. Thus, it is important to keep in mind those factors that lead to the accumulation of lipids in algae, such as nutrients, sunlight, and others (Schenk et al. 2008).

Microorganism	Oil content (% dry weight)
Mic	roalgae
Botryococcus braunii	25–75
<i>Cylindrotheca</i> sp.	16–37
Scenedesmus dimorphus	16–40
Chlorella emersonii	28–32
Schizochytrium sp.	50-77
Nitzschia sp.	45-47
Nannochloropsis sp.	31–68
Neochloris oleoabundans	35–54
Y	easts
Candida curvata	58
Cryptococcus albidus	65
Lipomyces starkeyi	64
Rhodotorula glutinis	72
Ba	cteria
Arthrobacter sp.	>40
Bacillus alcalophilus	18–24
Rhodococcus opacus	24–25
Acinetobacter calcoaceticus	27–38
F	ungi
Aspergillus oryzae	57
Mortierella isabellina	86
Humicola lanuginosa	75
Mortierella vinacea	66

Table 7.1 Oil Content of Microorganisms

Source: Meng, J. et al., Renewable Energy 34: 1–5, 2009; Singh, A. et al., Bioresource Technology 102: 10–16, 2011; Singh, A. et al., Bioresource Technology 102: 26–34, 2011.

The best performing microalgal strains can be obtained by screening a wide range of naturally available isolates, and the efficiency of those can be improved by selection, adaptation, and genetic engineering (Singh et al. 2011a). Liu and Saha (2008) showed that high iron concentrations could induce considerable lipid accumulation in the marine strain *Chlorella vulgaris*, and also suggested that some metabolic pathways related to the lipid accumulation in *C. vulgaris* are probably modified by high levels of iron concentration in the initial medium. Genetic engineering of key enzymes in specific fatty acid production pathways within lipid biosynthesis is a promising target for the improvement of both quantity and quality of lipids. The lipid quality is very important for biodiesel production, as the alkyl ester content dictates the stability and performance of the fuel, and this is also an important factor in meeting international fuel standards (Singh et al. 2011a).

Three distinct algae production mechanisms, photoautotrophic, heterotrophic, and mixotrophic can be used, all of which follow natural growth processes (Singh et al. 2011c). Nitrogen starvation is the most effective method for improving microalgae lipid accumulation. Also, it gradually changes the lipid composition from free fatty acids to triacylglycerol, which a more useful product for biodiesel production. The biodiesel production from algal biomass involves chemical reactions such as transesterification of the extracted oil. Manipulation of metabolic pathways can redirect cellular function toward the synthesis of specific products and even expand the processing capabilities of microalgae (Meng et al. 2009; Singh et al. 2011c).

Singh et al. (2011b) concluded, in a review on mechanism and challenges in commercialization of algal biofuels, that the integration of microalgae cultivation with fish farms, food processing facilities, wastewater treatment plants, and others, will offer the possibility for waste remediation through recycling of organic matter and at the same time provide the low-cost nutrient supply required for algal biomass cultivation.

#### 7.2.3 Biohydrogen

Hydrogen gas is seen as a strong candidate for future energy carrier by virtue of the fact that it is renewable, does not evolve the greenhouse gas  $CO_2$  in combustion, it liberates large amounts of energy per unit of weight in combustion, and it is easily converted to electricity by fuel cell (www.oilgae.com/algae/pro/hyd/hyd.html, 2012). Thus, biohydrogen can be considered as the cleanest biofuel because it is oxidized to water with no gas emission in the production and consumption process.

Progress in the late 1990s contributed to a breakthrough in terms of sustainable hydrogen production. However, the problem is with the way it was produced (Rathore and Singh 2013). Although hydrogen biogas can be efficiently produced at the laboratory level, there is no known commercially operating hydrogen from biomass production facility in the world today (Zhu and Beland 2006). Hydrogen has long been known to be produced as a final end product of fermentation, or as a side product in photosynthesis in multiple groups of microorganisms (Elshahed 2010).

Nature has created biological reactions for hydrogen production. Sunlight is used for the oxidation of water by oxygenic photosynthesis reactions, and electrons are used for the generation of  $H_2$  by hydrogenase enzymes (Rathore and Singh 2013). Oxygen adversely affects hydrogenase function and acts as a suppressor of hydrogenase gene expression (Florin et al. 2001; Ghirardi et al. 2000; Happe and Kaminski 2002). Hans Gaffron discovered hydrogen metabolism in unicellular green algae (Gaffron 1940). The first scientific investigation of hydrogen evolution by microalgae was reported by Gaffron and Rubin (1942). They demonstrated that after a period of dark anaerobic "adaptation," the green alga *Scenedesmus obliquus* produces hydrogen in the dark at low rates, and hydrogen production is greatly stimulated in the light, although only for relatively brief periods. Other noteworthy observations were that uncouplers and low  $CO_2$  concentrations stimulated light-driven hydrogen production in green microalgae. Gaffron observed that under anaerobic conditions, the green alga *S. obliquus* can use H<sub>2</sub> as an electron donor in the  $CO_2$  fixation process in the dark, and evolve H<sub>2</sub> in the light (Gaffron 1944; Gaffron and Rubin 1942).

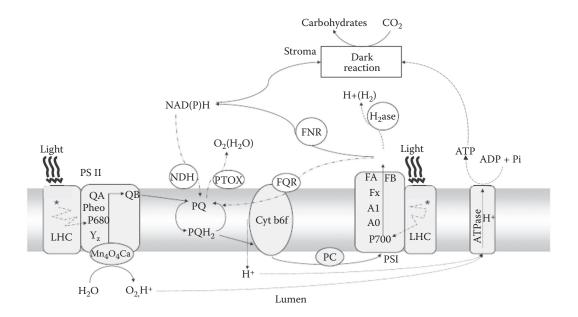
Algae structures are primarily for energy conversion without any development beyond cells. Their simple structure developments allow them to adapt to prevailing environmental conditions and prosper in the long term (Rathore and Singh 2013). Interest in green algae emanates from the fact that, in principle, they can use the highly efficient process of photosynthesis to produce hydrogen from sunlight and water as the most abundant of natural resources (Melis and Happe 2004).

Hydrogen has the largest energy content per weight of any known fuel, and can be produced by various means (Hallenbeck and Benemann 2002; Levin et al. 2004). Three main processes are commonly used for the production of biohydrogen. The most direct approach involves using photosynthetic microorganisms, for example, *Cyanobacteria* and green algae for biohydrogen production (Elshahed 2010). These microorganisms have the ability to split water, that is, produce electrons and oxygen from one molecule of water using sunlight as an energy source. The electrons are used for energy production through the electron transport chain as well as biomass production and sugar production using anabolic reactions, and they could also be converted to hydrogen by the action of hydrogenase enzymes (Elshahed 2010; Prince and Kheshgi 2005). A major problem with this approach is the extreme oxygen sensitivity of hydrogenases involved in hydrogen production. Therefore, the two processes, photolysis and hydrogen production, need to be temporarily uncoupled (Elshahed 2010).

The second approach uses nitrogenase enzymes in anoxygenic photoheterotrophic microorganisms for hydrogen production. The function of nitrogenase is to fix atmospheric  $N_2$  gas to ammonia to be incorporated into the cell's biomass. Thus, enabling nitrogenfixing microorganisms to grow in the absence of organic or inorganic nitrogen sources in growth media could lead to hydrogen production. However, nitrogenase enzymes are also capable of producing hydrogen from electrons and protons in the absence of oxygen and in the presence of light (Elshahed 2010).

The third approach is the production of hydrogen using fermentative bacteria from organic substrates, for example, sugar and lingocellulosic biomass, as well as industrial, residential, and farming wastes for anaerobic fermentation. Several groups of microorganisms are known to produce hydrogen as an end product of fermentation, for example, *Escherichia coli*, Enterobacteraerogenes, and *Clostridium butyricum* (Elshahed 2010).

The main hydrogen production pathways include electrolysis of water and thermocatalytic reformation of hydrogen-rich compounds, and usually require high-energy inputs obtained from nonrenewable resources (Levin et al. 2004). Biological production of hydrogen solves this problem by using microorganisms to convert biomass or solar energy into hydrogen gas (Das and Veziroglu 2001; Hallenbeck and Benemann 2002). Although Gaffron and Rubin (1942) considered both water and carbohydrates as the source of electrons for  $H_2$ evolution, they favored the latter as the main source. Spruit (1958) found evidence supporting water as the main electron donor in such a reaction. A schematic representation of the photosynthetic reactions in the chloroplast thylakoid membrane of green algae is demonstrated in Figure 7.2. In photosynthesis, excitation energy from the sun is transferred from antenna pigments of light-harvesting complexes of photosystem II to photoactive chlorophyll molecule (P680) of the reaction center. As a result, strongly oxidizing cation radical



*Figure 7.2* A scheme of the photosynthetic reactions in the chloroplast thylakoid membrane of green algae. Dash–dot line shows excitation migration from the antenna to the reaction center. The major electron transfer pathways are indicated by solid lines. The pathways involved in chlororespiration, cyclic electron flow around PS I, and the Mehler reaction are shown by dash–dot–dot lines. The dashed line designates the oxygen-sensitive electron transport route induced under anaerobic conditions. Reactions coupled to the generation of proton gradient and to the ATP synthesis are depicted by dotted lines. (From Antal, T.K. et al., *Applied Microbiology and Biotechnology* 89:3–15, 2011.)

P680<sup>+</sup> is formed, which catalyzes the oxidation of water in the oxygen-evolving complex of PS II through a series of redox-active components including the tyrozine Z residue (YZ), and the Mn<sub>4</sub>O<sub>4</sub>Ca cluster located at the luminal side of the thylakoid membrane (Barber 2008; Goussias et al. 2002; Kern and Renger 2007). Electrons released from P680<sup>+</sup> were accepted by a negatively charged radical of the Mg-free chlorophyll pigment pheophytin (Pheo). The reduced Pheo quickly passes an extra electron to the PS II primary quinone molecule (QA). The QA is tightly bound to PS II and acts as a single-electron carrier. From QA, an electron is transferred to the secondary quinone molecule (QB), a two-electron and two-proton acceptor. In double reduced and protonated form, QBH2 is loosely bound to PS II and thus it can be exchanged with the oxidized QB from the plastoquinone (PQ) pool. After that, an electron is transported further to PS I through the PQ pool, cytochrome b6f complex (cyt b6f), and plastocyanin (PC). Subsequent to light excitation, PS I passes an electron to the soluble protein ferredoxin (Fd) through a series of electron carriers including chlorophylls P700 and A0, quinone A1, and iron-sulfur clusters FX, FA, and FB. In the stroma, an electron is transferred from the reduced Fd to the NADP<sup>+</sup> with the formation of NADPH by a process catalyzed by the ferredoxin-NADP-reductase (FNR). In chloroplast, NADPH is used as a reducing power for fixation of carbon dioxide in the dark reaction (Bassham et al. 1950). This electron pathway corresponds to the linear electron transport (Antal et al. 2011). However, electrons can be alternatively redirected from the reduced Fd backward to the PQ pool through the putative ferredoxin-quinone-reductase (FQR) in the so-called cyclic electron transport (CET) around PS I (Moss and Bendall 1984). Such CET can be induced when FNR becomes inactive or when carbon fixation proceeds at a slow

rate, which takes place in dark-adapted plants or under different environmental stresses (Breyton et al. 2006; Golding and Johnson 2003).

Production of  $H_2$  from algae is mainly based on the photolysis of water molecules in the process of light reaction during photosynthesis. Hydrogen production is a property of many phototrophic organisms (Appel and Schulz 1998; Asada and Miyake 1999; Boichenko et al. 2001; Weaver et al. 1980), and the list of  $H_2$  producers includes several hundred species from different genera of both prokaryotes and eukaryotes (Boichenko and Hoffmann 1994). Oxygenic photosynthetic organisms, such as plants, algae (green, red, brown, and yellow), and cyanobacteria use water as a source of electrons and protons in photosynthesis. Among these organisms, only green microalgae and cyanobacteria have been shown to sustain hydrogen production (Hall et al. 1995; Melis et al. 2000; Sakurai et al. 2004). The enzyme mediating  $H_2$  production in green algae is the reversible (or bidirectional) hydrogenase that catalyzes the following ferredoxin (Fd)-linked reaction in the absence of ATP input (Boichenko and Hoffman 1994):

$$2H^+ + 2e^- \leftarrow^{Hydrogenase} \rightarrow H_2$$

The available  $H_2$  energy production processes from algal biomass can be divided into two general categories: thermochemical and biological processes. Combustion, pyrolysis, liquefaction, and gasification are the four thermochemical processes. Direct biophotolysis, indirect biophotolysis, biological water–gas shift reaction, photofermentation, and dark fermentation are the five biological processes (Ni et al. 2006).

Hydrogen was a vital energy source for organisms during the early stages of our planet, but under reducing atmospheric conditions, that is, in the absence of a substantial amount of oxygen (Rathore and Singh 2013). This process gradually lost its importance with the development of the photosynthetic machinery that was able to exploit light energy more efficiently. Hydrogen-dependent processes therefore lost their central role as a necessity for the survival of most cells. In consequence, there was no strong evolutionary pressure for the design of oxygen-resistant hydrogenases and the increasing oxygen content of the atmosphere produced by the water-splitting process led to these enzymes being switched off (Horner et al. 2002).

Nowadays, hydrogenases still exist in bacteria and microalgae. But their genes are mostly activated under anaerobic conditions and their main function is twofold, either to provide an alternative electron source to aid survival under suboptimal conditions or to capture electrons as a kind of security valve to prevent dangerous overreduction of the electron transport chain (Appel and Schulz 1998). Among the three principally different types of hydrogenases in nature—[Fe only]-type, [NiFe]-type, and [Fe-S cluster-free]-type hydrogenase—the [Fe only]-type are usually the simplest hydrogenases (only one subunit in the case of *Chlamydomonas*) and the most active hydrogenases known (up to 2,000  $H_2/s$ ) present in green algae (Girbal et al. 2005; Happe et al. 2002; Lyon et al. 2004).

Recent advances with respect to the identification of genes involved in maturation and regulation of [FeFe]  $H_2$ ase (Girbal et al. 2005; McGlynn et al. 2008; Posewitz et al. 2004) and [NiFe]  $H_2$ ase (Ludwig et al. 2009; Schubert et al. 2007). The development of heterologous expression systems for both classes of enzyme (Girbal et al. 2005; King et al. 2006; Lenz et al. 2005; Posewitz et al. 2004; Sybirna et al. 2008) make it feasible to biochemically characterize  $H_2$  ases by heterologous expression in organisms that do not possess endogenous  $H_2$  ase machinery. Recently, there has been considerable progress in identifying relevant bioenergy genes and pathways in microalgae, and powerful genetic techniques have been

developed to engineer some strains through the targeted disruption of endogenous genes or transgene expression (Rathore and Singh 2013). Some studies indicated that mutagenesis can be used to decrease the  $O_2$  sensitivity of the hydrogenase and thus eventually lead to a system that produces  $H_2$  under aerobic conditions (Flynn et al. 1999, 2002; Ghirardi et al. 1997; McTavish et al. 1995; Seibert et al. 2001). One of the most significant advances in algal genetics is the development of improved gene silencing strategies in *Chlamydomonas reinhardtii*. High-throughput artificial miRNA (amiRNA) techniques for gene knockdown, which are highly specific and stable, were recently reported (Molnar et al. 2009; Zhao et al. 2009).

## 7.3 Conclusion

The primary focus for the production of microbial biofuel depends on the feasibility of the microbial strains. Various microbial strains were found to be very rich in lipid content. The microorganisms can be used for the production of different biofuels such as bioethanol, biodiesel, and biohydrogen. The whole biomass of microalgae can be converted to biofuels with the biorefinery concept. Metabolic and genetic engineering can play a major role to make these production technologies more economical and sustainable.

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# chapter eight

# Microbial production of organic acids

# Sarafadeen Olateju Kareem and Temitope Banjo

# Contents

8.1	Intro	luction	170
8.2	Orgai	nic acids: An overview	170
	8.2.1	Acetic acid	170
	8.2.2	Ascorbic acid	170
	8.2.3	Citric acid	171
	8.2.4	Fumaric acid	171
	8.2.5	Gluconic acid	172
	8.2.6	Lactic acid	172
	8.2.7	Malic acid	172
	8.2.8	Pyruvic acid	173
	8.2.9	Succinic acid	
8.3	Micro	organisms involved in organic acid production	174
8.4	Upstr	eam processes	175
	8.4.1	Selection of microbial strains	175
	8.4.2	Immobilization	175
	8.4.3	Production medium and physicochemical conditions	176
		8.4.3.1 Moisture content	176
		8.4.3.2 Effect of temperature	176
		8.4.3.3 Control of pH	176
		8.4.3.4 Aeration and agitation	177
8.5	Ferme	entation processes	177
	8.5.1	Surface fermentation	177
	8.5.2	Submerged fermentation	177
	8.5.3	Solid-state fermentation	178
8.6	Dowr	stream processes	178
8.7	Prosp	ects of microbial fermentation of organic acids	179
8.8	Case	study: Comparative production of citric acid by immobilized A. niger	180
	8.8.1	Materials and methods	
		8.8.1.1 Immobilization of A. niger	180
		8.8.1.2 Citric acid production	180
		8.8.1.3 Citric acid determination	
		8.8.1.4 Biomass, residual sugars, and pH determination	181
	8.8.2	Results and discussion	
8.9	Concl	usion	184
Refe	rences		184

## 8.1 Introduction

Organic acids are commercially valuable products obtained through microbial fermentation. Generally, microbial fermentation is an aspect of microbial biotechnology that involves the use of microorganisms to metabolize various substrates to produce a particular metabolite. Microbial fermentation was dated to 6000 BC when the art was being used to produce food and beverages. In 1850, Louis Pasteur provided evidence of microbial involvement in beer fermentation when he discovered that yeasts were fermenting sugar into ethanol. He later turned his attention toward vinegar production and other acids of microbial origin.

Many organic acids possess a long chain of carbons attached to a carboxyl group and can be produced through microbial bioprocesses with the exception of a few that are produced commercially. Fermentation processes play a major role in the production of most organic acids such as citric, gluconic, itaconic, lactic, fumaric, and malic acids. Most organic acids are produced as intermediate metabolites of the tricarboxylic acid (TCA) cycle whereas other acids can be derived indirectly from the Krebs cycle such as itaconic acid, or directly from glucose. Acetic and lactic acids are formed as end products from pyruvate or ethanol. Many bacteria and fungi have the potential to produce a variety of organic acids with high yields. However, fungi have an intrinsic ability to accumulate many organic acids because of their ability to thrive at acidic pH from 2 to 5 (Goldberg et al. 2006).

Bioconversion of carbon substrate especially glucose is exploited in large-scale commercial production of a number of organic acids such as citric, gluconic, and itaconic acids. Other acids produced on a smaller scale are lactic acid, malic acid, and kojic acids. This chapter contains a review of the microbial production of organic acids, their biochemistry, upstream processes, fermentation processes, and downstream processes.

# 8.2 Organic acids: An overview

### 8.2.1 Acetic acid

Acetic acid is also known as ethanoic acid. It is a colorless organic compound that, in its undiluted form, is also called glacial acetic acid. Acetic acid is the main component of vinegar (4%–8% acetic acid). Vinegar is s a weak acid with a distinctive sour taste and pungent smell. It is often produced by fermentation and subsequent oxidation of ethanol. It is an important chemical reagent and industrial chemical, mainly used in the production of cellulose acetate for photographic film, and polyvinyl acetate for wood glue as well as synthetic fibers and fabrics. In the food industry, acetic acid is used as a condiment.

Acetic acid is used in the production of vinyl acetate polymers, which are components in paints and adhesives. Vinegar is typically 4% to 18% acetic acid by mass. Vinegar is used directly as a condiment, and in the pickling of vegetables and other foods (Bala 2003).

### 8.2.2 Ascorbic acid

Ascorbic acid or vitamin C is a weak sugar acid, white, soluble, and structurally related to glucose. In biological systems, it can be found only at low pH (acidic), but in neutral solutions above pH 5, it is predominantly found in ionized form (ascorbate). Ascorbate is easily and reversibly oxidized to dehydro-L-ascorbic acid, forming the ascorbyl radical as an intermediate (Schmidt and Moser 1985).

Ascorbic acid enhances iron absorption by forming soluble ferrous–ascorbate complexes that are readily absorbed in the cell. Ascorbic acid is involved in the development of connective tissue cells. It modulates neurotransmitter systems in the brain and stimulates

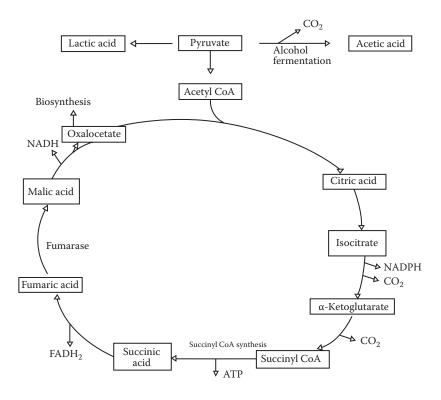


Figure 8.1 Biosynthesis of organic acids through the pyruvate and TCA cycle.

collagen synthesis, which is a requirement for muscle and bone formation (Ball 2004; Hallberg et al. 1996).

### 8.2.3 Citric acid

Citric acid is a TCA (2-hydroxy-1,2,3-propane tricarboxylic acid), is soluble in water, and has a pleasant taste. It is ubiquitous in nature and exists as an intermediate in the TCA cycle. It is naturally found in fruits such as lemons, oranges, pineapples, and pears as well as in the seeds of different vegetables, and in animal bone, muscle, and blood. The conversion of glucose or sucrose to citric acid by *Aspergillus niger* involves both the glycolytic pathway and the TCA cycle (Figure 8.1). Citric acid synthesis by *A. niger* from sucrose or sugar substrate has been attributed to the presence of the following enzymes; phosphofructokinase, pyruvate kinase, pyruvate decarboxylase, citrate synthase, aconitase and isocitrate dehydrogenase (Kapoor et al. 1983). Citric acid is the most important organic acid used in food industries as food preservatives and anticoagulant. The free acid is employed in pharmaceuticals as an acidulant and to enhance the flavor of syrups, solutions, and elixirs (Ali 2004; Sauer et al. 2007).

### 8.2.4 Fumaric acid

Fumaric acid is a naturally occurring organic acid that was first isolated from the plant *Fumaria officinalis*. Fumaric acid is also known as (*E*)-2-butenedioic acid or *trans*-1,2-ethylenedicarboxylic acid. Many microorganisms produce fumaric acid in small amounts

because it is a key intermediate in the citrate cycle. Fumaric acid is primarily an intermediate of the TCA cycle (Figure 8.1), but it is also involved in other metabolic pathways. Fumaric acid is used as food and in feed additives. It is also used in the production of polyester resins (Kautola and Linko 1989; Kenealy et al. 1986; Sauer et al. 2007).

### 8.2.5 Gluconic acid

Gluconic acid or pentahydroxycarboxylic acid is a noncorrosive, nonvolatile, nontoxic, and mild organic compound. The annual worldwide production capacity for gluconic acid and its derivatives is estimated to be 60,000 tonnes. The bulk of production (85%) is in the form of sodium gluconate and other alkali gluconate salts. In the oxidation of glucose to gluconic acid, beta-D glucose is oxidized to *d*-D-gluconolactone in the presence of oxy-gen. The gluconolactone is subsequently hydrolyzed nonenzymatically to D-gluconic acid. Gluconic acid and its salts are prepared exclusively by the oxidation of glucose or glucose-containing raw material through a simple dehydrogenation reaction catalyzed by glucose oxidase. Glucose is generally used as a carbon source for the microbial production of gluconic acid (Sankpal and Kulkarni 2002).

Gluconic acid is used in the chemical, pharmaceutical, food, beverage, textile, and other industries. In the food industry, it is used as a food additive and acidity regulator. It also serves as a cement additive in the construction industry because it enhances cement's resistance and stability under extreme climatic conditions (Ramachandran et al. 2006).

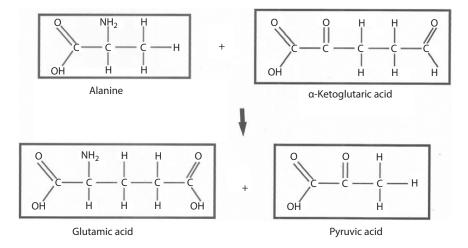
### 8.2.6 Lactic acid

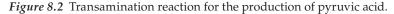
Lactic acid is a carboxylic acid with an adjacent hydroxyl group. It was first isolated in 1780 by the Swedish chemist Carl Scheele (VickRoy 2000). Lactic acid is miscible with water or ethanol, and is hygroscopic. Lactic acid is chiral and has two optical isomers. One is known as L-(+)-lactic acid or (S)-lactic acid and the other, its mirror image, is D-(–)-lactic acid or (*R*)-lactic acid (Tay and Yang 2002). Bacterial species of *Lactobacillus* are major producers of lactic acid and are able to hydrolyze cheap raw materials such as starchy and cellulosic materials into fermentable sugars before lactic acid production. This hydrolysis can be carried out simultaneously with fermentation by amylase-producing *Lactobacillus amylophilus* and *Lactobacillus amylovorus* (Hofvendahl and Hahn-Hägerdal 2000).

Lactic acid has many potential applications in the food, cosmetic, pharmaceutical, and chemical industries. It is used in food to enhance flavor, improve microbial quality, and regulate pH. Moreover, lactic acid is used commercially in processed meats as a preservative and as an acidulant in salads, dressings, and beverages. Other uses of lactic acid include cosmetic applications as moisturizers and pH regulators (Hofvendahl and Hahn-Hägerdal 2000).

### 8.2.7 Malic acid

Malic acid is a dicarboxylic acid. It was first isolated from apple juice by Carl Wilhelm Scheele in 1785. The main pathway for malic acid synthesis is the condensation of oxaloacetate and acetyl-coenzyme A (acetyl-CoA) to citric acid, followed by its oxidation to malate through the TCA cycle (Figure 8.1). If acetyl-CoA is generated by pyruvate dehydrogenase and oxaloacetate by pyruvate carboxylase, the conversion of glucose to malate through this oxidative pathway results in the release of two molecules of CO<sub>2</sub>. Malic acid is used in the treatment of the symptoms of fibromyalgia (Sauer et al. 2007).





### 8.2.8 Pyruvic acid

Pyruvic acid is an organic acid, a ketone as well as the simplest of the  $\alpha$ -keto acids. It is the end product of the anaerobic portion of glycolysis and is a central metabolite to several metabolic pathways. In aerobic respiration, pyruvate is converted to acetyl-CoA and carbon dioxide by the enzyme pyruvate carboxylase. Under limited amounts of oxygen, pyruvate is converted to lactic acid. It can further be converted to carbohydrates (gluconeogenesis), fatty acids, alanine, and ethanol.

Because glycolysis is the primary metabolic pathway used by microorganisms to generate energy, glucose is a widely used substrate for pyruvic acid synthesis. Prominent pathways for the production of pyruvic acid include glycolysis and transamination processes (Figure 8.2; Bao et al. 2011).

Glycolysis is the anaerobic catabolism of glucose to pyruvate. In this process, glucose passes through a series of metabolic reactions to generate two molecules of pyruvic acid. Transamination involves the transfer of an amine group from one molecule to another catalyzed by transaminases. Amino acids, which can be converted after several steps through transamination into pyruvic acid, include alanine, serine, cysteine, and glycine. In addition, pyruvic acid is also a substrate for the enzymatic production of amino acids such as L-tryptophan, L-tyrosine, D-/L-alanine, and L-dihydroxyphenylalanine through the transamination process.

Pyruvate has also been used in the production of crop protection agents, polymers, cosmetics, and food additives. Calcium pyruvate is useful in the food industry as a fat burner because it accelerates the metabolism of fatty acids in the human body. Clinical studies have demonstrated that pyruvate promotes fat and weight loss significantly, improves exercise endurance capacity, and reduces cholesterol levels effectively (Li et al. 2001; Stanko et al. 1994).

#### 8.2.9 Succinic acid

Succinic acid is another name for amber acid, which has been used in Europe as a natural antibiotic and general curative for centuries. Succinic acid is also a natural constituent of plant and animal tissues. Succinic acid is produced as an intermediate metabolite in the TCA cycle (Figure 8.1). Five key enzymes have been reported for succinic acid production.

These include phosphoenolpyruvate carboxykinase, malate dehydrogenase, malic enzyme, fumarase, and fumarate reductase. The higher CO<sub>2</sub> level resulted in an increased succinic acid production at the expense of ethanol and formic acid.

Succinic acid is known as a powerful antioxidant and has also been shown to stimulate neural system recovery and boost the immune system. In food and beverages, it is used as a flavoring agent and preservative. In pharmaceuticals, it is used in the preparation of active calcium succinate, which acts as an anticarcinogenic agent and as a deodorant for the removal of fishy odors (Sauer et al. 2007).

## 8.3 Microorganisms involved in organic acid production

Commercial production of many organic acids is usually achieved through fermentation processes. Citric, gluconic, itaconic, and lactic acids are manufactured through large-scale bioprocesses, whereas oxalic, fumaric, and malic acids have been produced in relatively small-scale bioprocesses (Table 8.1). A few other organic acids have been explored for the development of novel processes. The largest commercial quantities of fungal organic acids are those of citric acid and gluconic acid, both of which are prepared through the fermentation of glucose or sucrose by *A. niger*.

Many species of fungi have been evaluated for the production of organic acid (Kareem et al. 2010; Magnuson and Lasure 2004). The selection of a suitable fungus strain is critical

Organic acid	Chemical formula	Substrate	Microorganisms	References
Acetic	$C_2H_4O_2$	Fermented grains, malt, rice, potato mash, sorghum, apple cider	Acetobacter sp., G. oxydans, C. lentocellum	Sauer et al. 2007
Ascorbic	$C_6H_8O_6$	Glucose	Aspergillus flavus, A. tarmani, Acetobacter sp.	Ball 2004; Sauer et al. 2007
Citric	$C_6H_8O_7$	Glucose, sucrose, cassava whey, dairy whey, pineapple waste	A. niger, A. flavus, Candida spp., Saccharomyces spp.	Kareem et al. 2010
Fumaric	$C_4H_4O_4$	Glucose, sucrose, xylose, potato starch	<i>Rhizopus</i> sp., <i>Mucor</i> sp.	Sauer et al. 2007
Gluconic	$C_{6}H_{12}O_{7}$	Glucose, corn starch, cane molasses	A. niger	Roukas 2005
Lactic	$C_3H_6O_3$	Starch, glucose	Lactobacillus sp., L. lactis, Enterobacter faecalis, Rhizopus arrhizus	Wee et al. 2006
Malic	$C_4H_6O_5$	Glucose	A. flavus, Saccharomyces cerevisae, Zygosaccharomyces rouxii	Sauer et al. 2007
Pyruvic	$C_3H_4O_3$	Glucose	Proteus, Salmonella, Escherichia coli, Erwinia, S. cerevisae	Bao et al. 2011; Prescott et al. 2005
Succinic	$C_4H_6O_4$	Dairy whey, wood hydrolysate, glucose, lactose, arabinose, maltose, sorbitol	A. niger, A. fumigatus, Corynebacterium glutamicum, E. faecalis, A. succinogenes	Sauer et al. 2007

Table 8.1 Organic Acids at a Glance

to the production of citric acid. Strains with high sporulation, good growth in the substrates, short fermentation times, and high organic acid yields are used in industrial scale operations. Such strains are also expected to have long stability, and resistance to other microorganisms.

A number of microorganisms are involved in the production of acetic acid, which have different pathways or syntheses of production. These microorganisms include *Acetobacter* sp., *Gluconobacter oxydans*, and *Clostridium lentocellum*. Microorganisms that can produce lactic acid can be divided into two groups, bacteria and fungi. Prominent among bacterial species are *Lactobacillus delbrueckii* and *Enterococcus faecalis* (Sauer et al. 2007).

# 8.4 Upstream processes

### 8.4.1 Selection of microbial strains

Optimum production of organic acids in a fermentation medium is strongly dependent on the type, nature, and strain of microorganisms. Strain selection is a critical and primary stage of industrial microbiology. The search for a microbial strain with high commercial yields can be obtained from the natural environment or through strain improvement techniques. Strains can be selected after subjecting the genetic material to physical or chemical mutagenic agents or any other genetic manipulation. Mutant strains are able to utilize cheap substrates with high organic acid yield under reduced fermentation periods. They are also stable and resistant to unfavorable conditions. The size of the inocula is also a critical factor in organic acid yield because excess inoculum concentration leads to high biomass concentration and lower organic acid yield, whereas low amounts of inoculum lead to longer fermentation times (Yigitolu 1992).

### 8.4.2 Immobilization

Immobilization is a technique in which the biomass is immobilized onto the support. It enables repetitive use of the high biomass to carry out biochemical reactions rapidly leading to process economy and stability. Immobilization seems to be an attractive method for accomplishing high cell densities to achieve rapid carbohydrate conversion to organic acids. Immobilization can be carried out in the following ways:

- i. *Adsorption*: this process involves the binding of microbial cells directly to waterinsoluble carriers such as polysaccharide derivatives, synthetic polymers, and glass. It involves electrostatic interactions between the charged support and the charged cell. The major drawback of this method is cell leakage from the matrix.
- ii. *Entrapment*: this involves direct entrapment of microbial cells into polymer matrices such as agar, gelatin, calcium alginate, and others. This process is associated with a diffusion limitation in the transfer of either substrate or product through the matrix.
- iii. Covalent linkage: this technique involves the linkage of microbial cells or biocatalysts to the carrier through reactive groups on the matrix. The coupling reagents such as glutaraldehyde, diisocyanate, and others, are used to introduce a space group on the carrier surface which subsequently interacts with reactive groups on the cell surface. The use of this technique is limited by partial loss in activity.
- iv. *Microencapsulation*: it involves the immobilization of cells or enzymes within semipermeable microcapsules. These capsules are prepared from organic polymers such as cellulose, polystyrene, polyamine glycol, and others. This is an easy and economic technique; however, cell leakages also occur through microcapsules.

Generally the success of an immobilization process is determined by the type of support, cell retention, stabilization of cells or mycelia, and the amount of biomass. Immobilized cells are preferably used in the bioprocessing of organic acids because of cell stability, constant decrease in medium viscosity, enhanced nutrients and oxygen transfer (Anwar et al. 2009; Kareem et al. 2013; Najafpour 2007).

## 8.4.3 Production medium and physicochemical conditions

A suitable production medium for organic acid production must provide adequate carbon and nitrogen sources. Generally, microbial strains grow well in medium containing carbohydrates ranging from simple sugars to complex sugars. Simple sugars such as glucose, fructose, and maltose are efficiently utilized because organic acids are synthesized through the glycolytic pathway and the TCA cycle. However, complex carbohydrate substrates from agrowastes have also been exploited for organic acid biosynthesis.

Nitrogen is also another factor in the biosynthesis of organic acids because of its role in cell metabolism. Commonly used nitrogen sources are ammonium sulfate or ammonium nitrate (0.25%–0.5%). Adequate balance in C/N ratio is very critical to optimum biosynthesis of organic acids by microorganisms. Many divalent metals such as zinc, manganese, iron, copper, and magnesium at low concentrations (0.02%–0.025%) are known to play significant roles in the biosynthesis of many organic acids (Kareem et al. 2010).

### 8.4.3.1 Moisture content

Another major factor of great influence on the production of organic acid is moisture. Adequate moisture is required for the production of biomass and desired metabolites because it provides nutrients in soluble form for microorganisms. An insufficient moisture level gives a lower degree of swelling and higher water tension, and then reduces the solubility of the nutrients. Whereas excessive moisture levels decrease porosity, changes particle structure, promotes the development of stickiness, reduces gas volume and exchange, and decreases diffusion, which results in lower oxygen transfer.

### 8.4.3.2 Effect of temperature

Temperature is an important factor in the fermentation process (because microbial growth is confined to a narrow range of optimal temperatures). High organic acid yields are obtainable at an optimum growth temperature. Lower or higher temperature values to the optimum always exhibit signs of adverse growth and metabolic production. For instance, several researchers have reported that temperatures between 25°C and 30°C favored adequate sporulation and citric acid production by *A. niger.* Low temperatures lead to slow germination of the fungi and slow metabolic activity, but high temperatures lead to enzyme denaturation and reduced cell viability (Yigitolu 1992).

### 8.4.3.3 Control of pH

The success of microbial production of organic acids is critically dependent on the pH of the fermentation medium. The pH is known to influence important microbial growth and the synthesis of their metabolites. It affects the breakdown of substrates and the permeability of cell membranes, vis-a-vis substrate intake and product release. Optimum yields of many organic acids are obtainable at the optimum pH for microbial growth. It is important to maintain a constant pH during fermentation to ensure optimum yield and reduce the risk of contamination (Kareem et al. 2010). It is also important to note that organic acid

production is accompanied by a progressive decrease in the pH of the fermentation broth during the fermentation period.

### 8.4.3.4 Aeration and agitation

Aeration is a critical factor in the growth and performance of microbial cells by improving mass transfer characteristics with respect to substrate use and product formation. It fulfills four main functions in the fermentation processes: (i) maintain aerobic conditions, (ii) remove carbon dioxide, (iii) regulate the substrate temperature, and (iv) regulate the moisture level.

The amount of oxygen required in a fermentation setup depends on the type and nature of microorganisms (Kamzolova et al. 2003). In organic acid production, an adequate aeration rate is known to enhance the yield and reduce fermentation time. The construction of appropriate aeration devices in fermenters will fulfill the high oxygen demand, which depends on the viscosity of the fermentation broth. Sterile air serves as a source of oxygen intake in fermentation processes. The oxygen is normally dissolved in the liquid phase, which is measured by the amount of dissolved oxygen. In an aerobic fermentation, the bioreactor is usually equipped with a dissolved oxygen meter (Raimbault 1997). In addition to uniform distribution of oxygen, agitation is required to improve rheology, mass, and heat transfer of the fermentation medium.

# 8.5 Fermentation processes

## 8.5.1 Surface fermentation

The first fermentation system employed for the production of organic acid on an industrial scale utilized a surface fermentation process. *A. niger* forms a mycelium layer on the liquid surface of the aluminum or stainless steel trays, which are stacked in fermentation rooms supplied with filtered air that serves both to supply oxygen and to control the temperature of fermentation. These techniques make use of the fact that citric acid production occurs in cells not in the active stage of growth (Yigitolu 1992).

Surface fermentation is a simple technique that is easy to control and implement because it needs no aeration or agitation of the fermentation broth. The separation of citric acid from the mycelium is easy because the microorganism is not dispersed into the medium. Only the temperature and humidity of the fermentation chamber need controlling. It can be carried out easily in small plants and therefore is suitable for developing economies. With surface fermentation, the fermentation broth is concentrated due to the high evaporation rate during fermentation. Thus, expenses and losses during recovery and purification are low. However, fermentation time is long and therefore productivity is low (Kapoor et al. 1983).

### 8.5.2 Submerged fermentation

In submerged fermentation, the microorganism is grown in the fermentation broth. Submerged fermentation has the following advantages: lower total investment costs, higher yields of citric acid, improved process control, reduced fermentation time, reduced labor costs, simpler operations, and easier maintenance of aseptic conditions on an industrial scale. However, the use of submerged fermentation is limited by its high energy consumption and vulnerability to contamination (Kapoor et al. 1983).

### 8.5.3 Solid-state fermentation

Solid substrate fermentation involves the use of insoluble material as a substrate, both as a physical support and as a source of nutrients for the growth of microorganisms. The solid substrate material can be biodegradable but must provide suitable support for fungal growth and the exchange of gases. For citric acid production, oxygen acts as a direct regulator. As compared with submerged fermentation, solid substrate fermentation offers many advantages: limited consumption of water, low heat transfer capacity with easy aeration, high surface exchange air/substrate, low energy consumption for the process, lower water usage and wastewater treatment costs, limited contamination risks because of a lower moisture level, and simplified nutrient media composition.

On the other hand, the disadvantages of solid-state fermentation include the following: difficulty in controlling process parameters such as pH, heat, and nutrient conditions; problems with heat build-up; higher product impurity level; increased amount of solid waste to manage (including the fungal biomass); and higher product recovery costs (Raimbault 1997).

## 8.6 Downstream processes

Fermentation broths are complex mixtures of cells, fungal mycelia and spores, partially converted materials, desired products, and unwanted metabolites. These particles must be separated from the product through a series of unit operations known as downstream processing. The primary aim of downstream processing is to efficiently and safely recover the target product to the required specification while maximizing recovery yield and minimizing the process cost. It often accounts for up to 60% of the total production costs, excluding the cost of the purchased raw materials, and determines the economic feasibility of the process. The target products may be the cells (biomass), components within the fermentation broth (extracellular), or those trapped in cells (intracellular). It is often desirable to choose a suitable strain of microorganism that produces extracellular rather than intracellular products for easy recovery. If the product is intracellular, the cells must be ruptured to release the product after which extraction or purification is performed.

The desired product can be recovered by processing the cell or spent medium to a high degree of purity. The level of purity that must be achieved is usually determined by the prospective use of the products such as for food, health, research, and industrial purposes.

There are some fermentation factors that affect downstream processes such as the following:

- Type and nature of the microorganisms. Some properties of microorganisms such as morphology, flocculation characteristics, size, cell wall rigidity, mycelia formation, and sporulation can affect homogenization, sedimentation, and filtration processes.
- Presence of media impurities and unwanted metabolites can also add to downstream processing costs. Other fermentation additives and partially or unconverted materials may also interfere with the purification steps.
- The fermentation substrate is critical to the downstream process. A cheap substrate with many impurities may increase the downstream process cost whereas a more expensive but purer substrate may eventually cut the overall cost.
- 4. Process economics. Simple and available process equipment may be more cost-effective than sophisticated and complex techniques that might increase the overall cost. The process is normally kept to a minimum to reduce the cost and prevent the overall yield

loss of multistage purification processes. Therefore, the specific unit step in downstream processes is influenced by the economics of the process, the desired degree of purity, the unit yield attainable at each step, and the safety of the product (Najafpour 2007).

Bioseparation processes make use of many separation techniques commonly utilized in the chemical process industries. The first step in downstream processing is the separation of insoluble particles from the fermentation broth. The selection of a separation technique depends on the characteristics of the solids and the liquid medium, which can be accomplished by filtration or centrifugation. Filtration separates particles by forcing the fluid through a filtering medium on which the solids are deposited. The choice of filtration methods depends on the range of particle sizes removed and the pressure differences.

The fermentation broth has to be processed and pass through several stages for separation and purification. The product requires a sequence of operations for high purification. The usual steps are listed below.

Separation of biomass. This stage involves a solid–liquid separation to harvest cells from the spent medium and each fraction is subjected to further treatment depending on whether the target product is intracellular or extracellular. Several factors such as size or morphology of organisms, specific gravity, viscosity, and rheological property of the spent medium can influence the choice of solid–liquid methods. Separation techniques include filtration, sedimentation, and centrifugation.

*Cell disruption.* It is an appropriate method for the isolation and purification of intracellular products. Available methods for cell disruption can be classified into two groups: (i) mechanical and (ii) nonmechanical. Cell wall destruction is achieved by mechanical methods using solid-shear (bead mill) and liquid-shear forces (high-pressure homogenizer). Nonmechanical methods such as chemical and enzymatic disruption are mild and specific for cell wall degradation. Generally, increases in the viscosity of the suspension are noted as a challenge in cell disruption due to the liberation of DNA and other cellular inclusions.

*Concentration of broth.* This stage involves the isolation of microbial products through solvent extraction, absorption, precipitation, and ultrafiltration. During this process, target product concentration increases considerably.

*Product-specific purification*. To obtain a product of high purity, further product concentration techniques such as fractional precipitation, chromatography, and adsorption are used to remove some tiny impurities and further concentrate the product.

*Crystallization and drying.* Isolated product will be first crystallized and later dried by drum drying, spray drying, and freeze drying. Processes of centrifugation, freeze drying/lyophilization, or organic solvent removal are commonly used. The last step is aimed at presenting the desired product in a form suitable for final formulation (Najafpour 2007).

# 8.7 Prospects of microbial fermentation of organic acids

Organic acids are vital groups of chemicals that can be produced using fermentation processes. The natural microbial metabolites are intermediates in major metabolic pathways. Because of their functional groups, organic acids are useful as starting materials for the chemical industry. Some organic acids such as fumaric, malic, and succinic acids have the potential to replace petroleum-based chemicals such as maleic anhydride. The need for a sustainable technological development is shifting attention away from petroleum to the use of renewable resources through fermentation processes.

Fermentation technology has allowed the use of indigenously available and cheap substrates for the production of commercially valuable organic acids for sustainable development. Continuous success in the production of organic acids through advances in fermentation technology are envisaged to provide a sustainable industrial bioprocess. Such developments will lead to the evolution of biorefineries to ensure the production of biofuels as well as the building block chemicals from biomass.

# 8.8 Case study: Comparative production of citric acid by immobilized A. niger

Citric acid is an important metabolite produced by fungal fermentation, and is widely used in the food and pharmaceutical industries (Kareem et al. 2010). A large number of microorganisms including fungi, bacteria, and yeasts have been employed for citric acid production. Among the mentioned strains, the fungus *A. niger* has remained the organism of choice for commercial production because it produces more citric acid per time unit. The main advantages of using *A. niger* are its ease of handling, its ability to ferment a variety of cheap raw materials, and high yields (El-Holi and Al-Delamy 2003).

Several raw materials such as hydrocarbons, starch materials, and molasses have been employed as substrates for commercial submerged citric acid production, although citric acid is mostly produced from starch or sucrose-based media using submerged fermentation. Generally, citric acid is produced by fermentation using inexpensive raw materials such as starch hydrolysate, sugar cane broth, and by-products like sugar cane and beet molasses (Yigitolu 1992).

A variety of agro-industrial residues and by-products have also been investigated with solid-state fermentation techniques for their potential to be used as substrates for citric acid production such as coffee husk, wheat bran, apple pomace, pineapple waste, kiwifruit peel, grape pomace, citrus waste, and others (Haq et al. 2004).

The immobilization of microbial cells in polymeric matrices has been reported to enhance the microbial production of organic acids due to the speed and ease of recovery of the cells and ability to reuse the cells. The technique also makes repeated batch and continuous processes possible by allowing a consistent decrease in medium viscosity, and enhanced nutrient and oxygen transfer (Kareem et al. 2013). Cassava starch is abundantly available in the tropics as a cheap substrate. Therefore, this study compares citric acid production by immobilized *A. niger* cells.

### 8.8.1 Materials and methods

### 8.8.1.1 Immobilization of A. niger

Spores of *A. niger*  $(4.0 \times 10^3)$  were mixed with sodium alginate solution (2%) in a 1:1 ratio. The alginate mixture was added dropwise into a calcium chloride solution (0.2 M) with continuous shaking at 4°C. As soon as the drop of enzyme–alginate solution mixed with the CaCl<sub>2</sub> solution, Na<sup>+</sup> ions of sodium alginate were replaced by Ca<sup>2+</sup> ions from the CaCl<sub>2</sub> solution, which finally formed Ca-alginate beads. The beads thus formed were washed three to four times with deionized water and finally with 50 m Tris-HCl buffer at pH 7.5. These beads were dried and weighed for further studies.

#### 8.8.1.2 Citric acid production

Cassava starch (30 g) was added into 200 mL Erlenmeyer flasks and supplemented with nitrogen sources. Distilled water (200 mL) was added and the initial pH of the medium was adjusted to 6.0. The flask was cotton-plugged and autoclaved at 121°C for 15 min. After

cooling to room temperature, each medium was inoculated with immobilized Ca-alginate beads and incubated at 30°C in a rotary shaking incubator for 5 days. After incubation, the medium was filtered, and then the filtrate was used for the estimation of citric acid, residual sugar content, and pH.

#### 8.8.1.3 Citric acid determination

Citric acid was determined titrimetrically (Association of Official Analytical Chemists [AOAC] 1995) by using 0.1 M NaOH and phenolphthalein as an indicator, and calculated as a percentage according to the formula:

Citric acid g/L  $\frac{\text{Titer} \times 0.0064 \times 100 \times 10}{10 \text{ mL}}$  (where 0.0064 denotes citric acid factor)

### 8.8.1.4 Biomass, residual sugars, and pH determination

Biomass, sugar, and pH values were determined according to AOAC (1995). To determine the biomass, the whole fungal culture was filtered with sterile filter paper dried at 105°C to a constant weight. Results were expressed in grams per kilogram. Sugar was determined using a refractometer and pH was measured using a pH meter. Each analysis was conducted in triplicate.

### 8.8.2 Results and discussion

This study presents citric acid production using immobilized *A. niger* on raw cassava starch. The results showed that 26.84 g/L of citric acid was obtained in a medium containing cassava starch alone whereas the addition of nitrogen supplements enhanced citric acid production with the maximum citric acid yield (42.2 g/L) obtained in a cassava starch medium supplemented with peptone (Table 8.2). The results indicated that *A. niger* first hydrolyzed starch to glucose units to provide utilizable substrate for glycolysis and subsequent generation of citric acid in the TCA cycle. Higher citric acid accumulation by *A. niger* in medium containing high concentrations of glucose has been attributed to high hexokinase activity in the glycolytic pathway (Ali 2004). The results also indicate that nitrogen constituents not only enhance microbial growth but also stimulate organic acid synthesis. Kareem and coworkers (2010) reported that fermentation media for citric acid production should consist of substrates necessary for the growth of microorganisms; primarily carbon, nitrogen, and phosphorus sources.

	7 0
Supplements	Citric acid (g/L)
KH <sub>2</sub> PO <sub>4</sub>	32.81
$(NH_4)_2PO_4$	36.32
Peptone	42.12
Control	26.84

Table 8.2 Effect of Nitrogen Supplements on Citric Acid Production from Cassava Starch by A. niger

*Note:* Data are means of triplicates, SD = 0.052.

The effect of alcohol on citric acid production is shown in Table 8.3. The results showed that methanol at 1% concentration gave an inductive effect on citric acid production. This observation concurred with the results of Kareem and coworkers (2010) in which the addition of methanol to pineapple waste medium led to an increase in citric acid production by *A. niger*. The inductive effect of methanol for citric acid production may be due to the reduction of the inhibitory effects of metal ions (Kiel et al. 1981). The addition of alcohols to the fermentation media had been reported to increase citric acid yield by filamentous fungi (El-Holi and Al-Delamy 2003). However, in this study, methanol had a more inductive effect on citric acid production than ethanol. The exact mechanism of the alcohol effect, however, is unexplained, although it has been postulated that the addition of methanol increases the tolerance of fungi to  $Fe^{2+}$ ,  $Zn^{2+}$ , and  $Mn^{2+}$ .

The results of the effect of agitation on citric acid production by immobilized *A. niger* showed that a higher citric acid yield of 58.1 g/L was obtained under agitated conditions at the fourth day of fermentation (Table 8.4). Agitation and aeration are important in submerged fermentation, which is extremely important for adequate mixing, mass transfer, and heat transfer. In addition, agitation maintains homogeneous chemical and physical conditions in the culture by continuous mixing (Mantzouridou et al. 2002).

The effects of fermentation period on citric acid production, pH, and sugar consumption are presented in Figure 8.3. A steady increase in citric acid production was observed during the fermentation period, with an optimum yield (62.2 g/kg) after fermentation for 96 h, whereas a decrease in the consumption of sugar was observed. Previous reports have shown that the production of citric acid on pineapple waste approximately paralleled the consumption of sugar (Kareem et al. 2010). Based on the amount of fermentable sugar

	Citric aci	d (g/L)
Time (days)	Methanol	Ethanol
0	42.12	42.12
1	55.30	45.10
2	50.03	52.21
3	59.20	35.11

Table 8.3 Effect of Alcohol on Citric Acid Production from Cassava Starch by Immobilized A. niger

 Table 8.4
 Effect of Agitation on Citric Acid Production

 from Cassava Starch by A. niger

	Citric acid (g/L)		
Time (days)	Agitation	Static condition	
1	15.1	10.2	
2	36.2	28.4	
3	42.3	34.3	
4	58.1	51.2	
5	51.2	46.2	

*Note:* Data are means of triplicate determinations, SD = 0.033.

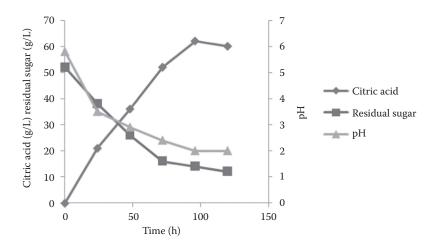
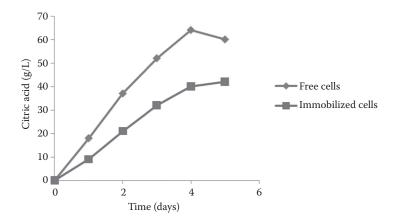


Figure 8.3 Effect of fermentation period on citric acid production, pH and sugar consumption.

consumed, the yield of citric acid was 90.62% under optimum fermentation conditions. A decrease in pH values from 5.0 to 2.2 was noted after 96 h of fermentation. The pH value maintained at the beginning of fermentation was important for a specific biomass formation. The progressive decrease in pH was noted as incubation time increased, which was due to the formation and accumulation of citric acid.

Comparison between fermentation with free cells and immobilized cells of *A. niger* (Figure 8.4) showed that immobilized *A. niger* cells gave an enhanced citric acid production and fermentation efficiency compared with free cells as a result of decreased medium viscosity and enhanced oxygen and nutrient transfer (Kareem et al. 2013). In addition, immobilized cells can be reused up to 20 repeated cycles. This will ensure the optimization of the citric acid bioprocesses and therefore reduce process cost.



*Figure 8.4* Comparative citric acid production by free and immobilized *A. niger* cells.

# 8.9 Conclusion

Organic acids are important building block chemicals and have a promising global market. Their biosynthesis through fermentation processes can make them available for global use and as well widening their application potential. The production of these commercially valuable organic acids is largely dependent on the choice of a suitable organism capable of utilizing cheap or locally available substrate with a view to reducing the process cost. Downstream processing is also critical to the production of organic acids of high purity. When adequate attention is given to the biosynthesis of these organic metabolites, the use of petrochemicals will be drastically eliminated and greenhouse gas emissions will be reduced.

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# chapter nine

# Microbial amino acids production

# Zafar Alam Mahmood

### Contents

9.1	Introd	luction	. 187
9.2	L-Glu	tamic acid	. 191
	9.2.1	Historical background and new challenges	. 191
	9.2.2	Biotechnological approach and production	. 192
	9.2.3	Industrial application and therapeutic role	. 194
9.3	L-Lysi	ne	. 194
	9.3.1	Historical background and new challenges	. 194
	9.3.2	Biotechnological approach and production	. 198
	9.3.3	Industrial application and therapeutic role	. 199
9.4	Methi	ionine	. 199
	9.4.1	Historical background and new challenges	. 199
	9.4.2	Biotechnological approach and production	. 201
	9.4.3	Industrial application and therapeutic role	.203
9.5	Trypt	ophan	. 203
	9.5.1	Historical background and new challenges	.203
	9.5.2	Biotechnological approach and production	
	9.5.3	Industrial application and therapeutic role	.205
9.6	Case s	study: production of L-lysine on glucose and molasses medium	
		emented with fish meal under submerged condition in mini jar fermenter	
	9.6.1	Cultivation of C. glutamicum FRL no. 61989	.205
	9.6.2	Isolation, extraction, and purification of L-lysine	.206
	9.6.3	Effect of media composition and fermentation parameters on L-lysine	
		production	.207
9.7	Concl	usion	.209
		lgments	
Refe	rences		. 209

# 9.1 Introduction

Amino acids are the constructive components (subunits) of proteins and peptides and are thus regarded as the building blocks of life. Amino acids contain a high percentage of nitrogen (~16%), which distinguishes them from fats and carbohydrates (Mahmood 2010). Amino acids are classified into two categories: essential amino acids, which are not synthesized in the body; and nonessential amino acids, which are synthesized by the body (Table 9.1). Amino acids have great demand and application in view of their importance in the food, feed, personal care, and pharmaceutical industries as nutrients, additives, rejuvenators, and drugs. Their roles are highly prominent and specific in different applications as

S. no	Essential amino acids	S. no	Nonessential amino acids
01	Arginine	01	Alanine
02	Histidine	02	Asparagine
03	Isoleucine	03	Aspartate
04	Leucine	04	Cysteine
05	Lysine	05	Glutamate
06	Methionine	06	Glutamine
07	Phenylalanine	07	Glycine
08	Threonine	08	Proline
09	Tryptophan	09	Serine
10	Valine	10	Tyrosine

Table 9.1 List of Essential and Nonessential Amino Acids

well as in therapeutic potential. During the last two decades, the global demand for some essential amino acids, for example, lysine and methionine, has tremendously increased because of their extensive use in the feed, food, and pharmaceutical industries. The addition of these amino acids in animal feed not only optimizes the growth of animals but also improves the quality and quantity of meat. The availability of such animal products has certainly helped a lot in overcoming the deficiency of essential amino acids, which is quite prominent in the primary foodstuffs of underdeveloped and overpopulated areas of the globe. The demand for nonessential amino acids, for example, glutamic acid, has also significantly increased in recent years due to its extensive application in the food industry. The global demand is further expected to increase in the next decades, requiring researchers to focus more on developing advanced manufacturing techniques utilizing biotechnological tools to meet the challenging demand of various amino acids.

The aim of this chapter is to present comprehensive information on some of the abovementioned amino acids, such as lysine, methionine, tryptophan, and glutamic acid (Figure 9.1) with respect to their historical backgrounds, new challenges along with the

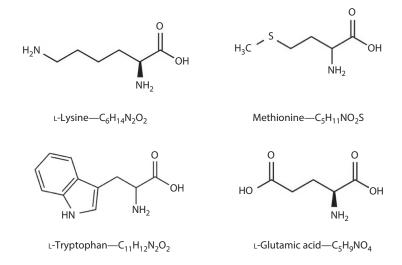


Figure 9.1 Structure and molecular formula of lysine, methionine, tryptophan, and glutamic acid.

utilization of biotechnological approaches and tools for increasing the production capability of bacterial strains, their industrial applications, and therapeutic roles.

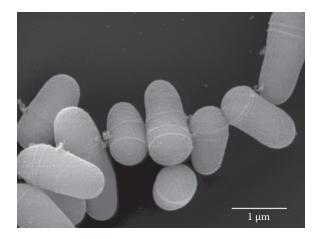
The utilization of microorganisms for the development of commercial commodities is supposed to have initiated in primordial times. This has resulted in the production of fermented beverages and food stuffs but without the true concept of fermentation. It was Pasteur who introduced the word *fermentation* for the first time and negated the theory of spontaneous generation. Subsequent studies on these lines have been performed by a large number of researchers using bacteria and filamentous fungi. However, the scientific interpretation of fermentation started only when the perception and understanding of microbial physiology and metabolic processes were fully understood (Mahmood 2010).

The biotransformation of organic molecules into useful products provided a new concept in the history of fermentation and, as a result, a number of metabolites such as antibiotics and amino acids were identified. The continued studies on the subject resulted in the discovery of an immense number of pharmaceutical and biochemical products. At the same time, advancements in fermentation engineering provided great support in the production of these useful products on a commercial scale. With advanced knowledge of microbial physiology, molecular biology, and genetic engineering, the fermentation industry has grown considerably in terms of introducing some novel products, which have great industrial and medical applications. These advancements have provided further significant knowledge about these microorganisms, their isolation and modification, and their growth conditions along with the possibility to improve the productivity of the desired product through process optimization or by the addition or subtraction of a particular nutrient.

Initial progress in terms of mutating the strains and utilization of microbial physiology to develop suitable media composition for enhanced yield of desired products was the beginning of modern biotechnology. The evaluation of new biological commodities that could be used as pharmaceuticals, nutraceuticals, or cosmeceuticals has produced phenomenal progress—with industrial microbiology on one side and fermentation engineering on the other. The intensified industrial exploitation of biological practices advocates that biotechnology will be a major growing industry in the near future and that this will alter the lives and prosperity of people all over the world.

Over the last 50 years, the knowledge of microbial physiology and molecular biology has increased substantially in terms of amino acid production and its utilization in human nutrition as well as in domestic animals. The diversified application of amino acids as a raw material in the food, pharmaceutical, and cosmetic industries has played a significant role in boosting the research activities in this particular field. Older methods, such as extraction of amino acids from natural sources or production through chemical synthesis, have largely been replaced by biotechnological processes such as production through fermentation or by enzymatic catalysis. Organized research on the microbial production of amino acids were being produced using microbial sources. The best example is of direct L-glutamic acid fermentation using *Corynebacterium glutamicum* (Figure 9.2) in Japan by Kyowa Hakka Kogyo Company (Kinoshita et al. 1957a).

This progress had a great economic effect in the field of amino acid production and, as a consequence, L-lysine was successfully produced and commercialized using a mutant strain of *C. glutamicum* (Kinoshita et al. 1958). This has resulted in a new concept of fermentative production of amino acids with the introduction of a series of artificially derived auxotrophic or regulatory mutants for the production of other amino acids as well, such as methionine, tryptophan, and others. During the last three decades, a large number of



*Figure 9.2* Electron micrograph of *C. glutamicum* ATCC 13032. (Adapted from Wittmann, C., and Becker, J. The L-lysine story: from metabolic pathways to industrial production. *Microbiology Monographs* DOI 10.1007/7171\_2006\_089/Published online: February 24, 2007.)

mutant strains of *C. glutamicum* have been developed to produce and meet the market demand for various amino acids. Biotechnology has certainly played a major role in the fermentative production of various amino acids.

With the development of new applications for amino acids in the feed, food, and pharmaceutical industries, enormous improvements have been made in production technology during the latter half of the twentieth century. Submerged fermentation technology plays a key role in this progress, and fermentative production of amino acids currently represent the leading products of biotechnology in terms of both volume and value. The field is very competitive and the cost of production is a major factor, which can certainly support the manufacturers to stay in the market with profitable business. For cost-effective fermentative production, many biotechnological methodologies have been applied to establish high productivity and improvement in product recovery processes. The role of genetic engineering relating to amino acid-producing strains is now being used for the development of biosynthetic and transport capacity of organisms for enhanced production. In addition, the rapid progress in genome analysis is expected to revolutionize microbial strain improvement techniques. With these advancements, the global production and consumption of amino acids is expected to increase more in the near future and thus, will have great effect on amino acid-based industries in terms of investment. Nowadays, the major producers of amino acids are based in Japan, the United States, South Korea, China, and Europe. However, some other Asian countries, such as Indonesia, Malaysia, Thailand, and Vietnam are also entering the amino acid business. The major manufacturing companies include Ajinomoto Group, Archer Daniels Midland Company, Cheil Jedang, COFCO Biochemical (Anhui) Co. Ltd., Daesang Corporation, Evonik, Global Bio-Chem Technology, Novus International Inc., Royal DSM N.V., Sekisui Medical Co. Ltd., Shandong Zhengda Linghua Biotechnology Ltd. Co., Toronto Research Chemicals Inc., Vedan, and VitaLys I/S. The global market for amino acids has been forecast to reach US\$11.6 billion by the year 2015. The demand for various amino acids is expected to increase in the production of animal feed, health foods, dietary supplement products, artificial sweeteners, and cosmetics in the subsequent years (Jose 2011; Wippler 2011). The world's consumption pattern in 2011 for lysine, methionine, and tryptophan is shown in Figure 9.3, which is self-explanatory

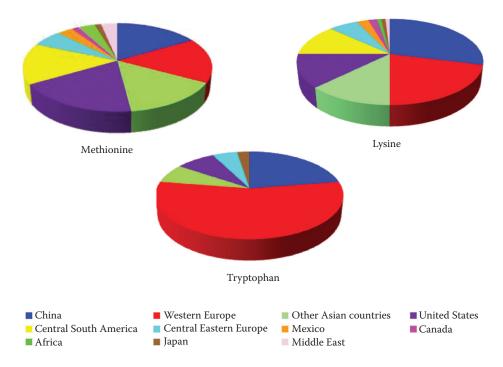


Figure 9.3 World's consumption of methionine, lysine, and tryptophan during 2011.

and provides a comprehensive idea about the consumption of these amino acids in different regions of the world (IHS Chemical 2012).

# 9.2 L-Glutamic acid

## 9.2.1 Historical background and new challenges

From an industrial application or commercial point of view, L-glutamic acid is one of the most important amino acids. The carboxylate ions and salts of glutamic acid are known as glutamates. Its initial discovery is quite old and was first reported in 1866 by a German scientist, Karl Heinrich Leopold Ritthausen, in wheat gluten treated with sulfuric acid (Plimmer and Hopkins 1912). In 1907, the first successful commercial production of glutamic acid was performed at Tokyo Imperial University by Kikunae Ikeda, who developed and patented a method to produce crystalline salt of glutamic acid, monosodium glutamate (MSG). Subsequently, MSG was patented by Ajinomoto Corporation of Japan in 1909. The turning point in the history of glutamic acid production was the discovery of *Micrococcus glutamicus* (later identified as *C. glutamicum* in 1957), which produces 30 g/L of L-glutamic acid in glucose medium through fermentation (Kinoshita et al. 1957b). The interest in largescale L-glutamic acid fermentation was primarily developed due to increased demand for MSG, which was and still is used as a flavor-enhancing agent. Glutamic acid and its salt MSG represent the largest product segment within the amino acid market. In 2007, the world product volume was indicated as 1.6 million tons (Demain 2007); in 2009, the figure reached 2.0 million tons (Sano 2009), and in 2012, the figure was reported to be 2.2 million tons (Abdenacer et al. 2012).

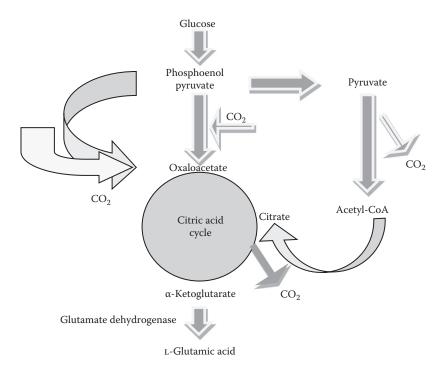


Figure 9.4 Biosynthesis of L-glutamic acid in C. glutamicum using glucose as a carbon source.

The market demand is continuously increasing and is creating a challenge for the manufacturers, either to increase the capacity of existing plants or to utilize advanced biotechnological tools to increase the production both in terms of strain improvement and modification in fermentation medium. The biosynthesis of L-glutamic acid using *C. glutamicum* involves relatively simple pathways of metabolism. Glucose is mainly metabolized through glycolytic pathways. Pyruvate and phosphoenol pyruvate play important roles in the production of oxaloacetate and then  $\alpha$ -ketoglutarate. The  $\alpha$ -ketoglutarate is converted to L-glutamic acid in the presence of enzyme glutamate dehydrogenase (Figure 9.4). The phosphorylation stage is quite important because the genes controlling this conversion process are located adjacent to the genes of cell wall synthesis or cell division, might inhibit the activity of  $\alpha$ -ketoglutarate dehydrogenase activity. As a result, the conversion of  $\alpha$ -ketoglutarate to glutamic acid might be affected (Eggeling and Sahm 2011).

### 9.2.2 Biotechnological approach and production

A great deal of biotechnological tools have been used by various researchers with an understanding of microbial physiology to modify the strains for increased production of glutamic acid, especially utilizing *C. glutamicum* (Ikeda and Takeno 2013). The progresses, with respect to the capabilities of the different strains of *C. glutamicum* to produce glutamic acid, have been reported from time to time over the last few decades. This was done having comprehensive knowledge of the biology, biosynthetic pathways, mutation, nutritional requirements, and putative mechanosensitive channels associated with this organism. Microbial production of L-glutamic acid has been extensively studied by a large number of

research investigators using different strains of *Coryneform* species and the process with specific strains have been patented. The most popular *Coryneform* species include *C. glutamicum*, *Corynebacterium lilium*, *Corynebacterium herculis*, *Brevibacterium flavum*, *Brevibacterium licon*, *Brevibacterium divarticum*, *Brevibacterium ammoniagenes*, *Brevibacterium thiogenetalis*, *Brevibacterium saccharoliticum*, and *Brevibacterium roseum* (Kinoshita 1999). Other glutamic acid-producing organisms include *Escherichia coli*, *Bacillus megaterium*, *Bacillus circulans*, *Bacillus cereus*, and *Sarcina lutea*. Industrially, glutamic acid is usually manufactured by batch/fed-batch submerged fermentation processes using genetically modified strains of *Corynebacterium* or *Brevibacterium*. Commercial-scale production is carried out in large fermenters equipped with all the facilities to control different parameters and optimize production such as provision of cooling, measurement of dissolved oxygen and pH, and others. A typical example of the submerged fermentation process on a laboratory scale for the production of L-glutamic acid utilizing a mutant strain of *C. glutamicum* has been summarized below.

Organisms are usually maintained on Hottinger slant agar containing beef trypsin digestion, sodium chloride, dipotassium phosphate, and agar at low temperature. After subculturing, a 24-h-old culture is used to inoculate 25 mL of sterile seed medium in 250 mL Erlenmeyer flasks in triplicate. The following seed medium composition can be used: glucose (8%), NH<sub>4</sub>Cl (0.5%), corn steep liquor (0.3%), K<sub>2</sub>HPO<sub>4</sub> (0.5%), KH<sub>2</sub>PO<sub>4</sub> (0.5%),  $MgSO_4$ ·7H<sub>2</sub>O (0.03%), CaCO<sub>3</sub> (1.0%), and deionized water to make 100%. The pH of the medium can be adjusted to 7.2 using NaOH. The inoculated flasks are grown in an orbital shaker incubator maintained at 30°C and 230 rpm for 15 h. The entire contents of the one flask is then transferred to a 2.0 L capacity Eyla Mini Jar fermenter with 500 mL of sterile nutrient medium containing molasses (20%), KH<sub>2</sub>PO<sub>4</sub> (0.5%), KH<sub>2</sub>PO<sub>4</sub> (0.5%), MgSO<sub>4</sub>·7H<sub>2</sub>O (0.3%), urea (0.8%), CaCO<sub>3</sub> (1.0%), and deionized water to make 100%. In most cases, the optimum pH of the medium was recorded as 7.0. The fermentation is usually initiated with continuous agitation and aeration for 48 h at 30°C. At the end of the incubation period, the culture broth is separated from the cells by centrifugation at 10,000 rpm for 5 min at  $4^{\circ}$ C and the supernatant is diluted 50-fold with 7% (v/v) of glacial acetic acid. The diluted sample is then further centrifuged at 10,000 rpm for 5 min at 4°C. The supernatant is then finally filtered through a nylon membrane of 0.22 µm pore size and then collected for further analysis and crystallization (Khan et al. 2013a,b; Pasha et al. 2011).

Since the discovery of the original strain of *C. glutamicum*, a large number of strains, which are capable of producing significant quantities of glutamic acid, have been developed through the process of genetic engineering. Yields have now exceeded 100 g/L and even more with modified strains compared with original wild strains, which produced only 10 g/L. Most original glutamic acid-producing strains of C. glutamicum were observed as biotin auxotrophs. Thus, growing such strains in a medium deficient in biotin was found to "trigger" glutamic acid production. The biotin-deficient medium has the property of altering the cell membrane of the organism because of suboptimal fatty acid biosynthesis and is thus responsible for increased production. In view of this hypothesis, it is believed that some other physicochemical factors such as fermentation at higher temperatures, the addition of surfactants (e.g., Tween 40) or antibiotics (e.g., penicillin) in the culture medium can also trigger the excretion of glutamic acid. These factors have been linked to a triggering mechanism resulting in a decrease or repression of the enzyme  $\alpha$ -ketoglutarate dehydrogenase. Thus, finally causing redistribution of metabolites at the branch point in the tricarboxylic acid (TCA) cycle leading from  $\alpha$ -ketoglutarate to succinyl-CoA or glutamate (Figure 9.4). However, this mechanism has limitations, therefore, a further increase or overproduction of glutamate can be achieved through optimization of the metabolic flux

by tuning the glutamate dehydrogenase activity (Asakura et al. 2007). L-Glutamic acid production was also achieved through immobilized cells of *C. glutamicum*. Using more than 93 g/L, good yields were achieved through batch fermentation but with low productivity (3.8 g/L/h), whereas 73 g/L of glutamic acid was recovered through continuous fermentation with high productivity of approximately 29.1 g/L/h (Amin and Al-Talhi 2007). A number of other studies have also been published, documenting the role of immobilized cells in the production of L-glutamic acid. It has been reported that immobilization of microbial cells in biological processes can occur either as a natural phenomenon or through artificial processes. Although attached cells in their natural habitat exhibit significant growth, the artificially immobilized cells also allow restructure of growth. The authors reported obtaining 48.5 g/L of L-glutamic acid through the immobilized cells of a mutant strain of *C. glutamicum* (Pasha et al. 2011; Prasad et al. 2009). The use of immobilized cells for the production of L-glutamic acid and some other amino acids is progressing rapidly and is expected to have some more valuable and useful data for commercial exploitation in the near future.

### 9.2.3 Industrial application and therapeutic role

The greatest application of glutamic acid and its salt is in the food industry as a flavor enhancer. A considerable quantity of free glutamic acid, or its salt MSG, is present in different food products, for example, in different kinds of cheeses and soy sauces. In addition, because glutamate is a key compound in the cellular metabolism, it therefore serves as a unique brain fuel and performs some other important functions such as detoxification of ammonia, as a hepatoprotective agent (glutamine), to aid in peptic ulcer healing, and others (Chaitow 1985; Zareian et al. 2012).

One of the leading roles of glutamic acid in pharmaceuticals is that of a neurotransmitter. L-Glutamic acid is widely distributed in the central nervous system (CNS) and is reported to act as a neurotransmitter. Furthermore, it may have a stimulating effect on the metabolism of the cerebral cortex thus improving mental performance and memory function. The concentration of glutamic acid in the CNS is higher than any other commonly recognized neurotransmitter. In addition, the biochemical and electrophysiological data also suggest that glutamic acid or its analogue acts as an excitatory neurotransmitter of CNS. There is a high concentration of *N*-methyl-D-aspartate (NMDA) receptor in the hippocampus (Bloom 2001). The blockage of NMDA receptors can greatly affect the memory and overall mental performance of an individual. Glutamic acid and aspartic acid have the capability to combine with NMDA receptors thus increasing cation conductance, depolarizing the cell membrane, and deblocking the NMDA receptors. The ultimate results of these biochemical features lead to improvements in memory function and mental performance (McEntee and Crook 1993).

# 9.3 L-Lysine

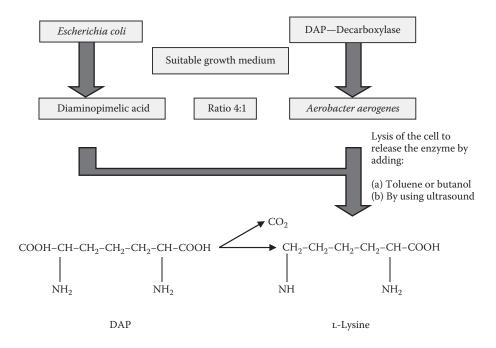
## 9.3.1 Historical background and new challenges

L-Lysine is one of the leading and most exploited amino acid among the essential amino acids list. It has vast applications and role in human and animal nutrition. The systematic research to explore the possibility of amino acid production through microorganisms started during the late 1940s, and during the later part of the 1950s, some fruitful results were reported (Mitchell and Houlahan 1948; Windsor 1951). In continuation, a number

of research investigators published their findings that L-lysine can be synthesized from  $\alpha$ -aminoadipic acid by yeast and *Neurospora* mold, or from diaminopimelic acid (DAP) by *E. coli* (Davis 1952; Dewey and Work 1952; Mitchell and Houlahan 1948; Windsor 1951). However, microbial production of L-lysine through the decarboxylation of DAP by Chas Pfizer and Company Inc. in the United States was regarded as the first commercial scale production (Casida and Baldwin 1956). A modified process (Figure 9.5) utilizing a single organism, which produced DAP and converted it to L-lysine, was then simultaneously discovered and reported (Kita et al. 1958).

The introduction of L-glutamic acid fermentation using *C. glutamicum* had a great economic effect in the field of L-lysine fermentation. Further research in this direction resulted in the development of an efficient mutant of *C. glutamicum* for the commercial production of L-lysine (Kinoshita et al. 1957b, 1958). This modified method was also successfully used to produce some other amino acids (Mahmood 2010). Despite the commercial challenges, the production of L-lysine using mutant strains brought about a new concept in the fermentative production of amino acids. With the emergence of new biotechnological approaches and tools, more focus was given to improving the strains and taking artificially derived auxotrophic and regulatory mutants that were resistant to feedback inhibition. Using these genetically modified organisms, increased production of L-lysine was achieved by controlling the biosynthetic pathways. The biosynthetic block most conducive for effective L-lysine accumulation in auxotrophic mutants was observed to be those requiring homoserine or threonine plus methionine for the growth of organisms (Figures 9.6 and 9.7).

Modifications in the biosynthetic pathway extended one of the biggest advantages by producing some other amino acids along with L-lysine using *C. glutamicum*, *B. flavum*, and *B. lactofermentum*.



*Figure 9.5* Production of L-lysine through the conversion of its immediate precursor DAP (two-step process).

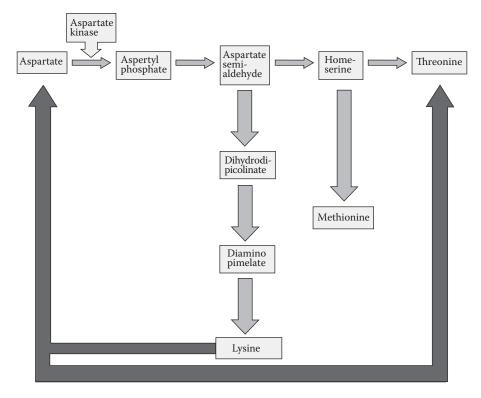


Figure 9.6 Regulatory pathways of L-lysine biosynthetic pathways in C. glutamicum.

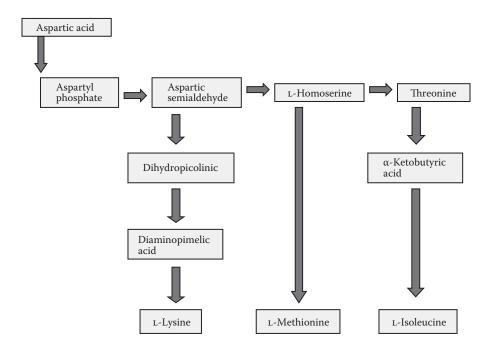


Figure 9.7 Production of L-lysine by the auxotrophic mutants (one-step process).

The isolation of auxotrophs can be carried out by using penicillin selection techniques followed by ultraviolet radiation (Adelberg and Myers 1953; Nakayama and Kinoshita 1961). The identification of growth factors (such as vitamins and amino acids) of the auxotrophs can be made auxanographically (Pontecorvo 1949). Further advancements in the production of L-lysine were achieved by introducing analogue-resistant C. glutamicum and B. flavum. An L-lysine analogue, S-(2-aminoethyl)-L-cysteine (AEC), is the best example of developing such analogue-resistant strains. The development of an AEC-resistant strain is primarily based on the treatment of a suitable homoserine auxotroph in a suitably complete culture medium containing meat extract, peptone, yeast extract, sodium chloride, and agar. The medium is supplemented with N-methyl-*N*-nitro-*N*-nitrosoguanidine in phosphate buffer. The treatment time is short (~30 min) and at a very low temperature (0°C). This is followed by the treatment of cells with AEC after washing with saline in a minimal medium supplemented with glucose as a carbon source, inorganic salts, and amino acids, for example, L-threonine and DL-methionine. It is now evident that auxotrophy and resistance to feedback inhibition, when genetically combined into a single strain (i.e., mutants resistant to AEC), results in increased production of L-lysine. The overproduction of L-lysine might be due to the starvation of threonine, which decreases the feedback inhibition of aspartate kinase, the first key enzyme in L-lysine biosynthesis (Figure 9.6). The genetic induction of an auxotropic or regulatory mutant that can overproduce L-lysine is of course extremely important. However, at the same time, process optimization (especially the medium composition and the addition of certain growth factors and culture conditions) is also indispensable for the commercial production of L-lysine.

The worldwide increased demand for L-lysine has certainly created a challenging situation for researchers to further improve the existing strains or develop new genetically modified organisms capable of producing exceptional quantities of L-lysine. The major producers of L-lysine in the world are Ajinomoto (with manufacturing facilities in Japan, France, Italy, United States, Brazil, China, and Thailand), Archer Daniels Midlands (United States), Changchun Dacheng (China), Cheil Jedang (with facilities in China, Indonesia, and Brazil), Global Biochem Technology (China), and COFCO Biochemical (China). In terms of tonnage, these major manufacturers along with other companies produced 1,755,300 tons of L-lysine in 2011 and the forecast for 2013 is around 2,376,300 tons, which is expected to reach 2,518,000 tons, corresponding to around US\$5.9 billion by the end of 2018. China is the biggest producer and will continue to dominate as the world's largest L-lysine-producing country accounting for approximately 65% of the world lysine production capacity (IHS Chemical 2012; Wippler 2011).

Based on the demand for L-lysine, which is still increasing, meeting the production demand with cost-effective raw materials and recovery processes is quite challenging and will play a major role in its success. In the fermentative production of L-lysine, all raw materials are natural or biologically available substances. No harmful by-products have been found in L-lysine fermentation. On the contrary, useful substances remain in the spent-broth from which many by-products could be recovered. The spent-broth still contains various useful substances including organic and inorganic nitrogen compounds, phosphorus compounds, and potassium salts, which could be used as animal feed additives or fertilizers. A recent process that involves the spray-drying of the whole fermented broth at the end of the incubation period has also been reported to produce feed-grade L-lysine hydrochloride very successfully (Eggeling and Sahm 2011). This will certainly reduce the cost of production of feed-grade L-lysine as well as the disposal of waste material.

### 9.3.2 Biotechnological approach and production

Despite tremendous advancements in the production of L-lysine, researchers are still trying to develop a cost-effective fermentation process using new biotechnological methodologies. Also, biotechnology companies are continuously seeking novel research developments and trying to use multifaceted management models and business approaches toward achieving market leadership in the field of L-lysine and other amino acids production. Along with highly productive strains, the use of cost-effective raw materials and recovery processes has now become highly essential for different manufacturers to compete and stay in the market. Genetically modified strains, which have been introduced for the production of L-lysine, can produce as much as approximately 170 g/L. With the use of metabolic engineering approaches, more advanced and targeted developments can be made to further improve the L-lysine biosynthetic pathway and its secretion. Control over these dynamics will certainly warrant effective precursor delivery and ensure the energy requirement of the cells. It has been indicated that along with the genetic engineering of central metabolic pathways, biosynthetic pathways, and transport systems, energy metabolism and osmoregulation are also gaining interest among researchers as new targets to be engineered for the overproduction of L-lysine. Furthermore, strains other than C. glutamicum, such as Bacillus methanolicus (methylotropic bacteria) are also gaining interest for the production of L-lysine using methanol and lignocellulose as an alternate raw material (Brautaset and Ellingsen 2011).

With the progress in molecular biology and genetic engineering, some focus has been driven to ensure an optimal carbon and energy flow within the central metabolism (CCM) of bacterial cells to achieve optimized metabolic production. A specific example is that of *C. glutamicum* in which the CCM involves glycolysis, the pentose phosphate pathway, and the TCA along with the anaplerotic and gluconeogenic reactions. It is expected that the in-depth knowledge of CCM along with the specific enzyme activities of the pathways will certainly provide an opportunity to create cell factories ideal for the production of not only L-lysine but also some other industrially important metabolites as well (Papagianni 2012).

In the industrial production of L-lysine, right from the selection of bacterial strains, the media composition, process optimization, and recovery of the end product have a great significance on the overall yield and economics of the process. A balance between carbon and nitrogen sources along with the nutritional requirements of the strain and the addition of certain precursors should be taken into consideration. In addition, a sufficient supply of oxygen to satisfy the cell's oxygen requirement is highly essential for increased yield of L-lysine. On an industrial scale, submerged fermentation is highly effective and is the most widely used method worldwide. The major raw materials such as starch, glucose, molasses, and others used in L-lysine fermentation are quite common and are used in other fermentation industries as well. The most commonly used organism for L-lysine fermentation is *C. glutamicum*. Therefore, to understand its production, a specific example has been mentioned below using a genetically modified strain of *C. glutamicum*.

Organisms are usually maintained at a low temperature, around 6°C to 8°C, on nutrient agar slants on a monthly transfer schedule. A 24-h culture, produced by subculturing from stock culture, is usually used for seed, followed by fermentation. A synthetic or natural medium that has the ability to fulfill the requirements of the particular strain can be used. In commercial-scale starches, molasses and glucose are mostly used as the carbon source. Care must be taken to create a balance between carbon and nitrogen sources such as corn steep liquor, soybean cake acid hydrolysate, yeast extract, peptone, and the like, along with the addition of inorganic salts such as  $KH_2PO_4$ ,  $K_2HPO_4$ ,  $MgSO_4/7H_2O$ , FeSO<sub>4</sub>·7H<sub>2</sub>O, ZnSO<sub>4</sub>·7H<sub>2</sub>O, MnSO<sub>4</sub>·7H<sub>2</sub>O, (NH<sub>4</sub>)<sub>2</sub> SO<sub>4</sub>, and others. Other necessary nutrients are biotin and vitamin B<sub>1</sub>. In some cases, the production of L-lysine can be enhanced by using the leucine fermentation liquor (0.2%–15%) in the production medium. In most cases, the optimum pH of the medium has been recorded as 7.2 and temperature at 30°C. The seed stage cultivation requires around 24 h, whereas the fermentation stage is complete by approximately 96 h. After this, harvesting is done and the product, L-lysine, is recovered using some suitable and economical method. More details on the subject can be obtained through the case study available at the end of the chapter.

### 9.3.3 Industrial application and therapeutic role

L-Lysine has one of most diversified applications and roles in feed, food, and pharmaceutical industries. It is the most important amino acid for monogastric animals and is regarded as the first limiting amino acid for pigs and the second limiting amino acid for poultry. Therefore, it is an important additive to animal feed for optimizing the growth of pigs and chickens. The metabolic fate of L-lysine is highlighted in Figure 9.8.

In the food industry, L-lysine is used in a number of dietary or nutritional supplements that are popularly used by athletes, weight lifters, bodybuilders, and even some individuals to boost their energy level and protect their muscles from deterioration. Nutritional supplements with a high quantity of L-lysine are available in various dosage forms such as syrups, film-coated tablets, capsules, and powder for instant drinks. L-Lysine is required by the body to synthesize L-carnitine, which is a substance required for the conversion of fatty acids into energy. L-Lysine also helps in calcium absorption and collagen formation, which are important for muscle and bone health. It also supports or acts as a precursor in the synthesis of enzymes, antibodies, and some hormones as well.

Additionally, L-lysine is also recommended for the treatment of some viral infections, for example, herpes simplex, cold sores, shingles, and human papillomavirus infections such as genital warts and genital herpes. It has also been reportedly used in the management of migraines and in some other types of pain and inflammation. Its deficiency may cause severe health problems, growth and development problems, formation of kidney stone, low thyroid hormone production, asthma, and chronic viral infection. Symptoms include nausea, fatigue, dizziness, anemia, loss of appetite and energy, inability to concentrate, irritability, bloodshot eyes, hair loss, growth retardation (especially in children), and reproductive disorders. The reported therapeutic dosage of L-lysine for the maintenance of an antiherpes effect is 500 to 1500 mg, and up to 3000 mg in active stages in a divided dosage with a low-arginine diet. The human body's daily requirement for L-lysine is 103 mg/ day for infants, 1600 mg/day for children, 800 mg/day for adult males, and 500 mg/day for adult females (Balch and Balch 1990; Chaitow 1985; Mahmood 1996, 2010).

## 9.4 Methionine

### 9.4.1 Historical background and new challenges

The discovery of glutamic acid-producing bacterium, *C. glutamicum*, eventually led to fermentation processes for producing various amino acids. However, during the early phase, not much attention was given to the fermentative production of methionine. There is a severe dearth of scientific publications relating to methionine production by microorganisms. Until 1967, very few references were available on the subject. In this period, a leucinerequiring strain of *Ustilago maydis* was reported to accumulate 6 g/L of methionine on

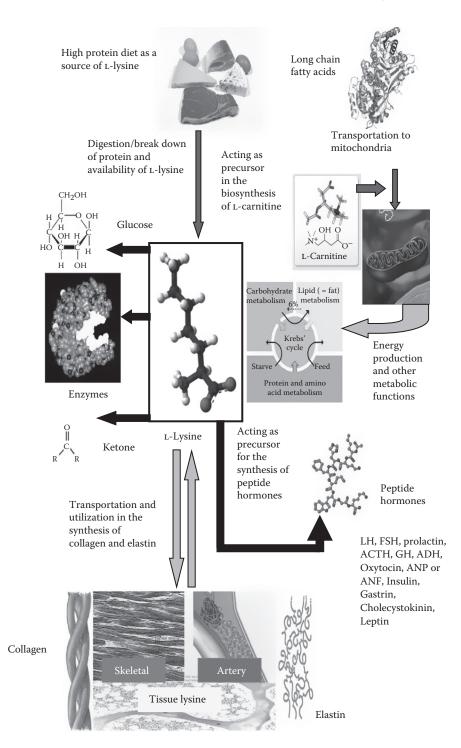


Figure 9.8 Metabolic fate: application and role of L-lysine.

a synthetic medium, and a strain of *Pseudomonas* sp. G-132-13 accumulated 13.2 g/L of methionine (Dulaney 1967).

During 1990 to early 2001, a number of constructive contributions were made by various researchers relating to the commercial production of methionine (Odunfa et al. 2001; Pham et al. 1992; Umerie et al. 2000). However, the fermentative process still needs more elaborate studies for its reproducibility and better yield before being implemented on a large scale. Currently, methionine is produced either by chemical synthesis or by enzymatic hydrolysis of proteins. Both processes are quite expensive. Chemical synthesis delivers a mixture of dextro and levo rotatory methionine, whereas enzymatic hydrolysis of proteins produces a complex mixture from which methionine needs to be separated through an expensive recovery process. Methionine isomers produced through chemical reactions can be resolved with the help of fungal enzyme aminoacylase using continuous-flow immobilized enzyme bioreactors (Tosa et al. 1967). Nevertheless, the chemical manufacturing process is still not suitable because of the hazardous nature of chemicals such as acrolein, methyl mercaotan, and ammonia, which are used during different manufacturing stages (Fong et al. 1981).

It is really challenging for the researchers as well as the manufacturers to overcome the problems associated with chemical synthesis. They need to find an optimum solution for these problems and look into the introduction of fermentative production using genetically modified organisms on a commercial scale. In bacteria, like other essential amino acids, methionine is synthesized from oxaloacetate-derived aspartate. Along with L-lysine, methionine is also a dominant amino acid used in animal feed. However, it is almost half of the L-lysine market volume. The global methionine market for animal feed has been reported as 850,000 tons in 2011, and is worth US\$2.85 billion (Roquette 2012). Two big companies, Cheil Jedang (South Korea) and Arkema SA (France), are establishing a huge plant in Malaysia to manufacture biomethionine. The operation is expected to start in the latter part of 2013 and will produce 80,000 tons per year (Saidak 2012).

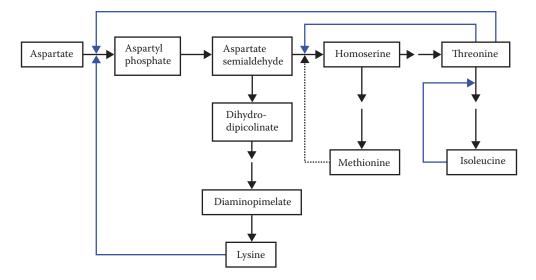
### 9.4.2 Biotechnological approach and production

Biotechnological tools have been applied for the production of methionine through auxotrophic, regulatory, or auxotrophic regulatory mutants. Auxotrophic mutants are generally less feasible for methionine (branched pathway amino acids) and therefore, auxotrophic regulatory are most suitable, because methionine itself will not inhibit or repress its own production. Microorganisms such as *Brevibacterium heali* can accumulate higher quantities (up to 25 g/L) of methionine (Mondal et al. 1994a,b). Regulation of methionine biosynthesis in *Corynebacterium* and *Brevibacterium* is shown in Figure 9.9.

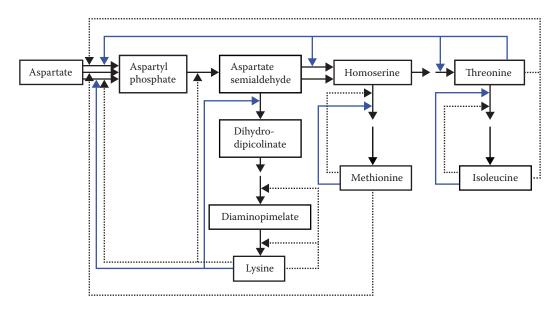
Regulatory mutants are also used for producing methionine (Kumar et al. 2003; Kumar and Gomes 2005). Recently, a new approach to screening for methionine-producing bacteria was reported using methionine auxotrophs of *E. coli* (Ozulu et al. 2012), whereas another study relating to the production of L-methionine using *B. cereus* isolated from soil samples was also reported (Dike and Ekwealor 2012). The regulation of methionine biosynthesis in *E. coli* is shown in Figure 9.10.

In *Corynebacterium* and *Brevibacterium*, the biosynthesis mechanism for methionine production is quite simple (Figure 9.9) as compared with the biosynthesis mechanism in *E. coli* (Figure 9.10) in which all enzymes are inhibited or repressed by the end product.

Most researchers reported approximately 10% glucose or 5% maltose as the carbon source supplemented with inorganic salts and different concentrations of biotin and vitamin  $B_1$  for the fermentative production of L-methionine (Banik and Majumdar 1975; Kase



*Figure 9.9* Regulation of methionine biosynthesis in *C. glutamicum*. The blue arrows show inhibition and dotted arrows represent repression.



*Figure 9.10* Regulation of methionine biosynthesis in *E. coli*. The blue arrows show inhibition and dotted arrows show repression.

and Nakayama 1975). Recovery of the product is based on treatment with ion exchange resins. Crystalline methionine can be obtained by concentrating under vacuum, treating with absolute alcohol, and drying at a low temperature of approximately 4°C for 24 h (Kumar and Gomes 2005).

#### 9.4.3 Industrial application and therapeutic role

The application of methionine is mostly reported in some pharmaceutical and livestock formulations. Because it is an essential amino acid, it is therefore required in the diets of humans and in livestock. Methionine is an excellent natural lipotrophic agent that processes and eliminates fats from the liver and acts as a natural detoxifying agent, removing heavy metals from the body and excess histamine from the brain. In addition, it has antioxidant properties as well and thus protects the body against free radicals (Chaitow 1985). Plant proteins are frequently deficient in methionine and, consequently, an exclusively vegetarian diet may fail to meet nutritional requirements. Methionine deficiency has been linked to the development of various diseases and complications. These include toxemia, childhood rheumatic fever, muscle paralysis, hair loss, depression, schizophrenia, Parkinson's liver deterioration, and impaired growth (Chaitow 1985). These symptoms appear due to deficiencies that can be overcome by incorporating methionine in the diet (Parcell 2002). Methionine is extensively used in the poultry and livestock industries (Funfstuck et al. 1997; Neuvonen et al. 1985).

## 9.5 Tryptophan

#### 9.5.1 Historical background and new challenges

Until 1967, no scientific reports were available relating to the microbial direct production of tryptophan. During this period, more attention was given by researchers looking into the possibility of tryptophan production through the conversion of anthranilic acid, coupling of indole and serine, transamination of 3-indolepyruvic acid, and conversion of  $\beta$ -indolyllactic acid (Dulaney 1967). The chemical method to produce tryptophan was the first method used for industrial-scale manufacturing (Sidransky 2001). This was followed by production through enzymatic reaction and fermentation (Potera 1991). A variety of microorganisms such as *Bacillus subtilis, Pseudomonas aeruginosa*, and others containing tryptophan synthase (TSase; EC 4.2.1.20) or tryptophanase (TPase; EC 4.1.pp.1) can be used for enzymatic reaction. *E. coli* contains both tryptophan synthase and tryptophanase (Austin and Esmond 1965; Hamilton et al. 1985). With the introduction of efficient strains of *Corynebacterium* and *E. coli*, now tryptophan is largely produced by fermentation. The market for tryptophan is not big. The 2012 figure indicates the production of 8,400 tons, and the main companies involved are Ajinomoto, Cheil Jedang, and Evonik. However, China is expected to dominate the market (Dong 2013; Gunderson and Koehler 2012).

#### 9.5.2 Biotechnological approach and production

The initial commercial production method used was based on chemical synthesis. However, by the end of 1980, other methods such as production through enzymatic reaction and fermentation were also included in view of the increased demand of tryptophan, especially by the feed and pharmaceutical industries. Thus, continuous attempts are now being made, especially utilizing the biotechnological approach to meet the desired production

demand. Among different options, metabolic engineering for increased L-tryptophan production has been given special attention. The results were encouraging when *E. coli* and *Corynebacterium* were used. The tryptophan biosynthetic pathway in *E. coli* is shown in Figure 9.11.

Overproducing strains of *E. coli* and *C. glutamicum* containing overexpressed genes have been developed, which can produce 40.2 g/L of tryptophan on suitable medium containing glucose as a carbon source in 40 h (Shen et al. 2012). A genetically engineered strain of *E. coli* was also reported to produce approximately 10.15 g/L of L-tryptophan in 48 h (Gu et al. 2012). It is believed that further research using more advanced biotechnological tools will be able to increase the productivity of tryptophan.

To understand the phenomenon of tryptophan fermentation, a brief method has been highlighted, which reflects the use of a genetically modified strain of *C. glutamicum* that is capable of producing tryptophan. Organisms are usually maintained on a monthly transfer schedule on nutrient agar slants supplemented with 0.5% sodium chloride. A seed medium containing glucose (20.0), peptone (5.0), yeast extract (1.0), sodium chloride (0.5), KH<sub>2</sub>PO<sub>4</sub> (0.5), K<sub>2</sub>HPO<sub>4</sub> (1.5), MgSO<sub>4</sub> (0.5), NH<sub>4</sub>SO<sub>4</sub> (5.0), and calcium carbonate, (5.0 g/L) can be used. Optimum pH was observed at 6.8. Cultivation can be done in Erlenmeyer flasks of 250 mL capacity containing 30 mL of the above sterilized medium. Inoculation can be done with a loopful of culture grown on nutrient agar slant for 24 h. The flask should be shaken at 200 rpm at 30°C for 24 h in an orbital shaker-incubator to prepare the seed culture.

Fermentation medium may be prepared from molasses (30%), corn-steep liquor (0.7%),  $KH_2PO_4$  (0.05%),  $K_2HPO_4$  (0.15%),  $MgSO_4$ ·7H<sub>2</sub>O (0.025%), ( $NH_4$ )<sub>2</sub>SO<sub>4</sub> (1.5%), and calcium carbonate (1%). A mixture of the following ingredients should be added in grams per liter: vitamin B<sub>1</sub> (1000.0), biotin (50.0), L-phenylalanine (200.0), and L-tyrosine (175.0). The pH of the seed and fermentation media remain unchanged, that is, adjusted to 6.8, whereas 1.0 mL of 20% silicon RD in deionized water is added as antifoam. For the laboratory scale, mini jar fermenters of 2 L capacity containing 1 L of the medium can be used. Serialization can be affected by autoclaving for 15 min at 15 psi. After cooling, the fermenter is inoculated

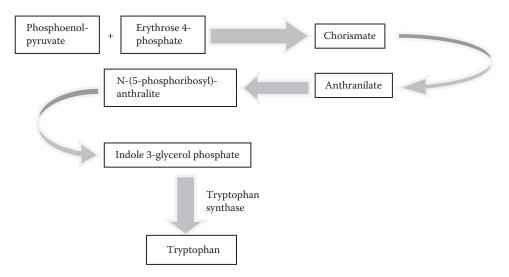


Figure 9.11 Tryptophan biosynthetic pathway in E. coli.

by one seed flask and stirred at 400 rpm with an air flow of 1.0 vvm (air volume/liquid volume per minute) at 30°C. The fermenters can be harvested after 72 h. Product recovery is usually done using ultracentrifugation at around 10,000 rpm, followed by treatment with cation exchange resin and decolorization with activated carbon. After further centrifugation, the mixture can be subjected to drying under a vacuum dryer. The product yield usually ranges between 10 and 25 g/L.

#### 9.5.3 Industrial application and therapeutic role

Tryptophan has a wide range of applications in the feed and pharmaceutical industries. As an essential amino acid with a unique indole side chain, which indicates its use as a precursor for a number of neurotransmitters in the brain, for example, serotonin, melatonin, and niacin associated with appetite, sleep, mood, and pain perception. Its application in the chemical synthesis of some antidepressant drugs and in the treatment of schizophrenia is quite prominent (Chaitow 1985; Porter et al. 2005; Van der Heijden et al. 2005).

## 9.6 Case study: production of L-lysine on glucose and molasses medium supplemented with fish meal under submerged condition in mini jar fermenter

The importance of L-lysine has been highlighted by many researchers based on its application in the food, feed, and pharmaceuticals industries. The deficiency of L-lysine can greatly affect the growth and development of bone in children; in adults, L-lysine helps calcium absorption and maintains nitrogen balance. Therefore, it may also help prevent bone loss associated with osteoporosis. Also, it plays a role in enhancing the defense mechanisms of the body, especially acting against cold sores and herpesviruses, and as an aid in antibodies, hormones, enzymes, and collagen formation; its benefits in the repair of tissues also cannot be ignored (Mahmood 2010).

The demand for L-lysine is thus continuously increasing, asking investigators to meet the challenge by increasing the production capacity using high-yield bacterial strains and the best-possible economical source of raw material, especially carbon and nitrogen for its production. This case study is also part of this program. In the current experiment, L-lysine production was evaluated with two different types of media using a homoserine auxotrophic mutant of *C. glutamicum* FRL no. 61989, which is also resistant to AEC. The auxotrophic and regulatory mutants are good options for increased L-lysine production. Isolation of the auxotrophic mutant is based on treatment of the wild strain of *C. glutamicum* with ultraviolet radiation followed by a penicillin selection method, whereas AEC-resistant strains were developed by treating the homoserine auxotrophic mutant with *N*-methyl-*N*-nitro-*N*-nitrosoguanidine, followed by growing the cells in a medium supplemented with AEC. Colonies that appeared on the surface of the agar plate during 2 to 7 days of incubation were picked up as AEC-resistant mutants (Mahmood 1996).

#### 9.6.1 Cultivation of C. glutamicum FRL no. 61989

The organism was maintained on a monthly transfer schedule on nutrient agar slants. A seed medium of following composition in grams per liter was used: glucose (30.0),

peptone (10.0), yeast extract (5.0), sodium chloride (3.0),  $KH_2PO_4$  (0.5),  $K_2HPO_4$  (1.5),  $MgSO_4$  (0.5),  $NH_4SO_4$  (0.5), sodium acetate (0.1), and 25.5 µg/L of biotin. The pH was adjusted to 7.2. Erlenmeyer flasks of 250 mL capacity containing 30 mL of the sterilized medium were inoculated with a loopful of culture grown on nutrient agar slant for 24 h. The flasks were shaken at 200 rpm at 30°C for 24 h in an orbital shaker-incubator to prepare the seed culture.

Fermentation media I and II with the following composition in grams per liter were used for comparative studies in terms of yield and cost affectivity. Medium I contains glucose (110.0), yeast extract (5.0), peptone (10.0), and meat extract (5.0). Medium II contains molasses (400.0), fish meal (25.0), and corn steep liquor (5.0). Both media were supplemented with the following ingredients in grams per liter:  $KH_2PO_4$  (0.5),  $K_2HPO_4$  (1.5),  $MgSO_4$ ·7H<sub>2</sub>O (0.5),  $(NH_4)_2SO_4$  (15.0), calcium carbonate (10.0),  $MnSO_4$ ·4H<sub>2</sub>O (0.01), ferrous sulfate  $4H_2O$  (0.01), sodium acetate (2.0), and vitamin B<sub>1</sub> (0.001). The concentration of biotin was 125.0 and 25.0 µg/L in media I and II, respectively. The pH was adjusted to 7.2 for both media, and 1.0 mL of 20% silicon RD in deionized water was added as antifoam.

One liter of media I and II was transferred separately to 2.0 L capacity mini jar fermenters and serialized by autoclaving for 15 min at 15 psi. The respective fermenters containing media I and II were inoculated with the contents of one seed flask and stirred at 400 rpm with an air flow of 1.0 vvm at 30°C. The fermenters were harvested after 96 h.

#### 9.6.2 Isolation, extraction, and purification of L-lysine

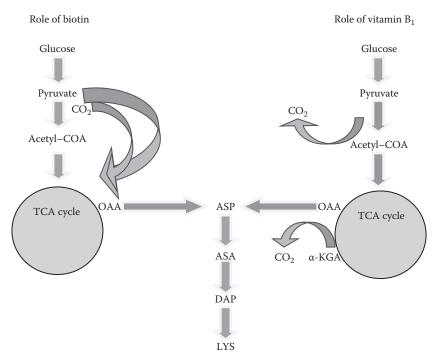
After the incubation period, the temperature of the fermenters was increased to 60°C for 15 min and the liquid culture was transferred to separate 3.0 L capacity Erlenmeyer flasks. Broths were centrifuged at 10,000 rpm using a high-speed centrifuge for 20 min to separate cells and other precipitated materials, and then subjected to ion exchange method for separation and purification. The supernatants from both fermenters were adjusted to pH 2.0 using 6N HCl and passed through a separate column 30 cm in length and 6 cm in diameter, packed with 250 g of Amberlite IR-120, a strong cation exchange resin. The rate of flow of supernatant through the column was fixed at approximately 100 drops/min to adsorb the L-lysine. After this, 250 mL of deionized water was passed twice through both columns and eluted with 500 mL of dilute ammonia. The whole liquid collected from each column was then passed through Amberlite IRC-50, a weak cation exchange resin. The column size, quantity of resin, and rate of flow of liquid were kept the same. Finally, columns were washed twice with a 250 mL quantity of deionized water and the solution was adjusted to pH 5.5 and treated with 10 g/L of activated carbon for 6 h. The slurry containing carbon and lysine was filtered under vacuum and concentrated to approximately 30% under reduced pressure and left overnight at  $10^{\circ}$ C to develop crystals of L-lysine. The material was filtered and, after washing with cold deionized water, the crystalline material was dried in a vacuum dryer.

The purity of the material was checked by performing thin-layer chromatography in a mixture of *n*-propanol and strong ammonia solution (67:33), using ninhydrin as a spray reagent as well as by comparing the infrared absorption spectrum with standard L-lysine. The purity was recorded as 98.9% and 98.2%, respectively. An assay for L-lysine was also performed and the endpoint was determined potentiometrically. The average of three experiments produced 67.80 and 65.20 g/L of L-lysine monohydrochloride by glucose and molasses medium with a corresponding yield ( $Y_{p/s}$ ) of 0.6116 gg<sup>-1</sup> (61.6% conversion efficiency) and 0.592 gg<sup>-1</sup> (59.2% conversion efficiency), respectively.

# 9.6.3 Effect of media composition and fermentation parameters on *L*-lysine production

A complex microbiological process is observed to be involved in the fermentation of L-lysine, which is greatly influenced by several biochemical and physical parameters. Therefore, in-depth knowledge of bacterial physiology and biochemistry is essential to achieve high yield of the product by choosing the most suitable carbon and nitrogen source along with the fermentation parameters optimization. In the present case study, after several sideline studies, the nutritional requirement and fermentation parameters were established for a strain of *C. glutamicum* FRL no. 61989 and accordingly applied for a final comparative study to observe the production capacity on glucose and molasses medium. The strain FRL no. 61989 was developed by treating a homoserine auxotroph with AEC.

During industrial production of L-lysine, the choice of raw material, especially the carbon and nitrogen source, largely depends on economic consideration. Therefore, it is essential to determine the optimum level of carbon and nitrogen source depending on the nutritional characteristics of the production culture. In the present study, a method that utilizes shake flask experiments with medium containing different concentrations of sugar supplemented with a sufficient quantity of nitrogen source to satisfy the demand imposed by increasing the carbon level was used. An analysis of the amount of L-lysine produced and the amount of sugar remaining after the process provides the data necessary to calculate the percentage yield of product. The sugar concentration, which provides the maximum percentage yield of L-lysine, can be treated as the optimum concentration of sugar to further proceed (Daoust 1976). In addition, the overall method used for the extraction and purification of the product including the fermentation time, yield, and treatment of waste material should also be taken into consideration. Similarly, the optimization of



*Figure 9.12* Effect of biotin and vitamin B<sub>1</sub> on L-lysine production.

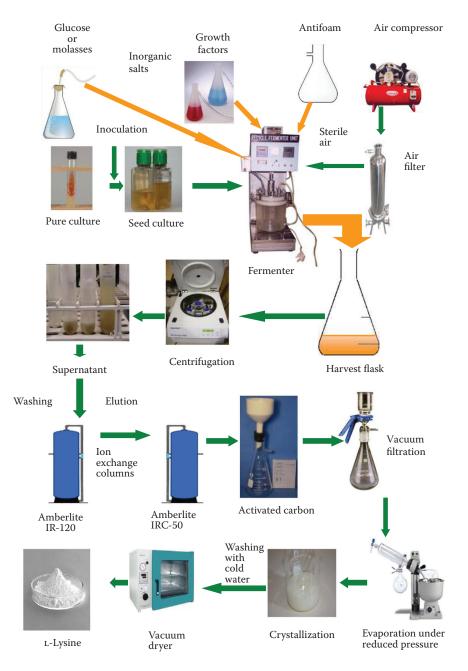


Figure 9.13 Laboratory-scale production of L-lysine.

culture conditions in relation to oxygen supply and carbon dioxide removal have played a key role in the scale-up and economic production of L-lysine. These factors determined both the rate of cell growth and product formation.

Because of the present case study, our finding is that the yield of L-lysine is more or less similar on glucose and molasses medium. Therefore, comparing the cost of glucose and molasses, the composition of medium II justified the use of molasses as a carbon source for the production of L-lysine. Furthermore, because molasses contain a sufficient quantity of biotin, the cost of the product can therefore be further reduced by decreasing the concentration of biotin in the molasses medium. Similarly, the use of fish meal as a nitrogen source was also justified in view of the high cost of soybean protein acid hydrolysate, yeast extract, peptone, and meat extract. Hence, a substantially less expensive L-lysine fermentation can be carried out using such reasonably low-cost materials.

The concentrations of biotin and vitamin  $B_1$  were established as 125 µg/L and 1.0 mg/L, respectively. The roles of biotin and vitamin  $B_1$  in amino acid fermentation are quite established. Biotin is reported to be involved in the oxidation of glucose and in the synthesis of proteins as well as in cell permeability (Tosaka et al. 1979), and as a carbon dioxide carrier of covalent bonds (Moss and Lane 1971). Furthermore, enzymes containing biotin provide aspartic acids, an important intermediate that leads to the synthesis of L-lysine (Tosaka and Takinami 1978). Vitamin  $B_1$  is reported to take part in the oxidative decarboxylation of pyruvate and  $\alpha$ -ketoglutaric acid in the bacterial system, leading to increased formation of acetyl-CoA and oxaloacetic acid, thus taking part in the increased formation of L-lysine (Koike and Reed 1960; Mahmood 1996). The effect of biotin and vitamin  $B_1$  is highlighted in Figure 9.12. Also, the overall production mechanism of L-lysine on the laboratory-scale is presented in Figure 9.13.

## 9.7 Conclusion

The microbial production of amino acids has gained significant attention in recent years because of its extensive application in food, feed, and pharmaceutical industries. Therefore, evaluation of the topic in view of recent biotechnological advancements is fully justified for increased production to meet the market demand. Some very specific amino acids such as glutamic acid, lysine, tryptophan, and methionine were selected for review in this chapter, and which are of considerable importance because of the role they play in our daily lives. It is expected that the selection of cost-effective raw materials and the utilization of a biotechnological approach will certainly bring a revolution with the introduction of new microbial strains capable of producing increased quantities of the desired amino acids. Also, an in-depth knowledge of submerged fermentation for the production of amino acids will have a significant contribution. With this approach, the chapter has been written to provide comprehensive information on the production, application, and market dimensions covering four important amino acids for both students and researchers interested or working on such topics. It is expected that the information cited will help in analyzing and understanding the dynamics of the topic and to overcome the challenges expected during process design and optimization for amino acid fermentation.

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## chapter ten

# Probiotics

## The possible alternative to disease chemotherapy

#### Adel M. Mahasneh and Muna M. Abbas

#### Contents

10.1	Introduction	. 213
	History of probiotics	. 215
	Functional properties of probiotics	. 216
	10.3.1 Tolerance to gastric transit and bile salt	
	10.3.2 Cholesterol reduction	. 216
	10.3.3 Antimicrobial activity	. 217
	10.3.4 Antibiotic susceptibility	. 219
	10.3.5 Adherence to intestinal epithelial cells	
10.4	Mechanisms of action of probiotics	
10.5	Oriental sources of probiotic strains	.222
10.6	Specific uses of probiotics	.223
10.7	Safety of probiotics	. 226
10.8	Case study: Characterization of probiotic lactobacilli isolates from fermented	
	camel milk	.226
	10.8.1 Detection of antibacterial activity of the bacterial isolates	. 226
	10.8.2 Cholesterol-lowering effect	.228
10.9	Conclusions	.229
Refe	rences	.230

## 10.1 Introduction

"Let food be thy medicine and medicine be thy food," as Hippocrates said, is the principle of today (Suvarna and Boby 2005). Probiotics are one of the functional foods that link diet and health. The term *probiotic* is derived from the combined Greek/Latin word "pro" and "bios," meaning for life (Chuayana et al. 2003; Gupta and Garg 2009). Probiotics are living microorganisms that, administered in adequate amounts, confer a health benefit to the host (Ayeni et al. 2011; de Los Reyes-Gavilán et al. 2011; Gupta and Garg 2009; Vijayendra and Gupta 2012).

The concept of probiotics was first postulated in 1908 by the Russian scientist and Nobel Prize winner Elie Metchnikoff who suggested that the long life of Bulgarian peasants resulted from their consumption of fermented milk products (Mercenier et al. 2002; Sonal et al. 2008; Tannock 2003). Metchnikoff thought that when the fermented milk products were consumed, the fermenting *Lactobacillus* positively influenced the microflora of the gut, decreasing toxic microbial activities there (Chuayana et al. 2003; Mercenier et al. 2002; Ouwehand et al. 2011). In 1965, the term "probiotic" was first used by Lilly and Stillwell for describing "substances secreted by one organism which stimulates the growth of another" (Gupta and Garg 2009; Saraf et al. 2010).

The concept of probiotics has been expanded to include bacteria from intestinal origin, those that are expected to beneficially affect the host by improving the intestinal microbial balance and hence are selected as probiotics besides the bacteria isolated from fermented dairy products (Ishibashi and Yamazaki 2001; Moreira et al. 2005; Zeng et al. 2010).

Lactic acid bacteria (LAB) are the most common type of microorganisms used as probiotics (Sonal et al. 2008). Strains of the genera Lactobacillus and Bifidobacterium are the most widely used and commonly studied probiotic bacteria (Ayeni et al. 2011; Yateem et al. 2008). Lactobacilli are Gram-positive, nonspore-forming rods, catalasenegative, and usually nonmotile and do not reduce nitrate (Bernardeau et al. 2008; Gill and Prasad 2008; Singh and Sharma 2009). They are either homofermentative or heterofermentative, aerotolerant or anaerobic, aciduric or acidophilic, and have complex nutritional requirements (Bernardeau et al. 2006). Lactobacilli are found in a variety of habitats, including human and animal mucosal membranes, on material of plant origin, in sewage, and in fermented milk products and spoiled food (Chiang and Pan 2012). The species of Lactobacillus most frequently used as probiotics are Lactobacillus acidophilus, Lactobacillus casei, Lactobacillus lactis, Lactobacillus helviticus, Lactobacillus plantarum, Lactobacillus bulgaricus, Lactobacillus rhamnosus, Lactobacillus johnsonii, Lactobacillus fermentum, Lactobacillus brevis, and Lactobacillus salivarius (Ishibashi and Yamazaki 2001; Sonal et al. 2008). Bifidobacterium genera include species of Bifidobacterium animalis, Bifidobacterium bifidum, Bifidobacterium longum, Bifidobacterium lactis, Bifidobacterium infantis, and Bifidobacterium breve, which are used as probiotics (Bhadoria and Mahapatra 2011). However, species belonging to the genera Enterococcus and Streptococcus are also considered as probiotic microorganisms (Ljungh and Wadström 2006; Saraf et al. 2010). Other bacteria that are nonlactic acid producers and considered as probiotics with less potential are isolates from the genus Propionibacterium and the genus Bacillus such as Bacillus lausii, Bacillus pumilus, Bacillus coagulans, Bacillus subtilis, and Bacillus cereus var. vietnami (Mahasneh and Abbas 2010; Marianelli et al. 2010; Saraf et al. 2010; Vinderola and Reinheimer 2003). Escherichia coli strain Nissle 1917 is also one of the nonlactic acid-producing bacteria used as a probiotic (Sonal et al. 2008). The yeasts Saccharomyces boulardii, Saccharomyces cerevisiae, and Kluyveromyces marxianus have been used as probiotics (Gupta and Garg 2009; Kurugol and Koturoglu 2005; Ljungh and Wadström 2006; Maccaferri et al. 2012). In addition, the filamentous fungi Aspergillus oryzae has been used as a probiotic (Parvez et al. 2006). The microorganisms used in probiotic preparations should be chosen according to several criteria, such as being nonpathogenic and nontoxic (Parvez et al. 2006; Suvarna and Boby 2005; Vijayendra and Gupta 2012). Probiotic preparations should be "Generally Recognized as Safe" (GRAS; Gupta and Garg 2009). Even if some of the microorganisms used in probiotics preparations are known to be GRAS or have the "Qualified Presumption of Safety" (QPS) status, the working groups of the WHO/FAO recommend proof that a given probiotic strain is safe (Ayeni et al. 2011). Moreover, some probiotics' safety have not been fully studied and understood scientifically. More information is especially needed on how safe they are for infants, children, the elderly, and those with compromised immune systems (Mahasneh and Abbas 2010). On the other hand, human origin is thought to be important for hostspecific interactions by the probiotics, although many of them are not of human origin (Dunne et al. 2001; Ouwehand et al. 2002). Probiotics should also produce a beneficial effect on the host by changing the composition of the normal intestinal microflora from

a potentially harmful composition toward a microflora that would be beneficial for the host (Ouwehand et al. 2002, 2011; Saraf et al. 2010; Suvarna and Boby 2005). Another important criteria that should be included is the ability to survive transit through the gastrointestinal tract (GIT), this means having the ability to resist the low pH of the stomach, bile, and pancreatic enzymes (Crittenden et al. 2005; Liong and Shah 2005; Ouwehand et al. 2011; Saito 2004; Sonal et al. 2008). Potential probiotics need to have technological properties so that they can be cultured on a large scale, have an acceptable shelf life, and in case of application in fermented products, to be of good taste by the production of flavoring compounds and other metabolites that will provide a product with the organoleptic properties desired by the consumer (Ouwehand et al. 2002; Parvez et al. 2006; Saraf et al. 2010; Sonal et al. 2008; Vijayendra and Gupta 2012). Effective properties also include viability during storage and use, although the requirement of viability has been questioned (Izquierdo et al. 2008; Sonal et al. 2008), because nonviable probiotics have been found to have a longer shelf life than the viable ones and are easier to store and transport, and exhibit less interaction with other compounds of food products during storage (Ouwehand et al. 2002; Tanaka et al. 2011).

### 10.2 History of probiotics

The history of good health associated with living microorganisms in food, mainly LAB, is as old as civilization. In a Persian version of the Old Testament (Genesis 18:8), it states that "Abraham owed his longevity to the consumption of sour milk" (Schrezenmeir and de Vrese 2001). In 76 BC, the Roman historian Plinius recommended the administration of fermented milk products for gastroenteritis treatment (Reddy et al. 2010).

The Ukrainian-born biologist and Nobel Prize winner Elie Metchnikoff was perhaps the first researcher to propose that fermented dairy products have beneficial properties (Azizpour et al. 2009). In 1894, he showed that cholera could be prevented by the presence of antagonistic organisms in the intestine (Reddy et al. 2010). In 1906, Henry Tissier, a French pediatrician, observed that children with diarrhea had in their stools a low number of bacteria compared with that in healthy children (Oyetayo and Oyetayo 2005). He suggested that these "bifid" bacteria could be administered to patients with diarrhea to help restore a healthy gut flora.

In 1908, while Metchnikoff was working at the Pasteur Institute in Paris, after discovering *L. bulgaricus*, he developed a theory that LAB present in Bulgarian yoghurt and in the GIT of Bulgarian farmers could, by preventing putrefaction, prolong life. This was based on his observation that Bulgarians lived longer than other people (Soccol et al. 2010). He devoted the last decade of his life to the study of lactic acid-producing bacteria as a means of increasing human longevity. The concept of probiotics was then coined and a new field of microbiology was opened and developed to what we know now as probiotic science as an offshoot of microbial biotechnology.

The term probiotic was first used by Lilly and Stillwell in 1965 to describe "substances secreted by one organism, which stimulates the growth of another" (Gupta and Garg 2009).

In 1971, Sperti applied the term to tissue extracts that stimulate microbial growth (Schrezenmeir and de Vrese 2001). In 1974, Parker was the first to use the term probiotic in the sense that it is used today: "Organisms and substances which contribute to intestinal microbial balance" (Fioramonti et al. 2003). Fuller (1989) broadened the definition of probiotics by including their effect on animals, in which he described probiotics to be "a live microbial feed supplement which beneficially affects the host animal by improving its intestinal microbial balance" (Goel et al. 2006).

## 10.3 Functional properties of probiotics

#### 10.3.1 Tolerance to gastric transit and bile salt

To secure functionality at best, the probiotic bacteria should overcome the many barriers of the host body. These barriers start with the oral cavity enzymes such as lysozyme, and the extremely low pH of the stomach, which may be as low as 1.5 (Bhadoria and Mahapatra 2011; Singhal et al. 2010). Bile and pancreatin that bacteria face in the small intestine are also among the main barriers that probiotics must overcome to arrive active and alive at the target action sites, the small and large intestine (Ayeni et al. 2011; Both et al. 2010). To entertain and to obtain the maximum efficiency of any probiotic bacterial strain, numbers should be no less than 10<sup>6</sup> to 10<sup>8</sup> CFU/g of intestinal contents (Charteris et al. 1998; de Los Reyes-Gavilan et al. 2011).

Probiotics should be able to tolerate the bile salt in the GIT to function properly (Klayraung et al. 2008). Bile, which is a yellow to green aqueous solution of organic and inorganic compounds, plays a major role in the emulsification and solubilization of lipids (Begley et al. 2005). The physiological concentration of the bile in the small intestine lies between 0.2% and 2.0% (Klayraung et al. 2008). Bile has the ability to affect phospholipids and proteins of cell membranes and disrupt cellular homeostasis. So, the ability of bacteria to tolerate bile is considered to be essential for their survival and subsequent colonization of the GIT (Begley et al. 2005; Mourad and Nour-Eddine 2006a,b; WHO 2006). Chang and coworkers (2010) isolated a potential LAB from kimchi that was resistant to biological barriers (acid and bile salts), whereas Klayraung and coworkers (2008) obtained acid-resistant and bile-resistant lactobacilli from traditional Thai foods. Ten strains of the LAB that have been isolated from Kung-Som, naturally fermented shrimp, showed a survival rate of more than 50% in simulated gastric juice and only four of them showed a survival rate of more than 50% in simulated small intestinal juices (Hwanhlem et al. 2010). In studying traditional Mongolian dairy products, 126 strains of the 543 LAB were assumed to be tolerant to bile acid, and 114 of the 126 were tolerant to low pH (Takeda et al. 2011). Resistance of potential probiotics to the stresses of the human GIT and their beneficial effects vary greatly between strains and species depending on the source of isolation (Both et al. 2010; Bhadoria and Mahapatra 2011; de Los Reyes-Gavilan et al. 2011). For example, eight different strains of B. longum showed a great variation in their resistance to GIT juices, but none of them showed viability or ability to grow after exposure to gastric juices at pH 1.5 (Izquierdo et al. 2008).

#### 10.3.2 Cholesterol reduction

Hypercholesterolemia is a risk factor for cardiovascular diseases (Zeng et al. 2010). Liong and Shah (2005) reported that for each 1 mmol above the normal cholesterol level, the risk for coronary heart disease is approximately 35% higher, and coronary death is 45% higher. Several studies have indicated that some *Lactobacillus* species such as *L. acidophilus* could reduce total cholesterol and low-density lipoprotein (LDL) cholesterol (Gilliland et al. 1985; Sanders 2000; Vijayendra and Gupta 2012). *In vitro* studies indicated the presence of a number of possible mechanisms that have been observed for their cholesterol-lowering action of probiotic bacteria (Pereira and Gibson 2002). One of these mechanisms is cholesterol assimilation by probiotics (Margolles et al. 2003). Others include enzymatic deconjugation of bile salts by bile salt hydrolases (Zeng et al. 2010). Furthermore, incorporation of cholesterol into the membrane of cells is not excluded and finally binding of cholesterol to the bacterial cell wall (Liong and Shah 2005; Pereira and Gibson 2002). Zeng and coworkers (2010) isolated *Lactobacillus buchneri* from pickled juice with a high cholesterol-reducing rate of 43.95%. Major portions of cholesterol were removed by cholesterol assimilation and the rest were coprecipitated with deconjugated bile salt and adsorbed to the cell surface. In addition to *L. buchneri*, *L. fermentum*, some strains of *L. aci-dophilus*, *L. plantarum*, *Lactobacillus paracasei* subsp. *paracasei*, and *Lactobacillus reuteri* have been found to reduce serum cholesterol level (Cardona et al. 2000; Chiang and Pan 2012; Gilliland et al. 1985; Gilliland and Walker 1990; Pereira and Gibson 2002; Zeng et al. 2010).

Chiang and Pan (2012) reported that milk fermented by *L. paracasei* subsp. *paracasei* NTU101, *L. plantarum* NTU102, and *L. acidophilus* BCRC17010 significantly decreased the levels of serum and liver total cholesterol in Syrian hamsters when fed a high-cholesterol diet. A similar trend was observed using Sprague-Dawley rat models after supplementation with *L. plantarum* 9-41-A and *L. fermentum* M1–16 (Jiang et al. 2010; Xie et al. 2011).

#### 10.3.3 Antimicrobial activity

Potential probiotic bacteria should present their antimicrobial actions against pathogens that are found as food-borne microorganisms that might cause infections or intoxication in the GIT (Klayraung et al. 2008). Primary metabolites of LAB such as lactic acid and acetic acid, ethanol and carbon dioxide, hydrogen peroxide and diacetyl are the main causes of their inhibitory action against Gram-positive and Gram-negative bacteria (Coeuret et al. 2004; Hernández et al. 2005; Kazemipoor et al. 2012; Khay et al. 2011; Klayraung et al. 2008; Mahasneh and Abbas 2010).

Four *Lactobacillus* isolates have been found to inhibit the growth of strains of *Listeria monocytogenes*, *Staphylococcus aureus*, *Salmonella* spp., and *E. coli* at 15°C more than at 37°C possibly by producing inhibitory compounds such as organic acids, hydrogen peroxide, and bacteriocins (Coeuret et al. 2004). Kazemipoor and coworkers (2012) have isolated LAB from home-made fermented vegetables, and they exhibited a remarkable inhibitory effect against both Gram-positive and Gram-negative pathogens due to the accumulation of many primary metabolites.

The inhibitory activities of LAB against Gram-positive pathogens are thought to be due to the different bacteriocins produced (Klayraung et al. 2008). Bacteriocins produced by Gram-positive bacteria have a bactericidal or bacteriostatic effect on other species and genera, but activity is usually limited to other genera (Corr et al. 2007). Bacteriocins have attracted great interest in the food industry due to their potential application in food preservation and in providing curative properties to these foods (Khay et al. 2011; Tiwari and Srivastava 2008). The bacteriocins of LAB are arranged into four classes in which classes I and II are the most thoroughly studied (Klaenhammer 1993). Class I bacteriocins, called lantibiotics, are small peptides that contain posttransitionally modified amino acid residues such as lanthionine (Batdorj et al. 2006); class II bacteriocins are small, heat stable, nonlanthionine-containing peptides that were recently organized into subgroups; class III bacteriocins are large, heat-labile nonlantibiotics; and class IV bacteriocins are complex bacteriocins containing chemical moieties such as lipid and carbohydrate (Khay et al. 2011; Savadogo et al. 2006).

The lantibiotic, nisin, which is produced by different *Lactococcus lactis* spp., is the most thoroughly studied bacteriocin and the only bacteriocin that is applied as an additive in food worldwide (Khay et al. 2011; Savadogo et al. 2006). *L. acidophilus* produces acidophilin, lactocidin, and acidolin (Vilà et al. 2010). They have demonstrated an *in vitro* inhibitory activity against *Bacillus, Klebsiella, Pseudomonas, Proteus, Salmonella, Shigella, Staphylococcus,* and *Vibrio* species and enteropathogenic *E. coli* (Brown 2011; Vilà et al. 2010). Various

bacteriocins produced by *L. plantarum* strains have been described such as plantaricin A, B, C, and others and glycocin F (Stepper et al. 2011; Tiwari and Srivastava 2008). Strains of *Enterococcus faecium* and *Enterococcus faecalis* are known to produce bacteriocins called enterocins, and generally belong to class II such as enterocin A, L50A, L50B and P, as well as lactococcin A from *L. lactis* ssp. *cremoris* (Batdorj et al. 2006). Moreover, the production of two synergistic enterocins and three enterocins from several *Enterococcus* spp. have been reported (Basanta et al. 2008; Casaus et al. 1997; Ghrairi et al. 2007).

Tiwari and Srivastava (2008) found that bacteriocin from a rhizospheric isolate of LAB identified as L. plantarum strain LR/14, inhibited not only related strains but also other Gram-positive and Gram-negative bacteria such as S. aureus, L. monocytogenes, and urogenic E. coli. However, the inhibitory effect of nisin, bacteriocin produced by L. paracasei, thermophilin produced by Streptococcus thermophillus, and some enterocins on Gram-negative bacteria, through their synergism with other antimicrobials, have gained increased interest (Kabuki et al. 2007; Khay et al. 2011; Li et al. 2005; Tolinački et al. 2010). Ennahar and coworkers (1998) found that enterocin 81 produced by E. faecium, and isolated from cheese, exhibited a narrow spectrum against Listeria spp. including L. monocytogenes. Batdorj and coworkers (2006) characterized two bacteriocins, enterocin A5-11A and enterocin A5-11B, produced by Enterococcus durans isolated from Mongolian airag, a traditional fermented mare's milk, which exhibited an antibacterial effect on a broad spectrum of bacterial species. Khay and coworkers (2011) studied the inhibitory activity of bacteriocin-like compounds produced by LAB such as L. lactis, Lactococcus cremoris, E. faecium, and E. durans isolated from Moroccan dromedary milk and they found that these isolates exhibited a strong inhibitory activity toward pathogenic bacteria and other closely related species. Wouters and coworkers (2002) investigated the production of bacteriocins and bacteriocinlike compounds by the wild lactococci in raw milk. They found nisin, diplococcin, lactococcin, and some unidentified bacteriocin-like compounds.

*Lactobacillus gasseri, L. rhamnosus,* and *E. faecium* are considered among the potential probiotic bacteria that have been isolated from human milk (Martín et al. 2004). Nisin-producing *L. lactis* ssp. *lactis* have been isolated from human milk during the early lactation period, and its antagonistic activity was determined against *Micrococcus luteus* (Beasley and Saris 2004).

The antagonistic activity of *Lactobacillus* and *Bifidobacterium* strains against various entero- and urinary pathogens have been determined. Lactobacilli strains *L. paracasei* 8700:2, *L. plantarum* 299v, and *L. fermentum* ME-3 were the most effective against *Salmonella enterica* spp. (Hütt et al. 2006).

Todorov and coworkers (2011) studied bacteriocins that were produced by *Lactobacillus curvatus* ET06, ET30, and ET31; *Lactobacillus delbrueckii* ET32; *L. fermentum* ET35; *Pediococcus acidilactici* ET34; and *E. faecium* ET05, ET12, and ET88 isolated from smoked salmon. Testing their antagonistic activity against several strains of *L. monocytogenes* of food origin and belonging to different serological groups, showed the ability of bacteriocins to suppress the growth of all tested *L. monocytogenes*. On the other hand, *L. salivarius* UCC 118 of human origin produces a potent broad spectrum class II bacteriocin, Abp 118, which is active against *L. monocytogenes* (Dunne et al. 2001). Corr and coworkers (2007) proved the ability of *L. salivarius* UCC 118 bacteriocin to protect mice against infection with *L. monocytogenes*. Mutants of *L. salivarius* UCC 118 that are unable to produce the Abp118 bacteriocin failed to protect mice against infection with two strains of *L. monocytogenes* (Corr et al. 2007; Sherman et al. 2009). Kaktcham and coworkers (2012) reported that the inhibitory spectrum of the bacteriocins produced by *Lactobacillus* species that have been isolated from Cameroonian traditional fermented

foods such as *L. rhamnosus*, *L. fermentum*, *L. plantarum*, and *Lactobacillus coprophilus* was impressive and included both Gram-positive and Gram-negative bacteria, several of which are classified by the World Health Organization as especially dangerous such as *S. enterica* subsp. *enterica*, *B. cereus*, *Streptococcus mutans*, and *Pseudomonas aeruginosa* as well as multidrug-resistant (MDR) strains of *S. aureus* and *E. coli*. Recently, Kazemipoor and coworkers (2012) have isolated LAB from home-made fermented vegetables and their isolates showed remarkable inhibitory effect against some Gram-positive and Gram-negative pathogenic bacteria.

#### 10.3.4 Antibiotic susceptibility

Bosch and coworkers (2011) studied two *L. plantarum* strains that have been isolated from the feces of healthy children to test their response to different types of antibiotics such as ampicillin, tetracycline, chloramphenicol, clindamycin, erythromycin, and gentamycin. They found that both strains were susceptible to all the antibiotics tested. Similarly, *L. plantarum* strains isolated from fermented olives were susceptible to most antibiotics tested including ciprofloxacin, rifampicin, clindamycin, vancomycin, chloramphenicol, ampicillin, and penicillin and showed resistance to others, for example, *L. plantarum* strains OL2, OL7, OL9, OL12, and OL15 showed resistance to kanamycin; OL12 and OL40 showed resistance to cefoxitine; and OL12, OL15, and OL40 showed resistance to oxacillin (Mourad and Nour-Eddine 2006b). Ortu and coworkers (2007) reported that all *Lactobacillus* strains that related to species of *L. paracasei*, *L. plantarum*, and *L. reuteri* and which were isolated from milk and Gioddu, traditional Sardinian fermented milk, were resistant to vancomycin. This resistance to vancomycin is commonly found in the genus *Lactobacillus* (Klayraung et al. 2008).

L. plantarum strains that were isolated from fermented vegetables showed variable behavior toward penicillin and tetracycline (Karasu et al. 2010). Variations in resistance of Lactobacillus to both of these antibiotics were also reported (Temmerman et al. 2003). Klayraung and coworkers (2008) isolated three strains of L. fermentum from Thai traditional food and found that they were sensitive to chloramphenicol, quinipristin, erythromycin, kanamycin, linezolid, rifampicin, streptomycin, and tetracycline but resistant to ciprofloxacin and vancomycin. Todorov and coworkers (2011) found an array of antibiotics that inhibited the growth of the probiotic strains of L. curvatus, L. fermentum, L. delbrueckii, and E. faecium with varying degrees. In this same study, amikacin, metronidazole, oxacillin, vancomycin, and nalidixic acid showed prominent activity against all probiotic bacteria tested. Lavanya and coworkers (2011) tested the response of different isolates of LAB originating from fermented milk to eight different types of antibiotics including penicillin, ampicillin, vancomycin, rifampicin, trimethoprim, kanamycin, streptomycin, and bacitracin and they found that almost all the strains tested were resistant to penicillin and only 10% were susceptible to ampicillin (β-lactam antibiotics) whereas Zhou and coworkers (2000) reported that some Lactobacillus and Bifidobacterium strains were susceptible to  $\beta$ -lactam antibiotics. Other studies showed that 20% of the tested strains were resistant to kanamycin and streptomycin, 70% were resistant to rifampicin, 20% were resistant to trimethoprim, and only 6% were resistant to bacitracin (Lavanya et al. 2011). Nawaz and coworkers (2011) isolated L. fermentum species and L. rhamnosus from breast-fed healthy babies in Pakistan. They were susceptible to penicillin, erythromycin, clindamycin, gentamycin, tetracycline, chloramphenicol, and streptomycin except L. fermentum NWS09 and NWS14, which were resistant to erythromycin and tetracycline, respectively.

#### 10.3.5 Adherence to intestinal epithelial cells

The ileum is the preferential site of colonization for intestinal lactobacilli, although they have been isolated from all portions of the human GIT (Charteris et al. 1998). To prevent the intestinal epithelium from pathogenic bacterial invasion by *Lactobacillus* strains, the latter should resist biological barriers such as gastric acid and bile salt, and then they should adhere to the host intestinal epithelium (Tsai et al. 2005). This may confer a competitive advantage and is important for bacterial maintenance in the human GIT (Chiu et al. 2008; Tsai et al. 2005). This property provides an interaction with mucosal surfaces, facilitating contact with gut-associated lymphoid tissue, mediating local and systemic immune effects, and allowing the competitive exclusion of pathogenic bacteria (Bosch et al. 2011; Chiu et al. 2008). Lactobacilli have been frequently observed to bind to epithelial cells and dissected tissue samples of the alimentary canal from humans and animals, to human vaginal epithelial cells, to intestinal mucus, to cultured human carcinomal intestinal cell lines, and to the components of the extracellular matrix (Izquierdo et al. 2008; Osset et al. 2001; Styriak et al. 2003; Todoroki et al. 2001; Yuki et al. 2000).

The role of exopolysaccharides and the involvement of lipoteichoic acids in the adherence of lactobacilli to intestinal and genital epithelia have been reported (Boris et al. 1998; Granato et al. 1999; Izquierdo et al. 2008). Furthermore, the reduced adhesiveness of lactobacilli treated with proteinases has led to the hypothesis that proteinaceous molecules mediate the adhesion of lactobacilli in the host intestine (Greene and Klaenhammer 1994; Izquierdo et al. 2008).

Few adhesions of lactobacilli have been identified and characterized at the molecular level. These include the collagen binding CnBP of *L. reuteri*, the collagen and laminin binding CbsA of *L. crispatus*, and fibronectin binding SlpA of *L. brevis* (Roos et al. 1996; Antikainen et al. 2002; Hynönen et al. 2002). Furthermore, a 29-kDa surface protein of *L. fermentum* that binds to pig small intestinal mucus and gastric mucins has been identified (Rojas et al. 2002). In preliminary studies, *L. acidophilus* expressed a 15-kDa protein that binds to fibronectin and 45 and 58 kDa proteins that bind to collagen type I (Lorca et al. 2002).

*In vitro* data have demonstrated that *B. longum* possessed high adherent activity to Caco-2 cell monolayers and suppressed the adherence of *P. aeruginosa* to the monolayers (Matsumoto et al. 2008). Izquierdo and coworkers (2008) showed the highest adhesion of *B. longum* human isolate to mucin compared with the commercial probiotic strains of *B. longum*. This adhesion was reduced by lysozyme, indicating the involvement of bacterial cell wall components in the adhesion process. Adhesion ability of different *Lactobacillus* strains to Caco-2 cells and MIM/PPK is variable depending on the strain and the cell line tested (Ortu et al. 2007). *L. plantarum*, which was isolated from pickled vegetables, was able to adhere to the mouse intestinal epithelial cells, whereas *Pediococcus pentosaceus* could not (Chiu et al. 2008). Bosch and coworkers (2011) revealed the high adhesion capacity of two *L. plantarum* strains isolated from the feces of healthy children to those of *L. rhamnosus* and *L. reuteri* commercial strains.

Tsai and coworkers (2005) found that *Lactobacillus* strains from animal origins were able to adhere strongly to human Int-407 and Caco-2 epithelial cells in addition to poultry and swine intestinal epithelium and to the columnar epithelial cells isolated from the BALB/c mouse intestine.

Ayeni and coworkers (2011) evaluated the adhesion of two *Weissella confusa* and two *L. paracasei* isolates to the epithelial intestinal cell lines Caco-2, HT-29, and HT-29-MTX, as well as to the epithelial vaginal cell line HeLa. Adhesion of the four selected isolates

to Caco-2 and HT-29 cell monolayers was poor when compared with the positive control *L. rhamnosus* GG. Adhesion percentages were better when HT-29-MTX was used as the matrix. *L. paracasei* showed similar or better adhesion than the positive control to HeLa cell monolayers, whereas the adhesion of the *Weissella* to this cell line was lower. Additionally, the ability of the four isolates to competitively inhibit the adhesion of *E. coli* to HT-29 or HeLa cell monolayers indicated that *L. paracasei* favored the adhesion of *E. coli* to the vaginal HeLa cell line, whereas the other three strains increased the *E. coli* adherence to the intestinal cell line HT-29 (Ayeni et al. 2011).

Maccaferri and coworkers (2012) demonstrated the highly adhesive ability of the probiotic yeast *K. marxianus* B0399 to Caco-2 cells and its modulation of the immune response. Tahamtan and coworkers (2011) studied the effectiveness of probiotic strains *B. lactis, B. animalis, B. longum*, and *B. bifidum* against the cytopathic effect of *E. coli* O157:H7 infection using a Vero cell line model. The bifidobacterial strains significantly reduced the adhesion of *E. coli* O157:H7 to the Vero cells and they inhibited the Shiga-like toxins (Stx-1 and Stx-2) activity produced by *E. coli* O157:H7 through neutralization of their cytotoxicity.

## 10.4 Mechanisms of action of probiotics

Probiotic bacterial strains have different mechanisms of action including enhancement of epithelial barrier integrity by increasing the secretion of mucus and by triggering inflammation in enterocytes of the intestines (Boirivant and Strober 2007; Boyle et al. 2006; Brown 2011). Chichlowski and coworkers (2007) found an increase in the number of goblet cells on chicken intestinal villi as a response to probiotics *L. casei, L. acidophilus, Bifidobacterium thermophilum,* and *E. faecium* mixture treatment, and suggested that metabolites produced during bacterial fermentation may play a role in the growth and maturation of goblet cells. Ohland and MacNaughton (2010) reported that several *Lactobacillus* species increased mucin expression in the human intestinal cell lines Caco-2 (MUC2) and HT29 (MUC2 and MUC3), thus blocking pathogenic *E. coli* invasion and adherence.

Probiotics also have the ability to enhance the intestinal barrier function through modulation of the epithelial tight junction and cytoskeleton architecture in the intestinal mucosa (Ng et al. 2009). Resta-Lenert and Barrett (2003) found that probiotic bacteria *Streptococcus thermophilus* and *L. acidophilus* enhanced the activation of tight junction proteins, therefore avoiding the development of a leaky intestine in HT29/cl.19A and Caco-2 cell lines exposed to enteroinvasive *E. coli* (EIEC 029:NM). Furthermore, live *S. thermophilus* and *L. acidophilus* alone increased transepithelial resistance accompanied by maintenance of cytoskeletal structures such as actin, zonula occludens-1, actinin, and occludin (Resta-Lenert and Barrett 2003; Ng et al. 2009). Mennigen and Bruewer (2009) reported that *L. acidophilus* increased the expression of occludin in the gut mucosa of animals with cecal ligation and perforation, leading to a reduced bacterial translocation. Similarly, Mennigen and coworkers (2009) found that VSL#3, a probiotic cocktail containing eight different bacterial strains protected the epithelial barrier by maintaining tight junction proteins, such as occludin, zonula occludens-1, and claudin-1, claudin-3, claudin-4, and claudin-5 expression and preventing epithelial apoptosis in a murine model of colitis.

Mechanisms of action also include probiotic competition with potential pathogens for nutrients and energy sources, thus preventing such pathogens from growth and proliferation in the gut. Probiotics compete with pathogens and toxins for adherence to the intestinal epithelium (Bosch et al. 2011; Brown 2011; Corcionivoschi et al. 2010).

The production of antimicrobial substances by probiotics have been reported to inhibit the growth of pathogenic bacteria such as antimicrobial effect of *Lactobacillus* species on *Helicobacter pylori* infection of the gastric mucosa (Bosch et al. 2011; Boirivant and Strober 2007). In the colon, probiotics are capable of producing short-chain fatty acids, such as acetic acid, propionic acid, and butyric acid, which lower the colonic pH, preventing overgrowth of pathogenic organisms (Vandenbergh 1993; Goel et al. 2006; Chichlowski et al. 2007).

Autoinducers are chemical signaling molecules that allow bacteria to communicate with each other and with the surrounding environment, a phenomenon called quorum sensing (Vilà et al. 2010; Sherman et al. 2009). Probiotics are thought to be able to degrade the autoinducers of pathogenic bacteria through enzymatic secretion or production of autoinducer antagonists (Brown 2011). *L. acidophilus* secretes factors that affect the quorum sensing signaling of *E. coli* O157:H7 by reducing the secretion of autoinducer 2 molecules (Medellina-Pena et al. 2007; Sherman et al. 2009). Villamil and coworkers (2012) observed that viable *L. acidophilus* and its extracellular products were able to inhibit violacein production, which is mediated by quorum sensing activity in *Chromobacterium violaceum*. Medina-Martínez and coworkers (2007) described the degradation of *N*-acyl-L-homoserine lactones (AHL) autoinducers by probiotic bacteria *B. cereus*.

Furthermore, probiotics have the ability to modulate the immune system against pathogens via antibody production and activation of lymphocytes (Ng et al. 2009). Roos and coworkers (2010) reported improved humoral immune response against *E. coli* and bovine herpes vaccines in lambs through feeding with probiotic *B. cereus* and *S. boulardii* strains.

Corcionivoschi and coworkers (2010) found that administrating *L. rhamnosus* HN001 and *B. lactis* HN019 stimulates the activity of cytotoxic lymphocytes that had a direct role in preventing the development of malignant tumors. Talib and Mahasneh (2012) reported that a combination of *B. longum* and some plant extracts were able to more efficiently target solid mammary gland tumors in mice.

Boirivant and Strober (2007) reported that *Lactobacillus* and *Bifidobacterium* probiotics induced macrophages to express more nitric oxide and inflammatory cytokines to increase antiviral activity. *In vitro* preincubation of epithelial cell line HT-29 with *L. salivarius* and *B. infantis* was able to down-regulate the proinflammatory response induced in epithelial cells by *S. typhimurium* (O'Hara et al. 2006).

#### 10.5 Oriental sources of probiotic strains

With the increased data about benefits of probiotics for human health and other applications and because most isolated and patented strains were mostly of western origin, it is greatly inviting to try and isolate such probiotics from the untapped sources exemplified by a wide variety of different fermented foods from the orient, hoping to find new and novel potential probiotic microorganisms (Figueroa-González et al. 2011; Mahasneh and Abbas 2010). For example, keshik, which is a Jordanian traditional fermented food madeup of parboiled dried wheat and buttermilk, is of interest (Tamime and O'Connor 1995). The product is similar to tarhana, Turkish traditional fermented food, which proved to be a rich source of probiotic LAB (Erdoğrul and Erbilir 2006; Sengun et al. 2009).

Among other potential foods is the fermented eggplant, locally called makdoos, which is made up of baby aubergines stuffed with ground walnut, garlic, parsley, and fermented in olive oil. Jameed, which is solar-dried curd of sheep or goat, naturally fermented milk prepared and used traditionally by Jordanian Beduins, the old desert dwellers, is another unique source of probiotic bacteria (Mahasneh and Abbas 2010). Fermented vegetables such as olives, kimchi in Korea, and khalpi in India are main sources of probiotics in addition to fermented cucumber, carrot, green beans, and cabbage (Karasu et al. 2010; Tamang 2001; Chang et al. 2010; Mourad and Nour-Eddine 2006a; Kazemipoor et al. 2012).

In Thailand there are famous traditional fermented products that are highly rich in probiotic bacteria as Kung-Som (traditional fermented shrimp), nham (traditional salt-fermented ground meat with garlic, pepper, salt, and cooked rice) and miang (fermented tea leaves; Hwanhlem et al. 2010; Klayraung et al. 2008). In the western highlands region of Cameroon, Kossam (fermented cow milk) and Sha'a (a maize-based beverage) are two traditional fermented foods that are natural reservoirs of LAB with high antimicrobial activity (Kaktcham et al. 2012). Reddy and coworkers (2007) studied the ability to isolate a potent probiotic LAB from kanjika, an Indian probiotic product, which is a lactic acid-fermented rice product and is prepared from raw material of plant origin and is devoid of dairy products.

Probiotic potential LAB were isolated from different Mongolian dairy products such as airag, a fermented mare's milk beverage; tarag, which is made from cow, yak, goat, or camel milk; and cow milk aaruul (Batdorj et al. 2006; Watanabe et al. 2008; Takeda et al. 2011).

LAB have been detected in various fermented products of camel's milk. These products vary according to the method of processing (Hassan et al. 2008). Shubat is camel's sour milk from Kazakhstan and China (Serikbayeva et al. 2005; Rahman et al. 2009). Gariss is a traditional Sudanese fermented camel milk that is widely consumed by the pastoralist communities living in arid and semiarid regions of Sudan (Abdelgadir et al. 2008; Ashmaig et al. 2009). Lehban is the fermented product from camel's milk in Syria and Egypt, whereas kefir is Caucasian fermented camel's milk (Hassan et al. 2008). Other studies on shmen or semma, a traditional butter made from camel's milk in the Sahara, Algeria, indicated the presence of different LAB genera (Mourad and Nour-Eddine 2006a,b). LAB strains have been found in suusac, a Kenyan traditional fermented camel milk product (Lore et al. 2005). Yaqoob and Nawaz (2007) reported that the most common products made from camel milk are dahi (yoghurt), lassi (sour milk), and kurth (cheese) in northeastern Baluchistan. All these products were main sources of LAB (Harun-ur-Rashid et al. 2007; Pawan and Bhatia 2007).

Some unique probiotic LAB strains *L. plantarum*, *L. pentosus*, and *L. lactis* ssp. *lactis* have been isolated from unpasteurized natural camel's milk with superior probiotic characteristics (Yateem et al. 2008). Similarly, LAB have been isolated from camel's raw milk in Morocco with the predominance of *Lactobacillus* and *Lactococcus* genera (Khedid et al. 2009). Benkerroum and coworkers (2003) also isolated different genera of LAB including *Pediococcus*, *Streptococcus*, *Lactococcus*, and *Leuconostoc* from raw camel's milk.

## 10.6 Specific uses of probiotics

Probiotics have been used for the prevention and treatment of a diverse range of disorders. Diarrhea cases are of the most common adverse events associated with bacterial infections and misuse of antibiotics (Saraf et al. 2010). McFarland and coworkers (1994) found that oral administration of the yeast *S. boulardii* significantly reduced the occurrence of diarrhea after administration of antibiotics. Other probiotics like *L. rhamnosus* GG, *L. acidophilus*, and *E. faecium* also seemed to be effective in preventing antibiotic-associated diarrhea (Saraf et al. 2010; Quigley 2011; Ouwehand et al. 2002).

Viral diarrhea, such as that induced by rotavirus, has been treated successfully by probiotics such as *L. casei* Shirota and *L. rhamnosus* GG (Quigley 2011; Ouwehand et al. 2002). A recent study demonstrated the effectiveness of *L. reuteri* in the treatment of rotaviral infection in a neonatal mouse gastroenteritis model (Preidis et al. 2012). *L. rhamnosus* strains 573L/1, 573L/2, and 573L/3 have reduced the duration of rotavirus diarrhea in children 2 months to 6 years of age (Szymański et al. 2006).

Traveler's diarrhea, which is a common health problem among travelers (Gupta and Garg 2009) and is caused by the ingestion of contaminated foods and water with pathogens such as *E. coli, Salmonella* species, and some protozoa like *Entamoeba histolytica* and *Giardia lambalia* (Lannitti and Palmieri 2010). Goel and coworkers (2006) reported that *S. boulardii, E. faecium*, and *L. rhamnosus* probiotics for the treatment of traveler's diarrhea were limited, whereas McFarland (2007) reported that *S. boulardii* and a mixture of *L. acidophilus and B. bifidum* had significant efficacy. Some studies revealed fewer or shorter episodes of traveler's diarrhea in subjects consuming the probiotic whereas others found no such effect (de Vrese and Marteau 2007).

Among other gastrointestinal disorders that have been treated with probiotics are inflammatory bowel diseases (IBD). Yan and coworkers (2011) described how a certain protein derived from *L. rhamnosus* GG reduced epithelial apoptosis and suppressed inflammation in mouse models of IBD.

VSL#3 probiotic cocktail was effective in the induction of remission among patients with ulcerative colitis, which is another form of IBD other than Crohn's disease (Bibiloni et al. 2005; Quigley 2011). This same cocktail alleviated symptoms in individuals with irritable bowel syndrome (IBS) such as decreasing flatulence, pain, and bloating (Quigley 2011; Yan and Polk 2011). Clarke and coworkers (2012) reported that *L. rhamnosus* GG variably reduced the frequency and severity of pain in children.

Saulnier and coworkers (2013) reviewed and discussed the role of gut microbiota and the implications for probiotics and prebiotics on epithelial cell function, gastrointestinal mobility, visceral sensitivity, perception, and behavior as well as their interaction with the central nervous system. Their remarks suggested the potential role of gut microbiota communities on brain functions such as psychological disorders, particularly in the field of psychological comorbidities associated with bowel disorders.

The gastrointestinal tract is a site of early HIV replication and CD4+ cell destruction, and HIV-associated enteropathy emerges as a critical outcome of the transition from HIV infection to AIDS (Yan and Polk 2011). *L. reuteri* and *L. rhamnosus* were found to resolve diarrhea and improve CD4+ cell number in patients infected with HIV (Anukam et al. 2008). Narwal (2011) reported that some *Lactobacillus* strains that were taken from the saliva of volunteers had produced proteins capable of binding mannose, a particular type of sugar found on the HIV envelope. This binding enables the *Lactobacillus* to stick to the mouth and digestive tract mucosal lining, thus initiating colonization. One strain secreted abundant mannose-binding protein particles into its surroundings, neutralizing HIV by binding to its sugar coating. They also observed that immune cells trapped by lactobacilli formed a clamp. This configuration would immobilize any immune cells harboring HIV and prevent them from infecting other cells.

*H. pylori* is a major cause of chronic gastritis and peptic ulcer and a risk factor for gastric malignancies (Gupta and Garg 2009; Medeiros and Pereira 2013). Cruchet and coworkers (2003) have shown that *L. johnsonii* La1 may interfere with *H. pylori* colonization in asymptomatic children and may be an effective alternative to modulate *H. pylori* infection. Furthermore, Hütt and coworkers (2006) found that both *L. paracasei* 8700:2

and *L. plantarum* 299v were antagonistic against *H. pylori*, which colonize the stomach mucosa.

Probiotics have been investigated for their role in the prevention of colon cancer (Saraf et al. 2010). Diets, especially those high in meat and fat or low in fiber, have a role in changing the composition of the intestinal microflora, which increases the fecal enzyme activity that converts procarcinogens into carcinogens (Ouwehand et al. 2002). Suvarna and Boby (2005) reported that *L. acidophilus* obviously reduced both the number and size of colon tumors induced by a carcinogen, 1,2-dimethyl hydrazine (DMH). Dietary administration of *B. longum* in laboratory animals significantly inhibited the incidences of colon adenocarcinomas and colon tumor multiplicity. Liong (2008) reported the effect of fermented milks on colon cancer cell growth. Milks were fermented with individual strains of *L. helveticus*, *L. acidophilus*, *Bifidobacterium*, or a mix of *S. thermophilus* and *L. delbrueckii* ssp. *bulgaricus* and when HT-29 cells were subsequently added into the fermented milk, 10% to 50% of these cells showed a decrease in growth. Goel and coworkers (2006) observed that consumption of *L. casei* would delay the recurrence of bladder tumors.

In dental practice, probiotics significantly lowered the rates of dental caries in children who were given *L. rhamnosus* GG (Goldin and Gorbach 2008). Saraf and coworkers (2010) reported the inhibitory effect of *L. rhamnosus* GG and *Bifidobacterium* isolates on salivary counts of *S. mutans* and yeast. Other studies showed the ability of *Bifidobacterium* to reduce gingival and periodontal inflammation (Reddy et al. 2010). Krasse and coworkers (2006) observed that *L. reuteri* was effective in reducing both gingivitis and plaque in patients with moderate to severe gingivitis.

Probiotic therapy has also been explored in nongastrointestinal disorders. There is ample evidence suggesting the use of probiotics to treat and prevent urinary tract infection (Velraeds et al. 1998). Reid and coworkers (2001) have shown that *L. rhamnosus* and *L. fermentum* inserted into the vagina can colonize and compete against uropathogens and reduce the risk of urinary tract infections. Petricevic and Witt (2008) found that vaginal flora can be restored after antibiotic treatment of bacterial vaginosis by exogenously applying *L. casei rhamnosus*.

Many studies have examined the efficacy of probiotics in the prevention and treatment of allergic disorders (Gill and Guarner 2004). Probiotics could have a good effect in infants at high risk of atopy and those having allergies to cow's milk (Viljanen et al. 2005), atopic eczema, and dermatitis (Sleator and Hill 2008; Prescott et al. 2005). This is linked to reports associating neonatal gut microecology with the development of atopic disease (Ouwehand et al. 2002). Therefore, it seems that intervention with probiotics before or immediately after birth is more effective in inducing immunological tolerance while the immune system is still immature, compared with older children with fully mature immune systems (Ouwehand 2007).

Moreover, LAB have antiobesity effects (Chiang and Pan 2012). Kadooka and coworkers (2010) showed that *L. gasseri* had lowering effects on abdominal adiposity, body weight, and other measures such as body mass index and waist and hip circumference. Takemura and coworkers (2010) proposed that *L. plantarum* no. 14 reduced adipocyte size in mice fed a high-fat diet.

Several studies reported the ability of probiotics to relieve constipation, prevent necrotizing enterocolitis, eliminate nasal pathogens, lower cholesterol and hypertension, and alleviate lactose intolerance. Furthermore, probiotics have been proposed to relieve the symptoms of anxiety (Azizpour et al. 2009; Culligan et al. 2009; Popova et al. 2012; Goel et al. 2006; Suvarna and Boby 2005; Gupta and Garg 2009).

#### 10.7 Safety of probiotics

Many different species and strains and preparations have been used as probiotics for the last decade with growing interest for the development of new and more active strains, and the question for their safety has increased with time (Quigley 2011; Marteau 2001). The criteria in place for the use of microorganisms as probiotics assures their safe use. However, more investigations are needed on how safe probiotics are for certain patients such as severely immunocompromised people, neonates, the elderly, and hospitalized patients, although ample studies support the safety of particular probiotic strains in particular high-risk populations (Sanders et al. 2010; Boyle et al. 2006).

Probiotics side effects, if they exist, would probably be mild digestive disturbances including gas and bloating (Mahasneh and Abbas 2010). Theoretically, probiotics may be responsible for more sophisticated and serious side effects (Marteau 2001). *Lactobacillus* species are a rare if any well-recognized cause of local or systemic infections including septicemia and endocarditis (Boyle et al. 2006). Many cases of fungemia have been reported in humans treated with the probiotic *S. boulardii* (Hannequin et al. 2000). This probably means that the risk of infection is not nil but is extremely low (Marteau 2001).

Among other probable side effects are the unhealthy metabolic activities of probiotics such as intestinal mucus layer degradation in which the accumulation of probiotics along the GIT might cause gastrointestinal disturbances including intestinal inflammation (Soccol et al. 2010; Ooi and Liong 2010).

Production of bile hydrolase and accumulation of deconjugated bile could be transformed into detrimental cytotoxic secondary bile acids, which could increase the risk of gastrointestinal diseases (Ooi and Liong 2010). Probiotics may also cause too much stimulation of the immune system (Boyle et al. 2006).

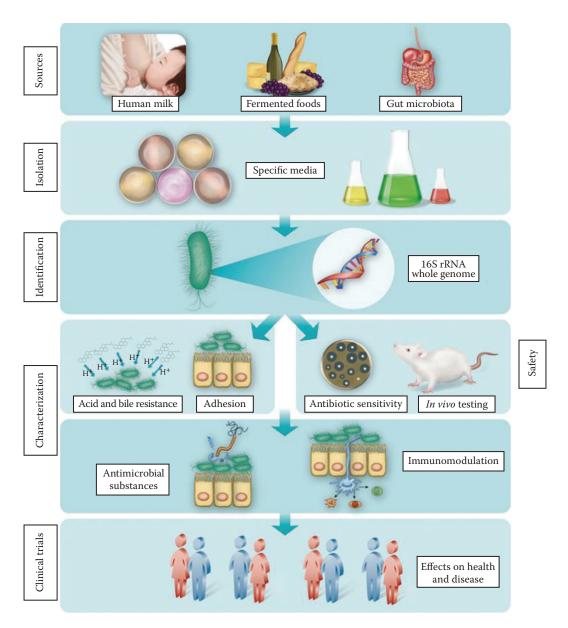
Another possible risk of using probiotics is transferring antibiotic resistance genes between microorganisms especially for more pathogenic bacteria in the intestinal microbiota (Lannitti and Palmieri 2010; Boyle et al. 2006). For example, *L. reuteri* and *L. plantarum* have been found to carry antibiotic resistance genes (Yan and Polk 2011). On the other hand, vancomycin-resistant genes of many *Lactobacillus* strains are chromosomally encoded and not inducible or transferable, which would increase the safety level of probiotic use (Marteau 2001; Boyle et al. 2006).

## 10.8 Case study: Characterization of probiotic lactobacilli isolates from fermented camel milk

In the quest for the isolation of probiotic lactobacilli from exotic sources, spontaneously fermented camel milk was used as one of the least studied oriental foods. The ongoing project uses well-documented materials and methods. Prospective probiotic candidates were isolated, identified, and tested to find out if they meet the criteria for probiotic use. Figure 10.1 shows the guiding protocol to isolate such probiotics as presented by Fontana and coworkers (2013), which could be modified to enhance any stage. In this section, our findings about antibacterial activity and the cholesterol-reducing effects of probiotic lactobacilli isolates from fermented camel milk will be described.

#### 10.8.1 Detection of antibacterial activity of the bacterial isolates

For the detection of antagonistic activities of the isolates, an agar spot procedure was used. The antibacterial activity of the selected *Lactobacillus* isolates was determined by the agar



*Figure 10.1* Flow chart describing the various steps to be followed for a bacterial strain to qualify as a novel probiotic. rRNA, ribosomal RNA. (From Fontana L. et al., *British Journal of Nutrition* 109 (2013): S35–S50.)

spot test described by Schillinger and Lücke (1989) with some modifications as follows: 5 mL of each overnight culture of *Lactobacillus* isolate were spotted onto the surface of MRS agar plates containing 0.2% glucose and then incubated under anaerobic conditions at 37°C for 48 h. An overnight culture of four indicator strains (*E. coli* ATCC 25922, *S. typhimurium* ATCC 14028, *B. cereus* toxigenic strain TS, and methicillin-resistant *S. aureus* [MRSA] clinical isolate) were grown in nutrient broth and were adjusted to 0.5 McFarland and then were diluted 1:10 using nutrient broth to reach 10<sup>7</sup> CFU/mL. Aliquots of 0.25 mL were

of the Selected Laciobactius Species					
	B. cereus	MRSA	E. coli	S. typhimurium	
M2	$28.5 \pm 0.7$	$54.5 \pm 2.1$	$28.5 \pm 0.7$	$32.5 \pm 0.7$	
M5	$36.0 \pm 1.4$	$36.5 \pm 0.7$	$34.0 \pm 4.2$	$30.0 \pm 2.8$	
M10	$54.0 \pm 1.4$	$44.0 \pm 1.4$	$39.0 \pm 1.4$	$28.5 \pm 3.5$	
M15	$42.5 \pm 4.9$	$48.5 \pm 3.5$	$36.5 \pm 0.7$	$32.0 \pm 2.8$	
M29	$41.0\pm0.0$	$42.0\pm1.4$	$36.5 \pm 2.1$	$29.5 \pm 6.3$	

Table 10.1 Antibacterial Activity (Inhibition Zones Diameter, in Millimeters) of the Selected Lactobacillus Species

*Note:* Low activity, <32 mm; moderate activity, 33–39 mm; and significant activity, >39 mm. Inhibition zone (mm) of indicator strains, mean ± S.D, *n* = 2.

inoculated into 7 mL of soft MRS containing 0.2% glucose and 0.7% agar. Inoculated soft MRS agar was immediately poured in duplicate over the MRS plate on which the tested *Lactobacillus* isolate was grown. The plates were incubated aerobically at 37°C for 24 h.

The antibacterial activity was detected by measuring the inhibition zones around the *Lactobacillus* bacterial spots. Inhibition was recorded as positive if the diameter of the zone around the colonies of the producer was 2 mm or larger (Mami et al. 2008).

Microbial growth inhibition by different fermented camel milk *Lactobacillus* isolates: M2, *L. fermentum*; M5, *L. brevis*; M10, *L. plantarum*; M15, *L. paracasei* ssp. *paracasei*; and M29, *L. rhamnosus*, would produce inhibition zones. The inhibitory activity of the five selected isolates varied depending on the *Lactobacillus* species tested and type of host microorganisms, which included both Gram-positive and Gram-negative bacteria.

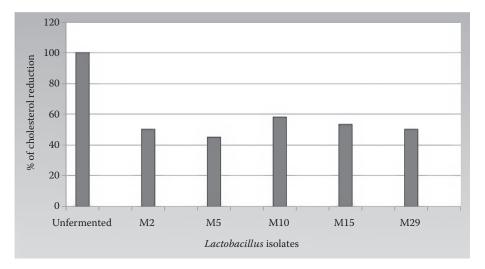
It can be seen from Table 10.1 that *Lactobacillus* isolates showed varying degrees of activity against four human pathogenic challenge indicators: two Gram-positive bacteria *B. cereus* toxigenic strain, TS and MRSA and two Gram-negative bacteria *E. coli* ATCC 25922 and *Salmonella typhimurium* ATCC 14028.

The highest activity was recorded for the isolate M2 (*L. fermentum*) against MRSA with an inhibition zone of 54.5 mm followed by M10 (*L. plantarum*) against *B. cereus* with an inhibition zone of 54.0 mm. The least inhibition zones were observed for isolate M2 (*L. fermentum*) against *B. cereus* and *E. coli* with inhibition zones of 28.5 mm as well as by M10 (*L. plantarum*) against *S. typhimurium* with an inhibition zone of 28.5 mm.

#### 10.8.2 Cholesterol-lowering effect

To screen for *Lactobacillus* with cholesterol-lowering effects, one fresh colony from each *Lactobacillus* isolate was inoculated into 5 mL MRS broths separately and incubated anaerobically for 24 h at 37°C. Then, they were inoculated in 1% MRS-THIO broth with 0.1 g/L filter-sterilized water-soluble cholesterol (polyoxyethanyl cholesteryl sebacate; Sigma-Aldrich) under anaerobic conditions at 37°C for 24 h. After the incubation period, cells were centrifuged at 12,000 × g at 4°C for 10 min, and the remaining cholesterol concentration in the broth was determined by using *o*-phthalaldehyde modified colorimetric method as described by Rudel and Morris (1973). One milliliter of the supernatant (broth containing the remaining cholesterol) aliquot was added with 1 mL of KOH 50% (w/v) and 2 mL of absolute ethanol, vortexed for 1 min, followed by heating at 37°C for 15 min. After cooling, 2 mL of distilled water and 5 mL of hexane were added and vortexed for 1 min.

The hexane layer of 2.5 mL was transferred into a glass tube. The hexane was evaporated from each tube at 60°C under the flow of nitrogen gas. The residue was immediately dissolved in 2 mL of *o*-phthalaldehyde reagent. The reagent contained 0.5 mg of



*Figure 10.2* Cholesterol-reducing rate of *Lactobacillus* isolates. The first column represents the unfermented MRS-THIO broth.

*o*-phthalaldehyde per milliliter of glacial acetic acid. After complete mixing, the tubes were allowed to stand at room temperature for 10 min, and then 0.5 mL of concentrated sulfuric acid was added and the mixture was vortexed for 1 min. After standing at room temperature for an additional 10 min, the absorbance was read at 550 nm against a reagent blank. The removal rate of every strain was computed by the following formula:

The cholesterol reducing rate =  $[(A_0 - A)/A_0] \times 100\%$ 

where  $A_0$ , absorbance of the unfermented broth; A, absorbance of the broth fermented for 24 h.

All *Lactobacillus* isolates (M2, M5, M10, M15, and M29) showed high ability of cholesterol reduction with no significant variation in cholesterol-reducing ability among the isolates. M10 (*L. plantarum*) had the highest cholesterol-reducing rate of 58.0% followed by M15 (*L. paracasei* ssp. *paracasei* 1) with a cholesterol reducing rate of 53.2%. M5 (*L. brevis*) had the lowest cholesterol-reducing rate of 45.0% (Figure 10.2).

## 10.9 Conclusions

Probiotics have shown their ability in promoting a wide variety of desired physiological effects for better human health. Traditional fermented foods are among the most promising sources for the isolation of unique probiotic bacteria, especially of the lactic acid group. These candidate probiotic microorganisms must meet the approved international criteria to include resistance to low pH, gastric and intestinal enzymes, and bile toxicity. The ability to adhere to epithelial cells and antimicrobial substances is also as important as the safety concerns. To substantiate the role of probiotics in disease therapy, more *in vivo* studies are needed and furthermore the need for specific designer probiotics should be thoroughly investigated and developed.

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# Microbial healthcare products

## Zafar Alam Mahmood and Saad Bin Zafar Mahmood

## Contents

11.1	Introduction	240
11.2	Antibiotics	240
	11.2.1 Historical background and new challenges	240
	11.2.2 Production and pharmaceutical presentation	
	11.2.3 Biotechnological approach	
	11.2.4 Therapeutic role	
	Antitumor agents	
	11.3.1 Historical background and new challenges	
	11.3.2 Production and pharmaceutical presentation	
	11.3.3 Biotechnological approach	
	11.3.4 Therapeutic role	
	Ergot alkaloids	
	11.4.1 Historical background and new challenges	
	11.4.2 Production and pharmaceutical presentation	
	11.4.3 Biotechnological approach	
	11.4.4 Therapeutic role	
	Vaccines	
	11.5.1 Historical background and new challenges	266
	11.5.1.1 Types of vaccines	
	11.5.1.2 Challenges	
	11.5.2 Production and pharmaceutical presentation	
	11.5.3 Biotechnological approach	270
	11.5.4 Preventive/therapeutic role	
	Flavonoids	
	11.6.1 Historical background and new challenges	271
	11.6.2 Production and pharmaceutical presentation	
	<ul><li>11.6.3 Biotechnological approach</li><li>11.6.4 Therapeutic role</li></ul>	
	Case study: Optimization of media composition for the production of ergot	2/4
11./	alkaloids from a mutant strain of <i>Botryodiplodia</i> sp. FRL-1955 through surface	
	culture technique	274
	11.7.1 Materials and methods	
	11.7.2 Results and discussion	
	Conclusion	
	iowledgments	
	0	276

## 11.1 Introduction

Microbial healthcare products have now gained huge application in both primary and secondary human healthcare systems irrespective of whether people are living in underdeveloped or developed countries. In the primary healthcare system, microbial products can play a role in immunization, prevention of locally endemic diseases, treatment of common diseases, and others, whereas their use and application is fully warranted in secondary healthcare systems for more organized treatment in hospitals for complicated infectious diseases and in chemotherapy. The vast majority of multinational companies are marketing a number of products of microbial origin and are generating huge revenues out of these products. The introduction of new products manufactured through cost-effective raw materials for both the primary and secondary healthcare sectors is still open and desirable both on the global and regional level. Healthcare products for human application can further be grouped under biopharmaceuticals, cosmeceuticals, and nutraceuticals, with prominent properties of prophylactic and therapeutic applications. Microorganisms have been utilized by human beings for centuries in view of their biochemical features and for the production of healthcare products. This can still be regarded as a very early phase of biotechnology, with the best possible economic use of microorganisms in food and allied industries. However, as knowledge of microbial physiology grew, the scope of utilizing microorganisms to develop healthcare products increased tremendously and their utilization was also expanded for the development of pharmaceuticals and, later on, by the personal and healthcare product manufacturing industries. The discovery of pathways for the development or biosynthesis of secondary metabolites such as antibiotics, alkaloids, flavonoids, antitumor agents, cholesterol-lowering drugs, and others, certainly provided a better understanding and platform for their role in the human healthcare system. A significant number of various products belonging to these diverse groups are now available commercially both for treatment and prophylaxis.

With the advancements and understanding in the fields of microbial physiology and molecular biology, followed by the extensive role of genetics, ultimately resulting in the isolation of DNA, has completely changed the concept and role of microorganisms in the production of healthcare products. Although the ultimate outcome of these findings resulted in the production of a host of products, the first one which came into the picture was "insulin," developed through genetic engineering. The in-depth studies involved in resolving the structure of DNA provided a great hope and toll to scientists to use advanced biotechnological techniques for the removal of DNA from one region of the genome and transport the DNA in a controlled way to another region of the same DNA, or DNA in a completely different organism for more economic gain and benefits by expanding the role and type of products in substantial quantities. In the current scenario, the use of microorganisms as biotechnological agents and the technology of profit have not only continued but have exclusively increased. Indeed, the biotechnology sector, as it is recognized today, is already a multibillion dollar sector worldwide. With these objectives, some of the major microbial healthcare products have been taken into consideration to study and to provide reasonable information to highlight their significance in the human healthcare system.

## 11.2 Antibiotics

#### 11.2.1 Historical background and new challenges

Microbial production of secondary metabolites and especially those with therapeutic importance such as "antibiotics" have been given significant attention worldwide in view

of their role in the management of infectious disease. It has been nearly a century since the first therapeutic compound "penicillin," from the fungal species, *Penicillium notatum*, followed by *Penicillium chrysogenum* (Figure 11.1), was discovered by Alexander Fleming in 1928. The actual name penicillin was given on March 7, 1929 and reported to be highly inhibitory effect against *Staphylococcus aureus*, a Gram-positive bacteria. Before this, Fleming also had the honor of discovering the antibacterial enzyme "lysozyme" in 1923. In 1924, actinomycetin was discovered, but was only used in the production of vaccines.

Despite the fact that the term antibiotics was used and noted in the literature before the discovery of penicillin (Burkholder 1952), it was not until 1942 when Selman Waksman suggested a proper definition of the term antibiotics as "a chemical substance obtained from certain microorganisms, capable of inhibiting the growth or destroying other microorganism in small concentrations." The term *low-molecular weight organic natural products* has been used in some recent literature in place of chemical substance, as originally used by Waksman (Sanchez and Demain 2011). By the end of 1940, Fleming was able to test the effect of penicillin (benzylpenicillin or penicillin G) on some other Gram-positive organisms responsible for causing scarlet fever, pneumonia, meningitis, and diphtheria and one Gram-negative organism causing gonorrhea, but no prominent place or identification was given to penicillin.

In 1941, Howard Florey and Emst Boris Chain completely revolutionized and changed the concept of infectious disease therapy with the successful commercial production of penicillin by surface culture techniques using thousands and thousands of glass bottles (Figure 11.2); this period was called the "Golden Age of Antimicrobial Therapy." This was followed by production through submerged fermentation. In 1945, Fleming, Florey, and Chain shared the Nobel Prize in Physiology or Medicine. Based on his discovery, Sir Alexander Fleming (after his death in 1955) was named one of the 100 top most famous important people of the twentieth century in 1999. Further success in the field was achieved with the introduction of the first oral penicillin (penicillin V) in 1952 by Hans Margreiter and Ernst Brandl (Greenwood 2008). The structures of penicillin G and penicillin V are shown in Figures 11.3 and 11.4, respectively.

After penicillin, subsequent research in the field of secondary metabolites, especially from fungi and actinomycetes (filamentous bacteria), have led to the discovery of a series of antibiotics for human and veterinary use, such as cephalosporins, tetracyclines,

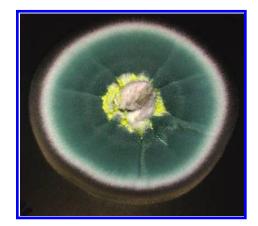


Figure 11.1 P. notatum colony.



Figure 11.2 Glass bottle used for the production of penicillin by surface culture technique.

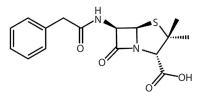


Figure 11.3 Benzylpenicillin (penicillin G).

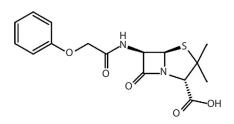


Figure 11.4 Phenoxymethylpenicillin (penicillin V).

aminoglycosides, chloramphenicol, macrolides, and glycopeptides. The actinomycetes are the most extensively studied organism and are capable of producing a large number of antibiotics (~75% out of the total), whereas its genus, *Streptomyces* has the largest contribution (~75%) and the specific species, *Streptomyces hygroscopicus* and *Streptomyces griseus*, have been observed to produce approximately 200 and 40 different antibiotics, respectively. The contribution of nonfilamentous bacteria is small, approximately 12% and that of filamentous fungi is approximately 20% out of total antibiotics reported thus far (Sanchez and Demain 2011). The discovery of tyrothricin by Dubos in 1939 has, to some extent, supported infectious disease therapy; however, the susceptible organisms are still Gram-positive. The introduction of streptomycin in 1944 certainly extended the antimicrobial spectrum as it was capable of inhibiting some Gram-negative organisms as well as *Mycobacterium* (Figure 11.5). The concept of broad-spectrum antibiotics was introduced for the first time after the

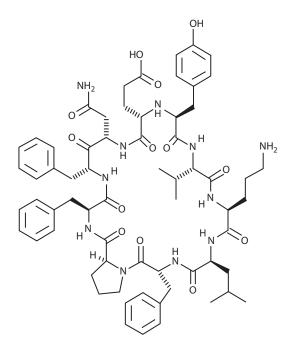


Figure 11.5 Tyrothricin.

discovery of chloramphenicol in 1947 and chlortetracycline in 1948, as these were quite effective against both Gram-positive and Gram-negative bacteria along with *Rickettsia*, large viruses, and certain protozoa. This was followed by the discovery of other members of the tetracyclines (Figures 11.6 and 11.7).

The subsequent discoveries of various antibiotics in about 25 years of duration after the commercial production of penicillin in 1941, such as "aminoglycoside" and "tetracycline,"

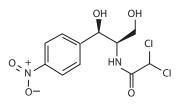


Figure 11.6 Chloramphenicol.

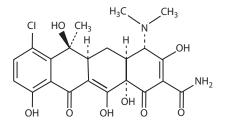


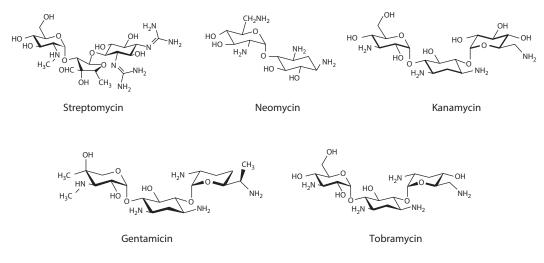
Figure 11.7 Chlortetracycline.

have had a great effect in the therapy of infectious disease because the spectrum of activity has significantly increased. The natural aminoglycosides (Figure 11.8) are bactericidal in action and act by inhibition of protein synthesis, for example, streptomycin, neomycin, kanamycin, gentamicin, and tobramycin are active against both Gram-positive and Gramnegative bacteria, but predominantly act on Gram-negative bacteria.

Because of nephrotoxicity and ototoxicity, significant precautions are required during therapy with aminoglycosides, which restricts their use to clinical applications when used alone, although it is still very commonly used along with other antibiotics, when used synergistically, by adjusting the dose. The introduction of semisynthetic aminoglycosides, such as netilmicin in 1976 (Figure 11.9) and amikacin in 1972 (Figure 11.10), further expanded the spectrum of activity; however, toxicity still remains the main factor for consideration.

Approximately 150 aminoglycoside antibiotics have thus far been discovered from various species of *Micromonospora*, the most well-known of these is gentamicin produced by *M. echinospora*. A series of other antibiotics (Figure 11.11) were discovered between 1950 and 1965, for example, nystatin (1951), erythromycin (1953), vancomycin (1955), novobiocin (1956), rifamycin (1959), and lincomycin (1962).

On the other hand, although tetracyclines (Figure 11.12) produced better results against both Gram-positive and Gram-negative, the spectrum of activity was further increased





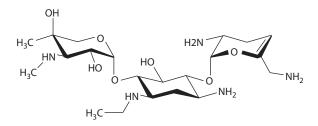


Figure 11.9 Netilmicin.

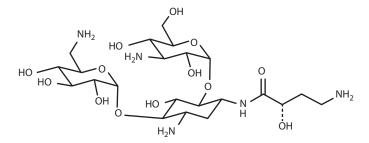


Figure 11.10 Amikacin.

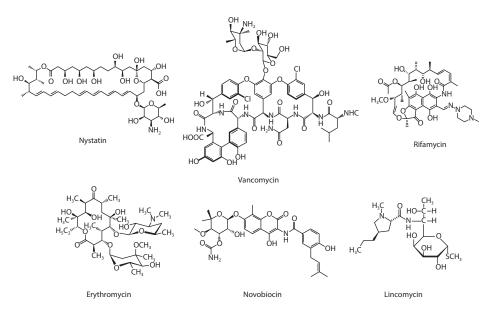


Figure 11.11 Antibiotics discovered between 1950 and 1965.

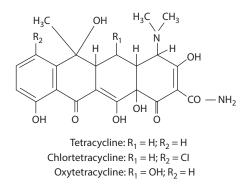
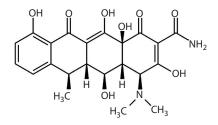


Figure 11.12 Tetracyclines.

with the introduction of semisynthetic tetracyclines such as doxycycline, minocycline, and tigecycline (Figures 11.13 through 11.15).

Macrolides, for example, erythromycin and its semisynthetic forms (Figure 11.16), have also found tremendous application, especially against respiratory tract infections. The macrolides contain a large lactone ring with multiple keto and hydroxyl groups, linked to one or more sugars. The modified macrolides derived from erythromycin, such as clarithromycin, azithromycin, roxithromycin, and the ketolides, for example, telithromycin, have been successfully used in clinical practice. These modified forms are active against penicillin and erythromycin-resistant *Streptococcus pneumonia*, *Haemophilus influenzae*, group A *Streptococci*, *Legionella* spp., *Chlamydia* spp., and *Mycoplasma pneumoniae*, and can be used both parenterally and orally.

The greatest moment in the history of antibiotics, which once again revolutionized therapy, was the introduction of semisynthetic penicillins and cephalosporins, classified under  $\beta$ -lactam antibiotics. The discovery of 6-aminopenicillinic acid (6-APA) followed by 7-aminocephalosporanic acid (7-ACA) by the action of enzyme penicillin amidase on penicillin G, and cephalosporin acylase on cephalosporin C, respectively, has lead to the production of a series of semisynthetic penicillins and cephalosporins with an extended spectrum of activity. Because the semisynthetic penicillins have shown remarkable



*Figure 11.13* Doxycycline.

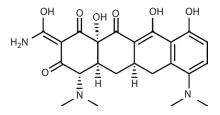


Figure 11.14 Minocycline.

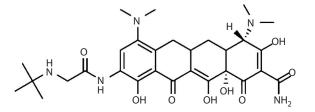


Figure 11.15 Tigecycline.

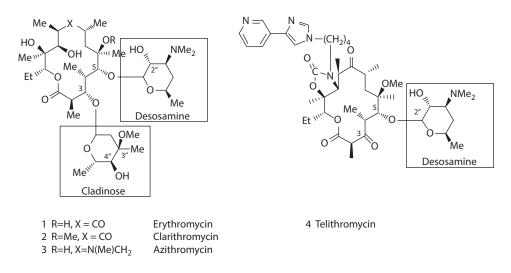


Figure 11.16 Structures of modified macrolides.

therapeutic activity, 6-APA and 7-ACA have naturally become very important pharmaceutical raw materials. Nowadays, a considerable number of semisynthetic penicillins and cephalosporins, all derived from 6-APA and 7-ACA (Figures 11.17 and 11.18), are in widespread clinical use, whereas a numbers of other novel structures are under clinical investigation.

The  $\beta$ -lactams are the most important class of antibiotics in terms of clinical use and safety profile. Members of the  $\beta$ -lactams constitute a major part of the antibiotic market today and include the penicillins, cephalosporins, monobactams, and carbapenems. Natural carbapenems such as thienamycin, which is resistant to  $\beta$ -lactamase, are made by *Streptomyces cattleya, Erwinia carotovora* subsp. *caratovora, Serratia* sp., and *Photorhabdus luminescens*; however, the commercial carbapenems, such as imipenem, meropenem, and ertapenem are manufactured synthetically (Sanchez and Demain 2011).

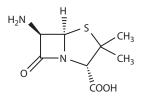


Figure 11.17 6-Aminopenicillinic acid.

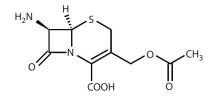


Figure 11.18 7-Aminocephalosporanic acid.

Other antibiotic groups include glycopeptides such as vancomycin, balhimycin, and chloroeremomycin (Figures 11.19 through 11.21) and lipopeptides such as daptomycin (Figure 11.22). Vancomycin has been the molecule of choice to treat infections caused by resistant organisms and is considered as the "antibiotic of last resort." However, lipo-glycopeptides (e.g., teicoplanin; Figure 11.23) and oxazolidinones (e.g., linezolid; Figure 11.24) were later found to inhibit vancomycin-resistant bacteria. This was followed by

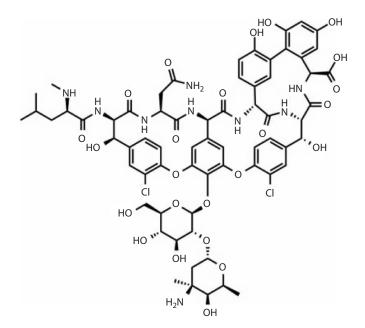


Figure 11.19 Vancomycin.

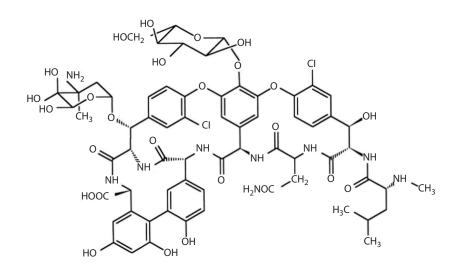


Figure 11.20 Balhimycin.

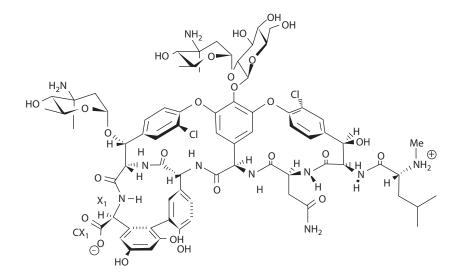


Figure 11.21 Chloroeremomycin.

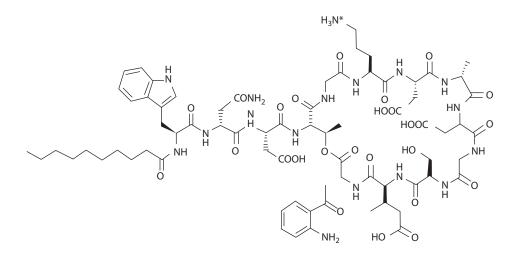


Figure 11.22 Daptomycin.

the discovery of streptogramins such as virginiamycin (Figure 11.25) and pristinamycin (Figure 11.26), as well as the derivatives of pristinamycin, for example, quinupristin/dal-fopristin, used in combination under the brand name of Synercid (Figure 11.27), with the hope and findings that it will treat even Gram-positive bacteria resistant to vancomycin and teicoplanin.

The world market for antibacterial drugs (covering natural, semisynthetic, and synthetic) was valued at more than US\$40 billion in 2012, having grown 1% since 2011. In both years, the leading submarket was cephalosporins, accounting for more than a quarter of the antibacterial drug revenue. In 2012, the United States was the leading national market for antibacterial drugs, with a market share of more than 25%. However,

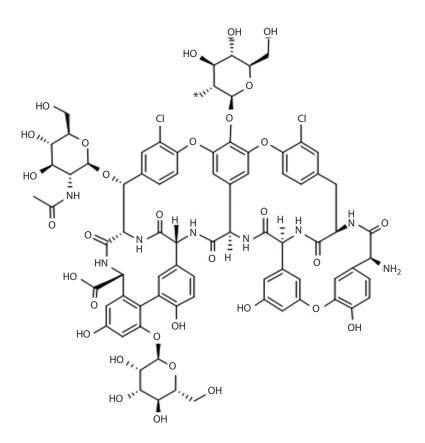


Figure 11.23 Teicoplanin.

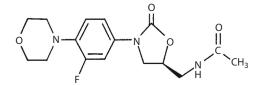


Figure 11.24 Linezolid.

emerging markets also account for a significant proportion of the global market. China's market share in 2012 was more than 15%. The leading antibacterial drug in 2012 was reportedly Zyvox (linezolid, oxazolidinone antibacterial from Pfizer), with a revenue of US\$1.35 billion. The top 10 antibacterial drugs had a combined revenue of US\$6.92 billion in 2012. There were approximately 10 new antibacterial drugs in phase III trials in 2013. Most of these could be launched in major markets by 2015 (Visiongain 2013). However, the pipeline for antibacterial drugs is much leaner than for other most important sectors of the pharmaceutical industry, despite unmet treatment needs existing for many bacterial infections.

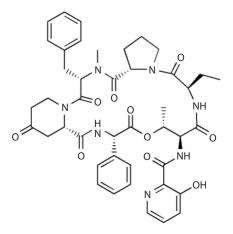


Figure 11.25 Virginiamycin.

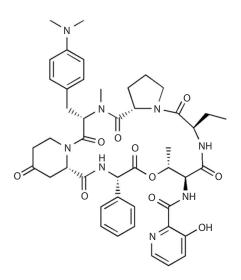


Figure 11.26 Pristinamycin.

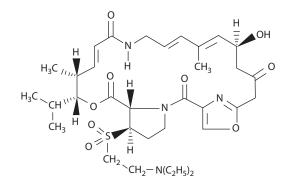


Figure 11.27 Synercid.

#### 11.2.2 Production and pharmaceutical presentation

The production of antibiotics is mainly based on the knowledge of microbiological and biochemical methods, which distinguish them from similar compound (antimicrobials), obtained through chemical synthesis. Most manufacturing processes have been patented by big multinational companies; however, the outline of these methods is fairly well known. The most feasible method for the production of antibiotics is through fermentation, which certainly has now reached tremendous advancement in terms of process optimization and extraction, and purification of the desired product. Classic fermentation processes include propagation (cultivation of microorganisms), isolation/extraction, purification, and bulk packaging.

Propagation starts with the preparation of inoculum from stock culture usually maintained on a suitable medium based on the type of organism used. This is followed by seed culture and then growth in large fermenters (Figure 11.28) with 100,000 to 150,000 L capacity containing suitable sterile liquid growth medium. The fermentation medium should be sufficient to provide carbon and nitrogen sources (such as molasses or soy meal for the growth of microorganisms and ammonium salts for the regulation of the metabolic cycle). The medium is supplemented with trace elements/inorganic salts (such as phosphorus, sulfur, magnesium, zinc, iron, and copper) and precursor according to the strain's requirements. Aeration, agitation, pH, and temperature are controlled through automatic/computerized systems and the optimum level of nutrients is maintained at all times. This is closely monitored and adjusted as and when necessary.

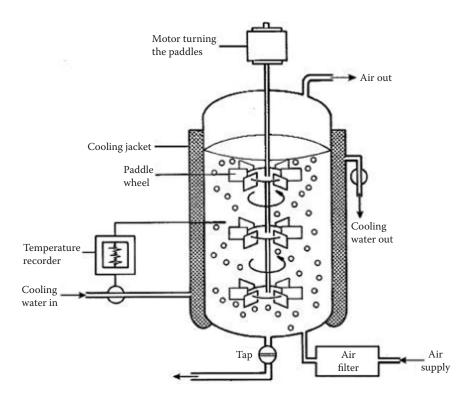


Figure 11.28 Structure of a modern fermenter used for submerged fermentation.

To prevent foaming during fermentation, a suitable concentration of antifoaming agents, such as octadecanol and silicones, are used. In most cases, the composition of media and the process conditions are slightly different for both seed and fermentation stages based on the objectives of the two stages. Process development in antibiotic fermentation is highly critical and thus needs comprehensive attention and consideration. At the end of the fermentation period, the fermenter is harvested and the contents are subjected to isolation, extraction, and purification of the product. During the entire manufacturing process, maintenance of sterile conditions is highly crucial because a slight contamination of antibiotic-resistant microorganisms or with phages will lead to complete loss of the production batch.

Pharmaceutical presentation of antibiotics is mainly based on its uses and application. Because antibiotics are used in many different forms, each of which enforces fairly different kinds of manufacturing requirements and packaging material. For bacterial infections on the skin surface, an antibiotic may be applied as an ointment, cream, or lotion packed in suitable collapsible containers. For eye and ear applications, the antibiotic is dispensed in sterile containers and the product is known as eye/ear drops. In addition, for some antibiotics, eye ointments are also available in sterile containers. If the infection is internal, the antibiotic can be swallowed in the form of immediate release or extended release filmcoated tablet formulations or injected directly into the body through the intramuscular or intravenous route dispensed in sterile containers (vials or ampoules). In these cases, the antibiotic is delivered throughout the body by absorption into the bloodstream.

#### 11.2.3 Biotechnological approach

Soon after the discovery and commercial production of penicillin, scientists started working on the genetic aspects of antibiotic formation, primarily because of the increased demand for penicillin to treat the victims of World War II. The initial studies were based on the development of high-vielding mutant strains using both physical and chemical treatment of the wild strains. The process of mutation is a classic approach in which strain improvement is usually observed stepwise after several rounds of mutagenization are required before the identification and selection of a good and efficient strain. Some great success was recorded with mutation, not only with antibiotics but for other secondary metabolites as well. As a result, penicillin, productivity significantly increased and a number of multinational companies started working to develop their own mutant strains, not only for the production of penicillin, but also to discover other antibiotics as well. It has been reported that mutation may increase yield by approximately 100-fold to 1000fold, and indeed it is the chief factor in antibiotic production. By the end of 1944, sufficient amounts of penicillin were produced with the help of mutant strains and, within 10 years' time, approximately 20 metric tons of penicillin was being produced every month (Porter 1976).

With the discovery of the complex mechanisms responsible for the regulation of antibiotics in various microbial strains, the classic mutation processes were further linked with genetic engineering techniques. This has not only shortened the strains' improvement and development time but has been able to reduce the unnecessary involvement of human resources during the mutant screening program. With the further advancements of new biotechnological approaches and tools, genetic engineering alone is quite sufficient to modify the strain, provided the biosynthesis mechanism is fully known for the desired antibiotic. Even if the biological mechanism is not fully known, protoplast techniques along with random mutagenization can be employed for strain improvement.

#### 11.2.4 Therapeutic role

In the history of medicine, antibiotics have had great therapeutic applications and a great role in the management of various infectious diseases. However, with the introduction of new antibiotics as well as new synthetic antimicrobials, significant revisions have been made in the selection of the drug of choice for treatment (since the initial discoveries). The choice or selection is usually made on the basis of clinical diagnosis and culture sensitivity. The mechanisms by which the antibiotics acts on greatly helps in assessing the therapeutic role of various antibiotics. Broad-spectrum (or extended spectrum) antibiotics can be used to treat a wide range of infections, whereas narrow-spectrum antibiotics have limited applications as these are only effective against a few types of organisms. Some antibiotics also act against anaerobic bacteria and are of significant importance. Along with the therapeutic role, some antibiotics have prophylactic roles as well and are also used to prevent infection, especially before abdominal and orthopedic surgeries. The therapeutic role of some antibiotics is also dependent on drug–drug and drug–food interactions.

Most antibiotics produce prompt effect against infection depending on the site and type of application/dosage form. The onset of action depends on the peak plasma concentration, which is very short in case of the intravenous route, and takes time, maybe hours, if delivered orally. The most important part of antibiotic therapy is to complete the whole course of medication to prevent the infection from relapsing. With incomplete therapy or misuse of drugs, both have significant effects in developing resistant strains, which are quite difficult to treat. In addition, antibiotic overuse is also an important factor that contributes toward the growing number of bacterial infections which are becoming resistant to antibacterial medications. According to the European Centre for Disease Prevention and Control (ECDE), antibiotic resistance is becoming a serious public health threat worldwide. The ECDE statement issued on November 19, 2012 highlighted that approximately 25,000 people die each year in the European Union from antibiotic-resistant bacterial infections.

## 11.3 Antitumor agents

Exploration of new and effective antitumor drugs perhaps remains one of the most important topics in recent years throughout the world. This is primarily because the number of new incidences of cancer have significantly increased and is approaching more than six million every year (Chandra 2012). The initial antitumor drug development was mostly based on the discovery of molecules that have significant cytotoxic or cytostatic activity on tumor cells. However, during the last two decades, a large number of chemical compounds (with some degree of success) have been developed through chemical synthesis, or isolated from natural sources (such as plants, animals, and microorganisms) after having a better understanding of the complexities of tumor biology. The secondary metabolites produced by plants, microorganisms, and marine organisms have been extensively investigated for such products. Therefore, there is hardly any doubt in highlighting the importance of secondary metabolites in the discovery of antitumor drugs produced from natural sources. However, the great biodiversity of microorganisms represents an excellent possibility of producing secondary metabolites, which can be used as antitumor drugs. Along with these dynamic sources, microorganisms, especially bacteria, have also been explored to deliver antitumor drugs. In-depth studies on bacterial metabolites, with the advanced knowledge in molecular biology and genetic engineering, have widened the scope of antitumor drugs from bacteria and other microorganisms. The antitumor drugs, as reported during recent decades, from microbial sources have proven quite effective and less toxic for chemotherapy (Cragg et al. 2009; Ma and Wang 2009; Ravelo 2004).

The role of antitumor drugs from microbial sources has significantly increased and are now very much dominating as chemotherapeutic agents. This is mainly because of the resistance to conventional antitumor therapies, especially in patients with advanced solid tumors. Based on these, different forms of bacteria (e.g., live, attenuated, nonpathogenic, or genetically modified) have now begun to appear as promising sources of antitumor drugs. In recent years, the pigments produced by some bacterial species such as *Serratia* have been extensively investigated to develop antitumor drugs. A natural red bioactive pigment, prodigiosin (Figure 11.29), produced by *Serratia marcescens* (Figure 11.30) has demonstrated potential as an antitumor drug by exerting immunosuppressive activity *in vitro* and apoptotic effect *in vivo*. More research on prodigiosin from other bacterial species, such as *Vibrio psychroerythrus, Hahella chejuensis*, and *Streptomyces coelicolor* is under investigation to establish the production system, fermentation strategies, purification, and identification processes. Certainly, the role and application of biotechnology will play a significant role in establishing these processes.

The global market for antitumor drugs is increasing very rapidly. The leading pharmaceutical companies involved in antitumor drug business are Roche, Novartis, AstraZeneca, Eli Lilly, Celgene and some other pharmaceutical companies. Based on 2011 business

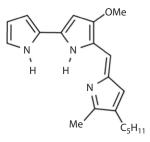
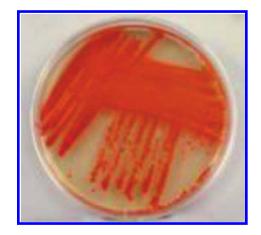


Figure 11.29 Prodigiosin.



*Figure 11.30 S. marcescens* culture.

figures for antitumor drugs (US\$68.8 billion), the proposed figure for 2012 was estimated at US\$75 billion and further forecasted to 139.3 billion by the end of 2022. The top 25 highest revenue-generating drugs dominate the world market, generating 83% of revenues in 2010. Avastin, Rituxan, Herceptin, and Glivec/Gleevec are a few examples out of the 25 leading drugs in the anticancer treatment market. The overall revenues for anticancer treatment are expected to increase strongly from 2012 to 2022. In addition, individual drugs are expected to benefit too. With the emerging advancement in pharmaceutical biotechnology and the ever-increasing demand in antitumor drugs because of the increase in cancer incidence and prevalence, both in developed and developing countries, the future of this particular field holds great promise. Research and development (R&D) pipelines in oncology are quite strong, and a number of new drug molecules are under critical investigation and some are waiting for Food and Drug Administration approval. Thus, high revenues are likely to be generated from many drugs—small-molecule products and biological agents (biologicals) by the end of 2022. The top 25 highest revenue-generating drugs dominated the world market, generating 83% of revenues in 2010.

#### 11.3.1 Historical background and new challenges

The true scientific approach to treating cancer was observed during the 1950s, after the discovery of vincristine and vinblastine (Figures 11.31 and 11.32) as alkaloids isolated from

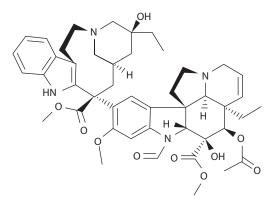


Figure 11.31 Vincristine.

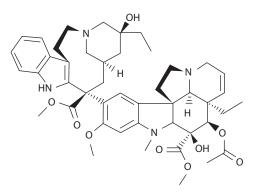


Figure 11.32 Vinblastine.

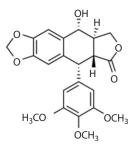


Figure 11.33 Podophyllotoxin.

*Catharantus roseus*. Vincristine was introduced successfully against non-Hodgkin's lymphoma, whereas vinblastine was introduced for Hodgkin's lymphoma, along with breast, testicular, and small lung cancer. Another nonalkaloidal drug, podophyllotoxin (Figure 11.33), isolated from the rhizomes of *Podophyllum peltatum* and *Podophyllum hexandrum*, was introduced for the treatment of genital warts caused by human papillomavirus (Harwell and Schrecker 1952).

By the end of 1990, after 40 years of struggle, scientists were able to identify a significant number of drugs suitable for treating different types of cancer belonging to different groups and chemical structures. Out of several classifications, the most acceptable was the one that included: DNA-damaging agents, antimetabolites, natural products and their analogues, hormonally directed agents, and biological response modifiers (American Medical Association 1995). A majority of the antitumor drugs (~60%) are reported to be of natural origin, that is, plants and microorganisms. Among the microorganisms, doxorubicin, dactinomicines, mitomycin, bleomycin, borrelidin,  $\beta$ -glucan, becatecarin, Eco-4601, NPI-0052, romidepsin, texol, spicamycine, amrubicin HCl, tartrolon D, elsamirucini, and prodigiosin are quite important. It is now well-documented that antitumor drugs from microbial sources have some reasonable advantages in terms of the diversity of the compounds and manipulation in production. One of the great challenges for researchers is to utilize bacteria to enhance the antigenicity of tumor cells using immunotherapeutic strategies, an emerging and effective approach in cancer therapy. Some work in this direction is underway using attenuated Salmonella typhimurium, which has demonstrated successful invasion of melanoma cells. Salmonella can infect the malignant cells both in vitro and in vivo, thus triggering the immune response (Patyar et al. 2010).

#### 11.3.2 Production and pharmaceutical presentation

Submerged fermentation is the method of choice for the production of microbial antitumor drugs. Primarily, the process can be divided into a number of steps such as maintenance of microorganisms, preparation of inoculum, propagation of organism, isolation/extraction, purification, and bulk packaging. All these steps are quite specific and depend on the type of microorganisms planned for use in the production of a drug. To understand the basic phenomenon, an example of prodigiosin produced by a strain of *S. marcescens* is described below.

The maintenance of a culture is usually done at low temperature using suitable maintenance media such as nutrient agar. For subculturing, nutrient broth or peptone glycerol broth can be used. The inoculum is usually prepared using shaken flask conditions for about 24 h. The inoculum is aseptically transferred to a seed vessel containing nutrient broth. The pH of the medium is maintained between 3 and 7, temperature at 28°C, and agitation at 200 rpm. After 12 h, the entire contents of the seed vessel is transferred to a fermenter containing nutrient broth (peptone 10 g/L, sodium chloride 5 g/L, yeast extract 3 g/L + 0.5% glucose or maltose) supplemented with glucose or maltose. Sesame seed medium is also equally effective in producing prodigiosin without adding glucose or maltose. Small amounts of thiamine and ferric acid can enhance the productivity of the product. The pH of the medium, fermentation temperature, and agitation rate is kept similar with that of the seed vessel. A temperature higher than  $37^{\circ}$ C and lower than  $20^{\circ}$ C can greatly inhibit the product of prodigiosin.

After 36 h of fermentation, the pigment can be isolated, extracted, and purified. The broth is filtered and then centrifuged at 10,000 rpm for 15 min using an ultracentrifuge. The supernatant is extracted with ethyl acetate and the pigment is extracted with acetone followed by ethyl acetate to remove the impurities. The two fractions after mixing are subjected to evaporation. Silica column of 80 to 100 mesh size can be used for separation of the noncolored impurities from the pigment. The dried powder at different concentrations was used for plotting the standard graph versus absorbance at 535 nm. The purified sample showing a single peak absorbance at 535 nm in the UV spectrophotometer was further analyzed for determination of molecular weight using a mass spectrophotometer.

Antitumor drugs are available in different dosage forms based on their chemical nature and the types of application required for therapy, such as topical, oral, subcutaneous, intramuscular, intravenous, intrathecal, intraperitoneal, intralesional, intrahepatic artery, intracavitary, intravesical, and isolated perfusion. Chemotherapy is mostly delivered through the intravenous route, although a number of drugs such as melphalan, busulfan, and capecitabine can be administered orally. In some special cases, such as melanoma, the isolated limb perfusion or isolated infusion of chemotherapy into the liver or the lung have been used. The main function and objectives of these approaches is to supply the required dose of antitumor drugs, which is usually very high at the site of the tumor, and protect from significant systemic damage. Infection risk is common during antitumor therapy. Therefore, to lower the risk of infection during continuous, frequent, or prolonged intravenous chemotherapy, various systems, such as the Hickman line, the Porta-Cath, and the PICC line are commonly inserted (following surgery) into the vasculature to maintain access. These systems have reasonably lower infection risk and are much less prone to phlebitis or extravasation, and eliminate the demand for repeated insertion of peripheral cannulae.

#### 11.3.3 Biotechnological approach

The future role of biotechnology in developing and understanding antitumor drugs looks quite promising. Recent advancements and approaches have significantly increased the possibility of exploring genetically modified microorganisms and the use of genedirected enzymes, especially bacteria, to deliver potent drugs for the selective destruction of tumors. Microbial toxins for the destruction of tumors along with the introduction of cancer vaccines are leading projects as well as objectives of various research institution and top multinational groups of companies, presently engaged and dominating the antitumor and vaccine market. For more targeted tumor therapies, some recent biotechnological approaches, such as conjugation of toxins with monoclonal antibodies or some polymeric carriers have also been suggested and are under investigation. Tumor-amplified protein expression therapy using genetically modified organisms, especially bacteria as a vector or vehicle for preferentially delivering drugs to solid tumors (e.g., brain tumors) is also under study and development. In addition, newer biotechnological techniques can also be applied during submerged fermentation of the organism to enhance the production yield of the desired product. One of the leading approaches is the immobilization of microorganisms in a special carrier to enhance the proliferation and productivity of the desired product.

#### 11.3.4 Therapeutic role

The antitumor drugs available today have diverse modes of action and thus vary in their effects on different types of cancer and normal cells. A single "cure" for cancer has proved subtle because of the large number of different types of cancer cells identified thus far. Furthermore, the biochemical differences between normal and cancerous cells is also one of the key factors in selective treatment and drug design. Therefore, the therapeutic role and application of many antitumor drugs, especially the traditional chemotherapeutic agents, are limited due to their toxicity to normal, rapidly growing cells in bone marrow, digestive tract, and hair follicles. The ultimate results/side effects of chemotherapy appear as myelosuppression (decreased production of blood cells, hence also immunosuppression), mucositis (inflammation of the lining of the digestive tract), and alopecia (hair loss). Some newer antitumor drugs such as monoclonal antibodies are not extensively cytotoxic, but rather focus proteins that are uncharacteristically expressed in cancer cells and are indicated as essential for their growth. The mechanism of these antitumor drugs is stated as targeted therapy, which is quite distinct from classic or traditional chemotherapeutic agents and are also used alone or in combination with traditional chemotherapeutic agents in antineoplastic treatment regimens.

## 11.4 Ergot alkaloids

Fungal metabolites, and especially the alkaloids, have been focused on and studied with great importance and interest in view of their significant biological activities. It symbolizes a cluster of fascinating and multifaceted chemical compounds containing at least one nitrogen atom in a ring structure in the molecule. These chemical compounds are produced as metabolites by the living organisms, including fungi in different biotopes. Fungal alkaloids have significant biopharmaceutical importance as some are used as drugs. The best example that can be given here is that of "ergot alkaloids," produced particularly by Claviceps purpurea, which is usually regarded as a parasite of cereals and grasses. Industrial production started in 1918 with ergotamine tartrate, and Sandoz (now Novartis) was the first company to start marketing it in 1921 and dominated the market until the 1950, when competitors began to emerge. Other reputable companies producing ergot alkaloids include: Boehringer Ingelheim (Germany), Galena (Czech Republic), Gedeon Richter (Hungary), Lek (Slovenia), Poli (Italy), Eli Lilly (USA), and Farmitalia (Italy). The 2006 figures indicated world production of ergot alkaloid between 5.0 and 8.0 metric tons, of which approximately 60% resulted from fermentation and the rest (40%) from field cultivation of a hybrid of wheat and rye (Schiff 2006). However, in 2010, the total production of these alkaloids significantly increased to about 20.0 metric tons, and the production ratio between field cultivation and fermentation became equal (50% each). Chemically, the identification of ergot alkaloids is based on the presence of a tetracyclic ergoline ring system and is derived from an amino acid, L-tryptophan (Figure 11.34). By nature, ergot alkaloids are indole compounds and signify the biggest group of nitrogenous fungal metabolites.

More than 80 different ergot alkaloids have been reported in the literature and *Claviceps* species were noted to contribute the highest number, more than 70 alkaloids.

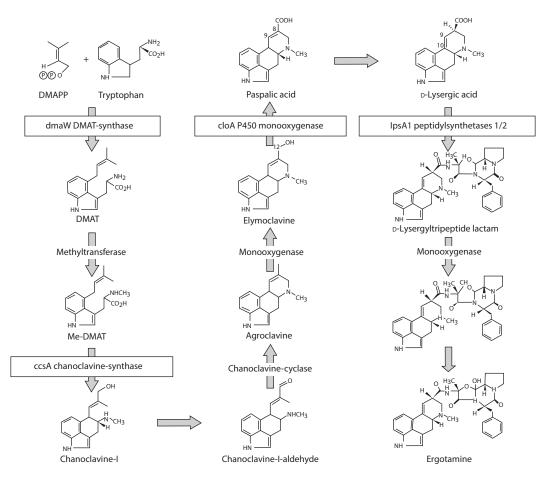


Figure 11.34 Biosynthesis of ergot alkaloids from tryptophan.

From a medicinal point of view, ergot alkaloids can be divided into two main groups, that is, the water-soluble amino alcohol derivatives and the water-insoluble peptide derivatives. The water-soluble amino alcohol contains 20% of the total alkaloid mixture, whereas the water-insoluble peptide derivatives constitute the remaining 80% of the alkaloids. However, based on their chemical nature, it will be more realistic to divide ergot alkaloids into three major groups: clavine alkaloids, D-lysergic acid and its derivatives, and ergopeptines (Hulvova et al. 2013). The clavine alkaloids (Figure 11.35) are substituted 6,8-dimethylergolines, which include approximately 35 different members, such as the ergolines, chanoclavines, agroclavines, and others, but none of the members of this group are of medicinal importance.

Lysergic acid and its derivatives (lysergic acid-derived amides) are significantly active pharmacologically. Ergonovine and its semisynthetic derivatives, such as methylergonovine and methysergide are quite important and have applications in medicine. The naturally occurring ergopeptines, such as ergotamine is the only example that is used medicinally. However, useful semisynthetic derivatives of peptide alkaloids (Figure 11.36) include: brominated ergocrytine (e.g., dihydroergotamine) and bromocriptine and dihydrogenated ergot alkaloids (e.g., ergoloid).

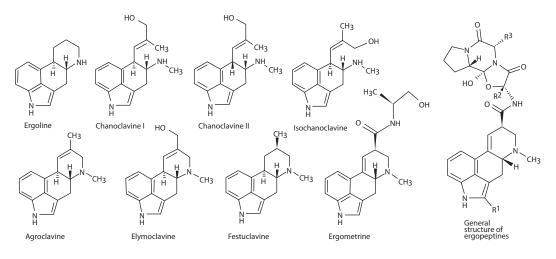
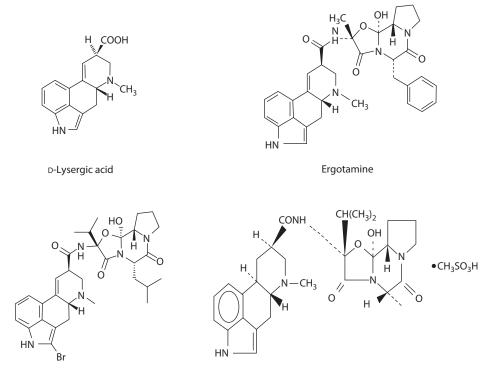


Figure 11.35 Structures of clavine alkaloids.



Bromocriptine

Ergoloid

*Figure 11.36* Semisynthetic derivatives of peptide alkaloids.

#### 11.4.1 Historical background and new challenges

The first microbial alkaloids were identified and reviewed in *C. purpurea*. Collectively, these are known as ergot alkaloids and can be isolated from the sclerotia produced after infection of the ovaries of the plant by *Claviceps* ascospores or conidia. Apart from *C. purpurea*, ergot alkaloids have also been identified in other fungal genera, such as *Aspergillus* and *Penicillium* (Rehacek 1984). The number of alkaloids found in fungi and other microorganisms is not large but it is believed that it would increase rapidly if more extensive research was carried out. In the case of ergot, it was observed that no comprehensive attempt was made until 1960 to obtain these alkaloids on a commercial scale for its proper utilization in medicine. In 1960, a number of studies were published that claimed to produce around 1,000 µg/mL of ergot alkaloids (lysergic acid hydroxyethylamide) through submerged fermentation using *Claviceps paspali*.

Continued research on the subject all over the world lead to the discovery of more important alkaloids such as lysergol, lysergine, and lysergene from the saprophytic culture of ergot fungi in 1961. This was the beginning of applying submerged fermentation technology for the production of ergot alkaloids. Finally, a group of scientists reported on the successful fermentative production of lysergic acid in 1964 using *C. paspali*, followed by ergotamine from *C. purpurea* in 1966 to 1967 (Amici et al. 1966, 1967; Kobel et al. 1964; Tonolo 1966). Further research in these directions, contributed to the fermentative production of ergocryptine and ergotamine, ergosine, and ergocristine with high yield using *C. purpurea* (Amici et al. 1969).

#### 11.4.2 Production and pharmaceutical presentation

The production of ergot alkaloids can be done through artificial parasitic cultivation on rye (traditional field cultivation) or by using fermentation, surface culture, or submerged fermentation techniques. Depending on the resources in different regions of the globe, these methods are still used for the production of alkaloids. In the traditional field cultivation process, a conidial suspension is sprayed onto field-cultivated rye and is observed for the development of various stages. The genus *Claviceps* has 36 members that can infect approximately 600 different species of monocotyledonous plants (Hulvova et al. 2013). Once the mycelium (mat-like body) of the fungus develops in the ovaries of the host plant; it eventually turns into a hard pink or purple body, the sclerotium, or ergot, which resembles a grain of rye in shape. The maximum alkaloidal content is usually observed after approximately 20 days. The sclerotium contains alkaloids that are subjected to extraction with a toluene/ethanol solvent mixture to obtain the primary extract. Further purification of the alkaloids is done by liquid–liquid extraction and a semipurified toluene extract is obtained. For further purification, partial evaporation of the toluene extract is done to obtain a crystalline product. A number of other solvents, such as methanol and chloroform, have also been used for extraction. The isolation and extraction of ergot alkaloids with efficient processes using nontoxic and environmentally friendly solvents are still desirable. According to some studies, approximately 10 to 20 kg of ergot alkaloids can be recovered from 1 to 2 metric tons of sclerotia. The climatic conditions can have great effect on the yield of the product (Tudzynski et al. 2001).

For surface culture, the *Claviceps* are grown on suitable artificial medium containing a carbon source (such as sorbitol, mannitol, or sucrose) supplemented with some inorganic salts, and maintained at a required temperature in flat glass bottles or using sterile plastic bags.

Submerged fermentation is now widely used in different regions to produce some ergot alkaloids through the production of paspalic acid, which is ultimately converted to D-lysergic acid through the process to some other semisynthetic alkaloid derivatives. Process parameters, media composition, and the addition of certain nutrients and precursors are key factors in the fermentative production of ergot alkaloids and have the capacity to affect overall yield of alkaloids. The cultures can be maintained on potato dextrose agar slants at low temperature. After extensive studies, the optimum pH for the production of alkaloids in saprophytic culture was reported to lie between pH 5 and 6. For surface culture technique, a medium containing yeast extract, glucose, mannitol, and peptone could be used. Strains of *C. purpurea* grown on a peptone-mannitol medium can produce high alkaloid content (fivefold increase), if concentrations were increased from 5% to 25%. The effect of various macroelements and microelements, along with carbon and nitrogen sources for the production of ergot alkaloids have been studied by various research investigators to optimize the production of alkaloids. The concentration of phosphate in the medium plays the most important role because it is essential for the production of secondary metabolites. The exhaustion of phosphate in the medium will affect the yield of alkaloids. KH<sub>2</sub>PO<sub>4</sub> added to the medium at a concentration of 1000 mg/L has been observed to produce the maximum yield of fermented broth, whereas the addition of glycols and Tween 80 has also been reported to increase the alkaloid yield (Mahmood 1983).

The pharmaceutical presentations of ergot alkaloids are highly significant in terms of achieving the desired effects. The regulatory issues and legislation are somewhat different in various countries; therefore, the availability of the finished dosage form and their applications differ from region to region. Various dosage forms, such as film-coated tablets, sublingual tablets, capsules, drops, and powders are available for oral administration. Injectable preparations, suppositories, and nasal sprays are also available. Ergoloid mesylate is available in 0.5, 1.0, 2.0, and 4.5 mg tablets, as well as 1 mg capsules, 1 mg/mL drops, and in 0.3 mg/mL injections in many countries. These dosage forms are recommended for very selective applications, for example, to increase cerebral metabolism and blood flow and in Alzheimer's disease, in which ergoloid may lower the scores on some cognitive and behavioral rating scales. Application in dementia needs more controlled studies to determine the risk-benefit profile. Ergotamine tartrate (Ergomar) is available as 2 mg sublingual tablets for the treatment of severe migraine headache and cluster headache. Ergotamine tartrate is also available in combination with caffeine (Cafergot) containing 1 mg ergotamine + 100 mg caffeine for the treatment of migraine attacks with or without aura. Another combination formulation containing 2 mg of ergotamine tartrate, 50 mg of cyclizine hydrochloride, 100 mg of caffeine hydrate (Migril) as well as 2 mg of ergotamine + 100 mg of caffeine as suppositories (Cafergot suppository) are also available for the same indication. Methylergonovine maleate in 0.2 mg tablets and 0.2 mg/mL injection (Methergine) are available for migraine. Dihydroergotamine is available as a 4 mg/mL nasal spray, and in 1 mg/mL injections. Bromocriptine mesylate (Parlodel) is available in 2.5 mg tablets and 5 mg capsules.

#### 11.4.3 Biotechnological approach

Despite the valuable information and material published on ergot alkaloids, less attention has been focused on the use of biotechnological tools for further improvements in yield through submerged fermentation by optimizing the process condition and manipulating the media composition. In fact, the field of fungal alkaloids is demanding that researchers pay more attention and explore new molecules to combat the new challenges emerging in the healthcare system. During the early phase of biotechnology, ergot alkaloids were reported as direct gene products. Thus, more emphasis was given to enzymatic transformation for their synthesis via small precursor molecules. However, the enzymes linked with the relevant structural and regulatory genes were not mapped, and although the isolation and cloning of the structural genes had been proposed, it was considered a difficult task. The possibility of gene transfer via protoplast fusion remained one of the leading topics during the early phase of *Claviceps* genetic engineering (Keller et al. 1980; Maier et al. 1980). Continuation of research in the field lead to the discovery of plasmids associated with mitochondria in a wild strain of *C. purpurea* (Tudzynski et al. 1983).

Later, the construction of a plasmid that replicates automatically, the development of a transformation system with drug resistance markers, and the development of genes controlling the regulation of the entire pathway were proposed (Rehacek 1984). In 1995, the first pathway gene cluster, *dmaW*, was reportedly cloned from *C. fusiformis* SD58, which encodes the determinant step in ergot alkaloid biosynthesis (Tsai et al. 1995), this was followed by the identification of the orthologue in a *C. purpurea* strain P1 (Tudzynski et al. 1999). Further research on this cluster has been extended and, thus far, more than 68.5 kb of the cluster has sequences containing 14 genes and is coordinately induced under ergot alkaloid production (Schardl et al. 2006, Wallwey and Li 2011).

Thus far, most of the randomly mutated strains of C. purpurea were employed in the production of ergot alkaloids on an industrial scale, but with the recently sequenced ESA cluster of 14 genes, it is expected that a guided overexpression of certain genes or targeted upregulation of the entire cluster could significantly increase the yield of ergot alkaloids (Hulvova et al. 2013). The gene *dmaW* is now being identified in all *Claviceps* species but, at the same time, *dmaW* has also been reported to pass through duplication and losses (Liu et al. 2009). Therefore, the possibility of loss of mutation and deletions of the *dmaW* genes associated with C. purpurea does exist, which may affect productivity and can be validated on a pseudogene of *dmaW* that has already been detected in an unspecified *C. purpurea* strain. Gene transfer via protoplast fusion, which was proposed in the early 1980s, has successfully been used to develop a C. purpurea mutant strain overexpressing genes easC and easG under a constitutive promoter of glycerol-3-phosphate dehydrogenase from Aspergillus nidulans; however, no significant increase in the production of ergot alkaloids was noted (Hulvova et al. 2010). Genetic manipulation of the proteins engaged in the signaling cascade and ultimately influencing biosynthesis has also been proposed as a possible way to increase the productivity of ergot alkaloids (Lorenz et al. 2009; Mey et al. 2002). Most recently, genome sequencing of some *Claviceps* species has also been reported and some work is also under progress in the Czech Republic where *Claviceps africana* is being sequenced for a possible increase in alkaloid productivity (Hulvova et al. 2013).

#### 11.4.4 Therapeutic role

Ergot alkaloids have a long history of benefits and are observed to play a key detriment to humans in different civilizations and, ultimately, are used primarily as an aid in childbirth and in the treatment of some neurological and cardiovascular disorders. In view of the prominent biological activity and well-established pharmacological activity, ergot alkaloids are of considerable importance in pharmaceuticals and were thus introduced into the first edition of the United States Pharmacopeia in 1820 and into the London Pharmacopeia in 1836. The alkaloid possesses adrenoblocking, antiserotonin, and dopaminomimetic effects because of the structure's similarities with receptors, such as adrenaline, serotonin, and dopamine, respectively. The best therapeutic application of ergot alkaloids is in the

management of migraine headache and hypertension (e.g., ergotamine and dihydroergotamine are responsible for vasodilatation of arteriovenous anastomoses and can affect  $\alpha_1$  and  $\alpha_2$  adrenergic receptors), Parkinsonism and hyperprolactinemia (bromocriptine derived from ergocryptine) to stimulate the contraction of uterus postpartum, mastopathy, and as a sedative. As an anti-Parkinson agent, the ergot alkaloids are not prescribed as first-line therapeutic agent due to the risk of fibrotic reaction. Some ergot alkaloids such as agroclavine and elymoclavine possess antibiotic activity as well, but are not used in clinical practice.

#### 11.5 Vaccines

Among the biologicals, one of the leading and most extensively studied products is vaccines. These are given to generate immune responses against a number of diseases. Although the process of immunization against certain infections has been observed and noted to be a centuries-old phenomenon, it was only given a proper status and recognition with the introduction of modern immunological processes and techniques. With the advancement in molecular biology and microbial physiology, the classic pragmatic approach for vaccines has now completely changed for anticipated immune response in controlling a particular infectious disease. The product presentation is usually done in the form of a pharmaceutical suspension of live, inactivated, or fractionated microorganism that has been depicted as nonpathogenic. Despite the fact that the safety and effectiveness of vaccines are highly questioned, its high benefit-to-risk ratio strongly supports its use in providing long-lasting, humoral and cellular immunity. The memory T-cells (CTL or Th1 cells) derived from normal T-cells play a significant role in the development of long-lasting immunity. The T-cells represent an extremely advanced wing of the adaptive immune system, which is able to differentiate between pathogens and is efficient at progressing or adapting during the lifetime of an individual such that immunity is enhanced with each consecutive contact with a pathogen.

The importance of immunization and the use of relevant products can further be supported by the fact that in a decade's time, the vaccine market has grown from US\$5.7 billion to approximately US\$27.0 billion and is further expected to grow at the rate of more than 8% through 2018 (Bryant 2012; Palmer and Bryant 2013). The success rate or efficacy of vaccines usually depends on the prevailing body immunity conditions or level (low, moderate, or high), which is again dependent on various factors such as diabetes, use of steroids, age, ethnicity, genetic predisposition, HIV infection, and others, and the capability of the cells of the host's immune system to generate antibodies to specific antigens. Apart from this, the packaging, adjuvants, and transportation of vaccines have significant implications on effectiveness and thus need special considerations to manage and maintain a high degree standard as well. The system used for storage and distribution of vaccines in a suitable condition is termed as cold chain, which consists of a series of storage and transport links, all of which are intended to retain the vaccine at an appropriate temperature until it reaches the end user.

The European companies and US-based companies are the biggest contributors supplying vaccines all over the world. Based on 2012 revenues, the top five vaccine manufacturers (which includes Sanofi, Merck, GSK, Pfizer, and Novartis) have captured 90% of the world's total vaccine business. The major part of the growth in vaccine sales has been recorded in the adult influenza market, with the huge emphasis on flu restraint and an acceptance of adult vaccines (Global Vaccine Market 2010; Palmer and Bryant 2013). According to reports, total time of development is usually 12 to 15 years with an investment of approximately US\$0.5 to 1 billion. This includes the selection of proper vaccine candidates, followed by phase I, phase II, and phase III trials. The successful completion of phase III trial data permits entry into a licensing program. Currently, approximately 145 pure vaccines and 11 combination vaccines are available for human use and the segment includes pediatrics, adolescents, adults, and the elderly (Castellani 2010).

## 11.5.1 Historical background and new challenges

The long and tedious history of vaccines began with the discovery of the smallpox vaccine in 1796 by Edward Jenner, a term derived from "vacca," meaning cow. Jenner used cowpox virus (vaccinia) to develop a vaccine that prevented smallpox infection. This was followed by a series of discoveries, which are highlighted below under types of vaccines along with year of discovery, the name of the discoverer, and the routes of administration. The types and classifications of vaccines usually depend on the strategies and applications, not only to reduce the risk of illness but also to induce a beneficial immune response.

#### 11.5.1.1 Types of vaccines

*Killed/inactivated vaccines*—these vaccines are produced from killed or inactivated microorganisms. Because the process of inactivation involves subjecting the pathogens to rigorous lethal conditions, there can be variations in the antigenicity of these vaccines. Therefore, repeated booster doses are recommended to maintain long-term immunity. Some specific examples include: vaccines against cholera (1879, Louis Pasteur, now not in use), rabies (1885, Louis Pasteur and Emile Roux), plague (1897, Alexandre Yersin), pertussis (1926, Thorvald Madsen), influenza (1945, Thomas Francis), polio (1952, Jonas Salk), anthrax (1930s in Russia; 1950s in the UK and the US), rabies HDCV (1967, Koprowski, Wiktor, and Plotkin), rabies inactivated vaccine (1971, Koprowski, Wiktor, and Plotkin), and hepatitis A (1992, Hilleman).

*Live, attenuated vaccines*—these vaccines contain live microorganisms that have been weakened (attenuated) to disable their virulent properties but to retain their antigenproducing capacity. Therefore, these can no longer be capable of producing any pathological complaints. However, some special precautions are required so that the attenuated forms could not mutate back into their virulent forms, producing the same disease that they were meant to prevent. Some specific examples include: typhoid (1896, Wright), tuberculosis BCG (1921, Calmette and Guerin), yellow fever (1926, Max Theiler), polio oral (1950, Koprowski), measles (1963, Enders; 1969, Hilleman), mumps (1967, Hilleman), rubella (1969, Hilleman), MMR (1971, Hilleman), chicken pox (1974, Hilleman), cholera (1980, oral), and rotavirus (1998).

*Toxoids*—these vaccines are manufactured using toxins produced by a pathogenic organism. After chemical treatment, the harmful effects of the toxin are removed and the inactivated toxin is used to produce a lower level of immunological protection. Some specific examples include: tetanus (1890, Behring) and diphtheria (1913, Behring).

*Subunit vaccines*—these represent antigens such as proteins to the immune system without introducing whole organisms. Specific examples include: vaccine against hepatitis B with protein subunits (1981, Hilleman), the virus-like particle vaccine against human papillomavirus (2006, Frazer and Zhou), and the hemagglutinin and neuraminidase subunits of the influenza virus.

*Conjugate vaccines*—these contain the polysaccharide outer coats with proteins or toxins. Specific examples include: *H. influenzae* type B vaccine (1977, Hilleman), pneumococcal (1977, Hilleman), meningococcal (1978, Hilleman), and typhoid oral (1975).

#### 11.5.1.2 Challenges

Despite the fact that the vaccine industry has been proved to be one of the fastest-growing sectors during the past decade, the vaccine business also has had great challenges, both from the development and marketing points of view. Manufacturers are trying to develop new and advanced methods, not only to produce cost-effective and safe vaccines but also at the same time looking into novel vaccine delivery systems, such as needle-free patches, edible vaccines, and nasal sprays. Currently, the US and European multinational companies are dominating the global market; however, they face a great threat with the emerging Chinese and Indian companies. With the introduction of new tools and the growing awareness for vaccine-preventable diseases, adults and the elderly are entering into focus as an appealing target population for future vaccine development. The most emerging and challenging area for big companies is to introduce new vaccines to combat malaria, dengue, tuberculosis, HIV/AIDS, and vaccines against cancer.

In addition, vaccines against diabetes, asthma, allergy, and hypertension are also under development. These vaccines are quite challenging and still need considerable work to move forward. Furthermore, it will be too early to predict any reasonable progress that has been noted in developing these specific vaccines. Some highly constructive developments have been noted to introduce peptide-based malaria vaccines capable of inducing antibodies against *Plasmodium falciparum*. Although the peptide sequence (Glu-Val-Leu-Tyr-Leu-Lys-Pro-Leu-Ala-Gly-Val-Tyr-Arg-Ser-Leu-Lys-Lys-Gln-Leu-Glu) derived from MSP-1, which is conserved in all parasite strains, but success remains questionable and more work is still required to develop malaria vaccine because *P. falciparum* is highly effective at avoiding detection by the immune system through different mechanisms (Lozano et al. 2006; Owens and Schweizer 2011; Urquiza et al. 1996).

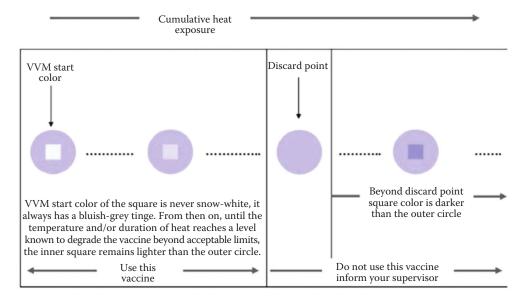
Transportation of vaccines is a highly gray area, which has great implications for stability and effectivity. The packaging conditions and temperature throughout the transportation period play a significant role in the success of the immunization process. The World Health Organization (WHO) provides reasonable guidelines for the packaging and shipping of vaccines. Any deviation from standard recommended packaging and the temperature is likely to result in an unsuccessful immunization process (WHO 2002). A timetemperature indicator known as a vaccine vial monitor (VVM) is absolutely vital to know that the vaccine has not been exposed to excessive heat and is suitable for immunization (Figure 11.37). This has now been successfully used for different vaccines. The VVM indicates three distinct stages based on color change—"use this vaccine," "discard point," and "do not use this vaccine" (Figure 11.38).



Figure 11.37 VVM. Manufactured by Temptime Corporation, Morris Plains, NJ.

In addition, the process of immunization itself is also quite challenging for some vaccines, and in a number of countries, because of some highly specific issues such as infertility, autism, and others, after the scientific data published by Slovakian researchers wherein polysorbate 80 (a nonionic surfactant), which is used as an emulsifier/solubilizing agent in some flu vaccines, was found to damage the vaginal and uterine lining, and induced hormonal changes, ovarian deformities, and infertility when injected into newborn female rats (Gajdova et al. 1993). After this, most manufacturers of flu vaccines using polysorbate 80 began highlighting "the manufacturer cannot guarantee that your fertility will be unharmed," in their package inserts. Another example of infertility and other side effects, such as anaphylactic shock, associated with polysorbate 80 present in HPV, have recently been highlighted in Australia and New Zealand through media and Internet campaigns, which asked parents not to allow their children to take part in HPV vaccination programs. Also, there is a general belief in some African countries, Afghanistan, and in northern parts of Pakistan that oral polio vaccine may cause infertility, and thus the situation remains challenging despite the fact that polio cases are still being reported from these areas. Perhaps the greatest misunderstanding or controversy developed during 2003 to 2004 when the State Government of Kano, Northern Nigeria detected the presence of two sex hormones, estrogen and progesterone, in polio vaccine. Although the concentration detected was very low and perhaps would not have been able to inhibit fertility, this caused tremendous concern among Nigerians and it was treated as a misdeclaration from the manufacturer and they launched a campaign, saying—"nobody wants their child to be crippled by polio, and nobody wants her child to be sterile, either." Although a clarification was made on the situation by saying that because the polio vaccine is developed in a culture made from monkey kidney, which contains these two sex hormones, and because the hormones are highly water-soluble, traces are bound to be found in the vaccine (Maiyali 2013).

However, it is still not clear, if by any means, the levels of these two sex hormones increased in monkey kidney due to some pathophysiological abnormality, then what will



*Figure 11.38* VVM showing three distinct stages based on color change—"use this vaccine," "discard point," and "do not use this vaccine." Manufactured by Temptime Corporation, Morris Plains, NJ.

be the consequences if the concentrations of these two sex hormones increased in the vaccine? Because generally, the batches of polio vaccines are released without monitoring or estimating the levels of these sex hormones.

Another challenging situation developed relating to polio vaccines when in 1992, an interview with Hilary Koprowski was published in *Rolling Stone Magazine*, mentioning oral polio vaccine (OPV) as a possible source of HIV and, in turn, the AIDS epidemic. The possible link between OPV and HIV was based on the information that chimpanzees' kidney cells were used to culture polio virus, which had been infected with simian immuno-deficiency virus (SIV), and thus a vaccine produced through such a cell culture could lead to HIV infection. Hilary Koprowski was a pioneering virologist who developed the first successful oral polio vaccine (he recently died at the age of 96). Although, at that time, a number of clarifications and explanations were given at significant levels, it was really difficult and it is still challenging to regain the confidence of the African people on the subject and this certainly needs to be addressed correctly with more justifications.

There is a huge list of ingredients (thimerosal, causing autism; aluminum, a neurotoxic; 2-phenoxyethanol, affecting organs and causing CNS disorders; formaldehyde, a nephrotoxic; monosodium glutamate, causing retinal degeneration; MRC-5 and WI-38 cellular protein taken from aborted human fetuses, serious ethical objections; and oxtoxinol, a vaginal spermicide) that are used for different purposes such as preservatives, stabilizers, and adjuvants in different vaccines, which certainly needs more attention and studies so that alternatives for these toxic chemicals and objectionable ingredients can be found.

#### 11.5.2 Production and pharmaceutical presentation

The large scale of vaccine production is generally an industry secret and most multinational companies have the patent rights to produce particular types of vaccines. However, a generalized production method with raw materials and area requirement is available to understand the basic phenomenon of vaccine production. A highly dedicated area and controlled conditions are required for vaccine manufacturing. However, sections and the modules of the manufacturing process are usually specific to either viral or bacterial vaccine production. Various manufacturing steps (such as propagation, isolation, purification, and formulation) are conducted under aseptic conditions and the procedure is strictly monitored for Good Manufacturing Practices (GMP) compliance throughout the whole process, which was designed and recommended by some international regulatory authorities.

The propagation step involves the growth and development of selected organisms in a suitable medium in a bioreactor. The nutritional requirements and fermentation conditions are maintained according to the type of microorganism used for a specific vaccine. This is followed by isolation, which involves the separation of microbial cells from the fermented media at the end of the fermentation process. The separation of microbial cells is usually done with the help of ultracentrifugation or some specific polysaccharide extraction method.

The purification step, which eliminates various impurities or materials sticking to the microbial cells in case of bacterial vaccine, is very specific to the antigens and some time may be required for chemical precipitation or fractionation, followed by ultrafiltration and chromatography. For the development of conjugated bacterial vaccines, carrier proteins may be conjugated at this stage to some polysaccharide vaccines, followed by the purification of conjugated vaccine by ultracentrifugation or chromatography. In case of viral vaccines, the purification step also involves centrifugation, followed by ultrafiltration, and chromatography. The viruses may be chemically inactivated to produce killed vaccine at this stage.

The pure cells thus obtained in the purification step are subjected to formulation, which is composed of mixing of the purified product in a suitable solution to achieve a required concentration followed by the addition of preservative to some vaccines, to warrant product stability in terms of microbial growth or contamination or to prevent cross-contamination from multidose vaccine vials. In case of bacterial vaccines, the purified products may be combined with several other antigens to develop the desired vaccine. Specific examples are of pneumococcal vaccine, which is developed by combining 23 different types of polysaccharides and diphtheria-tetanus-purtussis-*H. influenzae* type b-hepatitis B (DTP-Hib-HepB) is produced by combining bacterial vaccines with viral antigens. For manufacturing toxoids such as diphtheria and tetanus, exotoxins are used after subjecting to heat, UV light, or chemical treatment, which destroys the toxic part without altering their antigenic specificity and often boosting their immunogenicity.

Some highly specific methods are still under investigation to manufacture DNA vaccines and recombinant vector vaccines. The regulatory authorities are continuously monitoring the quality of vaccines produced and marketed worldwide. Various tests are performed by the regulatory authorities during the monitoring program, which includes safety tests (performed by injecting the product intraperitoneally into laboratory animals), identity test (based on the type and nature of vaccine), purity test (performed to check that the product meets the international standards of purity), potency testing (to check that vaccine confers protective immunity), and sterility test (performed to check that the vaccine is free from microbial contamination).

#### 11.5.3 Biotechnological approach

With the introduction of recombinant DNA technology, utilizing both microorganisms and mammalian cells have been exploited enormously for the production of vaccines in recent years. However, the research and development to introduce new vaccines is still mostly moving around a few advanced countries and their respective strong multinational groups. During the last 25 years, more than two-thirds of the new vaccines were developed and introduced only by US-based companies (Douglas et al. 2008). Recent advances introduced in molecular biology and genetic engineering techniques are now available, which grants the isolation and separation of strategic genetic sequences to produce vaccines for effective immune response. A wide range of host cells, such as bacteria, yeast, chicken eggs, and even plants have been examined as potential sources for engineered vaccines in substantial quantities. Tissue culture techniques are also being investigated in genetically modified host cells to produce large quantities of antigens.

Some vaccines, such as antihuman papillomavirus vaccine and hepatitis B vaccine are now manufactured through gene techniques. A single gene, usually a surface glycoprotein of the virus, can be expressed in a foreign host by cloning. Bacteria (e.g., *Escherichia coli* and *Bacillus subtilis*) and yeasts (e.g., *Saccharomyces cerevisiae*) are commonly used as expression hosts whereas viruses or plasmids are usually used as expression vectors on a commercial scale to produce large amounts of antigens, which are used as vaccines after purification and combination with an adjuvant.

The role of nanobiotechnology in the development of vaccines is also under investigation because nanoparticles play an important role in the drug delivery system for biological therapies. No adverse reactions were noted at the site of immunization in experimental animal modules when nanobeads were used as an adjuvant, instead of alum, the most commonly used adjuvant for vaccines. Nanobeads serve a dual purpose, that is, apart from stimulating antibody production, it also produces T-cells. The size of the nanobeads (measuring 40 nm) has critical value in vaccine production. Because most viruses are of a similar size (40 nm), the nanobeads are taken up profusely by the immune system and deceived into elevated levels of T-cell production. The baculovirus expression vector system, in recent years, has also been evaluated and observed as a feasible option for large-scale production of flu vaccines. In addition, further advances in virology also provide a platform for the production of virus-like particles as vaccines against emergent disease and viral vectors for gene therapy (Roldao et al. 2011).

Some highly useful biotechnological techniques are under investigation to develop new types of vaccines, such as DNA vaccines, which use genes that code for those all-important antigens. A few such DNA vaccines against influenza and herpes are also being tested in humans. Another experimental vaccine is the recombinant vector vaccine, which uses attenuated viruses or bacterium to introduce microbial DNA to the cells of the body. A number of vaccine research institutes and multinational companies engaged in developing new vaccines are working hard on both bacterial as well as viral recombinant vector vaccines for HIV, rabies, and measles. It is expected that with the introduction of more advanced biotechnological methodologies and approaches, the vaccine industry will grow more rapidly and will benefit from exploring and developing some novel vaccines helping mankind all over the world.

#### 11.5.4 Preventive/therapeutic role

Vaccines have both bioprophylactic and biotherapeutic roles and thus their applications have certainly generated value for mankind greater than any other biologicals or biopharmaceuticals available commercially in the pharmaceutical market. However, its established therapeutic role for the treatment of rabies and in the treatment of cancer, based on some recent preclinical and clinical studies, have delivered reasonable testimony that this distinctive therapeutic modality will lead to new theories and models in both clinical trial design and endpoints and in combination therapies. It is expected that after a successful launch, cancer vaccines will ultimately be employed for the therapy of many different types of cancer (Schlom et al. 2007).

The preventive roles of vaccines are quite predominant. According to a WHO report, every year, 3 million deaths are prevented and 750,000 children are saved from disability by vaccines (WHO 2009). The WHO has a comprehensive vaccine-preventable disease monitoring system, which highlights immunization schedules by disease covered with antigens all over the world, divided into six different regions: Africa, the Americas, Eastern Mediterranean, Europe, Southeast Asia, and the Western Pacific.

## 11.6 Flavonoids

The natural polyphenolic compounds with low molecular weight represent a diverse group of secondary metabolites, known as flavonoids, with distinct and pronounced biological activities. Their importance has now been explored comprehensively to ascertain their role as antimicrobials (antiviral, antibacterial, antifungal, antiprotozoal, etc.), antioxidants (responsible for beneficial effects on cardiovascular disease), anti-inflammatory, antiallergic, anticancer, antiobesity, and others. Because of the health-promoting effects, the usage and application of flavonoids and flavonoid-derived compounds have now become very popular and have drawn considerable attention all over the world to promote as nutritional supplements or as pharmaceutical products (Fowler and Koffas 2009; Shaik et al. 2006).

## 11.6.1 Historical background and new challenges

Historically, flavonoids were reported in plants in different forms such as free aglycones and glycosides. Bioflavonoids are responsible for the development of various colors. In

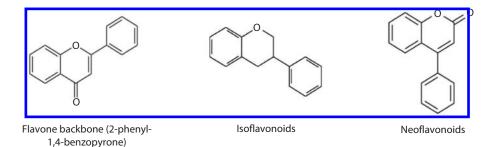


Figure 11.39 Classification of flavonoids under IUPAC nomenclature 1997.

view of their effects on the permeability of vascular capillaries, an additional name, "vitamin P" was also assigned; however, the name was later declared obsolete during the early 1950s. Based on the IUPAC (1997) nomenclature system, flavonoids are now classified under three categories: flavones, isoflavonoids, and neoflavonoids (Figure 11.39).

#### 11.6.2 Production and pharmaceutical presentation

The biosynthesis mechanism of flavonoids in plants is quite complex and involves a series of enzymatic reactions. The major skeleton is phenylpropanoids, which are synthesized from phenylalanine after the removal of an amino group by the enzymatic action of phenylammonium lyase to produce transcinnamic acid. The hydroxylation of the aromatic ring of transcinnamic acid by the enzyme cinnamate 4-hydroxylase thus produces *p*-coumaric acid, which can then be ligated to coenzyme A by another enzyme, 4-coumaroyl-CoA ligase.

A number of flavonoid-based oral preparations are available in the international market for the management of various diseases. However, these products are largely sold as nutritional or food supplements in the US and European markets, but in China, India, and in some South Asian countries, a proper place has been given to these products as a drug in the respective/traditional systems of medicine practiced in these countries. Although in some cases, comprehensive pharmacological investigation results and even the clinical studies data are available to support their usage and application. Yet more information on treatment, health benefits, and side effects with marketed flavonoid drugs products are desired for more authentications and for U.S. Food and Drug Administration consideration for approval as a drug. Currently, flavonoids are mostly available in the form of film-coated tablets/caplets, soft gel capsules, powders in a sachet, and creams for topical application.

#### 11.6.3 Biotechnological approach

The isolation, extraction, and purification of secondary metabolites from plants have always been posed as a difficult task in view of several factors affecting the final yield of the desired product. The seasonal and regional variations along with low concentrations of the pure compound certainly create challenges for researchers. Although, in some cases, chemical synthesis usually provides a reasonable alternative, but the intense reaction environments and disastrous chemicals required to synthesize flavonoids makes it less acceptable. Parallel to this, production through tissue culture has also been reported, but it is still difficult to organize the cell viability along with the engineering challenges affecting cultivation during large-scale production of flavonoids. Therefore, to meet the increased demand for flavonoids and to effectively utilize their therapeutic role, microbial production through genetically modified organisms is becoming progressively imperative from a recombinant manufacturing point of view. The field is interesting and open, but challenging for the biotechnologist looking for a successful commercial plan and roadmap because despite substantial developments in improving microbial strains and the yields of certain flavonoids, the overall fermentation process still needs considerable attention from researchers for more improvements in terms of process optimization and media composition.

A large number of genetically modified organisms have been used to biosynthesize various flavonoids, during which the precursor molecules play a significant role in the yield of the desired compound. *E. coli, S. cerevisiae, Streptomyces venezuelae,* and *Phellinus igniarius* have been extensively studied in the production of some important flavonoids such as pinocembrin, naringenin, apigenin, luteolin, genkwanin, eriodictyol, kaempferol, chrysin, galagin, genistein, flavanone, dihydroflavonol, flavone, flavonol, dihydrokaempferol, quercetin, and exopolysaccharides (Figure 11.40).

In view of the structural similarity of yeasts with plant cells, it has been extensively investigated as a module to observe the genes relating to the biosynthesis of flavonoids. There are some definite advantages with using yeast, which permits the posttranslational alterations of protein and endomembrane systems in eukaryotes into which plant P450 enzymes are typically embedded, together with their cytochrome P450 reductase partner.

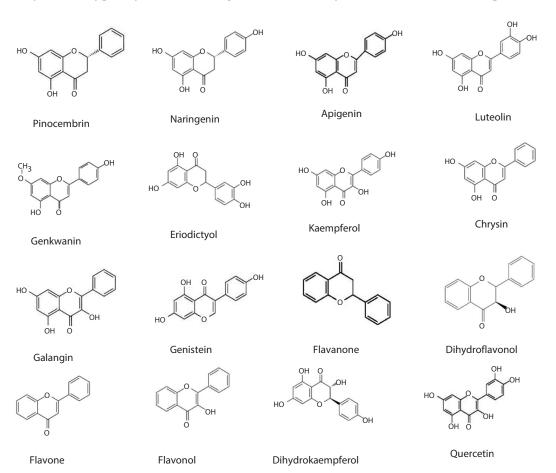


Figure 11.40 Some important flavonoids produced by microorganisms.



Figure 11.41 Flavonoids with antimicrobial properties.

## 11.6.4 *Therapeutic role*

aFlavonoids have now been identified in playing a significant therapeutic role in the management of various diseases. The role of flavonoids as antimicrobials, including controlling HIV, has been investigated and reported by a number of authors. Its antioxidant property ultimately confers its beneficial effect on cardiovascular disease (including reductions in low-density lipoproteins), allergies, and cancer. The antimicrobial activity is linked with the phenolic groups of flavonoids. Baicalein, karanjin, and galangin (Figure 11.41) are the most important flavonoids identified to possess antimicrobial activity.

## 11.7 Case study: Optimization of media composition for the production of ergot alkaloids from a mutant strain of Botryodiplodia sp. FRL-1955 through surface culture technique

With the increased medical importance of ergot alkaloids and its analogues, optimization processes for increased production of ergot alkaloids have been taken into consideration by research workers worldwide. However, major emphasis was given to several species of *Claviceps, Penicillium*, and *Aspergillus* for process optimization to increase yield.

In the present study, we report the production of ergot alkaloids in a mutant strain of *Botryodiplodia* species, which belongs to the class *Ascomycetes* and lives in a saprophytic way. It normally needs injured tissue to parasitize the plant. The presence of alkaloids has been reported in *Botryodiplodia* sp. during the earlier phase of the studies. The alkaloids have also been shown to possess antibacterial activity against some common pathogens when tested through bioautographic techniques (Mahmood 1983; Mahmood et al. 1986). However, more in-depth studies are required to further explore and establish the medicinal value of *Botryodiplodia* species in which this study is part of the program.

The generation of pycnidia, in which spores are formed, is a characteristic feature of *Botryodiplodia*. Pycnidia rarely develop on artificial media or may develop after a long time. If generated, it can be seen with the naked eye as black balls the size of a pinhead. The spores are elliptical and large. Young and immature spores are colorless and unicellular, whereas mature spores are brown, distichous, and thick-walled. The fast and constantly growing mycelium of the fungus is snow-white at first, turning black within 3 to 4 weeks. This discoloring is not due to the generation of spores.

## 11.7.1 Materials and methods

The fungus was grown by surface culture technique using 1-L capacity Roux bottles containing 100 mL of sterile media with glucose (25 g/L), mannitol (25 g/L), and triammonium citrate (12 g/L), which is supplemented with the following minerals:  $MgSO_4.7H_2O$  (2,000 mg/L),  $KH_2PO_4$  (250 mg/L),  $ZnSO_4.7H_2O$  (20 mg/L),  $CuSO_4.5H_2O$  (2.5 mg/L), and  $MnSO_4.H_2O$  (15 mg/L) at pH 5.6. One milliliter of spore suspension, prepared from a 5- to 7-day-old stock culture of *Botryodiplodia* sp. FRL-1955, grown on potato dextrose agar slant at 25°C, was used to inoculate each Roux bottle and were incubated at 30°C for 28 days. Selection of the media was based on several studies to optimize the production capacity of strain FRL-1955.

Upon completion of the incubation period, the mycelial cells were separated from the fermented medium by filtration using a suction pump, washed with sterile distilled water, and homogenized for extracting the alkaloids in 50 mL solvent (1:1 mixture of acetone and 2% tartaric acid) with 6 h contact time. The solids were removed by centrifugation and the extract was mixed with filtered fermented broth. The content was acidified with 2% acetic acid and treated with ethyl acetate to separate the impurities. The aqueous layer was basified with 20% NH<sub>4</sub>OH solution up to pH 10.0 and extracted three times with chloroform. The combined chloroform was concentrated under reduced pressure at low temperature in a rotary evaporator. The chloroform extract was fractionated on thin-layer chromatography plates using chloroform—methanol (8:2). The total alkaloids content was determined using Van Urk reagent (Van Urk 1929), as modified by Smith (1930). The amount of alkaloids present was calculated from a standard curve of ergotamine.

## 11.7.2 Results and discussion

Approximately 1.2 mg of alkaloid per milliliter of medium was obtained in this study, which is quite encouraging. The current study revealed that a medium composed of glucose in combination with mannitol and triammonium citrate supplemented with inorganic salts is an excellent source of carbon and nitrogen for the production of ergot alkaloids. The carbon-to-nitrogen ratio of 4.16:1 was recorded as optimum after several studies and was found in accordance with a previous study using Aspergillus fumigatus (Narayan and Rao 1982). During the initial optimization program, starch and molasses were also employed as the sole carbon sources, but were noted to be inferior in alkaloid production capacity of Botryodiplodia sp. Similarly, during the optimization program, it was also observed that an increase in nitrogen source level decreases alkaloid production. The increase in nitrogen level suppresses mycelial growth and, eventually the yield of alkaloids because a major part of the alkaloids accumulate in the mycelium. Different pH values between 4 and 8 were investigated and pH 5.6 was noted as optimum. Values lower than 3.5 and more than 7.5 significantly affected the growth and thus the alkaloid content. A lower concentration of potassium phosphate was used in the present study because high concentrations were reported to inhibit the secondary metabolites (Martin 1977). In conclusion, it was observed that different sources of carbon as well as the ratio between carbon and nitrogen, along with the concentration of inorganic phosphate and pH of the medium, can greatly affect the overall yield of ergot alkaloid.

## 11.8 Conclusion

The microbial healthcare products classified under secondary metabolites such as antibiotics, antitumor agents, vaccines, bioactive alkaloids, and flavonoids have received considerable attention in recent years because of their importance in our daily life, both from a therapeutic and prophylactic point of view. Therefore, reviewing the topic in light of recent biotechnological advancements is fully justified. The new dimension and biotechnological approach is expected to bring a revolution with the introduction of new antibiotics with a wider spectrum of activity, highly selective antitumor drugs, more safe and effective vaccines, and better understanding of submerged fermentation for the production of bioactive compounds or drugs through genetic manipulation. Based on this concept, this chapter was written to provide recent and updated information for both students and researchers, not only to observe the dynamics of the topic but also to analyze the challenges expected during the coming decades.

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## chapter twelve

## Microbial biomass production

## Ashok K. Rathoure

## Contents

12.1	Introduction	.279
12.2	Yeast	280
	12.2.1 Yeast metabolism	280
	12.2.2 Role of yeasts in baking	283
		283
12.3	*	286
	12.3.1 Baker's yeast as natural therapy for cancer	286
	12.3.3 Baker's yeast in genetics study	
12.4		288
	12.4.1 Production of SCP	289
	12.4.2 SCP from waste	. 291
	12.4.3 Advantages and disadvantages of SCP	292
12.5	8 8	.293
Ackr	nowledgments	293
Refe	rences	293

## 12.1 Introduction

Biomass refers to microbial mass, vegetable or plant matter, and sometimes animal matter that can be converted into an energy source. Microbial biomass refers to the biomass produced by the action of microbial strains, for example, baker's yeasts, single-cell proteins (SCP), biofertilizers, and biopesticides.

The history of yeast started 9000 years ago in the production of food. Bread, wine, sake, and beer are made with the essential contribution of yeasts, especially from the species *Saccharomyces cerevisiae*. Yeasts were first used in the Caucasian and Mesopotamian regions approximately 7000 BC (Lin and Tanaka 2006). However, there was no recorded information about yeast until Louis Pasteur discovered that it could produce  $CO_2$  and ethanol from sugar. After the discovery of fermentation by Pasteur, yeast fermentation became famous and by the end of the nineteenth century, the production of bread and beer become a common practice. Wine producers are more disinclined to change their traditional practices and only began using exogenous yeast inocula in the 1950s, especially in countries with less wine tradition. In the 1960s, yeast biomass production, which was used in European countries, contributed to the technology of producing large amounts of dry yeast (Reed and Nagodawithana 1988). Efficient and profitable processes have been developed to produce yeast biomass at the laboratory and plant scale. The standard process has

been developed from previous observations and experience for highest yield by increasing biomass production and decreasing costs (Daromola and Zampraka 2008; Dien et al. 2003).

## 12.2 Yeast

Yeasts are single-celled eukaryotic microorganisms classified in the kingdom Fungi, and are found on and around the human body. Although some species with yeast forms may become multicellular through the formation of a string of connected budding cells known as pseudohyphae or false hyphae, as in most molds. The fungi include edible mushrooms, common baker's yeast, and molds that ripen blue cheese as well as produce antibiotics for medical and veterinary use. Many consider edible yeast and fungi to be as natural as fruits and vegetables. Yeasts were identified as living organisms after the invention of the microscope, and it is also the agent responsible for alcoholic fermentation and dough leavening. With the knowledge that yeast is a living organism, and the ability to isolate yeast strains in pure culture form, the commercial production of baker's yeast started in the early twentieth century. Since that time, bakers, scientists, and yeast manufacturers have been working to find and produce pure strains of yeast that meet the specialized needs of the baking industry (Prescott and Dunn 1959; Van Hoek et al. 1998).

More than 600 different species of yeast are known and they are widely distributed in nature. They are found in association with other microorganisms as part of normal inhabitants of soil, vegetation, marine, and other aqueous environments. Baker's yeast is used to leaven bread throughout the world. Baker's yeast is produced from a species of yeast called *S. cerevisiae*. The genus in the scientific name of baker's yeast, *Saccharomyces*, refers to saccharo meaning sugar and myces meaning fungus. The species name *cerevisiae* is derived from the name Ceres, Roman goddess of agriculture. The size of the yeast cell equals that of a human red blood cell, and the shape varies from spherical to ellipsoidal (Bekatorou et al. 2006).

The reproduction in yeast takes place by budding (vegetative), a process in which a new bud grows from the side of an existing cell wall. This bud eventually breaks away from the mother cell to form a separate daughter cell. Each yeast cell undergoes this budding process 12 to 15 times before it is no longer capable of reproducing. The yeast should grow under controlled conditions on a sugar-containing medium during commercial production; yeast cells reproduce every 2 to 3 hours under ideal growth conditions.

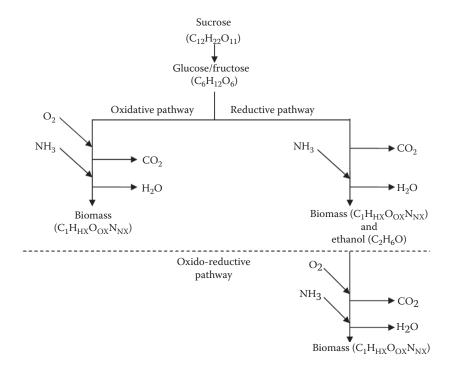
#### 12.2.1 Yeast metabolism

Metabolism refers to catabolic and anabolic pathways in the cell. Catabolic pathways are oxidative processes that remove electrons from substrates or intermediates used to generate energy, whereas anabolic pathways are a reductive process for the production of new cellular material. These processes use NADP or NAD as cofactors. There is a metabolic diversity in how organisms generate and consume energy from substrates. Although all yeasts derive their chemical energy from the breakdown of organic compounds (ATP). Knowledge of the underlying regulatory mechanisms is not only valuable in understanding the general principles of regulation but also of great importance in biotechnology, if new metabolic capabilities of particular yeasts have to be exploited. Now, it is well established that most yeasts employ sugars as their main carbon and hence energy source, but there are particular yeasts that can utilize nonconventional carbon sources. With regard to nitrogen metabolism, most yeasts are capable of assimilating simple nitrogenous sources to biosynthesize amino acids and proteins. Aspects of phosphorus and sulfur metabolism as well as aspects of metabolism of other inorganic compounds have been studied in detail (Nielsen and Jewett 2007; Ostergaard et al. 2000). The metabolic pathway of *S. cerevisiae* is presented in Figure 12.1.

The yeasts reoxidize NADH to NAD in a two-step reaction from pyruvate, which is first decarboxylated by pyruvate decarboxylase and the reduction of acetaldehyde is catalyzed by alcohol dehydrogenase in alcoholic fermentation. The alternative mode of glucose oxidation is the hexose phosphate pathway, which provides pentose sugars and cytosolic NADPH for biosynthetic reactions of the cell such as the production of fatty acids, amino acids, and sugar alcohols. Another function of this pathway is the production of ribose sugars, which serve in the biosynthesis of nucleic acid precursors and nucleotide coenzymes (Mormeneo and Sentandreu 1982).

Yeasts can be categorized into several groups according to their modes of energy production, respiration, or fermentation as regulated by environmental factors. Yeasts can adapt to varying growth environments and the prevailing pathways depend on actual growth conditions, for example, glucose can be used in different ways by *S. cerevisiae* depending on the presence of oxygen and other carbon sources (Table 12.1) as well as on the yeast's mode of respiration (Table 12.2; Birhanu et al. 1982; Gunasekaran and Chandra 1999). Yeasts use disaccharides as a substrate, which can be utilized by extracellular and intracellular enzymes, as presented in Table 12.3.

Catabolite repression is caused by glucose or the products of glucose metabolism, which represses the synthesis of various respiratory and gluconeogenic enzymes, whereas catabolite inactivation results in the rapid disappearance of enzymes upon the addition of glucose. Enzyme activity is lost by dilution with cell growth in catabolite repression,



*Figure 12.1* Metabolic pathways of *S. cerevisiae*. (From Sonnleitner, B., and Kappeli, O., *Biotechnology* and *Bioengineering* 28, 6: 927–937, 1986.)

Substrate	Intermediates	Enzymes	Products
Saccharose	None	Invertase	Glucose + fructose
Maltose	None	Maltase	Glucose
Melibiose	None	Melibiase	Glucose + galactose
Glucose	Glyceraldehyde-3-phosphate	Hexokinase	Pyruvate
Giucose Giyceraldenyde-3-phosphate		Phosphoglucose isomerase Phosphofructokinase Fructose bisphosphate aldolase Triosephosphate isomerase Glyceraldehyde 3-phosphate dehydrogenase Phosphoglycerate kinase Phosphoglycerate mutase Enolase Pyruvate kinase	
Ethanol	Acetaldehayde, acetyl-CoA, oxaloacetate	Alcohol dehydrogenase	Glucose by gluconeogenesis
Lactate	Pyruvate	Lactate dehydrogenase	Glucose by gluconeogenesis
Glycerol	Glycerol-3-phosphate, dihydroxyacetonphosphate	Glycerol kinase	Glucose by gluconeogenesis

Table 12.1 Carbon Sources for Growth of S. cerevisia	Table 12.1	Carbon Sources	for Growth	of S.	cerevisiae
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Source: Adapted from H. Feldmann. Yeast: Molecular and Cell Biology, 2012. Wiley-Blackwell: United Kingdom.

Types	Examples	Respiration	Fermentation	Anaerobic growth
Obligate respires	Rhodotorula sp. Cryptococcus sp.	Yes	No	No
Anaerobic respires	Candida sp. Kluyveromyces sp. Pichia sp.	Yes	Anaerobic in pregrown cells	No
Aerobic fermenters	S. pombe	Limited	Aerobic and anaerobic	No
Facultative aerobic fermenters	S. cerevisiae	Limited	Aerobic and anaerobic	Facultative
Obligate fermenters	Torulopsis sp.	No	Anaerobic	Yes

Source: Adapted from H. Feldmann. Yeast: Molecular and Cell Biology, 2012. Wiley-Blackwell: United Kingdom.

although enzymes are no longer synthesized due to gene repression by signals derived from glucose or other sugars. The nature of signals is currently not clear. Glucose repression in yeast describes a long-term regulatory adaptation to degrade glucose exclusively to ethanol and  $CO_2$ . Therefore, when *S. cerevisiae* is grown aerobically on high concentrations of glucose, fermentation will account for the bulk of glucose consumption. In batch cultures, when the levels of glucose decline, cells become gradually derepressed resulting in the induction of respiratory enzyme synthesis. This results in the oxidative consumption of ethanol, when cells enter a second phase of growth known as the diauxic shift.

Disaccharide	Extracellular hydrolysis	Intracellular hydrolysis	Products	Organism
Maltose	-	Maltase	2 Glucose	S. cerevisiae
Sucrose	Invertase	-	Glucose + fructose	S. cerevisiae
Melibiose	α-Galactopyranosidase	-	Glucose + galactose	S. carlsbergensis
Lactose	-	β-Galactosidase	Glucose + galactose	Kluyveromyces
Cellobiose	β-Glucosidase	-	2 Glucose	Brettanomyces

Table 12.3 Disaccharides as Substrates in Yeasts

Source: Adapted from H. Feldmann. Yeast: Molecular and Cell Biology, 2012. Wiley-Blackwell: United Kingdom.

Catabolite inactivation is faster than catabolite repression because glucose deactivates a limited number of key enzymes such as fructose 1,6-bisphosphatase. Inactivation occurs primarily by enzyme phosphorylation, followed by slower vascular degradation of the enzyme. The cAMP as a second messenger plays a central role in regulating catabolite repression and inactivation in *S. cerevisiae* (Cook 1958).

#### 12.2.2 Role of yeasts in baking

Yeast is a key ingredient for the production of bakery products and plays an important role in  $CO_2$  production, dough maturation, and fermentation flavor.  $CO_2$  is generated by yeast as a result of the breakdown of fermentable sugars in the dough. The evolution of  $CO_2$  causes an expansion of the dough trapped within the protein matrix. Dough maturation is accomplished by the chemical reactions of yeast producing alcohols and acids from the proteins of flour, and by the physical stretching of protein and  $CO_2$ , which results in the light and airy physical structure associated with yeast-leavened products. Yeast also gives bread its characteristic flavor. During dough fermentation, yeast produces many secondary metabolites such as ketones, higher alcohols, organic acids, aldehydes, and esters. Some of them escape during baking. Others react with each other and with other compounds to form new and complex flavor compounds. These reactions occur primarily in the crust and crumb of baked breads.

Yeasts can grow in the presence or absence of oxygen. Anaerobic growth is slow and inefficient, for example, in bread dough. The sugars that can sustain either fermentation or growth are used mainly to produce alcohol and  $CO_2$ . Only a small portion of the sugar is used for cell maintenance and growth. Under aerobic conditions, and in the presence of a sufficient quantity of dissolved oxygen, yeast grows by using most of the available sugar for growth and produces only negligible quantities of alcohol. If the concentration of sugar in the fermentation growth medium is greater than a very small amount, yeast can produce some alcohol with an adequate supply of oxygen or in large quantities. This problem can be solved by adding the sugar solution slowly to the yeast throughout the fermentation process. This type of fermentation is referred to as fed-batch fermentation (Sheoran et al. 1998).

#### 12.2.3 Yeast production

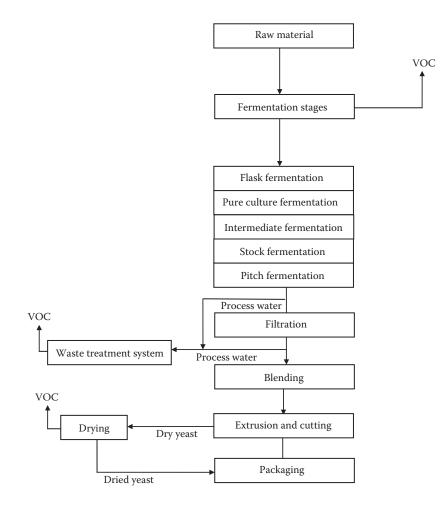
The basic sources of carbon and energy for yeast growth are sugars. Starch cannot be used because yeast does not contain the enzymes to hydrolyze starch. Beet and cane molasses

are commonly used as raw materials because the sugars present in molasses are readily fermentable. The molasses are a mixture of sucrose, fructose, and glucose. In addition, yeast also requires certain minerals, vitamins, and salts for growth. Some of these can be added to the blend of beet and cane molasses before flash sterilization whereas others are fed separately from the fermentation. Alternatively, a separate nutrient feed tank can be used to mix and deliver some of the necessary vitamins and minerals. Each of these nutrients is fed separately to the fermenter to permit better pH control of the process. The sterilized molasses are stored in a separate stainless steel tank. The sterilized molasses are referred to as mash or wort. The mash stored in this tank is used to feed sugar and other nutrients to the appropriate fermentation vessels (Atiyeh and Duvnjak 2003; Zamani et al. 2008). A variety of essential nutrients and vitamins is also required in yeast production. The nutrient and mineral requirements include nitrogen, potassium, phosphate, magnesium, and calcium, with traces of iron, zinc, copper, manganese, and molybdenum. Normally, nitrogen is supplied by adding ammonium salts, aqueous ammonia, or anhydrous ammonia to the feedstock. Phosphates and magnesium are added in the forms of phosphoric acid or phosphate salts and magnesium salts. Vitamins such as biotin, inositol, pantothenic acid, and thiamine are also required for yeast growth. Thiamine is added to the feedstock. Most other vitamins and nutrients are already present in sufficient amounts in the molasses malt (Chen and Chigar 1985).

Baker's yeast production starts with a pure culture, which serves as the inoculum for prepure culture tank. After growth, the contents are transferred to a larger pure culture fermenter in which propagation is carried out with aeration under sterile conditions. These stages should be conducted as set batch fermentations. In set batch fermentation, media and nutrients are introduced to tank before inoculation. From pure culture vessels, the grown cells are transferred to a series of progressively larger seed and semiseed fermenters. These later stages are conducted as fed-batch fermentations. During fed-batch fermentation, molasses, phosphoric acid, ammonia, and minerals are fed to yeast at a controlled rate. This rate is designed to feed enough sugar and nutrients to the yeast to maximize multiplication and prevent the production of alcohol. It is not economical to use pressurized tanks to guarantee the sterility of the large volumes of air required in these fermenters or to achieve sterile conditions during all the transfers through the many pipes, pumps, and centrifuges. Extensive cleaning of the equipment, steaming of pipes and tanks, and filtering of the air should be practiced to insure aseptic conditions. At the end of semiseed fermentation, the contents of the vessel are pumped to a series of separators that separate the yeast from spent molasses, and then the yeast is washed with cold water and pumped to a semiseed yeast storage tank where the yeast cream is held at 34°F, until it is used to inoculate the commercial fermentation tanks (Acourene et al. 2007).

The commercial fermentations should be carried out in large fermenters with working volumes of up to 200,000 L. To start the commercial fermentation, a volume of water referred to as set water, is pumped into the fermenter. At the start of the fermentation, the liquid seed yeast and additional water may occupy only about one-third to one-half the volume. Constant additions of nutrients during the course of fermentation brings the fermenter to its final volume. The rate of nutrient addition increases throughout the fermentation because more nutrients are supplied to support the growth of the increasing cell population. The number of yeast cells increases approximately fivefold to eightfold during this fermentation. Air is provided to the fermenter through a series of perforated tubes located at the bottom of the vessel. The rate of airflow is approximately one volume of air per fermenter volume per minute. A large amount of heat is generated during yeast growth, and cooling is accomplished by internal cooling coils or by pumping the fermentation liquid through an external heat exchanger. The regulation of pH, temperature, and airflow should be monitored carefully and controlled by computer systems during the entire process. Throughout fermentation, temperature should be maintained at around 30°C and pH at approximately 4.5 to 5.5. Yeast growth ranged from 120 kg in the intermediate fermenter, 420 kg in the stock fermenter, 2,500 kg in the pitch fermenter, and 15,000 to 100,000 kg in the trade fermenter (Figure 12.2; Birhanu et al. 1982; Zheng et al. 2005).

Once an optimum quantity of yeast has been grown, the yeast cells are recovered from the final fermenter. The centrifuged yeast solids should be concentrated by filter press or rotary vacuum filter. A filter press forms a filter cake containing 27% to 32% solids, whereas a rotary vacuum filter cake contains approximately 33% solids. The filter cake should be blended in mixers with small amounts of water, emulsifiers, and cutting oils to form the end product. In compressed yeast production, emulsifiers are added to give the yeast a white and creamy appearance. A small amount of oil is added to help extrude the yeast to form continuous ribbons. The ribbons are cut into pieces, wrapped, and cooled to less than 8°C. The yeast is extruded in thin ribbons and dried in a batch or continuous



*Figure* 12.2 Flow diagram for the production of baker's yeast.

drying system. The yeast should be packed under vacuum or nitrogen gas before heat sealing. The shelf life at ambient temperature is 1 to 2 years (Chen and Chigar 1985).

Volatile organic compound emissions, that is, ethanol and acetaldehyde are generated as by-products of the fermentation process. Other by-products such as butanol, isopropyl alcohol, 2,3-butanediol, organic acids, and acetates are also generated during the fermentation process. Volatile by-products form as a result of either excess sugar (molasses) present in the fermenter or because of an insufficient oxygen supply. When anaerobic fermentation occurs, 2 mol of ethanol and 2 mol of  $CO_2$  are formed from 1 mol of glucose. Under anaerobic conditions, the ethanol yield increases as the yeast yield decreases. Hence, it is essential to suppress ethanol formation in the final fermentation by incremental feeding of the molasses mixture with sufficient oxygen. The ethanol formation rate is higher in pure culture than in the final fermentation process. The earlier fermentation stages are batch fermenters, where excess sugars are present and less aeration is used during the fermentation process. These fermentations are not controlled to the degree that the final fermentations are controlled because the majority of yeast growth occurs in the final fermentation stages. The process wastewater is another potential emission from yeast production (Chen and Chigar 1985; Reed and Peppler 1973).

## 12.3 New researches in baker's yeast

## 12.3.1 Baker's yeast as natural therapy for cancer

In 2010, at a special conference on cell death mechanisms sponsored by the American Association for Cancer Research in San Diego, California, Dr. Mamdooh Ghoneum presented his findings about the role of yeasts in cancer. This conference included new complexities of cell death and cell survival, new technologies, and the clinical translational aspects necessary for the evolution of new therapeutic strategies. Dr. Ghoneum pursued a theory that cancer cells self-destruct when exposed to small quantities of yeast. Dr. Ghoneum said that in laboratory tests, cancer cells were ingested by yeast through phagocytosis, and then the cancer cells died. First, Dr. Ghoneum investigated this phenomenon *in vitro*, introducing yeast to breast, tongue, colon, and skin cancers. In later experiments on mice, a decrease in the size of the tumor mass was observed. Dr. Ghoneum's observations showed significant clearance of cancer cells from the lung and concluded that when the cancer cells eat the yeast, they die. These findings have been confirmed by similar studies at the U.S. Department of Health and Science, National Institutes of Health.

## 12.3.2 Model organism for somatic mutations and cancer study

Recently, new evidence has provided additional insight into cancer research in the field of epigenetics. According to this new theory, cancer is a genetic disease because most cancer cells do what they do, that is, proliferate at a rapid speed as a result of mutations in their genetic sequence. Mutations frequently occur as a normal part of cell life and most have a negligible effect on phenotype. Therefore, scientists suspect that most malignant transformations result from the accumulation of multiple mutations. Over the past 40 years, scientists have identified dozens of different cancer-causing mutations. Most of these mutations occur in genes that play some kind of role in regulating the cell cycle, or a series of events that a cell goes through as it replicates its DNA and divides. Dr. Leland H. Hartwell discovered some of these cancer-causing mutations and eventually received the Nobel Prize in 2001. Hartwell worked on the single-celled eukaryote *S. cerevisiae* for studying cancer and the cell cycle in his doctorate.

The same genes that control the cell cycle in baker's yeast and that malfunction in tumor cells exist in more or less the same capacity in human cells. Dr. Hartwell identified more than 100 genes involved in cell cycle control. These genes are known as cell division cycle (CDC) genes. Dr. Hartwell also determined the pathway of cell cycle regulation events. Many of the same genes that work to regulate cell division in yeast were also shown to work similarly in humans. Hartwell found that CDC genes regulate the cell cycle either by stimulating or by inhibiting cell division in response to the barrage of signals these cells constantly receive from their environment. However, in cancer cells, mutated genes that normally stimulate cell division only at the right time and place, the proto-oncogenes, start operating like stuck accelerators (turning into oncogenes). Meanwhile, other mutated genes that normally inhibit excessive cell division, so-called tumor suppressor genes simply stop working, similar to broken brakes. Because of their role in regulating cell division so precisely and because they are found to be mutated in tumor cells, these genes are frequently often referred to as cancer genes. Most oncogenes function as dominant mutations, that is, only one mutated copy of the gene is needed to start operating like a stuck accelerator, whereas most tumor suppressor genes function as recessive mutations, that is, two mutated copies of the gene are needed to lose its ability to apply the brakes. Despite this difference, both types of mutations, stuck accelerators and broken brakes, ultimately lead to out-of-control cell growth and reproduction. Hence, the yeast strains can be used to control the cancer cells.

## 12.3.3 Baker's yeast in genetics study

Baker's yeast such as *S. cerevisiae* is used as a model system for molecular research and genetics because the mechanics of replication, recombination, cell division, and metabolism are conserved between yeasts and eukaryotes including mammals. The complete genome sequence has proven to be extremely useful as a reference toward the human genome sequence and other higher eukaryotic genes. The ease of genetic manipulation of yeast allows its use for conveniently analyzing and functionally dissecting gene products from other eukaryotes. In yeast, direct transformation is unique and has become a model organism; synthetic oligonucleotides can be directly transformed into yeast allowing for easy production of altered forms of proteins. Also, yeast has the unique ability to recombine exogenous DNA with partially homologous segments directly to the specific genome location of your choice. High rates of gene conversion coupled with homologous recombination in yeast have led to the development of techniques for direct replacement of genetically engineered DNA sequences into their normal chromosome locations.

Yeasts are so easy to work with because

- It permits rapid growth in the laboratory, dispersed cells, ease of replica plating, and mutant isolation
- It has a well-defined genetic system and highly versatile DNA transformation system
- It can be handled with few safety measures; no special safety equipment required
- Remarkably inexpensive compared with higher eukaryote organisms
- It can be easily frozen and stored at -80°C for later use

Baker's yeast production practices are subject to yeast stress. The most critical phenotypes for baker's yeast improvement are those that tolerate high levels of sucrose, tolerate freeze–thawing stress, rapidly use maltose, and produce high  $CO_2$ . Dough can contain up to 30% sucrose per weight of flour, which applies harsh osmotic stress on yeast cells. Baker's yeast strains must efficiently use maltose, which originates from flour in lean doughs. *S. cerevisiae* strains that have high sucrose tolerance and also rapidly use maltose have been developed and are used commercially. A desired characteristic for baker's yeast is the development of strains that use the disaccharide melibiose. Raffinose found in molasses is hydrolyzed by yeast invertase to fructose and melibiose. Baker's yeast does not have the ability to use melibiose because it lacks  $\alpha$ -galactosidase. This enzyme is present in bottom-fermenting brewer's yeast strains, and this gene-encoding enzyme has been introduced into laboratory strains of baker's yeast. The resulting strains give increased biomass without alteration of growth rate in fermentations. One of the most desired properties of baker's yeast strains is a rapid fermentation rate. Free sugars in dough are sequentially consumed due to transport and catabolite repression regulatory mechanisms. Amylase found in dough releases maltose from starch, but many strains of baker's yeast use maltose. *Torulaspora delbrueckii* is highly tolerant to freezing and freeze–thawing and this yeast has been considered as a primary yeast or helper in frozen dough products.

## 12.4 Single-cell protein

The production of SCPs started in 1960. The term SCP was coined in 1966 by Professor Carol L. Wilson at the Massachusetts Institute of Technology (Ware 1977). SCP refers to mixed proteins from pure or mixed cultures of algae, yeasts, fungi, or bacteria used as a substitute for protein in human foods and animal feeds. Since 2500 BC, yeasts have been used in bread and beverage production. In 1781, processes for preparing highly concentrated forms of yeast were established. In 1919, a fat production process using *Endomyces vernalis* from sulfite liquor was developed, and in 1941, *Geotrichum* was introduced to improve this process. In the 1960s, researchers at British Petroleum developed the technology of SCP production by yeast from waxy *n*-paraffins, a product of oil refineries. The concept of food from oil became entirely popular in the 1970s. SCPs are primarily used as poultry and cattle feed. SCP production technologies arose as a promising way to solve the problem of a worldwide protein shortage (Haider et al. 2003; Nigam 2000).

Large-scale fermenters are required for the production of SCP and in high biomass production, oxygen transfer rate and respiration rate increases metabolic heat production, hence the need for an efficient cooling system. In continuous operation, the economics of production such as substrate cost, energy, operating costs, waste, safety, and the global market should be taken into account. The substrate cost is the largest single cost factor. Simplifying the manufacture and purification of raw material can save costs. Moreover, the manufacture of raw materials is more economical in larger plants. Factors involved in the raw materials' costs are raw material production, process capacity of the plant, and substrate yield. The energy for compressing air, cooling, sterilizing, and drying forms the next most important cost factor. Sites with cheaply available thermal, electrical, fossil, or process-derived energy are preferred. The capital-dependent costs are determined by the cost of the apparatus for the process, capacity of a plant, and its conditions (mainly the size of the plant). Small plants can be profitable only if they include simplifications of the processes and materials to a considerable degree. The greater expenditure on apparatus in processes with cheap, simple, and unpurified raw materials usually does not pay in comparison with more expensive pure substrates and simpler technology. The quality of the product is poorer for a low-value unpurified product of a rational mini process with varying compositions or one including numerous additional components than for upgraded products. Upgrading the product may consist of purification and separation into the components of the microbial biomass. Because of genetic variability, the possibility of

technical control during manufacture, and the simplicity of the process, microbial biomass is more suitable for such special products compared with biomass from plants or animals.

Various bacteria, mold, yeast, and algae are used for the production of SCPs. The bacteria include *Brevibacterium*, *Methylophilus methylitropous*, *Acromobacter delvaevate*, *Acinetobacter calcoacenticus*, *Aeromonas hydrophilla*, *Bacillus megaterium*, *Bacillus subtilis*, *Lactobacillus sp.*, *Cellulomonas sp.*, *Methylomonas methylotrophus*, *Pseudomonas fluorescens*, *Rhodopseudomonas capsulata*, *Flavobacterium* sp., *Thermomonospora fusca*, and others. Yeasts such as *Candida utilis* (torula yeast), *Yarrowia lipolytica*, *Candida tropicalis*, *Candida novellas*, *Candida intermedia*, and *S. cerevisiae* are all among the various organisms that have been used for the production of SCP. Some of the algae such as *Chlorella pyrenoidosa*, *Chlorella sorokiana*, *Chondrus crispus*, *Scenedesmus acutus*, *Porphyrium* sp., and *Spirulina maxima* are used for SCP production. The filamentous fungi include *Chaetomium cellulolyticum*, *Fusarium graminearum*, *Aspergillus fumigates*, *Aspergillus niger*, *Aspergillus oryzae*, *Cephalosporium cichhorniae*, *Penicillium cyclopium*, *Rhizopus chinensis*, *Scytalidium aciduphlium*, *Tricoderma viridae*, *Tricoderma alba*, and *Paecilomyces varioti* have been employed for SCP production (Bhalla et al. 2007). The desired microorganisms should be cultured on the medium under sterile conditions. The organisms to be cultured should be nonpathogenic to plants, humans, and animals.

## 12.4.1 Production of SCP

Large-scale SCP production has contributed greatly to present-day advancements. The future of SCP will be heavily dependent on reducing the production costs and improving quality by fermentation, downstream processing, and improvement in the producer organisms as a result of conventional applied genetics together with recombinant DNA technologies. The process of SCP production from any microorganism or substrate would have the following basic steps:

- i. Provision of a carbon source; it may need physical and chemical pretreatment
- ii. Sources of nitrogen, phosphorus, and other nutrients need to support the optimal growth of selected microbial strains
- iii. Prevention of contamination by maintaining aseptic conditions; the medium components may be heated or sterilized through filtration, and fermentation equipment may be sterilized
- iv. The selected microbial strain should be inoculated in a pure state
- v. SCP processes are highly aerobic, so adequate aeration must be provided; in addition, cooling is necessary as considerable heat is generated
- vi. The microbial biomass is recovered from the medium

The production of SCP takes place in a fermentation process. This can be done using selected microbial strains cultivated on suitable raw materials directed to the growth of culture and cell mass followed by separation processes. After microbial screening, suitable production strains are obtained from samples of soil, water, air, or from swabs of inorganic or biological materials and subsequently optimized by selection, mutation, or other genetic methods. In addition, process engineering and apparatus technologies are adapted to the technical performance of the process to prepare the product for use on a large technical scale. Safety demands and environmental protection are also considered in SCP production in relation to both the process and the product. Finally, safety concerns and protection of innovation introduce legal and control aspects, that is, operating licenses, product authorizations for particular applications, and the legal protection of new processes and

strains of microorganisms. The classic raw materials are substances containing monosaccharides and disaccharides because almost all microorganisms can digest glucose, other hexose and pentose sugars, and disaccharides. Other substrates for SCP include bagasse, citrus wastes, sulfite waste liquor, molasses, animal manure, whey, starch, sewage, and others (Nasseri et al. 2010).

The fermentation process requires a pure culture of the selected microbial strain, sterilized growth medium, production fermenter, cell separation, collection of cell-free supernatant, product purification, and effluent treatment. There are many methods available for concentrating the solutions such as filtration, precipitation, centrifugation, and the use of semipermeable membranes. The equipment used for these dewatering methods are expensive and thus would not be suitable for small-scale productions and operations. SCPs need to be dried to 10% moisture or they can be condensed and denatured to prevent spoilage. Therefore, an adequate technology that maintains appropriate growth conditions for a prolonged period must be implemented specifically for the purpose of obtaining high yield and productivity. Batch fermentation is inadequate for biomass production because the conditions in the reaction medium change with time. Fed-batch fermentations are better suited for the purpose of biomass production because they involve the control of carbon source supply through feeding rates. Fed-batch cultures are still in use for baker's yeast production using well-established and proven models. However, they have not been favored for the production of SCP at a large industrial scale. Prolonging a microbial culture by the continuous addition of fresh medium with the simultaneous harvesting of product has been implemented successfully in industrial fermentations destined for biomass production. The technical implications of chemostat culture are various and extremely relevant. The flow diagram for the production of SCP is illustrated in Figure 12.3.

A common problem for industrial fermentations is the appearance of a profuse foam on the headspace of the reactor. Among the various designs, deep-jet fermenters and airlift fermenters have been the most successfully applied. Airlift fermenters have had great success as the configuration of choice for continuous SCP production. The biomass from yeast fermentation processes is normally harvested by continuous centrifugation; filamentous fungi are harvested by filtration. Then, biomass is treated for RNA reduction and dried in a steam drum of spray driers. Drying is expensive, but results in a stabilized product with a shelf life of years. Solid-state fermentation (SSF) is the growth of microorganisms

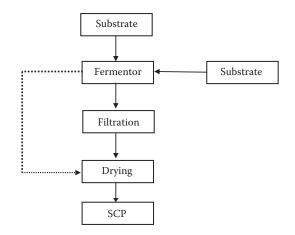


Figure 12.3 Flow chart for SCP production.

on predominantly insoluble substrate, where there is no free liquid. Generally, under combined conditions of low water activity and the presence of intractable solid substrate, fungi show luxuriant growth. Hence, proper growth of fungi in SSF gives much higher concentrations of the biomass and higher yield when compared with submerged fermentation. The advantage in SSF processes is the efficient use of waste as the substrate to produce commercially viable products. The process does not need elaborate prearrangements for media preparation. The process of SSF initially concentrated on enzyme production. Currently, there is worldwide interest for SCP production due to the dwindling conventional food resources. Large-scale production of microbial biomass has many advantages over traditional methods for producing proteins (Tipparat and Kittikun 1995).

## 12.4.2 SCP from waste

SCP production from waste and the effective advantages of microorganisms compared with conventional sources of protein is well known (Argyro et al. 2006). A number of agricultural and agro-industrial waste products are used for SCP production and other metabolites including orange waste, mango waste, cotton salks, kinnow-mandarin waste, barley straw, corn cops, corn straw, rice straw, onion juice, sugarcane bagasse (Nigam et al. 2000), cassava starch (Tipparat and Kittikun 1995), wheat straw (Abou Hamed 1993), banana waste (Saquido et al. 1981), capsicum powder (Zhao et al. 2010), and coconut water (Smith and Bull 1976). The use of such wastes as the sole carbon and nitrogen source for SCP production by microbial strains can be attributed to their abundance in nature and their cheap cost. The amount of agricultural and some industrial wastes used for SCP production can be locally very high and may contribute to a significant level of pollution in the water. Thus, the use of such materials in SCP processes serves two functions: the reduction in pollution and the creation of edible protein. Cellulose from agriculture and forestry sources constitutes the most abundant renewable resource in the planet as potential substrates for SCP production. Cellulose has emerged as an attractive substrate for SCP production, but in nature, it is usually found in complex form with lignin, hemicellulose, starch, and others. Therefore, if cellulose is to be used as a substrate, it must be pretreated chemically (acid hydrolysis) or enzymatically (cellulases) to make it a fermentable sugar. To use lignocellulose, pretreatment is usually necessary. Many of the pretreatment methods that have been reported vary from alkali or acid treatment, steam explosion, or x-ray radiation. To date, lignocellulosic wastes are used in mushroom production. Wood can also be cooked in a medium containing calcium sulfite with excess free sulfur dioxide. Lignin is converted to lignosulfonates and hemicellulose is hydrolyzed to monosaccharides and may be further broken down to furfurals. The amount of free sugars in the spent liquor varies with the type of procedure chosen as various cellulose fibers may be obtained with different degrees of degradation. Spent sulfite liquor has been used as a substrate for fermentation in Sweden since 1909 and later in many other parts of the world. S. cerevisiae is the first organism used for SCP production, although this organism is unable to metabolize pentoses that are found in considerable amounts in waste products. Later, other organisms better suited for the assimilation of all sugar monomers were chosen, that is, *C. tropicalis* and *C. utilis*.

Yeast biomass from sulfite liquor has been used as a food source during periods of war. However, in Finland, baker's yeast is produced from sulfite liquor using the Peliko process. The protein content of the fungus *P. variotii* exceeds 55% (w/w) and is officially approved as a food in Finland. The extracellular cellulases are commercially used in a celluloseseparating process. Cellulase is a complex of three enzymes including endocellulase, 1,4-β-cellobiohydrolase, and cellobiase. A number of efficient cellulase producers have been reported but Trichoderma viride continues to be the best known high cellulose-producing organism. C. cellulolyticum is another cellulolytic fungus that grows faster and forms 80% more biomass protein than Trichoderma. This means that C. cellulolyticum is suitable for SCP production, whereas T. viride is a hyperproducer of extracellular cellulases. The amino acid composition of C. cellulolyticum is generally better than T. viride, and is similar to alfalfa and soya meal protein. Starch is a cheaper and more amenable substrate for SCP production. Abundant amounts of carbohydrate may be obtained from rice, maize, and cereals. In tropical countries, cassava has been proposed as a good source of starch for SCP processes. The Symba process, developed in Sweden, uses starchy wastes combining two yeasts in a sequentially mixed culture; the amylase-producing Endomycopsis fibuligira and the fast-growing C. utilis. This process consists of three phases; the incoming starch waste is fed through heat exchangers and sterilized. The medium should feed to the first bioreactor where yeast grows and hydrolyzes starch. The hydrolyzed starch solution should feed to the second bioreactor where culture conditions favor the proliferation of *C. utilis*. Whey originates from the curding process in cheese production, but can now be obtained after ultrafiltration procedures for the production of spreading cheeses where the protein fraction corresponding to lactalbumins and lactoglobulins is incorporated into the casein fraction and all the proteins are in their native form, with the principal component being lactose (4%-6% w/v); although other nutrients can be found in significant amounts. Whey has been presented as an extremely suitable substrate for the production of SCP. In 1956, the French company, Fromageries Bel, produced yeast from whey using lactoseassimilating *Kluyveromyces marxianus*. Molasses is a by-product of the sugar manufacturing process. Biomass production from molasses requires supplementation with a suitable nitrogen source as well as phosphorus. Coffee-pressing wastes contain soluble carbohydrates and have a high chemical oxygen demand and soluble solid contents. In Guatemala, Trichoderma sp. is used to produce SCP on this substrate (Abou Hamed 1993; Prosser and Tough 1991; Rashad et al. 1990; Saquido et al. 1981).

#### 12.4.3 Advantages and disadvantages of SCP

Various factors such as nutrient composition, amino acid profile, vitamin and nucleic acid content as well as palatability, allergies, and gastrointestinal effects should be considered for nutritional value. The nutritive and food values of SCP vary with the microorganisms used. The method of harvesting, drying, and processing has an effect on the nutritive value of the finished product. SCP basically comprises proteins, fats, carbohydrates, ash, water, and other elements such as phosphorus and potassium. The composition depends on the organism and substrate used for growth. Proteins not only provide a nutritional component in a food system but also perform a number of other functions. SCP has high protein and low fat content. It has good amino acid composition, which is rich in lysine but poor in sulfur-containing amino acids such as methionine and cysteine. It is a good source of vitamins, particularly B-complex, thiamine, riboflavin, and folic acid. SCP from yeast and fungi has up to 50% to 55% protein and it has high protein–carbohydrate ratio compared with forage. SCP from bacteria has more than 80% protein, although it is poor in sulfur-containing amino acids and it has high nucleic acid content (Burgents et al. 2004; Campa-Cordova et al. 2002; Mahajan and Dua 1995; Oliva-Teles and Goncalves 2001; Tovar et al. 2002).

There are many problems in the adoption of SCP on a global basis such as a high concentration of nucleic acids (6%–10%), which elevates the uric acid level in serum and

results in kidney stone formation. Approximately 70% to 80% of the total nitrogen is represented by amino acids, whereas the rest occur in nucleic acids. This concentration of nucleic acids is higher than in other conventional proteins and is characteristic of all fastgrowing organisms. SCP from yeasts and fungi have high nucleic acid content. The filamentous fungi show slower growth rates than yeasts and bacteria, there is a high risk of contamination, and some strains produce mycotoxins; hence, they should be well screened before consumption. Similarly, SCP from bacteria has high ribonucleic acid content, and high risk of contamination during the production process and during the recovery of the cells. All these detrimental factors affect the acceptability of SCP as a global food (Nasseri et al. 2011).

## 12.5 Conclusion

Microorganisms such as algae, bacteria, yeasts, and fungi are used as protein sources. Yeasts can be considered as the oldest industrial microorganism. It is believed that the early fermentation systems for alcohol production and bread making formed through the natural microbial contaminants of flour, other milled grains, and from fruits or other juices containing sugar. Such microbial flora would have included wild yeasts and lactic acid bacteria, which were associated with cultivated grains and fruits. For hundreds of years, it was traditional for bakers to obtain the yeast to leave their bread as by-products of brewing and wine making. The baker's yeast industry is a major market. New strains of baker's yeast from recombinant DNA technology, which produces  $CO_2$  more rapidly and is more resistant to stress, or which produces proteins or metabolites that can modify the flavor of bread, dough rheology, or shelf life are now emerging. This low-value, high-volume product is produced under stringent environmental conditions to obtain the maximum product/biomass yield, which is dependent on process design, cost, and the strain of S. cerevisiae used. As with all biotechnological processes, research and development in this area is ongoing to create more beneficial strains and to optimize the fermentation and processing steps that make up the baker's yeast process. From many studies, it was found that yeast cells help control cancer. The same genes that control the cell cycle in baker's yeast (and that malfunction in tumor cells) exist in more or less the same capacity in human cells, and that the yeast strains can be used to control cancer cells. S. cerevisiae is used as a model system for molecular research because the basic cellular mechanics of replication, recombination, cell division, and metabolism are generally conserved between yeast and larger eukaryotes, including mammals. The SCP is another widely used microbial biomass, which has proteins, fats, carbohydrates, ash, water, and other elements such as phosphorus and potassium. SCP is not widely accepted as a global food due to its high nucleic acid content. Attempts to improve the acceptability of SCP products should be intensified. Further research and development will ensure the use of microbial biomass as SCP or as a supplement in the diet.

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## chapter thirteen

# *Microbial biofertilizers and their pilot-scale production*

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## Contents

13.1	Introduction	. 298
13.2	Microorganisms used as biofertilizers	. 299
	13.2.1 <i>Rhizobium</i>	. 299
	13.2.2 Azotobacter	. 299
	13.2.3 Azospirillum	. 299
	13.2.4 Mycorrhiza	
	13.2.5 Cyanobacteria	.300
	13.2.6 Plant growth-promoting rhizobacteria	.300
13.3	Biofertilizer application methodology	.301
	13.3.1 Seed treatment	
	13.3.2 Root dipping	.301
	13.3.3 Soil application	.301
13.4	Mass production of bacterial biofertilizer	.301
	13.4.1 Culturing of microorganisms	.302
	13.4.2 Preparation of biofertilizer packs	
13.5	Pilot-scale production of <i>Rhizobium</i> and <i>Azotobacter</i> biofertilizer in a rural area	
	of Eastern India	.303
	13.5.1 Stepwise calculation of pilot-scale production cost of <i>Rhizobium</i> and	
	Azotobacter biofertilizers (according to existing cost of utilities, wages,	
	and labor cost, etc., in eastern regions of India)	.305
	13.5.2 Quality control	.307
13.6	Cyanobacteria (BGA) biofertilizer production, quality control, and	
	popularization for entrepreneurship development in rural areas of India	
	13.6.1 Present status of the technology: Its strengths and weaknesses	.308
	13.6.2 Details of the appropriate technology	.309
	13.6.2.1 Production in polyhouse	
	13.6.2.2 Use of alternate carrier materials	.309
	13.6.2.3 Quality control parameters and testing methodologies	. 310
	13.6.3 Cyanobacteria biofertilizer production on a small scale; economics of	
	the technology and popularization for entrepreneurship development	. 310
	13.6.3.1 Usefulness of the technology for providing self-employment	. 312
	13.6.3.2 Economics (cost-benefit of the technology adapted as cottage	
	industry)	. 312

13.7	Constraints in the commercialization of microbial biofertilizer technology	313
	13.7.1 Extension programs	313
	13.7.2 Marketing constraints	
13.8	Conclusion	314
Ackr	nowledgments	314
	rences	

## 13.1 Introduction

Due to uncontrolled population growth in many developing countries, there is a heavy demand for food. Hence, the aim is to achieve sustained growth rates that are high enough to feed an enormous population without degrading the environment. Researchers are focusing on an evergreen revolution, which can be achieved through sustainable agriculture (Parr et al. 1990). However, slow poisoning of the soil due to chemical fertilizers is one of the major calamities of present-day agriculture. As a result of excessive fertilizer use, the quality of soil is continuously degraded year after year, making it difficult to sustain soil fertility. These chemicals, in addition to causing damage to the soil's health, also creates a chain of ecological and economic problems. The residual effects of these chemical fertilizers severely affect the soil microflora, especially on the nodulation of legume crops and other nitrogen-fixing free-living microorganisms that play a key role in productivity. Hence, the only alternative is to use biological nitrogen fixation technology for the maintenance of soil health and sustainable productivity in agriculture.

The goal of an evergreen revolution and sustainable agriculture to meet the needs of the present generation without endangering our natural resources is mainly dependent on soil organic matter for nutrient supply through efficient microorganisms. Microorganisms not only contribute to nitrogen fixation but also solubilize phosphate, produce growthpromoting substances, and inhibit the growth of plant phytopathogens resulting in better growth and yield of crops. The main advantages of such sustainable agriculture are ecological balance, low cost of cultivation, an increase of the soil microflora to maintain good and fertile soil, clean environment, and nutritious food without residues that are harmful to human health (Bhatnagar and Patla 1996).

The following have been given importance for sustainable agriculture:

- 1. Use of efficient N<sub>2</sub> fixing microorganisms and use of potentially indigenous (local) strains instead of exogenous ones.
- Because chemical fertilizers reduce soil organic matter, reduce soil porosity, impede oxygen flow, reduce the water-holding capacity of soil, and obstruct natural nitrogen fixation by soil bacteria, dependence on these chemical inputs needs to be lessened.
- 3. Agrochemical inputs are neither economically feasible nor environmentally desirable on a long-term basis. Therefore, improvement of the crop yield by inoculation with biofertilizers such as *Rhizobium*, *Azotobacter*, *Azospirillum*, Cyanobacteria, plant growth-promoting rhizobacteria (PGPR), and so on are to be adapted as ecofriendly technologies in agriculture.

Biofertilizers are defined as preparations containing living or dormant cells of efficient strains of nitrogen-fixing, phosphate-solubilizing, or cellulolytic microorganisms used for the application of seed, soil, or composting areas with the objective of increasing the number of such microorganisms and accelerating certain microbial processes to augment the availability of nutrients in a form that can be assimilated by plants. Hence, in present-day agriculture, biofertilizers are important natural resources in integrated nutrient management and are advocated for use as supplements to chemical fertilizers for increased productivity and sustainable agriculture.

## 13.2 Microorganisms used as biofertilizers

## 13.2.1 Rhizobium

*Rhizobium* is a soil bacterium able to colonize the roots of leguminous plants and fix atmospheric nitrogen symbiotically. They are non-spore-forming rods, motile, aerobic, Gramnegative, and capable of utilizing several carbohydrates. Initially, all root nodule-forming, nitrogen-fixing bacteria were assigned to a common name, Rhizobium, and were classified into species based on a cross-inoculation group concept (Subba Rao 1993). Based on a polyphasic approach despite splitting the genus into five different genera, they are still commonly referred to as rhizobia. These genera are Rhizobium with 14 root-nodulating nitrogen-fixing species and four nonnodulating non-nitrogen fixing species, Sinorhizobium with nine species, Mezorhizobium with eight species, Bradyrhizobium with four species, and a single species of Azorhizobium. Recently, two more genera, Methylobacterium with one species under alpha-proteobacteria and Burkholderia under B-proteobacteria have been included in the rhizobia group. The morphology and physiology of *Rhizobium* vary from free-living conditions to the bacteroid in the nodules of legumes. The symbiotic association is spread among a wide range of host species belonging to the family Leguminosae. Out of the 748 genera and nearly 19,000 legume species, nearly 49% of the genera and 16% of the species have thus far been examined for nodulation; of which 41% of the genera and 14% of the species have been found to bear rhizobial nodules (Yadav and Mowade 2005). They are the most efficient biofertilizers according to the quantity of nitrogen they fix. It has been estimated that 40 to 250 kg N/ha/year is fixed by different legume crops due to colonization of *Rhizobium*.

## 13.2.2 Azotobacter

*Azotobacter* is one of the most extensively used plant growth-promoting microorganisms because its inoculation benefits a wide variety of crops. These are polymorphic, possess peritrichous flagella, produce polysaccharides, sensitive to acidic pH, high salts, and temperatures above 35°C, grow on a nitrogen-free medium, and fix atmospheric nitrogen. Several types of azotobacteria groups have been found in the soil and the rhizosphere such as *Azotobacter chroococcum*, *Azotobacter nigricans*, *Azotobacter paspali*, *Azotobacter armenicus*, *Azotobacter salinestris*, and *Azotobacter vinelandi*, of which *A. chroococcum* is most commonly found in Indian soils (Subba Rao 1993). Besides nitrogen fixation, they produce siderophores, antifungal substances, and plant growth regulators (Suneja et al. 1996). Production of phytohormones such as auxin, cytokinin by *A. chroococcum*, have been reported (Verma et al. 2001). The bacterium also produces an abundant amount of slime, which helps in soil aggregation.

## 13.2.3 Azospirillum

Azospirilla are Gram-negative, vibroid to straight rods, sometimes curved, highly pleomorphic, with an abundant accumulation of poly beta hydroxybutarate in the cytoplasm, motile, with a preference for salts and organic acids as carbon source, oxidase positive, and are capable of nitrogen fixation. *Azospirillum* is comprised of seven species. These *are Azospirillum amazonense, Azospirillum brasilense, Azospirillum doebereinnerae, Azospirillum halopraeferens, Azospirillum irakense, Azospirillum largimobile,* and *Azospirillum lipoferum.* These are isolated from the root and aboveground parts of a variety of crop plants. The organism proliferates under both anaerobic and aerobic conditions but it is preferentially microaerophilic in the presence or absence of combined nitrogen in the medium. Apart from nitrogen fixation, production of growth-promoting substances, disease resistance, and drought tolerance are the additional benefits due to *Azospirillum* inoculation (Yadav and Mowade 2005).

## 13.2.4 Mycorrhiza

Mycorrhizal fungi are intimately associated with plant roots, and form a symbiotic relationship. The plant provides sugars to the host and the fungi provide nutrients such as phosphorus to the plants. Mycorrhizal fungi absorb, accumulate, and transport phosphate within their hyphae and release to plant cells in the root tissue. There are three major groups of mycorrhiza: ectomycorrhiza, endomycorrhiza, and ectendomycorrhiza, of which the former two are important in agriculture and forestry. Plants inoculated with endomycorrhiza have been shown to be more resistant to root diseases. Mycorrhiza increase higher branching of plant roots facilitating contact with a wider area of the soil surface, which results in an increase of the absorbing area for water and nutrients.

## 13.2.5 Cyanobacteria

Cyanobacteria are also known as blue-green algae (BGA), which obtain their energy through photosynthesis. Cyanobacteria include unicellular, colonial, and filamentous species. Some filamentous forms show the ability to differentiate into several different cell types: vegetative cells or photosynthesizing cells, which form under favorable growing conditions; akinetes or the climate-resistant spores, which form when environmental conditions become harsh; and thick-walled heterocysts, which contain the enzyme nitrogenase for nitrogen fixation. Heterocyst-forming species are specialized for nitrogen fixation and are able to fix nitrogen into ammonia (NH<sub>3</sub>) or nitrates (NO<sup>-3</sup>), which can be absorbed by plants and converted into proteins. Rice plants utilize fixed nitrogen from healthy populations of cyanobacteria as well as from the aquatic fern *Azolla* and the cyanobacterium *Anabaena*, as a symbiont occurring in waterlogged fields and hence are used as biofertilizers for rice.

## 13.2.6 Plant growth-promoting rhizobacteria

The group of bacteria that colonize roots or rhizosphere soil and that are beneficial to crops are referred to as PGPR. The PGPR inoculants promote growth through the following mechanisms: suppression of plant disease, improved nutrient acquisition, and phytohormone production. These PGPR are principally species of *Pseudomonas* and *Bacillus*, which produce biostimulants like indole-acetic acid, cytokinins, gibberellins, and inhibitors of ethylene production. They cause crops to have greater amounts of fine roots, which have the effect of increasing the absorptive surface of plant roots for uptake of water and nutrients.

Despite promising test results when applied to various crops, biofertilizers have not received widespread application in agriculture, mainly because of the variable response

of plant species to inoculation depending on the microbial strain used, the agroclimatic conditions, and cultivation practices that vary from region to region, especially in a tropical country like India with varied climatic regimes. On the other hand, good competitive ability with the indigenous flora is one of the major factors determining the success of an inoculated strain in establishing a biofertilizer. Studies into the synergistic activities and persistence of specific microbial populations in complex environments, such as the rhizosphere, need to be addressed to obtain efficient inoculants and to obtain appropriate formulations of microbial inoculants incorporating nitrogen-fixing, phosphate-solubilizing, and PGPR, which will help in promoting the use of such beneficial microbes as biofertilizers for sustainable agriculture.

## 13.3 Biofertilizer application methodology

Biofertilizers are applied in three different ways, that is, seed treatment, root dipping (or seedling treatment), or soil application depending on the crop for which it is used.

## 13.3.1 Seed treatment

Seed treatment is the most common method of application of bacterial inoculants. *Rhizobium, Azotobacter, Azospirillum,* along with phosphate-solubilizing microorganisms (PSM), and PGPR are used in this process. For small quantities of seeds up to 5 kg, the coating can be done in a plastic bag. For this purpose, the bag is filled with 2 kg or more of seeds and closed in such a way to trap as much air as possible. Then, the bag is squeezed for up to 5 min or until all the seeds are uniformly wetted. Then, the bag is opened, inflated again, and shaken gently. After ascertaining that each seed has received a uniform layer of culture coating, the bag is opened and the seeds dried under the shade before use. For large amounts of seeds, coating can be done in a bucket and inoculant mixed directly by hand.

## 13.3.2 Root dipping

For application of *Azospirillum*/PSM, seedlings of paddy and vegetable crops are raised in a nursery. The required quantity of *Azospirillum*/PSM biofertilizer is mixed with 5 to 10 L of water in which the seedlings are dipped for up to 30 min followed by transplantation.

## 13.3.3 Soil application

Almost all types of biofertilizers can be used for soil application and their recommended dose per acre varies from the biofertilizer type and the crop for which it is applied. For PSB, cow dung/farmyard manure along with rock phosphate are mixed, kept under shade overnight, and maintained at 50% moisture. The mixture is used as a soil application in rows or during the leveling of soil. This is similar with the method of application of *Azospirillum* and *Azotobacter* in addition to seed dipping; however, cyanobacteria are used for broadcasting after transplantation.

## 13.4 Mass production of bacterial biofertilizer

Biofertilizers are carrier-based preparations containing efficient strains of nitrogen-fixing, phosphate-solubilizing, or cellulolytic microorganisms. The organic carrier materials are

more effective for the preparation of the inoculants. The carrier-based inoculants carry more microbial cells and also support the survival of cells for longer periods of time. The mass production of carrier-based bacterial biofertilizers involves the following three steps:

- Culturing of microorganisms, the process of which is dependent on the type of organism used
- Type of carrier material it's processing before mixing with a particular type of microorganism
- Mixing the carrier with the microbial culture produced, its packing and storage

An ideal carrier material is to be chosen for the production of quality biofertilizer. Peat, soil, lignite, vermiculite, charcoal, and soil are used as carrier materials. However, neutralized peat soil/lignite are found to be better carrier materials especially for *Rhizobium*, *Azotobacter, Azospirillum*, and PSB biofertilizer preparations. The important criteria for selection of ideal carrier material are lower cost, local availability, high organic matter content, nontoxic to microbes, water-holding capacity of more than 50%, and ease of processing during packaging. The carrier materials are crushed into a fine powder that could pass through a 212-µm IS sieve. Peat soil or lignite are acidic, thus they need to be neutralized using calcium carbonate (1:10 ratio), followed by autoclaving to eliminate contaminants before being used.

## 13.4.1 Culturing of microorganisms

Microorganisms are produced in large quantities for a preparation of biofertilizers. For this purpose, nutrient broths are prepared by inoculating the seed material in flasks containing the appropriate medium and grown in an incubator shaker at ambient temperature until a cell population of approximately 10<sup>10</sup> colony-forming units per milliliter (CFU/mL) is produced. Under optimum conditions, this population level could be attained within 4 to 5 days for *Rhizobium*, 5 to 7 days for *Azospirillum*, 2 to 3 days for phosphobacteria, and 6 to 7 days for *Azotobacter*. For large-scale inoculant production, inoculum from starter culture broths are transferred to large flasks/seed tank fermentors and grown until the required cell count level is reached. During the entire process, the culture broth is checked for contamination, if any. After harvesting, the cultures are mixed with suitable carrier material and packed. After fermentation, the broth should not be stored for more than 24 h because even at 4°C storage temperature, the number of viable cells starts decreasing.

## 13.4.2 Preparation of biofertilizer packs

Biofertilizer packets are prepared by mixing the culture with carrier material after processing. First, the carrier material is spread onto a clean, dry, sterile metallic or plastic tray, and then bacterial culture is added and mixed well manually using sterile gloves or using a mechanical mixer. The culture suspension is added at 40% to 50% of the water-holding capacity of the carrier. Curing is done by spreading the inoculants on a clean floor/polyethylene sheet/by keeping it in open shallow tubs/trays with polyethylene covering for 2 to 3 days at room temperature before packaging. Using this material, biofertilizer packs of 200 g quantities are prepared and sealed. If polyethylene bags are used for packing, these should be of low density grade and the thickness should be approximately 50 to 75  $\mu$ m. Each packet should contain the name of the manufacturer, product name, details of the strain used,

storage instructions, batch number, the crop for which it is to be applied, mode of application, date of manufacture, date of expiry, price, full address of the manufacturer, and others. The biofertilizer packets should be stored in a cool place away from the heat or direct sunlight. The population of inoculants in the prepared packets should be checked at 15-day intervals to ascertain that they contain more than 10<sup>9</sup> cells/g at the time of preparation and 10<sup>7</sup> cells/g while using for various crops (Motsara et al. 1995).

# 13.5 *Pilot-scale production of* Rhizobium *and* Azotobacter *biofertilizer in a rural area of Eastern India*

For mass production of microorganisms on a pilot scale, the following equipments are required: (1) 20-L fabricated fermenter fitted with an air pressure pump, (2) horizontal rotary shaker, (3) vertical autoclave with pressure regulator, (4) incubator, (5) refrigerator, (6) inoculation chamber (with provision of UV-C), (7) distillation apparatus, (8) microscope, (9) polyethylene sealer, and (10) air-conditioning provision.

Selected region-specific bacterial strains isolated by Sethi and Adhikary (2009a,b,c) were grown in slants and transferred to a liquid broth to prepare the mother culture. The cultures were finally grown on a large scale in a fermenter for up to 5 days. The cultures were harvested in batch culture mode and then mixed with unsterile brown sterile soil/ charcoal in a ratio of 1:3. Two hundred grams of shade-dried, carrier-mixed bacteria were packed in polyethylene bags, sealed, and then stored for use as biofertilizer for the desired crop. The limited scale production of bacterial biofertilizers as standardized at the extension laboratory in a rural area has been presented photographically (Figure 13.1).



*Figure 13.1* Stepwise preparation of biofertilizer from isolation of *Rhizobium* from effective nodules till packing of biofertilizer carried out at the extension laboratory in the village Maniakati under Surada block of Odisha. (1) Uprooted *Vigna radiata* plant with effective nodules. (2) *Rhizobium* colonies on YEMA plates. (3) Growth of *Rhizobium* on plates with YEMA and YEMA + Congo red. (4) Routine maintenance of *Rhizobium* in screw cap test tubes. (5) Preparation of YEMA medium. (6) Sterilization of medium. (7) Growth of *Rhizobium* strains in incubator at 28°C  $\pm$  2°C. (8) Five-day-old culture of *Rhizobium* and the carrier material (charcoal + brown forest soil 3:1) used for preparation of biofertilizer. (9) Mixing of *Rhizobium* suspension with carrier material. (10) Drying of prepared biofertilizer in shade. (11) Weighing of 200 g each of biofertilizer per packet. (12) Packing of biofertilizer.

The different steps of production of the biofertilizer are also given schematically in the following:

## Bacterial culture slants (stored in refrigerator at 4°C)

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Culture in tubes (28°C for *Rhizobium* and 30°C for *Azotobacter* grown for 5 days)

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Culture in 500 mL conical flasks containing broth medium (incubation on rotary shaker for 3–5 days)

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Inoculation of actively growing broth culture into fermenter (incubation at 28°C–30°C depending on the organism for 5 days)

 $\downarrow$ 

Checking the purity of the culture (by microscope and streaking culture methods)

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Bacteria harvested and mixed with carrier (mix 5 kg of carrier grinded brown forest soil/charcoal = 1:3 for 1 L culture, and air drying)

 $\downarrow$ 

Biofertilizer prepared and stored in 200 g packs

For the production of biofertilizers, the fermenters were fabricated to make the production cost more economical. The details about the requirements for fabrication of the low-cost fermenters is given below:

- 1. Glass vessel (Borosil, cat. no. 1585) 10 L or 20 L capacity
- 2. Stainless steel fermenter head: 6-mm-thick SS plate cut into a circular piece with two extended lobes having holes for holding clamping nuts, and the setup is fitted to the mouth of two half-circular clamps around the bottle of neck. Three 6-mm tubes will be welded to the plate, of which one will have ceramic sparger for aeration
- 3. Membrane filter cartridges (47 mm; Hepa vent filter discs, Whatman no. 6723-5000) will be fitted to one of the outlet tubes
- 4. For air inlet and outlet filter, 500 mL capacity bottle (Tarson cat. no. 6-2020) fitted with glass wool will be fixed
- 5. One aerator (aquarium pump) will be fixed to each setup for bubbling/aeration

For limited-scale production of biofertilizer, two glass fermenters of 10 L capacity are used for each batch. For 1 L of bacterial culture, 5 kg of carrier material is required for biofertilizer preparation. Thus, 100 kg of carrier-based biofertilizer is produced from two fermenters in each batch or 500 packets (200 g each) of biofertilizer is produced. From

one-time harvest in a week (on average), 500 packets of carrier-based biofertilizer can be produced. Thus, in a month, production will be 2000 packets, and in a year, it will be 24,000 packets (4800 kg from 960 L of broth). Following the protocol, the cost–benefit of commercial production of bacterial biofertilizer (*Rhizobium* as well as *Azotobacter*) developed by us is as follows.

## 13.5.1 Stepwise calculation of pilot-scale production cost of Rhizobium and Azotobacter biofertilizers (according to existing cost of utilities, wages, and labor cost, etc., in eastern regions of India)

- 1. Fixed cost: includes capital investment on equipments
- 2. Variable cost: raw material cost, carrier material, broth (i.e., chemicals), polyethylene bags for package of biofertilizer
- 3. Indirect cost:
  - a. Salary of the staff, marketing costs (transport costs, power consumption per packet), publicity cost, marketing margin (wholesale and retail, subsidy or commission, if any, risk coverage against unsold packets)
  - b. Miscellaneous expenses

Fixed cost (nonrecurring)

Equipment	Cost (\$)
Glass fermenter (fabricated with aerator), ×2	3000
Autoclave (one electrically operated), ×1 (20 L capacity)	600
Glass double distillation set, 5 L/h capacity, ×1	200
Incubator, ×1	400
Compound microscope (binocular), ×1	300
Refrigerator, ×1	200
Inoculation chamber with UV, ×1 (fabricated)	100
Polyethylene sealer, ×1	100
Chemical balance, ×1	100
Total	5000

## Variable cost per year (recurring)

Material	Quantity	Cost of the material (\$)
Carrier material	10 quintal	80
Broth	1000 L, \$0.25/L	250
Polyethylene bag	30,000, \$1/1000 bags	30
Recurring expenses per year (consumables) plastic/polypropylene bottle + plastic wares + glass wares (conical flask, pipettes, test tube, measuring cylinder, beaker etc.)		100
Total cost		460

Requirements for chemicals in the production process:

For production of *Rhizobium* broth per 1000 packets require 40 L of broth, for example, Himedia chemical cost with VAT according to 2009–2010 price lists as follows:

Chemicals required	For 40 L (g)	Total cost for 40 L (\$)
Mannitol	400	7.5
Yeast extract	16	1.0
NaCl	4	0.5
MgSO <sub>4</sub> ·7H <sub>2</sub> O	8	1.0
K <sub>2</sub> HPO <sub>4</sub>	20	1.0
Total cost		11.0

Therefore, for 40 L of broth, the cost of required chemicals = \$11 For 1 L =  $11 \div 40 = 0.275$ 

#### Indirect cost

Salary and wages per annum (according to the existing wages in rural areas of eastern regions of India)	Salary component (\$)
One microbiologist (skilled), \$200 per month	2400
One production assistant (unskilled), \$100 per month	1200
Miscellaneous expenses per year, \$100	1200

(Maintenance of equipment, fuel charge, office expenses etc.), total cost = 4800Thus, for 24,000 packets of biofertilizer, the production cost is the variable cost + indirect cost = 5260 + 10% depreciation on fixed cost 5500 = 5760, for example, 6000. Therefore, per packet production cost is Rs.  $6000 \div 24,000$  packets = 0.25.

Price per packet of biofertilizer is calculated as follows:

Production cost per packet	\$0.25
Power consumption (0.046 kW/packet)	\$0.05
Profit margin	\$0.2

So, the sale price per packet is = \$0.5

Return:

Net sale value for 2000 packets per month of biofertilizer using 80 L of broth in four batches (one batch per week) is \$1000 or \$12,000 per annum. By subtracting the \$6000 toward investment for the production, the net profit per year through this pilot-scale production of bacterial biofertilizer is \$6000 or \$500 per month.

## Alternatively:

If the production capacity is multiplied by four times using eight fermenters (each of 10 L capacity), the cost–benefit calculation is as follows:

Fixed cost:

The cost would be \$12,000 (including cost of six more fermenters) + \$200 for other equipments = \$14,000. Thus, from eight fermenters, the production of biofertilizer is 96,000 packets at 200 g each.

Variable cost:

If the cost of carrier material, broth, and polyethylene will be four times higher, then the total cost is \$1840.

Indirect cost:

The additional manpower required with a salary component as follows:

One microbiologist, \$240 per month	\$2880
One production assistant, \$160 per month	\$1920
One marketing and sales personnel, \$160 per month	\$1920
One laboratory attendant cum peon, \$80 per month	\$960
Miscellaneous expenses	\$2000
Total cost	\$7680

#### Return:

Net sale value of $24,000 \times 4 = 96,000$ packets of biofertilizer per year with a price of \$0.5 per packet	\$48,000
Investment for production: 10% depreciation against fixed cost (\$1400) + variable cost (\$1840) + indirect cost (\$7680)	\$10,920
Gross profit per year = \$48,000 - \$10,920	\$37,080

After deduction of the cost of advertisement in print and electronic media of about 20% from the gross profit, the net profit is = 37,080 - 7416 = 29,664 per year or, for example, 30,000 or 2500 per month.

In the above production system, unsterilized carrier (forest soil/charcoal) is used in dried form after grinding to avoid the natural population of microorganisms from competing with the inoculated bacterial broth. However, if a sterile carrier is used, the cost would be higher. From our experience, it was found that the brown forest soil available in most of the central and southern regions of Odisha state, India is free from microbial load, hence it is used as a carrier even in nonsterile conditions.

## 13.5.2 Quality control

The quality of the inoculants in the biofertilizer pack is one of the most important factors resulting in their success or failure and acceptance or rejection by the farmers. Basically, quality means the presence of the right type of microorganism in active form and in desired numbers. The stages requiring quality control are during the mother culture stage, during carrier selection, during broth culture stage, while mixing of broth with carrier, during packing, and during storage. Testing of the culture is usually done by taking a sample from the finished product for comparison with standard specification at the time of mixing of broth with carrier. In India, the Indian Stand Institution has developed standards for *Rhizobium* and *Azotobacter* (ISI 1986). The standard prescribed is that the inoculants

shall contain a minimum 10<sup>7</sup> cells per gram of carrier on dry mass basis within 15 days before the expiry date marked on the packet when the inoculants are stored at 25°C–30°C (Matsara and Bhattacharyya 1994; Subba Rao 1993). If the hazards imposed by harsh storage and transport conditions under tropical climatic regime are minimized, either by decentralization of manufacturing units or by rapid transportation to farmers at the time of showing/planting, the quality of biofertilizers reaching the end users can be improved to a greater extent.

# 13.6 *Cyanobacteria (BGA) biofertilizer production, quality control, and popularization for entrepreneurship development in rural areas of India*

In India, De (1939) first suggested that BGA plays an important role in maintaining the fertility of tropical rice fields. Subsequently, Singh (1961) confirmed the role of these organisms in nitrogen fixation and emphasized the use of these organisms as biofertilizer for nitrogen fixation as well as for waste land reclamation. Then, a number of studies were carried out at different agriculture universities and Indian Council of Agricultural Research institutes to evaluate the benefit of BGA inoculation into rice crops. Venkataraman (1972) and his associates at the Indian Agricultural Research Institute (IARI), New Delhi termed the process as algalization, and through multilocation field trials, convincingly showed its beneficial effects on grain yield and nitrogen savings. Studies have shown that BGA, besides contributing nitrogen to the tune of 25 to 30 kg/ha per season, increases the productivity of rice by up to 10% to 15% (Venkataraman 1978, 1981). Even in the presence of chemical fertilizers, this biofertilizer can give supplementary effects leading to a reduction in the dose of chemical nitrogen by 25 to 30 kg. There are several other benefits that have been found due to cyanobacterial inoculation in the field. These include the addition of organic matter to the soil, mobilizing bound phosphate, improving aeration and water logging capacity of soil, and increasing soil aggregation and reclamation of saline and sodic soils (Kaushik and Subhasini 1995; Roger and Kulasooriya 1980).

## 13.6.1 Present status of the technology: Its strengths and weaknesses

Based on the natural ecology of BGA in the rice field ecosystem, a cost-effective and easily adaptable biofertilizer technology was developed by Venkataraman at IARI, New Delhi in 1978. This basic method involves the mass production of a mixture of cyanobacteria consisting of five nitrogen-fixing species: *Anabaena, Nostoc, Plectonema, Tolypothrix,* and *Aulosira* in shallow trays or cemented tanks in open air using soil as the carrier material. These organisms have been screened as better organisms in terms of growth rate and nitrogen-fixing capacity from among the several species isolated from more than 2000 soil samples collected from different parts of India through the All-India Coordinated Project on Algae at IARI.

In mass culture, for each square meter area of the cultivation tanks or trays, 2 kg of soil and 100 g of single superphosphate are added. To this, 5 mL of malathion or 25 mg of furadan (Carbofuran 3%) granules is added to prevent breeding of mosquitoes and other insects. The contents were thoroughly mixed and allowed to settle down. A BGA culture of all five species is sprinkled on the surface. These organisms multiply in the tanks exposed to direct sunlight and form a thick mat over the surface in 10 to 12 days. The contents were allowed to dry and the dried flakes are collected, packed, and used for inoculation at the rate of 10 kg/ha in the rice fields after transplantation.

Later on, this technology was further elaborated. Depending on the requirement for biofertilizers, four different production methods have been developed such as tank method, pit method (making pits of variable size with polyethylene lining at the base), field method, and the nursery-cum-field production method (cultivated in leveled fields covering large areas). All these methods have been practiced extensively since 1978; however, these have many limitations. The main limitation being essentially climatic as the production is affected by low temperatures in winter and by washing out of the organisms in the rainy season. Thus, the results of the product are highly unpredictable in terms of amount of BGA biomass. Furthermore, the open air tank culture method is easily subjected to contamination by other organisms. Thus, it was felt that although algalization technology certainly holds promise and the economics are also favorable, large-scale production and harnessing maximum benefits from BGA biofertilizers is possible only through the resolution of these major constraints during production (Adhikary 2000).

## 13.6.2 Details of the appropriate technology

#### 13.6.2.1 Production in polyhouse

In this method, BGA multiplication is undertaken in roller-compacted concrete (RCC) tanks inside a polyhouse. The dimensions and number of such production units are variable depending on the turnover. The smallest recommended size of such a polyhouse is  $12 \times 5 \times 3.5$  m. Polyhouses are ideal because they provide controlled light intensity and temperature as desired and also prevent contamination from other organisms. Here, five different species of BGA are grown in equal amounts, that is, one species is grown in few adjacent tanks in a row within the polyhouse. Even one polyhouse can be exclusively used for growing one species to avoid cross-contamination of the species. A starter culture of each species is grown in 500 mL flasks containing 250 mL of BG11 medium in the laboratory for 8 to 10 days. Five of these flasks containing approximately 0.5 g of wet BGA biomass are used to inoculate 50 L of medium in a single cement tank. These were grown in the tank inside the polyhouse for about 7 days at optimum conditions. A fully grown culture containing 10 g of wet BGA biomass is collected in a container and then mixed well with 1 kg of carrier material. Then, this is dried at room temperature, preferably inside the same polyhouses, for 2 to 3 days. Different BGA species on dried carriers are then mixed in equal proportions and then packed in polyethylene packets for distribution/marketing.

#### 13.6.2.2 Use of alternate carrier materials

Carrier is an important component in the biofertilizer formulation. The shelf life of the biofertilizer depends on the quality of the carrier. A few characteristics of the carrier were considered before selecting: it should not be toxic to the BGA, it should have good surface area, it should have high absorption and good buffering capacity, it should be able to carry high microbial load, and it should be easily available and cost-effective. As an alternative to soil carrier, wheat and rice straw, and coconut coir cut into small pieces have been successfully developed for use as a carrier. Individual cyanobacteria are grown separately in an alternative cheap medium in shallow tanks and the fresh harvested mass is properly mixed with a definite quantity of carrier. The BGA load is maintained at 10<sup>6</sup> cells/g carrier so that only 400 g of the biofertilizer is sufficient to inoculate a 1-acre rice field. Straw as a carrier has also been used in a different way in which desired organisms were grown in pulverized and sterilized straw under controlled conditions in polyalkene bags. In these methods, different BGA species are grown separately and then mixed with carriers in desired proportions that ensure an appropriate share of all the organisms used in the inoculum. Unlike soil-based inoculum, this biofertilizer is supplied in ready-to-use quality material for inoculation. Goyal et al. (1994) have reported that "multanimetti," a montmorillonite clay, can be used as a suitable carrier for BGA biofertilizer. It fulfills most of the qualities of a good carrier. It is a powder form of carrier with small particle size, good buffering capacity, high absorptive nature, and is microbiologically inert. Using this method, cultures were grown separately under controlled conditions in polyhouses. The BGA, while in a growing state, are taken out from the tanks and mixed with clay in a ratio of 1:1 and dried. The dried material is finally powdered and packed. This biofertilizer in multanimetti carrier possesses a longer shelf life and can be used even after 2 years of storage under ambient conditions.

#### 13.6.2.3 Quality control parameters and testing methodologies

Biofertilizers contain living or latent microbial cells and their metabolic activity after application to the field determines their efficiency. The quality of biofertilizers is the most important factor leading to the success and acceptance or failure and rejection. The quality of BGA biofertilizer can be controlled at various stages of its production and ascertaining their load in the packing material. The quality of the inoculum is assessed before packing in terms of colony-forming units, which should not be less than 10<sup>3</sup> CFU/g of the final product (Singh and Pabbi 1998). The packing should be done in plastic bags properly sealed to avoid moisture. The BGA biofertilizers can be stored for 2 years at ambient temperature with no appreciable loss in viability or efficiency.

The quality standard of the BGA biofertilizer is assured by the following techniques:

- 1. *Viable cell count*: In this method, a known quantity of BGA samples were spread on agar medium in petri dishes and incubated at 25°C to 30°C for about 10 days and the colonies that appeared were counted.
- 2. *Most probable number method*: BGA biofertilizer samples are diluted to known grades in culture tubes and incubated under similar conditions in a culture room.
- 3. *Acetylene reduction assay*: This assay quantifies the nitrogenase activity of the BGA cells in the biofertilizer packs, in which the process is carried out only in living cells. The assay is carried out using gas chromatography (GC) measuring the amount of acetylene reduced to ethylene which is an indicator of nitrogenase activity.

# 13.6.3 Cyanobacteria biofertilizer production on a small scale; economics of the technology and popularization for entrepreneurship development

We have isolated several cyanobacteria species from different agroclimatic zones of Odisha State in the eastern region of India and have maintained them in a germplasm collection since 1996 (Adhikary 1998). All these cyanobacteria have been documented with their strain history, for example, place of collection, habitat, mode of occurrence, and others (Nayak et al. 1996; Sahu et al. 1996; Tirkey and Adhikary 2005). Basing on their capability to tolerate several stress factors e.g., salinity, pH, pesticides, commercial inorganic fertilizers, desiccation, etc., which have been experimented in the laboratory (Rath and Adhikary 1995, 1996; Padhi et al. 1997; Das and Adhikary 1996; Sahu and Adhikary 2006). Five cyanobacterial species belonging to the genera *Trichormus, Nostoc, Cylindrospermum, Aulosira*, and *Westiellopsis*, of which some are free-floating (*Trichormus variabilis* UU 147 and *Nostoc carneum* UU25130), and others epiphytic, growing attached to rice culms (*Aulosira fritschii* UU 25118) or attached to soil (*Cylindrospermum indicum* UU131 and *Westiellopsis prolific* UU2542) were screened. They were grown year-round in bulk in cemented tanks in polyhouses (Figures 13.2 and 13.3). The performance of brown forest soil as a carrier-based inoculum containing region-specific strains (Figure 13.4) at



*Figure 13.2* Photograph of a semipolyhouse showing four cemented tanks covered with UV-free polyethylene where BGA was grown outdoors.



Figure 13.3 Close view of growth of BGA in cemented tanks in the polyhouse.



*Figure 13.4* One-kilogram BGA biofertilizer packs prepared with brown forest soil as carrier used for application in rice crops.

an application dose of 12 kg/ha was tested in the field for 3 consecutive years. The results showed an approximately 8% to 12% increase in the grain yield (Adhikary and Sahu 2000). Improvement in soil quality due to BGA inoculation was determined based on an increase in the porosity, water-holding capacity, build-up of organic carbon, and microbial population of soil. Thus, the previous methods of producing BGA biofertilizer in pits, trays, cemented tanks, or by field-cum-nursery methods were upgraded emphasizing quality biofertilizer production. Authentic region-specific strains were used for year-round production of BGA in outdoor polyhouses.

For popularization of the technology and its transfer from the laboratory to the field, a pilot-scale demonstration, production-cum-quality control of BGA biofertilizer has been set up in the village of Maniakati under Asurabandha Grampanchayat of Surada block in the Ganjam District (Odisha) since April 2000. Several training programs have been conducted for farmers, agricultural officials, and entrepreneurs including women at the project site in the village during the last 10 years. Already, a few cyanobacteria biofertilizer production units have been established by entrepreneurs in the rural areas of Odisha State in the eastern region of India with partial support through Rastriya Krshi Bikash Yojana of the State Government.

#### 13.6.3.1 Usefulness of the technology for providing self-employment

Five kilograms of soil-based biofertilizer or 1 kg of straw or coir-based biofertilizer, with the requisite number of viable cyanobacteria, is sufficient to use in 1 acre of land soon after transplantation of rice. The cost per kilogram of soil-based biofertilizer for selling in India is approximately \$0.5, which is easily affordable by small and marginal farmers. This technology has now been adapted as a cottage industry in rural areas of the eastern region of India, and unemployed youth and nongovernmental organizations produce this biofertilizer for income generation.

13.6.3.2	<i>Economics</i> (cost–benefit of the technology adapted as cottage industry)
Capital investm	ient:

RCC ground floor work (12 ft. × 4 ft. × 1 ft.) for 25 tanks with brick and cement	\$2000
Iron framework with UV-free polyethylene shading	\$1000
Light microscope for quality control	\$500
Total investment for infrastructure	\$3500

Recurring expenses (including cost of raw materials for 1 year):

Superphosphate, lime, insecticide, etc., for 25 tanks	\$500
Carrier material (soil/coir/straw)	\$500
Bags (polybag or jute bag) for packing of biofertilizer	\$500
Labor cost: \$5 per day for 100 days (approximately 8 days per month)	\$500
Publicity (in print, electronic media, display boards, etc.)	\$500
Total investment in recurring expenses per year	\$2500

#### Return:

Under normal conditions in the eastern region of India, 48 harvests are possible, yielding 12,000 kg of soil-based cyanobacteria biofertilizer in 25 tanks in a year

Selling price of biofertilizer of 1 kg pack	\$1
Cost of 12,000 kg biofertilizer	\$12,000
(-) Capital investment plus recurring expenses for production in a year	(-) \$6000
Net profit per year: \$12,000–\$6000	\$6000 or \$500 per month

The income would further increase by increasing the number of tanks for the production of biofertilizer. Hence, this is a viable technology for rice farming and is useful for the rural development sector.

## 13.7 Constraints in the commercialization of microbial biofertilizer technology

There are several constraints in biofertilizer production and its commercialization. These are physical, chemical, biological, technological, financial, and market-related issues.

- i. *Temperature*—soil temperature plays a pivotal role in the survival of microorganisms. The optimum temperature of soil is between 25°C and 30°C. The major constraint in the use of biofertilizer are the summer months as the soil temperature reaches 45°C to 50°C at 5 cm depth.
- ii. *Drought*—the number of microorganisms in soil declines drastically as the soil dries up. Microbial populations decrease rapidly when stored under dry conditions.
- iii. *Biological constraints*—the presence of numerous parasites and predators in soil/ fields pose problems in the establishment of biofertilizer. There is also the problem of competition with the native population. Some nematodes act as predators of most of these microbes in the soil.
- iv. *Technical constraints*—peat has been recognized as the most suitable and standard carrier for bacterial biofertilizers. Due to high cost, few manufacturers use it. Therefore, most production units use charcoal, which is easily available locally. Mixing charcoal to brown forest soil in a 3:1 proportion as a carrier is a viable alternative due to lower costs.

## 13.7.1 Extension programs

Most farmers do not have sufficient and clear knowledge of the use of biofertilizers, which is necessary for deriving maximum benefit. The farmers need to be educated through effective demonstrations on the use of biofertilizers and not simply that biofertilizers are cheap fertilizers. Once the farmers are convinced, they will adapt to the use of biofertilizers, which would help in the commercialization of biofertilizers on a large scale.

#### 13.7.2 Marketing constraints

This includes weakness in the marketing network, retail outlet, storage facility, and so on. Because biofertilizers contain living organisms, they have short shelf lives (usually 6 months; or ~2 years for cyanobacteria). Hence, there is a necessity to maintain these inoculants under proper storage conditions at the retail outlet.

## 13.8 Conclusion

Biofertilizer technology has an immense market potential considering the vast area of land under cultivation globally. Once appropriate production technologies for quality biofertilizer are adapted, and the farmers are convinced about their efficiency, the demand for biofertilizers will increase. Although biofertilizers are produced and also available in the market either strict quality control measures are not taken in most production units or the farmers are not apprised of the proper process of their application, including their crop specificity. As a result, in many instances, they are found ineffective as claimed. Hence, the production technology of different biofertilizers has been refined and upgraded in recent years, and quality biofertilizers are produced using inoculum from authentic sources together with quality assurance at every stage of production, and also by selecting appropriate carrier materials.

It is a primary necessity to test the biofertilizers according to their efficacy in the farmer's fields. Dissemination of knowledge for the production and quality control of biofertilizers to the production houses/industrial sectors is also important. Second, a strong network—by establishing quality control laboratories, biofertilizer-producing units at the regional level, maintaining region-specific bacterial isolates, and generation of trained manpower—is essential. Once farmers can come forward to apply different biofertilizers, being convinced on their performance in the field with the concept of "seeing is believing," it would lead to large-scale commercialization of the technology.

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## chapter fourteen

## Bacterial biocontrol agents

#### Rikita Gupta and Jyoti Vakhlu

#### Contents

14.1	Introduction	. 317
14.2	Biological control	. 318
14.3	BCA versus agrochemicals	. 318
14.4	Interaction between host-pathogen and BCA	. 319
	Mechanisms of biocontrol	
	14.5.1 Direct antagonism	. 320
	14.5.1.1 Hyperparasitism	
	14.5.1.2 Microbial metabolites	. 321
	14.5.1.3 Detoxification and degradation of virulence factors	. 323
	14.5.1.4 Competition for nutrients and niche	. 323
	14.5.2 Indirect antagonism	. 324
	14.5.2.1 Induction of host resistance	. 324
14.6	BCA with multiple mechanisms for pathogen suppression	. 325
	Limitations of BCAs	
14.8	Conclusion	. 326
Refe	rences	. 326

## 14.1 Introduction

There is an increasing pressure on agriculture for meeting the ever-increasing food demand of the world population. The world population is expected to grow by 2.3 billion people between 2009 and 2050, further increasing the demand (http://www.fao.org). However, growth in agriculture is not corresponding to the increasing food demand. This is attributed to crop loss due to biotic as well as abiotic factors. The biotic factors contributing to crop loss include pests, pathogens, and weeds. Although annual pesticide use has increased to approximately three million tons worldwide, crop loss from pests is more than 40% of potential world food production. According to the World Health Organization (WHO), three million pesticide poisonings occurred annually causing 220,000 deaths in 1992 (http://www.wikipedia.org). Besides the environmental hazard, pesticides select for resistant pests, further augmenting the problem.

Sustainable alternatives should be explored to reduce worldwide pesticide use. One such alternative is to use natural enemies for pest control, that is, biological control.

In this chapter, we describe biological control and its advantages over agrochemicals. The basic interactions that can take place between any two species, be it plant–microbe or microbe–microbe, and the mode of action of bacterial biocontrol agents (BCA) is described.

## 14.2 Biological control

Use of organisms for controlling insect pests, mites, weeds, and plant pathogens is referred to as biological control and abbreviated as biocontrol. These organisms are natural enemies of insect pests, mites, weeds, or plant pathogens. There are different descriptions for the term *biological control*. It has been broadly defined by the U.S. National Research Council as "the use of natural or modified organisms, genes, or gene products to reduce the effects of undesirable organisms and to favor desirable organisms such as crops, beneficial insects, and microorganisms" (Pal and Gardener 2006). According to Cook and Baker, "Biological control is the reduction of the amount of inoculum or disease-producing activity of a pathogen accomplished by or through one or more organisms other than man" (Cook and Baker 1983). This includes the use of hypovirulent or avirulent strains of pathogenic species such as nonpathogenic *Fusarium oxysporum* for biocontrol of pathogenic strains, microbial antagonists such as *Pseudomonas fluorescens* CHA0 and manipulation of host plant for effective defense, which may be done either by genetic engineering or by using a BCA that primes the plant defense system (Alabouvette et al. 2006). The term has also been classified on the basis of the biological field of its intended use (Pal and Gardener 2006). Biocontrol is concerned with two fields of biology, namely, entomology and plant pathology. In the field of entomology, biocontrol refers to the control of pest insects. Similarly, in plant pathology, it refers to the control of pathogen and weed populations. Therefore, the BCA, in terms of entomology, will include predatory insects, entomopathogenic nematodes, and microbes such as entomopathogenic bacteria, fungi, and viruses. While in case of plant pathology, microbial antagonists or host-specific pathogens will be the candidates. Microbial antagonists of plant pathogens include bacteria, fungi, and viruses. However, for the scope of this chapter, we will talk only about bacterial antagonists involved in the biocontrol of plant pathogens. Biocontrol may be achieved by introducing a known BCA or by using native microorganisms that are potentially suppressive to the pathogen.

## 14.3 BCA versus agrochemicals

Plant growers rely on chemical fertilizers and pesticides besides good agricultural practices for improving crop yield, but these agrochemicals come packed with a bundle of drawbacks. These agrochemicals are usually environmental hazards and have a broadspectrum action that can harm beneficial microflora of plants, for example, chemical fungicides act on essential fungal functions such as respiration, sterol biosynthesis, or cell division, and have risk of pathogens developing resistance and showing toxicity to plants/ crops (Chandrashekara et al. 2012).

BCAs, being natural enemies of the pathogen, offer the following advantages over chemical agents:

- Effective against specific plant diseases
- Not toxic to plants and some show plant growth promoting properties in addition to biocontrol
- The products of microbial origin are biodegradable and hence environmentally safe because no residues are left in the environment

BCAs are not as popular as chemical agents. They suffer from a major disadvantage of short shelf life and are, at the moment, less potent than chemical agents because many

factors need to be taken into consideration before using a BCA. An important factor to be taken into consideration is the amount of biocontrol bacteria that is needed to control a pathogen in the field. This is a problem because the amount of pathogen present in the environment cannot be established. In addition, the number of different pathotypes, that is, strain variants of the pathogen in the field are not known. Due to such variations in the environment, the success of BCA in the field does not exactly reflect the success in the laboratory (Erdogan and Benlioglu 2010). However, sustainable agriculture being the need of the hour, developments should be made to overcome the shortcomings of biocontrol.

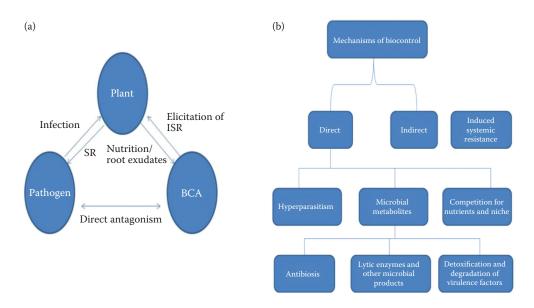
## 14.4 Interaction between host–pathogen and BCA

Association of bacteria with plants can result in three different types of interactions, namely, positive, negative, and neutral. Bacteria entering into a positive interaction, that is, beneficial bacteria enhance plant growth and protect it from infection, thus acting as plant growth-promoting bacteria or BCA. Such bacteria helps plants by nitrogen fixation as in case of *Rhizobium* associated with legumes, solubilization of nutrients, for example, iron solubilization and competition with pathogen by siderophore production, production of plant growth regulators, such as indole acetic acid by plant growth-promoting rhizobacteria, or by inducing a primed state of defense in the plant and production of antifungal antibiotics such 2,4-diacetylphloroglucinol (2,4-DAPG) by *Pseudomonas* species (Bhattacharyya and Jha 2012). These bacteria may be mutualists, that is, they also derive benefit from this positive interaction and hence utilize plant exudates such as organic acids, amino acids, and sugars as a source of nutrition, or they may be facultative mutualists as in the case of protocooperation (Pal and Gardener 2006).

Interaction of pathogens with host plant falls into the category of negative interactions because they derive nutrition for their growth and survival at the expense of host plant that is harmed by the interaction. These may be facultative because most saprophytic soilborne fungi can survive on dead and decaying organic matter besides the living host or obligate parasites such as viruses that require living plant host for their survival.

There is yet another type of bacteria that does not affect plants at all and lives commensally to the plant. Such microorganisms can be considered to have neutral interactions with plants. They may, however, be involved in positive or negative interactions with other microorganisms associated with the plant.

Biological control of a plant disease involves three major parties: BCA, pathogen, and host plant (Figure 14.1). The host is attacked by the pathogen, whose success relies on its virulence as well as the plant defense response. In addition to these two interacting parties, there are some bacteria that can interact with the plant and enhance its defense capability by inducing systemic resistance (ISR) or negatively interact with pathogen (antagonism) and reduce disease severity. Biocontrol is also affected by other microorganisms present in the plant's microenvironment. Some bacteria are regarded as commensals, because they neither harm nor benefit the plant. However, these bacteria may compete with BCA, reducing its efficiency (Berendsen et al. 2012). Nutritional composition of the plant's microenvironment may also affect biocontrol by either favoring or impeding the growth of BCA (Rudrappa et al. 2008). Therefore, biocontrol can be summed up as the net positive result of interactions between host plant and the associated microorganisms. The multiple plant–microbe as well as microbe–microbe interactions involved in the entire process can be simplified as the interaction between host plant, pathogen, and BCA.



*Figure 14.1* (a) Interactions between plant–pathogen and BCA and (b) mechanisms of action of BCA.

## 14.5 Mechanisms of biocontrol

Biocontrol conclusively is a result of plant-BCA and BCA-pathogen interactions. Several mechanisms have been put forward for the mode by which microbial BCA control plant pathogens (Figure 14.1; Table 14.1). Broadly, these mechanisms can be divided into two main categories, direct antagonism and indirect antagonism.

#### 14.5.1 Direct antagonism

Direct antagonism refers to a situation in which BCA directly interacts with the plant pathogen leading to its suppression. This includes biocontrol caused by hyperparasitism, production of antibiotics and lytic enzymes, and competition for nutrients and niche.

#### 14.5.1.1 Hyperparasitism

Parasitism is a negative interaction in which a parasite grows at the expense of its host. A plant pathogen is the primary parasite of its host plant and derives nutrition by harming its host. However, there are some bacteria (secondary parasites) that parasitize plant pathogens and hence the interaction is known as hyperparasitism. Hyperparasitism ranges in severity from simple attachment to complete lysis and death of the host. For example, attachment of *Enterobacter cloacae* cells to the fungal hyphae of *Pythium ultimum* and complete lysis and degradation of *Pythium debaryanum* hyphae by *Arthrobacter* (Whipps 2001).

Bacterial hyperparasites may be classified as opportunistic, facultative, or obligate parasites. Opportunistic hyperparasites have saprophytic capabilities and act on the fungal cell wall by producing lytic enzymes. Obligate hyperparasites are specific to the host and have more controlled mechanisms and cause less destruction. Facultative hyperparasites are intermediaries that have adapted to colonize host tissue besides their saprophytic capability. For example, *Pasteuria penetrans*, which is a hyperparasite of root-knot nematodes *Meloidogyne* spp. Because an obligate parasite has a high specificity toward its host, its commercialization

Biocontrol bacteria	Mode of action	Target pathogen	Reference
E. cloacae	Hyperparasitism	P. ultimum	Whipps 2001
Arthrobacter		P. debaryanum	Whipps 2001
P. penetrans		Meloidogyne spp.	Davies 2009
L. enzymogenes		Fungi, peronosporomyctes, nematodes, and bacteria	Islam 2011
<i>B. laterosporus</i> strain G4 and <i>Bacillus</i> sp. B16		H. glycines, T. colubriformis, and P. redivius	Tian et al. 2007
<i>Serratia</i> sp. G3	Lytic enzymes (chitinase and protease), antibiosis (pyrrolnitrin), competition for iron (siderophore)	B. cinerea, Cryphonectria parasitica, Rhizoctonia cerealis, and Valsa sordida	Liu et al. 2010
B. subtilis CPA-8	Antibiosis (lipopeptides)	<i>Monilinia</i> spp.	Yánez-Mendizábal et al. 2012
Pseudomonas CMR12a	Antibiosis (lipopeptides and phenazine)	R. solani	D'aes et al. 2011
<i>Pseudomonas</i> sp. LBUM300	Antibiosis (2,4-DAPG and HCN)	Clavibacter michiganensis	Lanteigne et al. 2012
Pseudomonas putida NH-50	Antibiosis (pyoluteorin)	C. falcatum	Hassan et al. 2011
P. aeruginosa	Lytic enzymes (chitinase)	S. sclerotiorum	Gupta et al. 2006
Bacillus sp. A24	Quorum quenching	Erwinia carotovora	Dong et al. 2002
P. fluorescens	Degradation of BS toxin	B. sorokiniana	Aggarwal et al. 2011
B. subtilis CAS15	Siderophore (bacillibactin) ISR	<i>F. oxysporum</i> Schl. f.sp. <i>capsici</i>	Yu et al. 2011
Pseudomonas sp.	Siderophore (pyoverdine and pyochelin) ISR	B. cinerea	Verhagen et al. 2010

Table 14.1 Examples of Biocontrol Bacteria and Their Modes of Action

is prevented by the inability to culture outside the host (Davies 2009). *Lysobacter enzymogenes* as facultative parasites have the ability to produce lytic enzymes as well as antibiotics besides hyperparasitism as a mode of action (Islam 2011). *Brevibacillus laterosporus* strain G4 and *Bacillus* sp. B16 are examples of opportunistic parasites of nematodes such as *Heterodera glycines, Trichostrongylus colubriformis* and *Panagrellus redivius* (Tian et al. 2007).

#### 14.5.1.2 Microbial metabolites

14.5.1.2.1 Antibiotics. Antibiotics are low-molecular weight organic compounds produced by microorganisms to kill and poison other microorganisms. Most of the microorganisms produce one or more antibiotics that enable them to survive and compete under severe environmental conditions. Bacteria produce a diverse range of different types and number of antibiotics (Raaijmakers and Mazzola 2012). Production of several antibiotics by a single BCA may account for the suppression of specific or multiple pathogens. For example, pyrrolnitrin is a broad-spectrum antibiotic produced by *Pseudomonas* and *Burkholderia* 

sp. that shows antibiotic activity against a wide range of Basidiomycetes, Deuteromycetes, and Ascomycetes, including several economically important pathogens like Rhizoctonia solani, Botrytis cinerea, Verticillium dahliae, and Sclerotinia sclerotiorum (Raaijmakers et al. 2002). Members of the genus Bacillus and Pseudomonas produce cyclic lipopeptides of three different families, namely, surfactin, iturin, and fengycins (Raaijmakers et al. 2010). Surfactins are heptapeptides interlinked with  $\beta$ -hydroxy fatty acid to form a cyclic lactone ring. These are biosurfactants that interfere with membrane integrity and display antibacterial and antiviral activity but lack antifungal activity. Iturins, on the other hand, display fungicidal activity by affecting membrane permeation. They form ion-conducting pores in the membrane and lead to osmotic imbalance. Fengycins are fungitoxic lipodecapeptides with less activity compared with the other two classes (Ongena and Jacques 2007). Another class of antibiotics produced by the genus Pseudomonas is phenazines, which are heterocyclic nitrogen-containing pigments. For example, phenazine-1-carboxylic acid, 2-hydroxyphenazine-1-carboxylic acid, and 2-hydroxyphenazine, which show antifungal activity. Phenazines have a broad-spectrum activity and are also produced by Streptomyces, Nocardia, Sorangium, Brevibacterium, and Burkholderia species besides Pseudomonas. Other antibiotics include phloroglucinols, pyoluteorin, polymyxin, circulin, colistin, and zwittermicin A (Compant et al. 2005; Raaijmakers et al. 2002).

14.5.1.2.2 Lytic enzymes and other microbial products. The fungal cell wall is composed of three different polymeric compounds, namely, chitin, proteins, and glucans. The lytic enzymes such as chitinase, proteases, and glucanases act to hydrolyze polymeric compounds of the pathogen such as chitin, proteins, and glucan in the fungal cell wall. These lytic enzymes help in predation or parasitism exhibited by various BCA against fungal plant pathogens.

Chitin is a homopolymer of *N*-aceylglucosamine with residues having  $\beta$ (1-4) linkage and accounts for 10% to 20% of the fungal cell wall (dry weight) and is also found in the exoskeleton of insects. Its microfibrillar structure is formed by interchain hydrogen bonding, which provides tensile strength and integrity to the cell wall. Chitinases belong to the glycosyl hydrolases family 18 or 19 and hydrolyse chitin component of fungal cell wall. Many bacteria with potential biocontrol activity produce chitinases. For example, *Pseudomonas aeruginosa* showing activity against *S. sclerotiorum* and fluorescent *Pseudomonad* strain active against *F. oxysporum* f.sp. *dianthi* and *Alternaria solani* produce chitinases (Gupta et al. 2006; Singh and Shanmugam 2012).

Glucans form the major structural polysaccharide of the cell wall constituting 50% to 60% of its dry weight, itself being composed of glucose monomers linked by  $\beta(1-3)$  linkage, although  $\beta(1-4)$ ,  $\beta(1-6)$ ,  $\alpha(1-3)$ , and  $\alpha(1-4)$  are also found. Glucanases are hydrolytic enzymes with lytic activity against glucans. For example,  $\beta(1-3)$  glucanase produced by *Bacillus sub-tilis* and *Bacillus pumilus* (Essghaier et al. 2009).

All fungal cell walls contain a protein component tightly interwoven within the polysaccharide components, that is, chitin and glucan, accounting for 30% to 40% of the cell wall's dry weight. Traditional cell wall proteins are glycoproteins such as mannoproteins, proteins glycosylated with mannans as in the cell wall of *Saccharomyces cerevisiae* and *Candida albicans*, and proteins glycosylated with galactomannans as in the case of *Aspergillus fumigatus* (Bowman and Free 2006). Thus, extracellular proteolytic enzymes play a role in antagonism by BCA. For example, *Stenotrophomonas maltophilia* inhibits the growth of *P. ultimum* by extracellular proteolytic activity (Dunne et al. 1997).

Besides the hydrolytic enzymes described previously, bacteria also produce enzymes or metabolites that suppress pathogen. Quorum quenching is one such strategy opted by some bacteria, for example, *Bacillus* sp. A24 produces lactonase enzyme, which is capable of

degrading *N*-acyl homoserine lactones (AHLs). AHLs are signaling compounds produced by bacteria for quorum sensing that coordinate gene expression in a density-dependent manner. *Variovorax paradoxus* neutralizes AHLs by the production of an aminoacylase that cleaves the fatty acid tail and mineralizes the homoserine lactone ring (Molina et al. 2003; Dong et al. 2002; Leadbetter and Greenberg 2000). Products of BCA are capable of degrading or neutralizing AHLs to suppress plant pathogenic bacteria by blocking their regulation of gene expression and hence affecting their virulence.

#### 14.5.1.3 Detoxification and degradation of virulence factors

Virulence factors produced by pathogens are important for causing infection. Some bacteria act as BCA by detoxifying these virulence factors. For example, detoxification of albicidin toxin produced by *Xanthomonas albilineans*. *Klebsiella oxytoca* and *Alcaligenes denitrificans* produce a protein that reversibly binds to albicidin toxin, whereas *Pantoea dispersa* detoxifies it with the help of an esterase. *Burkholderia cepacia* and *Ralstonia solanacearum* can detoxify fusaric acid, a toxin produced by *F. oxysporum*, which is important for disease development (Walker et al. 1988; Basnayake and Birch 1995; Zhang and Birch 1997; Toyoda and Utsumi 1991; Toyoda et al. 1988). Bacterial strains belonging to the genera *Bacillus, Paenibacillus, Microbacterium, Staphylococcus*, and *Pseudomonas campestris* involved in the regulation of virulence (Newman et al. 2008). BS toxin is produced by *Bipolaris sorokiniana*, which causes spot blotch in wheat and barley. *P. fluorescens* reduces the severity of symptoms induced by BS toxin on wheat tissues (Aggarwal et al. 2011). *P. fluorescens* strains have also been shown to detoxify phytotoxin produced by sugarcane red rot pathogen *Colletotrichum falcatum* (Malathi et al. 2002).

#### 14.5.1.4 Competition for nutrients and niche

Because microorganisms in the environment are mostly exposed to nutrient-limiting conditions, they compete for whatsoever nutrients available. Plants secrete 40% of their photosynthates as root exudates, making the rhizosphere a nutrient-rich environment favorable for microorganisms compared with the nonrhizosphere (bulk) soil and thus microbial density is higher in the rhizosphere; this is known as the rhizosphere effect (Berendsen et al. 2012). This gives rise to the competition between different species for available nutrients and space. Some bacteria act by competing with the pathogen for its infection sites and nutrients. A BCA with a better ability to colonize plant roots and utilize root exudates will have a competitive advantage over the pathogen.

Iron is an essential element for growth in all microorganisms, being a part of the enzymes involved in electron transport. The availability of iron in soil depends greatly on the pH of the soil, which has low solubility at neutral pH. In a well-aerated soil, iron is oxidized to its ferric form and is insoluble in water (pH 7.4), and the available concentration (10<sup>-18</sup> M) is quite low as compared with that required (10<sup>-6</sup> M). Most microorganisms produce low-molecular weight ferric ion chelators called siderophores, which form a complex with ferric ions (Fe<sup>3+</sup>). Siderophores then bind to iron limitation-dependent receptors at the bacterial cell surface and subsequently release iron into the cytoplasm. Inside the cytoplasm, Fe<sup>3+</sup> is reduced to its active form, Fe<sup>2+</sup>. Bacteria producing greater amounts of high-affinity siderophore compared with the pathogen will have a competitive advantage under iron-limiting conditions. Such bacteria will have better ability to colonize the available root niche competing with the pathogen. For example, *Pseudomonas syringae* produces pyoverdin (peptide type), achromobactin (citrate type), and yersiniabactin siderophores, which enhance its colonization and biocontrol ability (Wensing et al. 2010). *B. subtilis* CAS15 is shown to control *Fusarium* wilt of pepper. It produces a catecholic siderophore called bacillibactin (Yu et al. 2011).

#### 14.5.2 Indirect antagonism

#### 14.5.2.1 Induction of host resistance

When a pathogen attacks its host plant, the plant responds by activating a series of reactions called hypersensitive response (HR) as innate immunity in case of animals. A successful HR is followed by systemic resistance called systemic acquired resistance (SAR) toward a broad spectrum of pathogens. HR response is triggered when a plant recognizes the pathogen, which involves an interaction between the pathogen's virulence gene product and the plant's resistance (R) gene product. This interaction determines the plant's susceptibility toward the pathogen. R genes in plants are highly polymorphic, leading to several different products that help the plant in recognizing the pathogen, a cascade reactions starts. This includes an efflux of hydroxide and potassium ions and an influx of calcium and hydrogen ions. Reactive oxygen species (ROS) are generated, which lead to oxidative burst of the cells involved in HR. Oxidative burst causes the death of these cells and the formation of local lesions. Further deposition of lignin and callose around these lesions forms a barrier between the infected and unaffected tissues. However, a virulent pathogen evades the plant defense response and successfully infects the host.

Thus, SAR is developed in response to pathogen attacks and is characterized by the development of necrotic lesions at the site of the attack, accumulation of salicylic acid (SA), increased the expression of pathogenesis-related (PR) genes, and a slowly acquired resistance against a broad spectrum of pathogens. ISR differs from SAR as it is induced by non-pathogenic bacteria. It does not lead to a necrotic lesion and is mediated by Jasmonic acid (JA) or ethylene (ET). However, some bacteria have been shown to induce resistance mediated by SA (De Meyer et al. 1999). Some bacteria activate both pathways, that is, SA and JA dependent. For example, *Bacillus cereus* AR156 simultaneously activates both SA and JA/ET-mediated pathways in *Arabidopsis thaliana* (Niu et al. 2011). The two pathways are antagonistic such that SA leads to the expression of PR genes whereas JA primes the plant for enhanced defense gene expression upon challenge by a broad spectrum of pathogens.

Many bacterial compounds have been illustrated as elicitors of ISR in plants (Table 14.2). For example, in the elicitation of ISR in *Arabidopsis* by *P. fluorescens*, 2,4-DAPG has been found to be a major determinant because 2,4-DAPG minus mutants showed significantly reduced activity (Weller et al. 2012). SA, pyochelin, and pyoverdine (siderophore) have been shown to be important in the elicitation of ISR by *Pseudomonas* sp. against *B. cinerea* in grapevines (Verhagen et al. 2010). Thus, components of direct antagonism may provide additional advantage to biocontrol by elicitation of plant resistance.

Elicitor	Bacterial strain	Plant (host)	Reference	
Peroxidase, chitinase and β-1,3-glucanase	Bacillus mycoides strain Bac J B. pumilus 203-6	Sugar beet	Bargabus et al. 2002, 2004	
2,3-Butanediol	B. subtilis GB03 and IN937a	Arabidopsis	Ryu et al. 2004	
Antibiotics (DAPG)	P. fluorescens CHA0	Arabidopsis	Weller et al. 2012	
Lipopolysaccharide	Pseudomonas putida	Arabidopsis	Meziane et al. 2005	
Z,3-hexenal	Pseudomonas putida BTP1	Bean	Ongena et al. 2004	
Siderophore	Serratia marcescens 90-166	Cucumber	Press et al. 2001	
SA, pyochelin and pyoverdine	Pseudomonas sp.	Grapevine	Verhagen et al. 2010	
Acetoin	B. subtilis FB17	A. thaliana	Rudrappa et al. 2010	

Table 14.2 Some Bacterial Elicitors Involved in Induction of Plant Resistan	<i>Table</i> 14.2	Some Bacterial	Elicitors	Involved in	Induction	of Plant Resistant
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## 14.6 BCA with multiple mechanisms for pathogen suppression

Most biocontrol bacteria combine two or more mechanisms, providing efficient suppression of the plant pathogen. For example, *Paenibacillus ehimensis* IB-X-b produces two lytic enzymes, namely, chitinase and  $\beta$ -1,3-glucanase, which show synergistic antifungal activity against several plant pathogenic fungi (Aktuganov et al. 2008). *Lysobacter antibioticus* HS124 shows antifungal activity against *Phytophthora capsici*, which can be attributed to both the production of lytic enzymes such as chitinase, glucanase, and protease as well as 4-hydroxy phenylacetic acid (Ko et al. 2009). Biocontrol bacteria *Pseudomonas protegens* produces two antibiotics, namely, 2,4-DAPG and pyoluteorin (Ramettea et al. 2011).

## 14.7 Limitations of BCAs

Limitations of BCAs include the following:

- The main limitation of microbial products for use in biocontrol is the lack of consistency. Narrow specificity of the biological control agent limits its efficacy where a diversity of pathotypes is present in the natural population of a pathogen (Schisler et al. 2000).
- To ensure the efficacy of BCA, it is necessary to study the effect of inoculum type, application rate, and time of application. Time and place of application is more important when the mode of action is direct such as antibiosis because these second-ary metabolites are produced in small quantities and are not transported a great distance. The size of inoculum must be comparable or larger to that of the pathogen when competition is the mode of action.
- Achieving the optimum number is much more difficult in a natural environment where the pathogen population density is not known.
- In addition to the diversity in pathogen population and variability in climatic conditions, the efficacy of BCA is also subject to its interaction with the natural microflora associated with the plant.

Their use is more common in the United States, Australia, and New Zealand in comparison with other countries (Table 14.3).

Microorganisms	Product	Plant, pathogens, or pathosystems	Company
Ageobacterium radiobacter 84	Galtrol	Ornamentals, fruits, nuts	AgBioChem, US
B. subtilis FZB24	FZB24 li, TB, WG, RhizoPlus	Potatoes, vegetables, ornamentals, strawberries, bulbs, turf, and woods	ABiTEP, Germany
Pseudomonas chlororaphis	Cedomon	Leaf stripe, net blotch, <i>Fusarium</i> sp., sot blotch, leaf spot, etc., on barley and oats	BioAgri AB, Sweden
Serratia plymuthica HRO-C48	RhizoStar	Strawberries, oilseed rape	Prophytaiologischer Pflanzenschutz, Germany
Streptomyces griseoviridis K61	Mycostop	Phomopsis spp., Botrytis spp., Pythium spp., Phythophora spp.	Kemira Agro Oy, Russia

Table 14.3	Commercial	Biocontrol	Products
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## 14.8 Conclusion

The role of microbes, bacteria in particular, has been proven beyond a doubt for suppressing various plant diseases and there are many success stories from the laboratory to the market. In the present-day scenario, BCA are not capable of completely replacing agrochemicals. However, some studies have demonstrated a reduction in the percentage of chemical fungicide use, when used in conjunction with BCA. For example, a study on integrated control of fruit rot and powdery mildew of chili has shown a 50% reduction in use of the fungicide Azoxystrobin when used in combination with BCA *P. fluorescens* (Anand et al. 2010). Several areas of research need to be explored for further development in biological control. These include:

- 1. Study of distribution of pathogens including pathotypes and their antagonists in the environment
- 2. Effect of BCA on the natural microflora of plants and vice versa
- 3. Management practices that can aid biocontrol
- 4. Dynamics of plant defense induction, and
- 5. Effective combinations of BCA to control multiple pathogens/pathotypes

Thus, biological control agents have the potential to substitute hazardous agrochemicals. However, much research is required to this end to answer several questions that arise about the efficiency of biological control.

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## chapter fifteen

# Microbial biodegradation of polycyclic aromatic hydrocarbons

#### Pankaj Kumar Jain

#### Contents

Introduction	
Toxicity of PAHs	
Technologies for removal of PAHs	
15.3.2.1 Types of bioremediation	
Molecular techniques used for identification of PAH degrading	
microorganisms	
Conclusion	
rences	
	<ul> <li>Toxicity of PAHs</li> <li>Technologies for removal of PAHs</li> <li>15.3.1 Physicochemical treatments for removal of PAHs</li> <li>15.3.2 Biological treatments for removal of PAHs</li> <li>15.3.2.1 Types of bioremediation</li> <li>Molecular techniques used for identification of PAH degrading microorganisms</li> <li>Role of plasmid in degradation of PAHs</li> <li>15.5.1 Bacterial metabolism of naphthalene and phenanthrene</li> <li>15.5.2 phn genes for degradation of phenanthrene</li> </ul>

## 15.1 Introduction

Petroleum is a heterogeneous mixture of hydrocarbons, including aliphatic (*n*-alkanes), alicyclic, and aromatic hydrocarbons. It varies in compositional and physical properties according to the reservoir's origin (Van Hamme et al. 2003). The two major groups of aromatic hydrocarbons are the monocyclic aromatic hydrocarbons (MAHs) such as benzene, toluene, ethylbenzene and xylene (BTEX), and the polycyclic aromatic hydrocarbons (PAHs) such as naphthalene, anthracene, phenanthrene, and benzo(*a*)pyrene. PAHs are the most widespread contaminants in the environment. PAHs are a class of organic compounds that consist of two or more fused benzene rings that are arranged in various structural configurations. These hydrocarbons are organic compounds containing carbon and hydrogen, which are highly insoluble in water.

PAHs originate from two main sources: (i) natural (biogenic and geochemical), and (ii) anthropogenic (Mueller et al. 1996). Natural sources of PAHs are volcanic eruptions and forest fires (Blumer 1976). However, the major causes of accumulation in nature are anthropogenic activities such as the combustion of organic materials, for example, fossil fuels, coal, diesel, wood, and vegetation. These are also released in the environment by waste disposal, accidental spills during transportation and storage, and as pesticides, but are also formed during incomplete combustion such as with coal (Freeman and Cattell 1990; Lim et al. 1999).

Interest in the biodegradation of polycyclic aromatic hydrocarbons and compounds (PAHs/PACs) is motivated by their ubiquitous distribution, low bioavailability, high persistence in soils, and potentially deleterious effects to human health.

## 15.2 Toxicity of PAHs

PAHs have a major ecological effect on contaminated marine and terrestrial ecosystems (Santos et al. 2011). Many PAHs are recognized as having acute carcinogenic, mutagenic, and teratogenic properties. Benzo(*a*)pyrene is recognized as a priority pollutant by the U.S. Environmental Protection Agency (U.S. EPA; Renner 1999). This compound is identified as the most potently carcinogenic of all known PAHs and is used as an environmental indicator for PAHs (Table 15.1; Juhasz and Naidu 2000).

PAH concurrent exposure is common. After several experiments, it was proven that exposure to mixtures of PAHs in coke production, roofing, oil refining, and coal gasification are carcinogenic to humans (Integrated Risk Information System [IRIS] 1994; Agency for Toxic Substances and Disease Registry [ATSDR] 1995). Exposures to environmental mixtures, coal combustion effluent, vehicle exhaust, used motor lubricating oil, and tobacco smoke are also thought to be carcinogenic due to the PAH component of the mixtures (IRIS 1994). The effects on health depend on exposure time and the mixture involved. Some studies have suggested that exposure to some of the weakly carcinogenic or noncarcinogenic PAHs such as benzo(e)pyrene, benzoic(g,h,i)perylene, fluoranthene, pyrene with

PAHs	U.S. EPA	IARC	DHHS
Acenaphthene			
Acenaphthylene	Not classifiable		
Anthanthrene	Not classifiable	Not classifiable	
Benz(a)anthracene	Probably carcinogen	Probably carcinogen	Animal carcinogen
Benzo(a)pyrene	Probably carcinogen	Probably carcinogen	Animal carcinogen
Benzo(b)fluoranthene	Probably carcinogen	Probably carcinogen	Animal carcinogen
Benzo(e)pyrene		Not classifiable	
Benzo(g,h,i)perylene	Not classifiable	Not classifiable	
Benzo(j)fluoranthene	Not included	Possibly carcinogen	Animal carcinogen
Benzo(k)fluoranthene	Possibly carcinogen	Possibly carcinogen	
Chrysene	Probably carcinogen	Not classifiable	
Dibenz( <i>a</i> , <i>h</i> )anthracene	Probably carcinogen		Animal carcinogen
Fluoranthene	Not classifiable	Not classifiable	
Fluorene	Not classifiable	Not classifiable	
Ideno(1,2,3-cd)pyrene	Probably carcinogen		Animal carcinogen
Phenanthrene	Not classifiable	Possibly carcinogen	-
Pyrene	Not classifiable	Not classifiable	

Table 15.1 Classification of PAHs by the U.S. EPA, IARC, and DHHS on the	he
Basis of Carcinogenicity	

Source: B.-K. Lee and V.T. Vu. Sources, distribution and toxicity of polycyclic aromatic hydrocarbons (PAHs) in particulate matter. In *Air Pollution*, edited by V. Villanyi, 99–122, 2010. Sciyo Janeza Trdine: Rijeka, Croatia.

*Note:* DHHS, Department of Health and Human Services; IARC, International Agency of Research on Cancer; PAHs, polycyclic aromatic hydrocarbons; U.S. EPA, U.S. Environmental Protection Agency.

benzo(*a*)pyrene, or other carcinogenic PAHs enhances the carcinogenic potential. The noncarcinogenic PAHs induce the initiation and promotion of tumor properties (ATSDR 1995). However, concurrent exposure to other noncarcinogenic PAHs, for example, benzo(*a*)fluoranthene, benzo(*k*)fluoranthene, and chrysene has also been reported to lower the potential carcinogenic effects of benzo(*a*)pyrene (ATSDR 1995). Concurrent exposure to particulate matter and chemicals other than PAHs may also enhance the carcinogenic potential of a mixture (ATSDR 1995). Table 15.2 represents the emission of PAHs in various cities of various countries.

PAHs are recognized as carcinogens but are not genotoxic unless activated by mammalian enzymes, cytochrome P450 monooxygenase, to reactive epoxides and quinones. This enzyme oxidizes the aromatic ring to form epoxide and diol-epoxide reactive intermediates. It has been reported that these intermediates may undergo one of at least four different mechanisms of oxidation and hydrolysis before the intermediates combine with or attack DNA to form covalent adducts with DNA. DNA adducts can lead to mutations of the DNA, resulting in tumors (Harvey 1996). For example, benzo(*a*)pyrene is converted into epoxide by cytochrome P450 monooxygenase, forming different epoxides through the addition of one atom of oxygen across a double bond. The epoxides are short-lived compounds and may rearrange spontaneously to phenols or undergo hydrolysis to dihydrodiols. The dihydrodiols may also act as a substrate for cytochrome P450 once again to form new dihydrodiol epoxides, for example, *trans-7*,8-dihydroxy-7,8-dihydrobenzo(*a*)pyrene-9,10-oxide, which unfortunately are poor substrates for further hydrolysis. These dihydrodiol epoxides may react with DNA and cause mutations and possibly cancer (IARC 1983).

PAHs are present in the environment through anthropogenic or natural activities and exist almost everywhere. These may present in gaseous or particulate phases in ambient air. Due to the persistent nature of PAHs, living beings are always exposed to them and acquire various types of diseases. So, elimination of PAHs from the environment is indispensable.

	*		0	
Country	Area	ΣPAHs	PAHs concentration (ng/m <sup>3</sup> )	References
North Chinese	Urban	10	870	Liu et al. 2008
Plain	Rural	10	710	
Flanders, Belgium	Rural	16	114	Ravindra et al. 2006
Seoul, Korea	Urban	16	89	Park et al. 2002
Chicago	Urban	16	13–1865	Li et al. 2008
New Delhi, India	Urban	12	1049–1344	Sharma et al. 2008
	Urban	12	672	Sharma et al. 2007
Chennai, India	Urban	11	326–791	Mohanraj et al. 2011
Agra, India	Urban	16	72.7	Lakhni 2012
Campo Grande, Brazil	Campus	14	8.94	Poppi and Silva 2005
Tai Chung,	Urban	21	220	Fang et al. 2004
Taiwan	Rural	21	831	Ū
	Industry	21	1650	
Brisbane, Australia	Urban	16	0.4–19.73	Lim et al. 2005

Table 15.2 Comparison of Total PAHs (ng/m<sup>-3</sup>) in Various Cities of Various Countries

Remediation method	Benefits	Challenges	Examples
Bioremediation	Cost-effective, ecological	Possibly a slow process, Possible high monitoring costs	Natural attenuation Biostimulation Bioaugmentation
Phytoremediation	Cost-effective, ecological	Possibly a slow process, Possible high monitoring costs	Phytoextraction Phytostabilization Phytodegradation Phytotransformation Phytovolatilization
Containment	Fast, cheap	Potential leakage, does not reduce contaminant volume or toxicity	Encapsulation Capping
Physical treatment	Comprehensive	Loss of soil characteristics, costs, contaminants have to be discarded or remediated	Heat treatment Soil washing In situ flushing Air sparging Soil vapor Extraction Solidification Vitrification
Chemical treatment	Cheap, easy	Loss of soil characteristics, Contaminants have to be discarded or remediated	Liming <i>In situ</i> oxidation Permeable reactive barriers

Table 15.3 Remediation Technologies and their Benefits and Challenges

Source: Penn, I. et al., Eco-Industrial Strategies and Environmental Justice: An Agenda for Healthy Communities: Intergrating Brownfields and Eco-Industrial Development. National Center for Eco-Industrial Development, Center for Economic Development, School of Policy, Planning and Development, University of Southern California, 2002.

## 15.3 Technologies for removal of PAHs

Removal of PAHs from the environment is indispensable due to the carcinogenicity and toxicity of these compounds. For the treatment of PAH-contaminated soil, physicochemical and biological remediation is used (Table 15.3).

#### 15.3.1 Physicochemical treatments for removal of PAHs

The physicochemical treatments include incineration, thermal desorption, coker, flushing, soil vapor extraction, solidification/stabilization, cement kiln, solvent extraction, pyrolysis-gasification, and landfilling (Figure 15.1).

Currently, physicochemical treatments are dominant over biological treatments; however, these methods lack some desired properties. Incineration is a very effective treatment method, but it is costly and after burning, the soil will lose most of its nutritional value and structure. Landfilling does remove the contaminants, but only relocates the problem (Lageman et al. 2005). Furthermore, in Europe, legislation requires a reduction of the number of landfills. In 2004, the number of landfills with untreated contaminated material in England and Wales was reduced from more than 200 to only 11 (EA 2006). In Finland, the number of landfills has decreased from 232 in 1999 to 80 in 2005 (FEI 2006). As a result, the

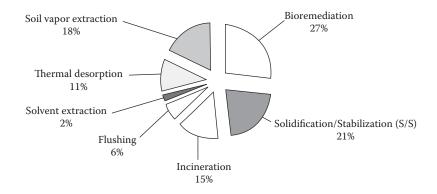


Figure 15.1 Methods used for remediation of PAHs.

Table 15.4 Physicochemical Remediation Technology of Petroleu	n Sludge Treatment
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Tachnology	Comments
Technology	Comments
Landfill	Very old but still one of the extensively used technologies, most landfills do not have energy production facilities
Incineration	High-temperature treatment, air pollution risks, expensive control equipment, high capital cost
Thermal desorption	High-temperature oil removal and recovery method from oily solids, high capital and material preparation costs, nonhazardous residues
Pyrolysis gasification	Pyrolysis is the thermal degradation of waste in the absence of air to produce gas, liquid, or solid, which generally takes place between 400°C and 1000°C. Gasification takes place at higher temperatures than pyrolysis (1000°C–1400°C) in a controlled amount of oxygen. Pyrolysis and in particular gasification is obviously very attractive to reduce and avoid corrosion and emissions by retaining alkali and heavy metals
Coker	Complicated sludge preparation for coker feed, some oil recovery, high capital and transportation costs
Cement kiln	Complicated sludge preparation for use of fuel, high material preparation, transportation, and disposal costs
Solvent extraction	Uses solvents and centrifugation or filtration for the separation of oil from sludges, safety hazard with solvent use, high capital cost

cost of dumping contaminated soils into landfills has increased considerably. It is therefore evident that new, innovative methods are needed to treat contaminated soils. Table 15.4 presents the physicochemical treatment technologies used for the removal of PAHs.

#### 15.3.2 Biological treatments for removal of PAHs

Before 1946, Zobell recognized that many microorganisms had the ability to utilize hydrocarbons as the sole source of carbon and energy. He further recognized that the microbial utilization of hydrocarbons were highly dependent on the chemical nature of the components in the petroleum mixture and environmental determinants (Atlas 1981).

Microbial biodegradation of pollutants has intensified in recent years as mankind strives to find sustainable ways to clean-up contaminated environments (Diaz 2008). Biodegradation of hydrocarbons by natural populations of microorganisms represent

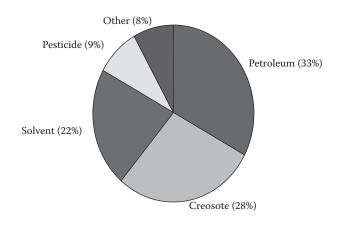


Figure 15.2 Representing bioremediation technology is applicable to all major types of wastage.

one of the primary mechanism by which petroleum and other hydrocarbon pollutants are eliminated from the environment. The effects of environmental parameters on microbial degradation of hydrocarbons, the elucidation of metabolic pathways, genetic basis for hydrocarbon dissimilation by microorganisms, and the effects of hydrocarbon contamination on microbial communities have been areas of intense interest and the subjects of several reviews (Atlas 1981, 1984).

Biodegradation is a process of metabolic ability of microorganisms to transform or mineralize organic contaminants into less harmful or nonhazardous substances. It is a natural process and is influenced by several factors such as nutrients, oxygen, pH value, composition, concentration, bioavailability, chemical, physical characteristics, and the pollution history of the contaminated environment.

Microbial remediation of a hydrocarbon-contaminated site is accomplished with the help of a diverse group of microorganisms, particularly the indigenous bacteria present in soil. These microorganisms can degrade a wide range of target constituents present in oily sludge (Eriksson et al. 1999; Barathi and Vasudevan 2001; Mishra et al. 2001; Figure 15.2). Therefore, we can apply this technology to various contaminants.

Hydrocarbon-degrading bacteria and fungi are widely distributed in marine, freshwater, and soil habitats. Similarly, hydrocarbon-degrading cyanobacteria have been reported, although contrasting reports indicated that the growth of mats built by cyanobacteria in the Saudi coast led to the preservation of oil residues (Barth 2003; Lliros et al. 2003; Chaillan et al. 2004). Typical bacterial groups already known for their capacity to degrade hydrocarbons include *Pseudomonas* spp. (Brito et al. 2006). Molds belonging to the genera *Aspergillus* spp., *Penicillium* spp., *Fusarium* spp., *Amorphoteca* spp., *Neosartorya* spp., *Paecilomyces* spp., *Talaromyces* spp., *Graphium* spp., and the yeasts *Candida* spp., *Yarrowia* sp., and *Pichia* spp. have been implicated in hydrocarbon degradation (Chaillan et al. 2004). Leahy and Colwell (1990) reported the biodegradation of petroleum oil. Biodegradation of oil by the fungi *Rhodotorula*, *Sporobolomyces*, *Aspergillus*, and *Penicillium* has been also studied (Head and Swannell 1999; Table 15.5).

#### 15.3.2.1 Types of bioremediation

Broadly, we can classify bioremediation into two types. Figure 15.3 represents various types of bioremediation methods.

	Ba	cteria	
Achromobacter	Enterobacter	Sinorhizobium	Sphingomonas
Acidovorax	Flavobacterium	Sphingomonas	Micrococcus
Acinetobacter	Herbaspirillum	Staphylococcus	Cellulomonas
Actinomyces	Methylobacter	Stenotrophomonas	Gordonia
Aeromonas	Microbacterium	Streptomyces	Dietzia
Alcaligenes	Methylococcus	Streptococcus	Sarcina
Aquaspirillum	Micrococcus	Spirilum	Serratia
Arthrobacter	Mycobacterium	Variovorax	Paenibacillus
Arthrobacter	Moraxella	Vibrio	Rhizobium
Bacillus	Nocardia	Xanthomonas	Rhodomonas
Boseaa	Erwinia	Yokenella	Rhodococcus
Brevibacterium	Norcadia	Marinobacter	Cytophaga
Burkholderia	Proteus	Alcanivorax	Corynebacterium
Capnocytophag	Providencia	Curtobacterium	Corynebacterium
Cellulomonas	Pseudomonas	Clavibacter	Chromobacterium
Paenibacillus	Microbulbifer		
	F	ungi	
Aspergillus	Neosartorya	Talaromyces	Sporobolomyces
Penicillium	Paecilomyces	Graphium	Amorphoteca
Fusarium	Candida	Yarrowia	Pichia
Rhodotorula			
	P	ants	
Agropyron smithii	Andropogon gerardi	Bouteloua curtipendula	Populus deltoides
Bouteloua gracilis	Buchloe dactyloides	Elymus canadensis	Sorghastrum nutans
Festuca rubra	Schizchyrium scoparious		-

 Table 15.5
 Genera of Hydrocarbon Degrading Bacteria, Fungi, and Plants which Grow in the Presence of PAHs, BTEX, or Alkanes

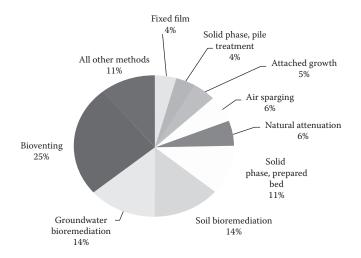


Figure 15.3 Representing various types of bioremediation methods.

- i. In situ bioremediation—in this method, excavation of soil is not needed. This usually leads to considerable savings in the costs of excavation and transport. According to the calculations of the Federal Remediation Technologies Roundtable (FRTR 2000), the costs of biological *in situ* treatments were approximately 8 to 80  $\in$ /m<sup>3</sup>. Table 15.6 shows types of *in situ* bioremediation.
- ii. *Ex situ bioremediation*—this type of bioremediation is generally used only when the site is threatened for some reason, usually by the spill that needs to be cleaned up. *Ex situ* bioremediation is only used when necessary because it is expensive and damaging to the area as the contaminated land is physically removed. Table 15.7 represents the various types of *ex situ* bioremediation.

Selection of the best treatment method depends on the type and quantity of the contaminants, treatment costs, the soil type, and the environmental conditions on the site,

	Table 15.6 Types of In situ Bioremediation
Intrinsic bioremediation (natural attenuation)	<ul> <li>Intrinsic bioremediation uses microorganisms already present in the environment to biodegrade harmful contaminants</li> <li>There is no human intervention involved in this type of bioremediation, and because it is the cheapest means of bioremediation available, it is the most commonly used</li> <li>When intrinsic bioremediation is not feasible, scientists turn next to accelerated bioremediation</li> </ul>
Land treatment	<ul> <li>It is a full-scale bioremediation technology in which contaminated soils, sediments, or sludges are periodically turned over (tilled) and allowed to interact with the soil and climate at the site</li> <li>The advantage of land treatment is that conventional farming equipment is usually the only equipment required</li> </ul>
Bioventing	• Contaminated soil is treated by drawing oxygen through it to stimulate microbe growth
Biosparging	<ul> <li>Biosparging involves the injection of air under pressure below the water table to increase groundwater oxygen concentrations and enhance the rate of biological degradation of contaminants by naturally occurring bacteria</li> <li>It increases the mixing in the saturated zone and thereby increases the contact between soil and groundwater</li> </ul>
Bioaugmentation	• The addition of a specific laboratory cultivated microbes or genetically modified organisms to the contaminated sites to increase the rate of biodegradion
Biostimulation	<ul> <li>Use of indigenous microbes to promote the growth of native microbes by providing nutrients</li> <li>Depends on necessary native microbial and organic material to be present</li> </ul>
Electroremediation	<ul> <li>Electrokinetic remediation is traditionally used to remove metals and polar organic compounds from soils, sludges, and sediments</li> <li>This method uses electrodes with a low-level direct current electric field (usually &lt;10 V/cm or mA/cm<sup>2</sup>) installed into the contaminated soil</li> <li>The current mobilizes and transports charged chemicals in the soils liquid phase toward the electrodes. Negatively charged anions and organic compounds will move to the anode, whereas positively charged chemicals such as metals will move toward the cathode</li> </ul>

Table 15.6 Types of In situ Bioremediation

	<i>Tuble 15.7</i> Types of <i>Ex situ</i> bioremediation
Landfarming	<ul> <li>Landfarming is a simple technique in which contaminated soil is excavated and spread over a prepared bed and periodically tilled until pollutants are degraded</li> <li>Adaptation of traditional farming techniques (aerating, ploughing) to</li> </ul>
	contaminated areas to increase microbes activity
Composting	<ul> <li>Composting is a technique that involves combining contaminated soil with nonhazardous organic compounds such as agricultural wastes</li> <li>The presence of these organic materials supports the development of a rich migrabil population and elevated termerature characteristics of compositing</li> </ul>
Biopiles	<ul><li>microbial population and elevated temperature characteristic of composting</li><li>This is a hybrid of landfarming and composting</li></ul>
Diopires	<ul> <li>Essentially, engineered cells are constructed as aerated composted piles</li> <li>Typically used for treatment of surface contamination with petroleum hydrocarbons, this is a refined version of landfarming that tends to control physical losses of the contaminants by leaching and volatilization</li> <li>Biopiles provide a favorable environment for indigenous aerobic and anaerobic microorganisms</li> </ul>
Bioreactors	<ul> <li>Slurry bioreactors or aqueous bioreactors are used for <i>ex situ</i> treatment of contaminated soil and water pumped up from a contaminated plume</li> <li>Bioremediation in reactors involves the processing of contaminated solid material (soil, sediment, sludge) or water through an engineered containment system</li> </ul>

Table 15.7 Types of Ex situ Bioremediation

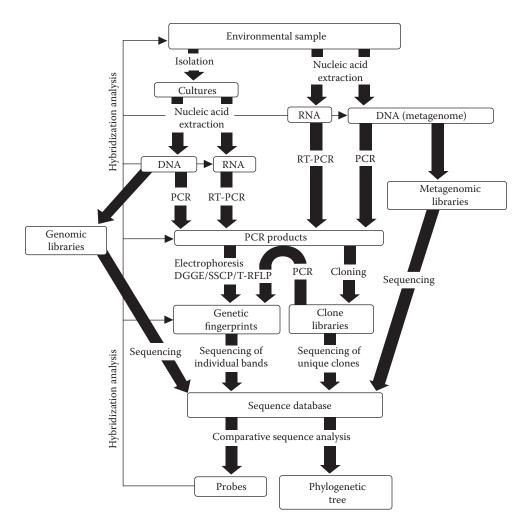
among other things (Khan et al. 2004; U.S. Department of Defense 1994). Before the selection of the best available method or methods, it is necessary to conduct a thorough investigation of the properties of the soil. This is especially true for *in situ* remediation because performance of the remediation process is more difficult to monitor and control during treatment than it is in traditional *ex situ* remediation processes (Jeltsch 1990; Morgan and Watkinson 1992).

# 15.4 Molecular techniques used for identification of PAH degrading microorganisms

Currently, molecular approaches are used to identify microorganisms in PAH contaminated soil and water (Figure 15.4). These techniques are more effective than our classic techniques (Table 15.8). Identifying the diversity of microorganisms that degrade PAHs/ PACs can be utilized in the development of bioremediation techniques.

## 15.5 Role of plasmid in degradation of PAHs

Plasmids are small, circular, or linear extrachromosomal parts of DNA. They contain approximately 2% of the genetic information of a cell and are separate from chromosomes. These can replicate independently from the main chromosome of bacteria. Plasmids are not essential to the life of a bacterium but they determine a cell's resistance to antibiotics (R factor or resistance factor) or degradation to any compounds. These have specific genes that are known as modules. Modules provide the plasmid resistance to a particular antibiotic and degradation of high-molecular weight compounds. The degradative plasmids are known as catabolic plasmids (Table 15.9). These are large ( $80 \rightarrow 1100$  kb) circular DNA and



*Figure 15.4* Molecular approaches for detection and identification of PAHs degrading bacteria and their catabolic genes. (From Muyzer, G., and Smalla, K., *Antonie Leeuwenhoek* 73: 127–141, 1998; Widada, J. et al., *Applied Microbiology and Biotechnology* 58: 202–209, 2002.)

Table 15.8 Advantages of Molecular Approaches over Classic Approaches
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Molecular approaches	Classic approaches
No need for culture microorganisms. Therefore, it can be applied to all types of microorganisms	A culture of microorganisms is essential. So, we cannot apply these techniques on unculturable microorganisms. These unculturable microorganisms may be involved in biodegradation
Molecular methods preserve the <i>in situ</i> metabolic status and microbial community composition because samples are frozen immediately after acquisition	Incubation in the laboratory produces artificial changes in the microbial community structure and metabolic activity
Direct extraction of nucleic acids from environmental samples can be used for a very large proportion of microorganisms	We can apply these techniques only for a small population of microorganisms due to handling problems

Bacterial strain	Plasmid	References
P. putida G7	NAH7	Yen and Serdar 1988; Menn et al. 1993; Simon et al. 1993
P. putida NCIB9816-4	pDTG1	Yen and Serdar 1988; Simon et al. 1993; Resnick et al. 1996
P. putida NCIB9816	pWW601	Yen and Serdar 1988; Yang et al. 1994
Pseudomonas sp. C18	Nah (G7)	Denome et al. 1993
Ralstonia sp. U2	Nag (U2)	Fuenmayor et al. 1998; Kiyohara et al. 1982
A. faecalis AFK2	Phn (AFK2)	Zhou et al. 2001
Pseudomonas fluorescens 5R	pKA1	King et al. 1990
Pseudomonas fluorescens 18H	pUTK21	King et al. 1990
Pseudomonas fluorescens HK44	pUTK21	King et al. 1990
Pseudomonas fluorescens 5RL	pUTK21	King et al. 1990
Sphingomonas aromaticivorans F199	pNL	Romine et al. 1999
<i>Staphylococcus</i> sp. PN/Y	pPNY	Mallick et al. 2007
<i>Terrabacter</i> sp. DBF63	pDBF1	Habe et al. 2005
Sphingomonas sp. HS362	p4	Hwa et al. 2005
<i>Beijerinkia</i> sp.	pKG2	Kiyohara et al. 1983
Pseudomonas fluorescens strain LP6a	pLP6a	Foght and Westlake 1996

*Table 15.9* Bacteria and their PAHs Degrading Plasmids

can be transferred by cell-to-cell contact to a wide range of bacteria. The role of plasmids in the evolution of bacterial populations in the environment cannot be overemphasized.

The large catabolic gene clusters carried by these plasmids play a central role in the Earth's carbon cycle. Many have a broad host range, enabling their hosts to degrade and recycle complex naturally occurring molecules such as lignin, the second most abundant plant biopolymer on Earth after cellulose. Furthermore, these catabolic gene clusters have evolved the ability to degrade and recycle high-molecular weight petroleum hydrocarbons, which are the most toxic and recalcitrant environmental pollutants, providing a biological solution to environmental pollution. These genes are used for the degradation of PAHs. Many of the genes involved in the degradation of PAHs are often located on plasmids (Johnsen et al. 2005). Thus, a bacterium needs the appropriate catabolic genes to degrade a compound (Table 15.10).

#### 15.5.1 Bacterial metabolism of naphthalene and phenanthrene

Naphthalene is the bicyclic aromatic hydrocarbon. It is abundant in soils. Several bacterial species are able to grow on naphthalene as the sole source of carbon and energy (Shamsuzzam and Barnsley 1974; Bosch et al. 1999; Kulakov et al. 2000). It has often been used as a model compound of PAH degradation. The bacterial degradation of naphthalene is well understood. The degradation of naphthalene in soil pseudomonads was first reported by Davies and Evans (1964).

The genes encoding the enzymes required for the catabolism of naphthalene are often carried on plasmids such as the NAH plasmids (Zakharian and Yen 1982). The naphthalene dioxygenase enzyme, encoded by NAH7 plasmid, is known today to be a highly versatile enzyme system, encoding a wide range of reactions (Ensley et al. 1983; Parales et al. 2002). The first evidence of the versatility of the NAH plasmid encoded genes was provided

Strain	Substrate	Gene	Function
P. putida	Naphthalene	nahAa	Reductase
	(upper pathway)	nahAb	Ferredoxin
		nahAc	Iron sulfur protein large subunit
		nahAd	Iron sulfur protein small subunit
		nahB	cis-Naphthalene dihydrodiol dehydrogenase
		nahC	1,2-Dihydroxynaphthalene oxygenase
		nahD	2-Hydroxychromene-2-carboxylate isomerase
		nahE	2-Hydroxybenzalpyruvate aldolase
		nahF	Salicyaldehyde dehydrogenase
	Salicylate	nahG	Salicylate hydroxylase
	(lower pathway)	nahT	Chloroplast-type ferredoxin
		nahH	Catechol oxygenase
		nahI	2-Hydroxymuconic semialdehyde dehydrogenase
		nahN	2-Hydroxymuconic-semialdehyde dehydrogenase
		nahL	2-Oxo-4-pentenoate hydratase
		nahO	4-Hydroxy-2-oxovalerate aldolase
		nahM	Acetaldehyde dehydrogenase
		nahK	4-Oxalocrotonate decarboxylase
		nahJ	2-Hydroxymuconate tautomerase
	Regulator for both operons	nahR	Induced by salicylate
Ralstonia	Naphthalene	nagAa	Ferredoxin reductase
sp. U2	-	nagAb	Ferredoxin
		nagAc	$\alpha$ Subunit of ISP (iron sulfur protein)
		nagAd	β Subunit of ISP
		nagB	cis-Dihydrodiol dehydrogenase
		nagF	Aldehyde dehydrogenase
A. faecalis	Phenanthrene	phnAa	Ferredoxin reductase
AFK2		phnAc	α Subunit of NDO (naphthalene dioxygenase)
		phnAd	β Subunit of NDO
		phnB	cis-Dihydrodiol dehydrogenase
		phnC	Dihydroxyphenanthrene dioxygenase
		phnD	Isomerase
		phnE	Hydratase-aldolase
		phnF	1-Hydroxy-2-naphtholdehyde dehydrogenase
		phnG	1-Hydroxy-2-naphthoate dioxygenase
		phnH	trans-2-Carboxybenzaldehyde dehydrogenase
		phnI	2-Carboxybenzaldehyde dehydrogenase
		gst	Glutathione-S-transferase

Table 15.10 Plasmid Encoded PAHs Degrading Gene Clusters of Bacteria

Source: Habe, H., and Omori T., Bioscience, Biotechnology, and Biochemistry 67: 225–243, 2003; Van Hamme, J.D. et al., Microbiology and Molecular Biology Reviews 67(4): 503–549, 2003.

independently by two research groups (Menn et al. 1993; Sanseverino et al. 1993). This led Sanseverino and coworkers (1993) to conclude that maintaining and monitoring one catabolic bacterial population may be sufficient for the degradation of a significant fraction of PAHs in contaminated soil. However, some organisms, such as *Pseudomonas stutzeri* AN10, carry these genes on their chromosome (Bosch 1999). Transcription of the genes in this operon is regulated by a salicylate-dependent transcription regulator such as NahR for the operon located on the NAH plasmid (Schell 1986; You and Gunsalus 1986).

*Pseudomonas putida* PpG7 have the capacity to degrade naphthalene because it has plasmid NAH7, which regulates the mineralization of naphthalene. Yen and Serdar (1988) reviewed the biochemistry and genetics of the naphthalene degradation pathway contained on plasmid NAH7. Dunn and Gunsalus (1973) established that the genes involved in naphthalene degradation are located on plasmids and were transferable by conjugation. The size of the *nah* gene is 25 kb and it is located on NAH7 (Grund and Gunsalus 1983; Schell 1983).

It has also been established that the same genes are responsible for partial degradation of phenanthrecene and anthracene. The metabolism of naphthalene has been well studied genetically in *P. putida* strain *G7. P. putida* PMD-1 can utilize naphthalene and salicylic acid as the sole carbon sources due to the presence of plasmid. It has been shown that the plasmid is required for the degradation of salicylate during the conversion of naphthalene (Zuniga et al. 1981).

Three operons are located on this plasmid. The first operon includes genes nahABCDEF. Naphthalene is converted into salicylate by encoding these genes. The second operon (sal) encodes the lower pathway's genes. Salicylate is converted into a TCA cycle intermediate through meta-ring cleavage (oxidation of salicylate via the catechol meta-cleavage pathway) by encoding these genes. This operon includes genes nahGHIJK. The third operon encodes a regulatory enzyme (NahR), which regulates both the upper and lower operons by a transacting positive control. It is located immediately upstream of the nahG gene NahR is needed for the high-level expression of the nah genes and their induction by salicylate (Yen and Gunsalus 1982; Grund et al. 1983; Yen and Gunsalus 1985; Schell 1985, 1986; Schell and Wender 1986).

Phenanthrene is a three condensed rings-fused PAH. These three rings fused in an angular fashion. It is often used as a model substance for microbial metabolism of "bay region" and "K region" containing carcinogenic PAH, such as benzo(*a*)pyrene, benzo(*a*)anthracene and chrysene (Samanta et al. 1999). Recent studies suggest that phenanthrene can be degraded by several Gram-negative and Gram-positive bacterial species (Evans et al. 1965; Kiyohara and Nagao 1978; Kiyohara et al. 1982; Barnsley 1983; Gibson and Subramanian 1984; Houghton and Shanley 1994; Samanta et al. 1999), for example, *Pseudomonas* (Yang et al. 1994; Balashova et al. 1999; Tian et al. 2002), *Mycobacterium* (Kelly and Cerniglia 1991; Churchill et al. 1991; Miller et al. 2004), and *Nocardioides* (Iwabuchi et al. 1998; Saito et al. 2000).

Kiyohara and coworkers (1983) reported that phenanthrene, biphenyl, and other PAH compounds can be degraded by *Beijerinkia* sp. This bacterium has two plasmids, PKG1 and PKG2, with 147 and 20.8 KDa, respectively. The genes for degradation of phenanthrene and biphenyl are localized on PKG2. Also, Okpokwasili et al. (1984) reported a phenanthrene utilizing *Flavobacterium*. In this bacterium, genes for degradation of phenanthrene are present on a 34 MDa plasmid. Guerin and Jones (1988) reported a phenanthrene degrading *Mycobacterium* strain BG1. In this bacterium, genes for degradation of phenanthrene are present on three plasmids 21, 58, and 77 KDa. Cho and Kim (2001) reported a 500-kb plasmid in *Sphingomonas* strain K514, and it is capable of mineralizing phenanthrene to carbon dioxide. This plasmid is much larger than the NAH plasmid. Hwa et al. (2005) reported the phenanthrene degrading bacterium *Sphingomonas* sp. HS362. It has five plasmids and degradative ability is localized on plasmid P4.

## 15.5.2 phn genes for degradation of phenanthrene

Phenanthrene biodegradation by bacteria has been documented, and various genes are involved in this. *Burkholderia* sp. strain RP007 degrades low-molecular weight PAHs like naphthalene and anthracene. The naphthalene and phenanthrene are degraded through a common upper pathway via salicylate and 1-hydroxy-2-naphthoic acid, respectively. phn operon of *Burkholderia* sp. strain RP007 located on an 11.5-kb *Hind*III fragment, which contained nine ORFs (Table 15.11; Iwabuchi et al. 1998; Samanta et al. 1999; Doddamani and Ninnekar 2000; Prabhu and Phale 2003).

It has been found that the *AaAb* genes, which encode iron sulfur protein  $\alpha$  and  $\beta$  subunits of the PAH initial dioxygenase (phn Ac and Ad) are not present on the *phn* locus. Two putative regulatory genes (phnR and phnS) are upstream of the phn catabolic genes. The phnS is a LysR-type transcriptional activator, and phnR is a positive transcriptional regulator. However, the detailed regulatory mechanism of the phn genes has not been described.

*Alcaligenes faecalis* strain AFK2 can utilize phenanthrene as the sole carbon source through the *o*-phthalate pathway. The phn genes are located in 13 ORF (Table 15.12). However, details have not been reported.

S. no.	Protein	Gene name
1	Regulatory protein	phnR
2	Regulatory protein	phnS
3	Aldehyde dehydrogenase	phnF
4	Hydratase aldolase	phnE
5	Extradiol dioxygenase	phnC
6	Isomerase	phnD
7	ISP a subunit of initial dioxygenase	phnAc
8	ISP b subunit of initial dioxygenase	phnAd
9	Dihydrodiol dehydrogenase	phnB

Table 15.11 Plasmid Encoded PAHs Degrading Gene Clusters of Bacteria

<i>Table</i> 15.12	Plasmid	Encoded	PAHsT	Degrading	Gene	Clusters	of Bacteria
11010 10.12	1 Iuomia	Littoucu	111110 L	Chinania	OUTIC	Clubicity	JI Ducteriu

S. no.	Protein	Gene name
1	Ferredoxin	phnAb
2	Ferredoxin reductase	phnAa
3	Cis-dihydrodiol dehydrogenase	phnB
4	A subunit of NDO	phnAc
5	B subunit of NDO	phnAd
6	Putative 2-hydroxychromene-2-carboxylate isomerase	phnD
7	Glutathione-S-transferase	Gst
8	Trans-2-carboxybenzalpyruvate hydratase-aldolase	phnH
9	1-Hydroxy-2-naphthoate dioxygenase	phnG
10	2-Carboxybenzaldehyde dehydrogenase	phnI
11	3,4-Dihydroxyphenanthrene dioxygenase	phnC
12	1-Hydroxy-2-naphthoal-dehyde dehydrogenase	phnF
13	Putative-trans-o-hydroxybenzylidenepyruvate hydratase-aldolase	phnE

## 15.6 Conclusion

Contamination of the environment by PAHs has become a significant environmental concern due to their low bioavailability, toxicity, mutagenicity, and carcinogenic properties. PAHs are persistent, mainly due to their high hydrophobicity. The fate of PAHs and other organic contaminants in the environment is related with both abiotic and biotic processes. Abiotic processes include volatilization, photooxidation, chemical oxidation and biotic process including bioaccumulation, and microbial transformation. Microbial action has been recognized as the most influential and significant cause of PAH removal. A variety of microorganisms such as bacteria and fungi can degrade PAHs. Thus, there is a major interest in studying the bacteria present in contaminated environments for bioremediation. Understanding the mechanisms of bacterial populations to adapt in the presence of pollutants, and the identification of key functional genes, will allow the exploitation of microbial PAH-degradative capabilities. This will augment the success of bioremediation strategies.

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Page numbers followed by f and t indicate figures and tables, respectively.

## A

Abiotic habitats, 46 Acetaldehyde, 281 Acetic acid, 170, 174t Acetogenic bacteria, 157 Acetogens, 158 Acetylene reduction assay method, 310 Achromopeptidase, 47 Aconitases, 119 Acremonium chrysogenum, 21 Actinobacteria, 13, 14, 16, 17 bioactive compounds of, 18t-19t Actinomycetes, 13, 16 Actinomycetin, 241 Activation energy, 126 Active compounds, formation of, 136 Adsorption, 175. See also Immobilization Aeration, 133, 177. See also Upstream processes Aerobic organisms, 115 Agitation, 177. See also Upstream processes Agrochemical inputs, 298 Agro-industrial waste products, 291 Air-drying, 119 Airlift fermenters, 290 Air separation, 139 Alcohol oxidase, 90 Algae, 23 production mechanisms, 160 structures, 161 Algal cells, 159 Algalization technology, 309 α-linolenic acid (ALA), 83 Alkaline extracellular protease (AEP), 73 Alkaloid, 143t Alpha-proteobacteria, 299 Alteromonas rubra, 19 Amikacin, 245f Amino acid, 116, 173, 187, 292 for monogastric animals, 199 7-Aminocephalosporanic acid (7-ACA), 247f Aminoglycosides, natural, 244f 5-Aminolevulinic acid, 144t 6-Aminopenicillinic acid (6-APA), 247f

Ammonium hydroxide, 69 Amylases, 71, 288 Anaerobic fermentation, 286 Anaerobic growth, 283 Anhydrobiosis defined, 113 induced. See Induced anhydrobiosis in microorganisms, 114-115 Anhydrobiosis quotient, 123 Anhydrobiotic cells of biocontrol agent (case study), 123-126, 124f, 125f Anoxygenic photoheterotrophic microorganisms, 161 Antibacterial activity of bacterial isolates, detection of, 226-228, 227f, 228t Antibiotic resistance, 10, 23, 226, 254 Antibiotics. See under Microbial healthcare products Antibiotic susceptibility, 219. See also Probiotics Antifreeze proteins, 72 Antifungal activity primary screening for, 26. See also Screening of microbial metabolites/bioactive compounds in vitro assay for, 26-28, 27f, 27t Antimicrobial activity, 217-219. See also Probiotics Antioxidant(s), 174 defense systems, 116 Anti-Parkinson agent, 265 Antitumor agents. See also Microbial healthcare products biotechnological approach, 258-259 historical background, 256-257, 256f, 257f overview, 254-256, 255f production/pharmaceutical presentation, 257-258 therapeutic role, 259 Antiviral drug, 16 Apigenin, 273f Arachidonic acid (AA), 83 Arrhenius kinetics, 126 Arrhenius plot, 125 Artificial miRNA (amiRNA) techniques, 164 Ascomycetesares, 142 Ascorbate, 170 Ascorbic acid, 170-171, 171f, 174t Aspergillus fumigatus, 275 Aspergillus gorakhpurensis, 22

Aspergillus niger, 171 Aspergillus ustus, 22 Attenuated vaccines, 266 Autoinducers, 222 Autonomously replicating sequence (ARS), 78, 88 Auxotrophic mutants, 196f, 201 Auxotrophies, 65 *Azorhizobium*, 299 *Azospirillum*, 299–300, 301, 302. *See also* Microorganisms species of, 300 *Azotobacter*, 299. *See also* Microorganisms pilot-scale production of, 303–307

## В

Bacillus thuringiensis, 20 Bacteria, bioactive compounds from, 12-21, 15t-16t, 18t-19t. See also Bioactive compounds from microorganisms Bacterial antibiotics, 21 Bacterial artificial chromosome (BAC), 49 Bacteriocin-like substances (BLIS), 21 Bacteriocins, 21, 217, 218 Baculovirus expression vector system, 271 Baicalein, 274f Baker's yeast. See also Microbial biomass production in genetics study, 287-288 model organism for somatic mutations/cancer study, 286-287 as natural therapy for cancer, 286 production of, 285f Baking, role of yeast in, 283. See also Yeast Balhimycin, 248f Basal salt medium (BSM), 69 Basidiomycota, 142 Batch fermentation, 290 Bead beating, 47 Beet, 283-284 Benzylic hydroxylation, 135, 136 Benzylpenicillin (penicillin G), 242f Beta-d glucose, 172 Bile, 216 Bile hydrolase, 226 Bile salt, 216 Bioactive compounds of actinobacteria, 18t-19t antagonistic activities of, 15t-16t from microorganisms, 10-24, 12t from bacteria, 12-21, 15t-16t, 18t-19t from cyanobacteria, 23, 24t secondary metabolites of fungi, 21-23 Bioactive metabolites, 45 Bioalcohols, 157–158, 158f. See also Microbial biofuels production Biocatalysts, metagenome-derived, 50, 50t Bioconversion of carbon substrate, 170 Biodiesel, 158-160, 159t. See also Microbial biofuels production feedstock, 156

Biodiversity of marine microbes, 12 Bioethanol from biomass, 158f Biofertilizer(s) carrier-based preparations, 301-302 defined, 298 drying of, 302f microbial. See Microbial biofertilizers packing of, 303f production, steps of, 304 Biofertilizer application methodology. See also Microbial biofertilizers root dipping, 301 seed treatment, 301 soil application, 301 Biofertilizer packets, preparation of, 302-303. See also Mass production of bacterial biofertilizer Bioflavonoids, 271 Biofuels, 180. See also Microbial biofuels production defined, 155 importance of, 155 Biohydrogen, 160-164, 162f. See also Microbial biofuels production Biological detoxification of herbs, 136 Biological water-gas shift reaction, 163 Biomass, 6, 7, 181 separation of, 179 Biophotolysis, direct/indirect, 163 Bioprocess parameters, 69 Bioprospecting, 45 Bioreactors of SSF. See also Solid-state fermentation (SSF) classification of SSF technology, 145f mixed beds with forced aeration, 146 unmixed beds with forced aeration, 145-146 unmixed beds without forced aeration, 145 Bioseparation processes, 179 Biosynthesis, 272 of ergot alkaloids, 260f of organic acids, 171f Biosynthetic markers, 67 Biotechnological approach antibiotics, 253 antitumor agents, 258-259 ergot alkaloids, 263-264 flavonoids, 272, 273f Biotechnology, defined, 1 Biotic habitats, 46 Biotin, 207f, 209 Biotin-deficient medium, 193 Biotransformation of herbs, 135-136 of organic molecules, 189 Blasticidin, 67 Blue-green algae (BGA), 300, 308 biofertilizer, quality standard of, 310 growth in cemented tanks, 311f prepared with brown forest soil, 311f Botryodiplodia species, 274, 275 Bound water, 145 B-proteobacteria, 299

Bradyrhizobium, 299 Brassica napus, 77 Broad-spectrum antibiotics, 242, 254 Brominated ergocrytine, 260 Bromocriptine mesylate (Parlodel), 263 Broths concentration of, 179 fermentation, 178 Burkholderia, 299 (E)-2-butenedioic acid, 171

# С

Caffeine (Cafergot), 263 Calcium pyruvate, 173 Candida antarctica lipase B (CalB), 82 Cane molasses, 283-284 Capital investment, 312 Carbapenems, natural, 247 Carbon catabolite repression, 86 Carbon-neutral biofuels, 155 Carbon sources, 144 Carboxylic acid, 172 Carotenoids, 73 production, 83-84 Carrier materials alternate, 309-310 ideal, 302 organic, 301 Cassava starch, 180, 182t Catabolic pathways, 280 Catabolite inactivation, 283 Catabolite repression, 281 Caulerpa peltata, 19 Cell disruption, 179 Cell division cycle (CDC), 287 Cell membranes, 124 Cellulases, 71 Cellulose, 157, 291 Cell wall destruction, 179 Central metabolism (CCM), 198 Central nervous system (CNS), 194 Cetrimonium bromide (CTAB), 47 Cetyltrimethylammonium bromide, 47 Chemical fertilizers, 298 Chemical synthesis, 201 Chemotherapy, 258 Chinese herbs multilevel utilization of, 138 overall multistage conversion/directional fractionation of, 146 preparation culture medium with, 144-145 solid-state fermentation of, categories, 133 SSF for concoction of, 135-136 SSF for extraction of, 136-137, 137t Chinese herbs, fermented. See also Solid-state fermentation (SSF) Hongqu, 134 Shenqu, 134

Chinese herbs, pretreatment process of. See also Solidstate fermentation (SSF) cutting process technology, 140, 141f drying process technology, 141 selecting and washing, 139-140, 140f Chinese herb solid-state fermentation (SSF) processing, 131. See also Solid-state fermentation (SSF) advantages of, 137-138, 138f based on steam explosion technology, 146-147, 147f history of, 133-135, 134t natural SSF of herbs, 135 principle of, 135, 137t, 138f process and technologies in, 139t process engineering of, 148-150, 148f, 149f prospects of, 146 SSF for concoction of Chinese herbs, 135-136 SSF for extraction of Chinese herbs, 136-137, 137t Chinese medicine compounds, 138 Chloramphenicol, 243f Chloroeremomycin, 249f Chloroform, 275 Chloroplast, 162 Chlortetracycline, 243f Cholesterol-lowering effects, 228-229, 229f. See also Probiotics Cholesterol reduction, 216-217. See also Probiotics Chromatography, 28, 269 Chrysin, 273f Citric acid, 171, 174t. See also Organic acids, microbial production of agitation on, effect of, 182t alcohol on, 182t biomass/residual sugars/pH determination, 181 determination, 181 fermentation period on, 183f by immobilized A. niger, comparative production of (case study), 180-183 materials and methods, 180-181 results and discussion, 181-183, 181t, 182t, 183f nitrogen supplements on, 181t production, 180-181 Claviceps paspali, 262 Claviceps purpurea, 259 Clavine alkaloids, 260f structures of, 261f Clones removal, 52 Cloning-based screening, 54 Cluster iron-sulfur (redox sensor), 119 Coenzyme Q10, 144t Cofactor sensor, 119 Cofeeding of sorbitol, 70 Coffee-pressing wastes, 292 Colon cancer, 225 Column DEGC flame ionization detector, 28 CoMed, vector system, 88 Commercialization of biofertilizer technology, constraints in biological constraints, 313 drought, 313

extension programs, 313 marketing constraints, 313-314 technical constraints, 313 temperature, 313 Conjugate vaccines, 266 Cordycepin, 136 Corynebacterium glutamicum electron micrograph of, 190f Corynebacterium glutamicum FRL no. 61989, cultivation of, 205-206 Coryneform species, 193 Cost of pilot-scale production of Rhizobium/ Azotobacter biofertilizer, 305-307 fixed cost, 305 indirect cost, 305, 306 variable cost, 305 Counter-selectable marker, 67 Covalent linkage, 175. See also Immobilization Cranberry pomace, 136 Cryoprotector, 92 Crystallization and drying, 179 Cutting process technology, 140, 141f. See also Chinese herbs, pretreatment process of Cyanobacteria, 117, 300. See also Microorganisms bioactive compounds from, 23, 24t. See also Bioactive compounds from microorganisms Cyanobacteria biofertilizer production. See also Microbial biofertilizers appropriate technology, details of alternate carrier materials, use of, 309-310 production in polyhouse, 309 quality control parameters and testing methodologies, 310 popularization of cost-benefit of technology, 312-313 for self-employment, 312 on small scale, 310-313 technology, status of, 308-309 Cyanobacterial species, 310 Cyanobacterium, 119 Cyanobactins, 50 Cyclic electron transport (CET), 162 Cyclomarin, 16

## D

Daptomycin, 249f Dark fermentation, 163 Data-mining system, 56 Deep-jet fermenters, 290 Dehydration, 115, 116 Dehydrogenase, 67 Deionized water, 206 Desiccated microorganisms, stability of, 122–123 Desiccation damage mechanisms, 115–117, 115f Desiccation tolerance defined, 114 mechanisms of, 117–119, 118f

Desiccation-tolerant cells, 116 Detergents, 47 2,4-diacetylphloroglucinol (Phl), 20 Diaminopimelic acid (DAP), 195, 195f Diaporthe helianthi, 22 Diarrhea, 223 Dicarboxylic acid, 172 Diets, 225 Dihydroergotamine, 263 Dihydroflavonol, 273f Dihydrokaempferol, 273f 1,2-dimethyl hydrazine (DMH), 225 Dimethyl sulfoxide (DMSO), 27 Diphtheria-tetanus-purtussis-H. influenzae type b-hepatitis B (DTP-Hib-HepB), 270 Direct DNA extraction, 47. See also Metagenomics Direct fermentation, 157 Disaccharides, 117, 120, 281 DmaW, 264 DNA, 240 DNA hydration, 116 DNA vaccines, 271 Docosahexaenoic acid (DHA), 83 Dough maturation, 283 Downstream processes bioseparation processes, 179 fermentation factors in, 178-179 in microbial biotechnology, 6-7, 7f primary aim of, 178 purification steps cell disruption, 179 concentration of broth, 179 crystallization and drying, 179 product-specific purification, 179 separation of biomass, 179 Doxycycline, 246f Dried herbs, 140 Drought on microbial populations, 313 Drought tolerance, 114 Drying of microorganisms, 121-122, 121t Drying process technology, 141. See also Chinese herbs, pretreatment process of DuPont, 73, 83

## Е

Ectendomycorrhiza, 300 Ectomycorrhiza, 300 Eicosapentaenoic acid (EPA), 73 Electron micrographs of *T. paurometabola* C-924 cells, 125f Electron transport, 162 Electroporation, 65 Endomycorrhiza, 300 Enterocins, 218 *Enterococcus durans*, 218 Entrapment, 175. *See also* Immobilization Enzyme activity, 281 Enzyme replacement therapies (ERT), 73

Ephedra herb SEM of, 147f untreated, 147f Ereky, Karl, 3 Ergonovine, 260 Ergopeptines, 260 Ergot alkaloids. See also Microbial healthcare products biosynthesis of, 260f biotechnological approach, 263-264 historical background, 262 overview, 259-261, 260f, 261f production/pharmaceutical presentation, 262-263 therapeutic role, 264-265 Ergotamine tartrate (Ergomar), 263 Eriodictyol, 273f Erlenmeyer flasks, 206 Essential amino acids, 187, 188t Ethanoic acid, 170 Ethanol, 66, 157, 282t, 286 Ethanol production, SSF for, 149 Ethyl acetate, 17, 27 Eukaryotic microorganisms, single-celled, 280 European Centre for Disease Prevention and Control (ECDE), 254 European Federation of Biotechnology, 3 Excitation energy, 161 Exopolysaccharides, 220 Extension programs, 313. See also Microbial biofertilizers

## F

FAME. See Fatty acid methyl esters (FAME) Fatty acid analysis of streptomyces parvulus DOSMB D105 (case study), 24-32 antifungal activity, primary screening for, 26 fatty acids using FAME method, analysis of, 28-32, 29f, 29t-31t, 32t isolation of Streptomyces spp., 26 in vitro assay for antifungal activity, 26-28, 27f, 27t Fatty acid methyl esters (FAME), 28-32 Fatty acids, 222 using FAME method, analysis of, 28-32, 29f, 29t-31t, 32t. See also Screening of microbial metabolites/bioactive compounds Fed-batch fermentation, 283, 284, 290 Fermentation, 89, 93 batch, 290 broths, 178 commercial, 284 industrial, 290 in microbial biotechnology, 5, 6f processes in microbial biotechnology, 5-6, 6f solid-state, 178 submerged, 177, 190, 252f, 257 surface, 177 Fermented tea, SEM of, 138f Fermenter large-scale, 288 structure of, 252f

Ferredoxin-NADP-reductase (FNR), 162 Ferredoxin-quinone-reductase (FQR), 162 Fe-superoxide dismutase (SodF), 119 Fibromyalgia, 172 Fibrous residues, 150 Filter press, 285 Filtration, 179 Fixed cost, 305 Flavanone, 273f Flavone, 272f, 273f Flavonoids biotechnological approach, 272, 273f classification of, 272f historical background, 271-272, 272f production/pharmaceutical presentation, 272 therapeutic role, 274, 274f Flavonol, 273f Fleming, Alexander, 241 Fluidized bed drying, 121, 121t, 122 Fluorescence-activated cell sorting (FACS), 48, 53 Fodder yeast, 73 Food-borne microorganisms, 217 Forced aeration. See also Solid-state fermentation (SSF) mixed beds with, 146 unmixed beds with, 145-146 unmixed beds without, 145 Formaldehyde, 67 Formaldehyde dehydrogenase (FLD), 66, 67, 92 Fossil fuels, 155 Fourier transform infrared spectroscopy, 116 Free acid, 171 Free sugars in dough, 288 Freeze-drying, 114, 121, 121t, 122 Fuel ethanol, 93 Fumaric acid, 171–172, 174t. See also Organic acids, microbial production of Function-based screening, for novel metabolites heterologous complementation of host strains/ mutants, 51-52 phenotypical detection of desired activity, 52 substrate-induced gene expression, 52-53 Fungal metabolites, 259 Fungi, sponge-associated, 23 Fusarium moniliformis, 83

## G

Galangin, 273f, 274f Gariss, 223 Gas chromatograph, 28 Gastric transit/bile salt, tolerance to, 216 Gastritis, 224 Gastrointestinal tract (GIT), 215 Gene knockout method, 78 Generally regarded as safe (GRAS), 72, 214 Genetic engineering, 160 Gene transfer, 264 Gene transfer via protoplast fusion, 264 Genistein, 273f Genkwanin, 273f Gentamicin, 244 Geotrichum, 288 γ-glutamylo-l-cysteinyloglycine, 91 Giga base pairs (Gbps), 55 Glass transition temperature, 120 Glassy additives, 120 γ-linolenic acid (GLA), 83 Gluconic acid, 172, 174t. See also Organic acids, microbial production of Gluconolactone, 172 Glucose, 192, 282t catabolite repression, 86-87. See also Hansenula polymorpha repression in yeast, 282 Glutamates, 191 Glutamic acid. See also Microbial amino acids production biotechnological approach, 192-194 historical background, 191-192, 192f industrial application/therapeutic role, 194 structure/molecular formula, 188f Glutathione, 91–93 Glycan, 70 Glycerol, 69, 282t Glycerol batch phase (GBP), 69 Glycerol-induced G3P promoter, 81 Glycoengineering of Yarrowia lipolytica, 82-83. See also Yarrowia lipolytica GlycoFi/GlycoSwitch, 70 Glycolysis, 173 Glycopeptides, 248 Glycoproteins, 70, 82 N-glycoproteins, 82 GlycoSwitch vectors, 68 Glycosylation, 68 Glycosylphosphatidylinositol (GPI) anchor, 77 Glycosylphosphatidylinositol (GPI) motifs, 88 Glycosyltranferases, 83 Golden Age of Antimicrobial Therapy, 241 Golgi apparatus, 90 Good Manufacturing Practice (GMP) compliance, 269 guidelines, 76 Gram-negative bacteria, 117, 217, 241, 242, 243 Gram-positive bacteria, 13, 17, 20, 117, 214, 241, 243, 249 Green fluorescent protein (GFP), 52, 77

# Η

Half-life time (HLT), 125 Hand-cutting process, 140 *Hansenula polymorpha*, 84–94. *See also* Nonconventional yeasts, genetic engineering of available strains, 85 expression of own and heterologous proteins, 87–89

genetic methods/genome research, 85-86 glucose catabolite repression, 86-87 methanol induction, genetic control of, 86-87 processes and products heterologous protein, 89-91 low-molecular weight active compounds, 91-94 as producer of fuel ethanol from lignocellulose, 93-94 Harsh lysis approach, 47 Healthcare products for human application, 240 Heat shock proteins (HSP), 118 Helicobacter pylori, 222, 224–225 Heme-thiolate peroxygenase, 136 Hemicelluloses, 157, 158 Hepatitis B surface antigen (HBsAg), 89 2-heptylquinolinol, 19 Herbs extraction of, 137t and impurities, 139 particle size of, 140 Heterobasidiomycetes, 142 Heterocyst-forming species, 300 Heterologous protein, 64, 65, 89-91. See also Hansenula polymorpha Hexane, 228 Hodgkin's lymphoma, 257 Homobasidiomycetes, 142 Hongqu, 134 Host strains/mutants, heterologous complementation of, 51-52. See also Function-based screening, for novel metabolites Hottinger slant agar, 193 Hp4d-driven expression, 76 Human cathepsin K protease inhibitor, 72 Human gelatin, 72 Humanized pharmaceutical protein, production of, 82 - 83Hybridization, 85 Hydrocarbonoclastic bacteria, 84 Hydrocarbons, biosynthetic, 84 Hydrogenases, 163 Hydrogen gas, 160 Hydrogen peroxide, detoxification of, 92 Hydrolysis, 172 Hydroxyectoine, 117 Hydroxylated fatty acids (HFA), 84 Hypercholesterolemia, 216 Hyperglycosylation, 91 Hypermannolysation in yeasts, 68

# Ι

Immobilization. *See also* Upstream processes adsorption, 175 covalent linkage, 175 entrapment, 175 microencapsulation, 175 Immunization, 267, 268 importance of, 265

Immunomodulator, 92 Inactivated vaccines, 266 Indian Stand Institution, 307 Indirect cost, 305, 306, 307 Indirect DNA extraction, 47-49. See also Metagenomics Indirect fermentation, 157 Induced anhydrobiosis anhydrobiotic cells of biocontrol agent (case study), 123-126, 124f, 125f desiccation damage mechanisms, 115-117, 115f desiccation tolerance, mechanisms of, 117-119, 118f drying of microorganisms, 121-122, 121t in microorganisms, 114-115 overview, 113-114 preservation of microorganisms, method for, 119-120 stability of desiccated microorganisms, 122-123 Industrial fermentation, 290 Industrial microbiology, 2-4, 3f. See also Microbial biotechnology timeline of, 3f Industrial microorganisms, preservation of, 113 Inflammatory bowel diseases (IBD), 224 Inhibitor-free metagenomic DNA, 48 Inorganic salts, 144 Insulin-like human growth factor II (IGF-II), 90 Intestinal epithelial cells, adherence to, 220-221 In vitro assay for antifungal activity, 26–28, 27f, 27t. See also Screening of microbial metabolites/ bioactive compounds Iron absorption, 170 Irritable bowel syndrome (IBS), 224 Isoflavonoids, 272f Isopropyl-α-d-thiogalactopyranoside (IPTG), 52

# J

Jameed, 222

## Κ

Kaempferol, 273f Karanjin, 274f Kefir, 223 Keshik, 222 α-ketoglutarate, 192 Ketone, 173 Killed/inactivated vaccines, 266 *Kluyveromyces marxianus*, 94 Koprowski, Hilary, 269 Kossam (fermented cow milk), 223 Kung-Som (traditional fermented shrimp), 223

## L

Laccases, 72 β-lactams, 247 Lactate, 282t Lactic acid, 172, 174t. See also Organic acids, microbial production of Lactic acid bacteria (LAB), 214, 218 Lactobacillus buchneri, 217 Lactobacillus paracasei, 120 Lactobacillus plantarum strains, 219 Lactobacillus species, 226 Lantibiotics, 217 Larrea tridentata powder, 136 Late embryogenesis abundant (LEA) proteins, 118, 119 Lehban, 223 Life-threatening diseases, 9 Lignite, 302 Lignocellulose, 93 Lignocellulosic biomass, 157 Lignocellulosic wastes, 291 Linezolid, 250f Lipases, 71 Lipoglycopeptides, 248 Lipoteichoic acids, 220 Lithium acetate, 78 Live/attenuated vaccines, 266 L-lysine. See also Microbial amino acids production biotechnological approach and production, 198-199 historical background, 194-197, 195f, 196f industrial application and therapeutic role, 199 laboratory-scale production of, 208f producers of, 197 L-lysine on glucose and molasses, production of (case study) cultivation of C. glutamicum FRL no. 61989, 205-206 isolation/extraction/purification of l-lysine, 206 media composition/fermentation parameters on, 207-209, 207f, 208f Long terminal repeats (LTR), 78 Low-density lipoprotein (LDL) cholesterol, 216 Low-molecular weight active compounds. See also Hansenula polymorpha glutathione, 91–93 H. polymorpha as producer of fuel ethanol from lignocellulose, 93-94 penicillin, 91 Low-molecular weight organic natural products, 241 Luteolin, 273f Lysergic acid, 260 Lysergic acid-derived amides, 260 Lysine, 292 consumption of, 191f Lysozyme, 47

### М

Machine-cutting process, 140 Machine drying, 140 Macrolactins, 20 Macrolides, 246 structures, 247f Magnesidins, 20 Maillard reaction, 117 Makdoos, 222 Malic acid, 172, 174t. See also Organic acids, microbial production of Maltose, 282t Mangrove defined, 24 ecosystem, 24-25 mud, 25 Mangrove sediments of Andaman sea coast, 24. See also Fatty acid analysis of streptomyces parvulus DOSMB D105 (case study) Mannose-type N-glycans, 82 α-1,3-mannosyl transferase, 91 Marine bacteria, 13 Marine-derived fungi, 21 Marine microorganisms, 14 Marine sediment of India, 25 Marine sources, importance of, 13 Marketing constraints, 313-314. See also Microbial biofertilizers in biofertilizer technology, 313-314 Mass production of bacterial biofertilizer, 301-303. See also Microbial biofertilizers biofertilizer packets, preparation of, 302-303 culturing of microorganisms, 302 Mechanical cell disruption, 179 Medicinal fungi, 133, 142. See also Solid-state fermentation (SSF) types of, 142, 143t Mega base pairs (Mbps), 55 MEGAN4, 56 Mehler reaction, 162f Melibiose, 282t Metabolic flux analysis (MFA), 70 Metabolite screening, 49 Metagenomic data analysis, 55-56, 55f Metagenomic DNA extraction, 46-49. See also Metagenomics direct DNA extraction, 47 indirect DNA extraction, 47-49 Metagenomic library, 49, 52, 53 Metagenomics defined, 43 overview, 43-44 prospecting from metagenomes, 45-56, 45f function-based screening, 51-53 metagenomic data analysis, 55-56, 55f metagenomic DNA extraction, 46-49 prospects, 56 sampling, 46 screening for novel metabolites, 49-51, 50t sequence-based screening, 53-55 as source of novel metabolites, 44-45 Metchnikoff, Elie, 213, 215 Methanol, 27, 27t, 64, 65, 66, 69, 86, 87, 89, 90, 182 Methanol induction, genetic control of, 86-87. See also Hansenula polymorpha

Methanol induction phase (MIP), 69 Methanol/sorbitol mixture, 70 Methicillin-resistant Staphylococcus aureus (MRSA), 10, 227f Methionine. See also Microbial amino acids production biosynthesis, regulation of, 202f biotechnological approach and production, 201-203, 202f consumption of, 191f deficiency, 203 historical background, 199-201, 200f industrial application/therapeutic role, 203 structure/molecular formula, 188f Methomyl, 67 Methylergonovine maleate, 263 Methylobacterium, 299 Methylotrophic yeasts, 65, 84, 86, 92 Mezorhizobium, 299 Miang (fermented tea leaves), 223 Microalgae, 159 Microbial amino acids production glutamic acid biotechnological approach, 192-194 historical background, 191-192, 192f industrial application/therapeutic role, 194 l-lysine biotechnological approach and production, 198-199 historical background, 194-197, 195f, 196f industrial application and therapeutic role, 199 methionine biotechnological approach and production, 201-203, 202f historical background, 199-201, 200f industrial application/therapeutic role, 203 overview, 187-191, 188f, 188t, 190f, 191f production of l-lysine on glucose and molasses (case study) cultivation of Corynebacterium glutamicum FRL no. 61989, 205-206 isolation/extraction/purification of l-lysine, 206 media composition/fermentation parameters on, 207-209, 207f, 208f tryptophan biotechnological approach and production, 203-205, 204f historical background, 203 industrial application/therapeutic role, 205 Microbial biofertilizers biofertilizer application methodology root dipping, 301 seed treatment, 301 soil application, 301 constraints in commercialization of extension programs, 313 marketing constraints, 313-314

cyanobacteria biofertilizer production appropriate technology, 309-310 popularization for entrepreneurship development, 310-313 on small scale, 310-313, 311f technology, status of, 308-309 mass production of, 301-303 biofertilizer packets, preparation of, 302-303 culturing of microorganisms, 302 microorganisms used as biofertilizers Azospirillum, 299-300 Azotobacter, 299 Cyanobacteria, 300 Mycorrhizal fungi, 300 plant growth-promoting rhizobacteria (PGPR), 300 Rhizobium, 299 overview, 298-299 pilot-scale production of Rhizobium/Azotobacter biofertilizer, 303-305, 303f calculation of cost of, 305-307 quality control, 307-308 Microbial biofuels production advancements in, 156 bioalcohols, 157-158, 158f biodiesel, 158-160, 159t biohydrogen, 160-164, 162f microbial oil, 156-157 overview, 155 Microbial biomass production baker's yeast in genetics study, 287-288 model organism for somatic mutations/cancer study, 286-287 as natural therapy for cancer, 286 overview, 279-280 single-cell protein (SCP), 288-293 advantages/disadvantages, 292-293 production of, 289-291, 290f from waste, 291-292 yeast production of, 283-286, 285f role in baking, 283 yeast metabolism, 280-283, 281f, 282t, 283t Microbial biotechnology defined, 1 history, 2 industrial microbiology/microbial biotechnology, 2-4, 3f overview, 1 processes in downstream processes, 6-7, 7f fermentation processes, 5-6, 6f upstream processes, 4, 5f Microbial cells, 44, 115 Microbial diversity, 44 Microbial fermentation of organic acids, 179-180. See also Organic acids, microbial production of Microbial growth inhibition, 228

Microbial healthcare products antibiotics amikacin, 245f 7-aminocephalosporanic acid, 247f 6-aminopenicillinic acid, 247f balhimycin, 248f benzylpenicillin (penicillin G), 242f biotechnological approach, 253 chloramphenicol, 243f chloroeremomycin, 249f chlortetracycline, 243f daptomycin, 249f discovered between 1950 and 1965, 245f doxycycline, 246f historical background, 240-251, 241f-251f linezolid, 250f macrolides, structures, 247f minocycline, 246f natural aminoglycosides, 244f netilmicin, 244f Penicillium notatum, 241f phenoxymethylpenicillin (penicillin V), 242f pristinamycin, 251f production/pharmaceutical presentation, 252-253, 252f synercid, 251f teicoplanin, 250f tetracyclines, 245f therapeutic role, 254 tigecycline, 246f tyrothricin, 243f vancomycin, 248f virginiamycin, 251f antitumor agents biotechnological approach, 258-259 historical background, 256-257, 256f, 257f overview, 254-256, 255f production/pharmaceutical presentation, 257-258 therapeutic role, 259 case study materials/methods, 274-275 results/discussion, 275 ergot alkaloids biotechnological approach, 263-264 historical background, 262 overview, 259-261, 260f, 261f production/pharmaceutical presentation, 262-263 therapeutic role, 264-265 flavonoids biotechnological approach, 272, 273f historical background, 271-272, 272f production/pharmaceutical presentation, 272 therapeutic role, 274, 274f vaccines biotechnological approach, 270-271 historical background, 266-269, 267f, 268f overview, 265-266

preventive/therapeutic role, 271 production/pharmaceutical presentation, 269-270 types of, 266 Microbial lipids, 156 Microbial strains, selection of, 175. See also Upstream processes Microbial toxins, 258 Microbiological quality control, 113 Microbodies, 77. See also Yarrowia lipolytica Micrococcus luteus, 218 Microencapsulation, 122, 175. See also Immobilization Micromonospora strain, 16 Microorganisms anhydrobiosis in, 114-115 bioactive compounds from. See Bioactive compounds from microorganisms in biochemical reactions, 141 culturing of, 302. See also Mass production of bacterial biofertilizer drying of, 121-122, 121t preservation of, 113, 119-120 used as biofertilizers. See also Microbial biofertilizers Azospirillum, 299–300 Azotobacter, 299 Cyanobacteria, 300 Mycorrhizal fungi, 300 plant growth-promoting rhizobacteria (PGPR), 300 Rhizobium, 299 Microorganisms, SSF, 141-144. See also Solid-state fermentation (SSF) medicinal fungi, 142 photosynthetic bacteria (PSB), 142-143, 143t, 144f Microwave-assisted extraction (MAE), 137t Migril, 263 Minimum inhibitory concentrations (MIC), 19 Minocycline, 246f Modern industrial microbiology, 2 Moisture content, 176. See also Upstream processes Molasses, 284, 292 biomass production from, 292 sterilized, 284 Mongolian dairy products, 223 Monosaccharides, 93 Monosodium glutamate (MSG), 191, 194 Mortierella alpina, 83 Most probable number method, 310 Multicopy integrant selection, 87 Multidrug-resistant (MDR) strains, 219 Multiple displacement amplification (MDA), 48 Multiple overlapping, 54 Multistage purification processes, loss of, 179 Mutations, 286 Mycelia, 28 Mycobacterium, 242 Mycorrhizal fungi, 300. See also Microorganisms Myelosuppression, 259

# Ν

Nanobeads, 270 Nanobiotechnology, 270 Naringenin, 272f Natural aminoglycosides, 244f Natural products, 11, 12t Natural therapy for cancer, baker's yeast as, 286. See also Baker's yeast Neoflavonoids, 272f Nephrotoxicity, 244 Netilmicin, 244f Next generation sequencing (NGS) technology, 54, 55 Nham (fermented products of Thailand), 223 Nisin, 217, 218 Nitrogen, 176 fixation, 298, 300 sources, 67, 144 starvation, 160 Nitrogen-fixing species, 299, 308 N-methyl-d-aspartate (NMDA), 194 Non-Arrhenius behavior, 126 Nonconventional yeasts, genetic engineering of Hansenula polymorpha, 84-94 available strains, 85 expression of own and heterologous proteins, 87-89 genetic methods/genome research, 85-86 glucose catabolite repression, 86-87 methanol induction, genetic control of, 86-87 processes and products, 89-94 Pichia pastoris biopharmaceutical/medical uses, recombinant proteins for, 70-71 bioprocess operations and optimization, 69-70 genetic and molecular tools, 65-69 historical background, 64-65 industrial uses, recombinant proteins for, 71-72 Yarrowia lipolytica characteristics/industrial applications, 72-73, 74t genetic/molecular tools, 73-82, 75t, 80t processes and products of, 82-84 Nonessential amino acids, 187, 188t Non-Hodgkin's lymphoma, 257 Nonmechanical cell disruption, 179 Non-nitrogen fixing species, 299 Nonpeptide antibiotics, 21 Nuclear magnetic resonance spectra, 17 Nucleic acids, 47, 293 Nursery-cum-field production method, 309 Nutrition, 144 Nutritional supplements, 199

# 0

Oil content of microorganisms, 159t Oleaginous microalgae, 156 Oleosins, 78 Oleosomes, 78

O-mannosyl transferase, 90 O-oligosacharides, 68 Oral cavity enzymes, 216 Oral polio vaccine (OPV), 269 Organic acids biosynthesis of, 171 defined, 170 Organic acids, microbial production of acetic acid, 170 ascorbic acid, 170-171, 171f citric acid, 171 citric acid by immobilized A. niger, comparative production of (case study), 180-183 materials and methods, 180-181 results and discussion, 181-183, 181t, 182t, 183f downstream processes, 178-179 fermentation processes solid-state, 178 submerged, 177 surface, 177 fumaric acid, 171-172 gluconic acid, 172 lactic acid, 172 malic acid, 172 microbial fermentation of organic acids, 179-180 microorganisms in, 174-175, 174t overview, 170 pyruvic acid, 173 succinic acid, 173-174 upstream processes immobilization, 175–176 microbial strains, selection of, 175 production medium/physicochemical conditions, 176-177 Organic carrier materials, 301 Ototoxicity, 244 Oven-drying, 124 Oxazolidinones, 248 Oxidative stress, 115, 119 Oxygenic photosynthetic organisms, 163 Oxygen in citric acid production, 178 Oxyrane, 77

#### Р

Parlodel, 263 Pasteur, Louis, 279 Peat soil/lignite, neutralized, 302 Pectinase, 71 Peliko process, 291 Penicillin, 91 biosynthesis, 91 Penicillin G, 242f Penicillin V, 242f Penicillin V, 242f Pentahydroxycarboxylic acid, 172 Peptic ulcer, 224 Peptide alkaloids, 260, 261f Peptide antibiotics, 21 Peroxisome, 65 pH, control of, 176-177 Pharmaceutical industry, 9 Phaseolus vulgaris phytohemagglutinin, 68 pH determination, 181 Phenazine-1-carboxylic acid (PCA), 20 Phenoxymethylpenicillin (penicillin V), 242f Phenylpropanoids, 272 Phillips Petroleum Company (PPC), 64 Phosphatases, 71 Phosphate, exhaustion of, 263 Phosphate-solubilizing microorganisms (PSM), 301 Phosphoenol pyruvate, 192 Phosphoenolpyruvate carboxykinase, 174 Phosphorylation, 192 Photofermentation, 163 Photosynthesis, 160, 161, 163, 300 Photosynthetic bacteria (PSB), 142-143, 143t, 144f, 144t. See also Solid-state fermentation (SSF) Phylogeny, 10 Phytases, 72 Phytoestrogen, 147 Phytohormones, 299 Pichia pastoris. See also Nonconventional yeasts, genetic engineering of bioprocess operations and optimization, 69-70 genetic and molecular tools, 65-69 historical background, 64-65 recombinant proteins for biopharmaceutical/ medical uses, 70-71 recombinant proteins for industrial uses, 71-72 Pichia pastoris acid phosphatase, 68 Pilot-scale production of Rhizobium/Azotobacter biofertilizer, 303-305, 303f calculation of cost of, 305-307 quality control, 307-308 Pinocembrin, 273f Pit method, 309 Plant growth-promoting rhizobacteria (PGPR), 300. See also Microorganisms Plant oleosins, 78 Plant proteins, 203 Plasmodium falciparum, 267 Plastocyanin, 162 Plastoquinone, 162 Podophyllotoxin, 257, 257f Polio vaccines, 269 Polygonum cuspidatum, 147 Polyhouse method, 309, 311f Polyhydroxyalkanoates (PHA), 84 Polyhydroxyalkanoate synthase, 77 Polypeptide, 143t Polyphasic approach, 299 Polyphenols, 142 Polysaccharides, 142, 143t, 270 Polyvinylpolypyrrolidone (PVPP), 47 Porphyrin, 144t Positive clones, 53 Potassium phosphate, 275

Prefiltering, 56 Preservation of microorganisms, method for, 119-120 Primary drying, 122 Pristinamycin, 251f Probiotics, 114 functional properties of adherence to intestinal epithelial cells, 220-221 antibiotic susceptibility, 219 antimicrobial activity, 217-219 cholesterol reduction, 216-217 tolerance to gastric transit/bile salt, 216 history of, 215 mechanisms of action of, 221-222 oriental sources of probiotic strains, 222-223 overview, 213-215 probiotic lactobacilli (case study) antibacterial activity of bacterial isolates, detection of, 226-228, 227f, 228t cholesterol-lowering effects, 228-229, 229f safety of, 226 specific uses of, 223-225 Probiotic strains, oriental sources of, 222-223 Probiotic therapy, 225 Prodigiosin, 20, 255, 255f Production/pharmaceutical presentation antibiotics, 252-253, 252f antitumor agents, 257-258 ergot alkaloids, 262-263 flavonoids, 272 vaccines, 269-270 Product-specific purification, 179 Prokaryotes, 44 Promoter of glyceraldehyde 3-phosphate dehydrogenase gene (pGAP), 66 Promoters (source gene), 74-77, 75t. See also Yarrowia lipolytica Proteases, 71 Protectant additives, 119 Proteinase K, 47 Proteins, 292 dehydration, 116 engineering, 82 intracellular, 118 oxidation, 116 recombinant. See also Pichia pastoris for biopharmaceutical/medical uses, 70-71 for industrial uses, 71-72 secretion, 64 Proteobacteria, 51 Pseudomonas sp., 19 Pueraria (Radix puerariae), 148 Pu-erh tea, 135, 137 Pulse field electrophoresis, 85 Purification methods, 6, 48 Purity level, use of products, 178 Pycnidia, 274 Pyoluteorin (Plt), 20 Pyrrolnitrin (Prn), 20 Pyruvate, 192, 281

Pyruvate decarboxylase (PDC), 94 Pyruvic acid, 173, 174t. *See also* Organic acids, microbial production of

## Q

Qualified Presumption of Safety (QPS), 214 Quercetin, 273f Quinone molecule, 162

## R

Radiation-tolerant bacterial species, 117 Radix puerariae, 148, 149, 149f Raw herbs, 146 Raw protein materials, 144 Reactive oxygen species (ROS), 115, 116 Red biotechnology, 4 Redox sensor, 119 Rehydration, 117, 120 Residual sugars, 181 Resveratrol (3,5,4'-trihydroxystilbene), 147 Rhizobium, 299. See also Microorganisms classification of, 299 pilot-scale production of, 303-307 soil bacterium, 299 Ribosomal RNA (rRNA), 227f Rice plants, 300 Rice straw, 309 Root dipping, 301. See also Microbial biofertilizers Rotary cutting machine, 141f

# S

Saccharides, 117 Saccharomyces cerevisiae, 157, 279 carbon sources, 282t metabolic pathways of, 281f Saccharomycetes, 72 Saccharose, 282 Safety of probiotics, 226 Salinospora, 14 Salinosporamide A, 16 Salk Institute Biotechnology/Industrial Associates Inc., 64 S-(2-aminoethyl)-l-cysteine (AEC), 197 Sanger's method, 54 Saturated fatty acid, 32t Scanning electron micrograph (SEM), 137 of ephedra herb, 147 of tea, 138f Scenedesmus obliguus, 161 Screening methods, conventional, 56 Screening methods for novel metabolites. See also Metagenomics function-based, 51-53 heterologous complementation of host strains/ mutants, 51-52 phenotypical detection of desired activity, 52 substrate-induced gene expression, 52-53

for novel microbial metabolites, 49-51, 50t sequence-based, 53-55 Screening of microbial metabolites/bioactive compounds bioactive compounds from microorganisms, 10-24, 12t from bacteria, 12-21, 15t-16t, 18t-19t from cyanobacteria, 23, 24t secondary metabolites of fungi, 21-23 fatty acid analysis of streptomyces parvulus DOSMB D105 (case study), 24-32 antifungal activity, primary screening for, 26 fatty acids using FAME method, analysis of, 28-32, 29f, 29t-31t, 32t isolation of Streptomyces spp., 26 in vitro assay for antifungal activity, 26-28, 27f, 27t overview, 9-10 Secondary drying, 122 Secondary metabolites, 10, 11 of fungi, 21-23. See also Screening of microbial metabolites/bioactive compounds Secretion signal, 75t, 77 Seed treatment, 301. See also Microbial biofertilizers Semipolyhouse, 311f Semisynthetic aminoglycosides, 244 Semisynthetic penicillins, 246 Separation techniques, biomass, 179 Sequence-based screening, for novel metabolites, 49, 53-55 Serialization, 204 Serratia marcescens, 255, 255f Set batch fermentation, 284 Sha'a (maize-based beverage), 223 Shelf-life time (SLT), 125 Shenqu, 134 Shock screening machine, 140 Shotgun sequencing, 54 Shubat, 223 Simian immunodeficiency virus (SIV), 269 Simple sugars, 176 Simultaneous saccharification and fermentation (SSF), 93 Single-celled eukaryotic microorganisms, 280 Single-cell oils (SCO), 73, 83 Single-cell proteins (SCP), 64, 288-293. See also Microbial biomass production advantages/disadvantages, 292-293 production of, 289-291, 290f from waste, 291-292 Sinorhizobium, 299 Soaking process, 140 Sodium dodecyl sulfate (SDS), 47 Soft lysis method, 47 Soil actinobacteria, systematic screening of, 17 Soil application, biofertilizer, 301. See also Microbial biofertilizers Soil bacterium, 299 Soil temperature, 313

Solid-state fermentation (SSF), 178, 290 Chinese herb SSF processing advantages of, 137-138, 138f based on steam explosion technology, 146-147, 147f history of, 133-135, 134t natural SSF of herbs, 135 principle of, 135, 137t, 138f process and technologies in, 139t process engineering of, 148-150, 148f, 149f prospects of, 146 SSF for concoction of Chinese herbs, 135-136 SSF for extraction of Chinese herbs, 136–137, 137t for ethanol production, 149 overall multistage conversion/directional fractionation of Chinese herbs, 146 overview, 131-132 pretreatment process of Chinese herbs cutting process technology, 140, 141f drying process technology, 141 selecting and washing, 139-140, 140f SSF microorganisms, 141-144 medicinal fungi, 142 photosynthetic bacteria (PSB), 142-143, 143t, 144f SSF technologies bioreactors of SSF, 145-146, 145f preparation culture medium with Chinese herbs, 144-145 versus submerged fermentation (SMF), 132-133, 132t Somatic mutations, model organism for, 286-287. See also Baker's yeast Sonication, 47 Soybean lipoxygenase, 84 Spent sulfite liquor, 291 Sponges, 22 Spray-drying, 121, 121t, 124, 124f, 197 SSF. See Solid-state fermentation (SSF) Staphylococcus aureus, 241 Starch, 283, 292 hydrolyzed, 292 Steam explosion technology, 146-147, 147f Sterile air, 177 Sterol, 143t Strain cultivation, 93 Strains, 81-82. See also Yarrowia lipolytica Straw, 309 Streptomyces, 14, 242 Streptomyces avermitilis, 55 Streptomyces griseus, 17 Streptomyces spp., isolation of, 26. See also Screening of microbial metabolites/bioactive compounds Stress proteins, 118 Submerged fermentation (SMF), 177, 190, 252f, 257, 262, 263 versus SSF, 132-133, 132t

Substrate cost, 288 Substrate-induced gene expression (SIGEX), 52-53. See also Function-based screening Substrate starvation, 92 Subunit vaccines, 266 Succinic acid, 173–174, 174t. See also Organic acids, microbial production of Sucrose, 120, 124, 124f Sugar-nucleotide donors, 83 Sugars, 118 residual, 181 Sulfuric acid-treated rice straw hydrolysate (SARSH), 156 Sunlight, 160 Supercritical fluid extraction (SFE), 137t Supernatant, 193 Superoxide dismutase (SOD), 119 Surface culture technique, 242f Surface display GPI anchor signals for, 74t systems, 77 Surface fermentation, 177 Surface glycoprotein, 270 Sustainable agriculture, importance of, 298 Symba process, 292 Synercid, 251f Syngas, 158

# Т

TCA (2-hydroxy-1,2,3-propane tricarboxylic acid), 171 TCM. See Traditional Chinese medicine (TCM) Tea, SEM of, 138 Teicoplanin, 250f Temperature on microbial populations, 313 Terpene, 143t Tetrabromopyrrole, 19 Tetracyclines, 244, 245f Therapeutic role of antibiotics, 254 of antitumor agents, 259 ergot alkaloids, 264-265 flavonoids, 274, 274f vaccines, 271 Thermoprotector, 84 Thermotolerance, 84 Thiamine, 284 Thin-layer chromatography, 206 Tigecycline, 246f Tissier, Henry, 215 Tissue culture techniques, 270 Toluene extract, 262 Toxins, microbial, 258 Toxoids, 266, 270 Traditional Chinese medicine (TCM) chronology of important events related to, 134t defined, 133 Transamination, 173 Transcinnamic acid, 272

Transcriptome analysis, 86 Trans-1,2-ethylenedicarboxylic acid, 171 Transgenic organisms, 10 Transmission electron microscopy, 124 Traveler's diarrhea, 224 Trehalose, 117, 120 Tricarboxylic acid (TCA) cycle, 193 Trichosporon fermentans, 155 Trioxaccerains, 17 Tryptophan, 260f. See also Microbial amino acids production biotechnological approach and production, 203-205, 204f historical background, 203 industrial application/therapeutic role, 205 structure/molecular formula, 188f Tsukamurella paurometabola C-924, 123, 124 electron micrographs, 125f Tumor-amplified protein expression therapy, 258 Tumor suppressor genes, 287 Tyrosol, 22 Tyrothricin, 243f Tyrozine Z residue, 162

## U

Ultracentrifugation, 269 Ultrafiltration, 269, 292 Ultrasound-assisted extraction (UAE), 137t Unculturable bacteria, 51 Unsaturated fatty acid, 32t Untreated tea, SEM of, 138f Upstream processes. See also Organic acids, microbial production of immobilization adsorption, 175 covalent linkage, 175 entrapment, 175 microencapsulation, 175 in microbial biotechnology, 4, 5f microbial strains, selection of, 175 production medium/physicochemical conditions aeration, 177 agitation, 177 moisture content, 176 pH, control of, 176-177 temperature, effect of, 176 Ustilago maydis, 200

# V

Vaccines biotechnological approach, 270–271 historical background, 266–269, 267f, 268f overview, 265–266 preventive/therapeutic role, 271 production/pharmaceutical presentation, 269–270 transportation of, 267 types of, 266

Vaccine vial monitor (VVM), 267, 267f stages based on color change, 268f Vancomycin, 248f Van der Waals interactions, 116 Van Urk reagent, 275 Variable cost, 305, 307 Vegetables, fermented, 223 Viable cell count method, 310 Vibrio harveyi, 82 Vinblastine, 256f, 257 Vincristine, 256f, 257 Vinegar, 170 Vinyl acetate polymers, 170 Viral diarrhea, 224 Virginiamycin, 251f Vitamin B<sub>1</sub>, 209 Vitamin B<sub>12</sub>, 144t Vitamin C, 170 Vitamins, 284 Vitrification, 126 Volatile organic compound emissions, 286

## W

Water, 144 Water-soluble amino alcohol, 260 Weibull model, 122 White biotechnology, 4 Whole-cell biocatalyst, *Yarrowia lipolytica* as. *See also Yarrowia lipolytica* for carotenoid production, 83–84 for SCO production, 83 Williams–Landel–Ferry theory, 126 Wine producers, 279 Wood, 291 World Health Organization (WHO), 267

## Χ

Xenobiotic detoxification, 92 Xylanases, 71 Xylitol dehydrogenase (XDH), 94 Xylose reductase (XR), 93 Xylulokinase (XK), 94

## Υ

Yarrowia lipolytica. See also Nonconventional yeasts, genetic engineering of characteristics/industrial applications, 72-73, 74t genetically modified (GM) strains of, 73, 74t genetic/molecular tools, 73-82, 75t promoters, 74, 76-77 strains, 81-82 targeting signals for secretion/surface display, 77 targeting to microbodies, 77-78 vectors/methods/amplification strategies, 78-81, 80t glycoengineering of, 82-83 as high-throughput expression platform for protein engineering, 82 prospects of, 84 as whole-cell biocatalyst for carotenoid production, 83-84 for SCO production, 83 Yeasts, 64. See also Microbial biomass production alkane-degradating properties of, 72 disaccharides as substrates in, 283t metabolism, 280-283, 281f, 282t, 283t modes of respiration in, 282t nonconventional. See Nonconventional yeasts, genetic engineering of production of, 283-286, 285f role in baking, 283 thermotolerant, 93 xylose-fermenting, 93

# Ζ

Zymomonas mobilis, 157